Q.1) The IUPAC name of $\stackrel{\bullet}{\mathcal{H}}$ is
[1] neonane
[2] 3-ethylpentane
[3] tetra ethyl methane
[4] 3,3-diethylpentane
Q.2) Chlorination is an example of
[1] addition reaction
[2] substitution reaction
[3] elimination reaction
[4] rearrangement reaction
Q.3) Heterolysis of C-Cl bond leads to the formation of
[1] carbanion
[2] carbocation
[3] carbon free radical
[4] carbene
Q.4) Which molecule has highest dipole moment?
[1] CH ₃ F
[2] CH ₃ CI
[3] CH ₃ Br
[4] CH ₃ I
Q.5) Presence of Chlorine in an organic compound shows
[1] –I effect
[2] +I effect
[3] Electromeric effect
[4] Mesomeric effect
Q.6) Which of the following is a nucleophile?
[1] AICI ₃
[2] H ₃ O ⁺

[3] BF ₃ [4] CN ⁻
Q.7) A fuel has the same knocking property as a mixture of 70% iso-octane and 30% nheptane by volume. The octane cumber of the fuel is[1] 100[2] 70[3] 30[4] 40
Q.8) Number of carbon atoms in gasoline ranges from [1] C ₃ -C ₅ [2] C ₈ -C ₁₂ [3] C ₆ -C ₁₀ [4] C ₁₀ -C ₁₆
Q.9) Ethyne upon addition with water in presence of HgSO ₄ and H ₂ SO ₄ gives [1] Propanone [2] Ethanal [3] Ethanol [4] Diethyl ether
Q.10) Ethylene and acetylene can be distinguished by following reagent: [1] alkaline KmnO ₄ [2] bromine solution in CCl ₄ [3] conc. H ₂ SO ₄ [4] ammonical Cu ₂ Cl ₂
Q.11) Hydrolysis of calcium carbide produces [1] methane [2] ethane [3] acetylene [4] ethylene

- Q.12) What are the respective products of electrolysis of sodium salt of ethanoic acid, succinic acid and maleic acid?
- [1] ethane, ethene and ethyne
- [2] ethane, propane and ethyne
- [3] ethane, propyne and propene
- [4] ethane, ethanol and ethanal
- Q.13) Alkenes are isomeric with
- [1] alkane
- [2] cycloalkane
- [3] alkyne
- [4] arene
- Q.14) Saytzeff rule is invalid to
- [1] primary alcohol
- [2] secondary alcohol
- [3] tertiary alcohol
- [4] all of above
- Q.15) Ethane is obtained from ethylene by
- [1] Wurtz reaction
- [2] Darzen's process
- [3] Sabatier-Senderen's reaction
- [4] Dehydrogenation
- Q.16) Which of the following give methane on hydrolysis?
- [1] Fe₃O₄
- [2] Al₂O₃
- [3] CaC₂
- [4] Al₄C₃
- Q.17) Find X and Y in the reaction: $C_2H_2 + H_2 \xrightarrow{\frac{Pd}{BaSO_4}, quinoline} X \xrightarrow{O_2, Ag} Y$
- [1] Ethane and Benzene
- [2] Ethene and Ethylene oxide

- [3] Ethyne and Ethanol
- [4] Ethene and ethylene glycol
- Q.18) 2 methylpropene $\stackrel{O_3}{\rightarrow}$ A $\stackrel{Zn/H_2O}{\longrightarrow}$ B, what is B?
- [1] Acetone and Propanone
- [2] Formaldehyde and Propanal
- [3] Acetone and Formaldehyde
- [4] Formaldehyde and Propanone
- Q.19) Quality of diesel is measured in terms of Cetane Number. Cetane is
- [1] naphthalene
- [2] hexadecane
- [3] n-octane
- [4] ethyl heptane
- Q.20) Metamerism is not shown by:
- [1] Alcohol
- [2] Ether
- [3] Ester
- [4] Amine
- Q.21) In Friedel Crafts acylation, electrophile is
- [1] Carbonium ion
- [2] Carbanion
- [3] Acyl Chloride
- [4] Acylium ion
- Q.22) Nitration reaction occurs quickly in
- [1] Benzene
- [2] Toluene
- [3] Phenol
- [4] Nitrobenzene
- Q.23) The electrophile in sulphonation of benzene is

