## Unit 5

## 1. Quantum Well, Quantum Wire, Quantum Dot

#### **Quantum Well:**

- A quantum well is a nanostructure where the motion of carriers (electrons or holes) is confined in one dimension but free to move in two dimensions.
- It is created when a thin layer of a semiconductor material (with a small bandgap) is sandwiched between two layers of materials with a larger bandgap.

## **Key Features**:

- Electrons are confined in the thickness (z-axis) direction.
- Energy is quantized along the confined direction but continuous along the free directions.
- Example: GaAs quantum well between AlGaAs layers.
- Applications: Lasers, photodetectors.

## **Quantum Wire:**

- In a quantum wire, the carriers are confined in two dimensions and free to move in only one dimension.
- Fabricated by further reducing the size of a quantum well into a narrow wire-like structure.

#### Key Features:

- Electron motion is allowed only along the wire axis (say x-axis).
- Energy levels are quantized along the y and z axes.
- **Example**: Semiconductor nanowires like InP or GaN nanowires.
- **Applications**: High-speed transistors, nanodevices.

#### Quantum Dot:

- A quantum dot confines carriers in all three spatial dimensions.
- Often referred to as "artificial atoms" due to their discrete energy levels.

## • Key Features:

- o Completely quantized energy levels.
- Size typically 2-10 nm.
- Example: Colloidal quantum dots (CdSe nanoparticles).
- Applications: Quantum computing, LEDs, solar cells.

# 2. Density of States (DOS) in 0D, 1D, and 2D Systems

**Density of States (DOS)** describes the number of available electronic states at a particular energy level.

## 0D (Quantum Dot):

- In 0D structures, carriers are confined in all directions.
- DOS:
  - o Represented by delta functions at discrete energy levels.
  - Only specific discrete energies are allowed.
- Expression:

$$D(E) \propto \delta(E - E_n)$$

• **Graph**: Sharp spikes at allowed energy levels.

# 1D (Quantum Wire):

- In 1D structures, carriers are free in one direction.
- DOS:
  - $\circ \quad \text{Increases sharply near the sub-band edges}.$

$$D(E) \propto \frac{1}{\sqrt{E - E_n}}$$

- Follows
- Graph: Sharp increase at the start of each sub-band.

## 2D (Quantum Well):

- In 2D structures, carriers are free to move in two directions.
- DOS:
  - o Constant for each sub-band.
  - Step-like graph.

## • Expression:

D(E)=constant for each energy sub-bandD(E) = \text{constant for each energy sub-band} D(E)=constant for each energy sub-band

# 3. SEM (Scanning Electron Microscope) and TEM (Transmission Electron Microscope)

#### Scanning Electron Microscope (SEM):

- SEM scans the surface of a specimen using a focused beam of high-energy electrons.
- The electrons interact with atoms at the surface, producing signals like secondary electrons, backscattered electrons, and characteristic X-rays.

#### • Working:

- Electron beam is raster-scanned over the sample.
- Secondary electrons are collected to form an image.

#### • Features:

- 3D-like surface morphology.
- Resolution: ~10 nm.
- Sample preparation is simple (coating with conductive material if nonconductive).
- Applications: Surface studies, fracture analysis, material science.

## **Transmission Electron Microscope (TEM):**

- TEM passes a beam of electrons through an ultra-thin specimen.
- The interaction of electrons with the specimen forms an image.

#### Working:

Electrons transmitted through the sample are collected and magnified.

#### • Features:

- High-resolution images revealing internal structures.
- Resolution: ~0.1 nm (atomic scale).
- Sample must be very thin (~100 nm or less).
- Applications: Crystallography, defect analysis, nanotechnology.

#### **Difference Table:**

Property	SEM	TEM
Principle	Scattered electronS	Transmitted electrons
Resolution	~10 nm	~0.1 nm
Image	Surface morphology	Internal structure
Sample preparation	Easier	Harder (ultrathin sections needed)

## 4. PVD (Physical Vapor Deposition) and CVD (Chemical Vapor Deposition)

## **Physical Vapor Deposition (PVD):**

- A vacuum deposition method where material goes from solid to vapor phase and back to solid as a thin film on the substrate.
- Types:
  - Evaporation
  - Sputtering
- Process:

Material is heated/ionized  $\rightarrow$  vaporized  $\rightarrow$  transported  $\rightarrow$  condensed on substrate.

- Features:
  - o Low pressure environment.
  - Good for metallic coatings.
- Applications: Hard coatings, optical coatings, semiconductor device fabrication.

## **Chemical Vapor Deposition (CVD):**

- A chemical process in which gaseous reactants react on or near the heated substrate to form a solid thin film.
- Types:
  - Atmospheric Pressure CVD (APCVD)
  - Low Pressure CVD (LPCVD)
- Process:

Chemical reactions occur at the substrate  $\rightarrow$  solid material deposits.

#### Features:

- Higher temperatures required.
- o Produces conformal, high-purity films.
- Applications: Semiconductor industry (SiO<sub>2</sub> layers, TiN coatings), Solar panels.

## 5. CNT (Carbon Nanotubes)

#### Introduction:

- Carbon Nanotubes are cylindrical structures composed of carbon atoms arranged in a hexagonal lattice.
- They can be thought of as rolled-up sheets of graphene.

## Types:

• Single-walled Carbon Nanotubes (SWCNT):

One layer of graphene rolled into a cylinder.

Multi-walled Carbon Nanotubes (MWCNT):

Multiple layers of graphene cylinders nested inside each other.

#### **Properties:**

- Extremely high tensile strength (100x stronger than steel).
- High electrical conductivity (can act as metallic or semiconducting).
- High thermal conductivity.
- Lightweight and flexible.

## **Synthesis Methods:**

- Arc discharge method.
- Laser ablation.
- Chemical Vapor Deposition (CVD).

## Applications:

- Nanoelectronics (transistors, diodes).
- Sensors (gas sensors, bio-sensors).
- Energy storage (batteries, supercapacitors).
- Reinforcement in composites (for aerospace, sports goods).
- Drug delivery systems.



