³⁶Ar(p,d) 1968Jo04,1968Ko11

 J^{π} =0+ for ³⁶Ar ground state.

1968Jo04: a 27.5(1)-MeV proton beam was produced by the 1.3-m FFAG cyclotron at the Nuclear Physics Laboratory at the University of Colorado. Targets were natural argon gas (99.6% 40 Ar) and 99.0% enriched 36 Ar gas. Deuterons were detected using a telescope of a 0.228-mm transmission surface barrier detector and a 0.37-cm lithium-drifted stopping counter with FWHM=120 keV. Measured $\sigma(E_d,\theta)$. Deduced levels, L, and spectroscopic factors from finite-range DWUCK-DWBA analysis of the measured $\sigma(\theta)$.

1968Ko11: a 33.6-MeV proton beam was produced by the 64-in. sector-focusing cyclotron at Michigan State University. The target was >99% isotropically enriched 36 Ar gas. Deuterons were detected using a telescope of a 279- μ m silicon surface barrier detector and a 3-mm lithium-drifted silicon counter with FWHM=130 keV. Measured $\sigma(E_d,\theta)$. Deduced levels, J, π , L, and spectroscopic factors from zero-range Macefield-DWBA analysis of the measured $\sigma(\theta)$.

³⁵Ar Levels

Spectroscopic factor $C^2S = \sigma(\theta)_{exp}/\sigma(\theta)_{DWBA}/N$, where N=1.65*3/2 is a normalization factor adopted by 1968Jo04. The DWBA calculation in 1968Ko11 assumed neutron-well radius parameter r_{0n} =1.15 F ($d_{5/2}$) and 1.35 F ($d_{3/2}$), respectively, to fit the forward-angle *J*-dependence observed in the L=2 angular distributions. 1968Ko11 states that a large change in radius results in a large change in the magnitude of the calculated DWBA cross-section and this gives rise to problems in extracting absolute spectroscopic factors, therefore, the C^2S from 1968Ko11 are not adopted but are given in the comments. However, there is some evidence that the radius parameter corresponding to the best DWBA fit to the data might also result in the most trustworthy value for the spectroscopic factor.

E(level)	\mathbf{J}^{π}	<u>L</u> †	C^2S^{\ddagger}	Comments
0	3/2+	2	2.92	C ² S: other: 3.03 for 1.15 F and 1.76 for 1.35 F (better fit) (1968Ko11).
1180 20	1/2+	0	2.50	E(level): from 1968Jo04 and 1968Ko11.
				C ² S: other: 1.29 for 1.15 F and 1.05 for 1.35 F (1968Ko11).
1700 <i>30</i>	5/2+			E(level): from 1968Ko11.
				C ² S: 0.1 for 1.15 F from 1968Ko11.
2615 20	$(3/2^+)$	2		E(level): weighted average of 2630 20 (1968Jo04) and 2600 20 (1968Ko11).
				$L,C^2S: S=0.42 \text{ for } 1.15 \text{ F and } 0.28 \text{ for } 1.35 \text{ F (better fit) for } L=2 \text{ from } 1968Koll \text{ Other:}$
				$C^2S=0.12$ for L=1 from 1968Jo04. 1973Be26 determined L=2 from $^{36}Ar(^3He,\alpha)^{35}Ar$
				and states that L=1 is probably in error as this level corresponds well with the known
				3/2 ⁺ state at 2695 keV in ³⁵ Cl and there are no nearby negative-parity states in ³⁵ Cl.
2970 20	$(5/2^+)$	2	2.47	E(level): weighted average of 2990 20 (1968Jo04) and 2950 20 (1968Ko11).
				C^2S : assuming $5/2^+$ (1968Jo04). Other: 3.10 assuming $3/2^+$ (1968Jo04). 2.31 for 1.15 F
				(better fit) and 1.53 for 1.35 F from 1968Ko11.
3200 20	$7/2^{-}$	3	0.63	E(level): weighted average of 3210 20 (1968Jo04) and 3190 20 (1968Ko11).
1==< =0	4 (2 ±		0.4=4	C ² S: other: 0.64 for 1.15 F and 0.37 for 1.35 F (better fit) (1968Kol1).
4756 28	1/2+	0	0.172	E(level): weighted average of 4770 20 (1968Jo04) and 4700 40 (1968Ko11).
7 10 2 2 0	- 12		0.45	C ² S: other: 0.05 for 1.15 F and 0.04 for 1.35 F (1968Ko11).
5102 20	7/2-	3	0.46	E(level): weighted average of 5110 20 (1968Jo04) and 5070 40 (1968Ko11).
				L: from 1968Jo04. L is not reported in 1968Ko11. 1973Be26 determined L=2 from
				36 Ar(3 He, α) 35 Ar and states that L=2 gives a much better account of the data than
5400 <i>50</i>				does L=3. E(level): from 1968Ko11.
5598 20	$(3/2,5/2)^+$	2	2.93,2.37	E(level): Holl 1906K011. E(level): weighted average of 5610 20 (1968Jo04) and 5570 30 (1968Ko11).
3390 20	(3/2,3/2)	2	2.93,2.31	C ² S: for $3/2^+$ and $5/2^+$, respectively (1968Jo04). Other: 1.77 for 1.15 F (better fit) and
				1.25 for 1.35 F (1968Ko11).
6024 20	$(3/2,5/2)^+$	2	1.58,1.31	E(level): weighted average of 6030 20 (1968Jo04) and 6010 30 (1968Ko11).
002.20	(5/2,5/2)	_	110 0,110 1	C^2S : for $3/2^+$ and $5/2^+$, respectively (1968Jo04). Other: 1.18 for 1.15 F and 0.83 for
				1.35 F (better fit) (1968Ko11).
6620 <i>30</i>	$(1/2^+)$	0		E(level),L: from 1968Ko11.
				C ² S: 0.24 for 1.15 F and 0.19 for 1.35 F (1968Ko11).
6700 20	7/2-	3	2.60	E(level),L,C ² S: from 1968Jo04.

³⁶Ar(p,d) 1968Jo04,1968Ko11 (continued)

³⁵Ar Levels (continued)

 E(level)
 J^{π} L^{\dagger} C^2S^{\ddagger} Comments

 6820 30
 $(3/2,5/2)^+$ 2
 E(level),L,C²S: from 1968Ko11 only.

 $C^2S: 0.72$ for 1.15 F (better fit) and 0.51 for 1.35 F (1968Ko11).

 7030 20
 $(3/2,5/2)^+$ 2
 2.20,1.79
 $C^2S:$ for $3/2^+$ and $5/2^+$, respectively (1968Jo04).

2

 $^{^\}dagger$ From 1968Jo04 and 1968Ko11, unless otherwise noted.

[‡] From 1968Jo04.