[1] SO ₃ ⁺ [2] SO ₃ [3] HSO ₃ [4] HSO ₄ ⁻
Q.24) Typical reaction of aromatic compound is [1] EAR [2] NAR [3] ESR [4] NSR
Q.25) Which is correct for deactivation of benzene ring? [1] -OH>-NO ₂ >-CN>-CH ₃ [2] -OH<-NO ₂ >-CN>-CH ₃ [3] -OH>-NO ₂ >-CN<-CH ₃ [4] -OH<-NO ₂ <-CN>-CH ₃
Q.26) Chloroform when reacts with Zn dust and water and Zin dust and HCl respectively forms [1] CH ₄ and CH ₂ Cl ₂ [2] CH ₂ Cl ₂ and CCl ₄ [3] CH ₂ Cl ₂ and CH ₄ [4] CCl ₄ and CH ₂ Cl ₂
Q.27) When propylene reacts with HCl in presence of <i>Organic peroxide</i> , it follows [1] Markonikov's rule [2] Anti-Markonikov's rule [3] Chain reaction [4] Free radical reaction
Q.28) When a given organic compound is treated with propanone along with base potassium hydoxide, it results to form a hypnotic product. What is given compound? [1] CH ₃ CH ₂ OH

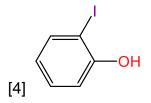


- [3] CHCl₃
- [4] DDT
- Q.29) $C_6H_6 \xrightarrow{\text{HCl }/O_2} C_6H_5Cl + H_2O$, This reaction is called
- [1] Gattermann reaction
- [2] Sandmayer's reaction
- [3] Raschig process
- [4] Schiemann reaction

Q.30)
$$P \xrightarrow{alc.KCN} CH_3CH_2CN \xrightarrow{Stephen\ reduction} Q$$
, P=? and Q=?

- [1] Bromopropane and ethanal
- [2] Bromoethane and ethanal
- [3] Bromoethane and propanal
- [4] Bromoethane and propanamide
- Q.31) lodoform test is given by
- [1] CH₃-CO-OH
- [2] CH₃-OH
- [3] CH₃-CH₂-OH
- [4] Both a and c
- Q.32) Drinking of which alcohol may cause blindness
- [1] Wood spirit
- [2] Ethanol

- [3] absolute alcohol
- [4] rectified alcohol
- Q.33) Formation of alcohol and carboxylic acid from ester is an example of
- [1] Esterification
- [2] Trans-esterification
- [3] Ester-hydrolysis
- [4] Oxidation



Q.35) Which of the following does not form dative bond with ether?

- [1] BF₃
- [2] AICI₃
- [3] RMgX
- [4] NaBH₄

- Q.36) The major product obtained when tert-butyl chloride is reacted with sodium ethoxide is
- [1] Ethane
- [2] Diethyl ether
- [3] 2-methyl-1-propene
- [4] tert-Butyl methyl ether
- Q.37) Which is most acidic among the following?

[4]

- Q.38) Phenol and benzoic acid can be ditinguished by
- [1] NaHCO₃
- [2] Tollen's reagent
- [3] Bayer's reagent
- [4] CuSO₄

$$\begin{array}{c} CH_3 \\ H_3C-CH \end{array} \longrightarrow \begin{array}{c} Air \\ H_2O \end{array} \longrightarrow \begin{array}{c} X \end{array} \longrightarrow Y$$
Q.39)

- [1] Cresol, Sodium benzoate
- [2] Phenol, Sodium benzoate
- [3] Acetone, Benzoic acid
- [4] Phenol, No reaction

Q.40)
$$H_3C-NH_2 \xrightarrow{NaNO_2 + conc.HCI} X$$
, What is X?

- [1] CH₃OH
- [2] CH₃OCH₃
- [3] CH₃ONO
- [4] All of above
- Q.41) Which is the most suitable technique to separate formaldehyde and benzaldehyde?
- [1] Fehling's solution
- [2] Tollen's solution
- [3] I₂ + NaOH
- [4] NH₃
- Q.42) The reagent used to detect sugar in urine is
- [1] Bayer's reagent
- [2] Grignard's reagent
- [3] Schiff's reagent
- [4] Benedict's solution
- Q.43) The product of Rosenmund's reacton is heated with alc. KCN which results the formation of

- Q.44) The product of Gattermann-Koch reaction and Ozonolysis of ethene are treated with conc. NaOH, Which of the following is not obtained as product?
- [1] HCOONa
- [2] C₇H₆O
- [3] H₂O
- [4] CH₃OH
- Q.45) The reaction of formic acid with conc. H₂SO₄ gives
- [1] CO
- [2] CO₂
- [3] Oxalic acid
- [4] Glycerol
- Q.46) Which of the following is amphoteric?
- [1] Picric acid
- [2] Ethanamine
- [3] Ethanamide
- [4] Ethanoyl chloride
- Q.47) Which is/are example(s) of Hunsdiecker reaction?
- [1] RCOOAg + $Cl_2 \rightarrow RCl + CO_2 + AgCl$
- [2] $2RCOOAg + I_2 \rightarrow RCOOR + CO_2 + 2AgI$

- [3] CH3COOC2H5 $\xrightarrow{C_2H_5ONa/H^+}$ CH₃COCH₂COOC₂H₅ + C₂H₅OH
- [4] Both a and b
- Q.48) Nitrobenzene when treated with Glucose in presence of base yields
- [1] Aniline
- [2] Azobenzene
- [3] Hydrazobenzene
- [4] Azoxybenzene
- Q.49) Which of the following statement is true?
- [1] When nitrobenzene is heated with alkali, meta derivative is formed.
- [2] When nitrobenzene is heated with alkali, ortho and para derivative are formed.
- [3] Nitrobenzene responds Friedel-Crafts reaction.
- [4] Nitrobenzene is called oil of artificial wintergreen.
- Q.50) Arrange the order of basicity in aqueous medium.
- [1] Ammonia>Methylamine>Ethylamine>Aniline
- [2] Ammonia<Methylamine<Ethylamine>Aniline
- [3] Ammonia>Methylamine>Ethylamine<Aniline
- [4] Ammonia<Methylamine<Ethylamine<Aniline
- Q.51) Primary and secondary amines cannot be distinguished by:
- [1] Nitrous acid test
- [2] Hisenberg test
- [3] DNP test
- [4] Both b and c
- Q.52) Gabriel Phthalimide synthesis gives
- [1] Primary amine and secondary amine
- [2] Primary amine but not Secondary amine
- [3] Secondary amine and tertiary amine
- [4] Aniline and primary amine only

Q.53) Aniline is warmed with chloroform and alcoholic caustic alkali. Offensive odor is due to formation of

- [1] Phenyl cyanide
- [2] Methyl cyanide
- [3] Amino nitrile
- [4] Phenyl carbylamine

Q.54) Cellulose is the polymer of

- [1] Maltose
- [2] Fructose
- [3] β -glucose
- [4] α-gulcose

Q.55) Bordeaux mixture contains

- [1] DDT
- [2] Zinc phosphate
- [3] Zinc sulphate
- [4] Copper sulphate

Q.56) Which one is chromophore?

- [1] -C=C-
- [2] -NO₂
- $[3] \; N_2$
- [4] -CN

Q.57) Name the compound indicated by above molecular diagram.

- [1] Aspirin
- [2] Savlon

[4] Paracetamol
Q.58) Biuret test is done in order to detect
[1] Sugar
2] Protein
[3] Fat
[4] Oils
Q.59) Na on fusion with organic compound containing nitrogen element forms
[1] NaONO
[2] Na ₄ C
[3] NaCN
[4] NaNO ₂

[3] Salol