

[Total No. of CO's - 05]

Seat No:

[Total No. of Pages: 02]

G. H. Raisoni College of Engineering and Management, Pune.
(An Autonomous Institution affiliated to Savitribai Phule, Pune University)
F.Y Year B. Tech (Mechanical, Civil & E&TC) (Term-**II**)
ESE SUMMER 2021
Environmental Chemistry (UBSL 102)

[Time: -- 2Hours]

[Max. Marks – 50]

COURSE OUTCOME:

1. Explain various methods of water treatment for domestic and industrial use
2. Differentiate various categories of waste and its disposal techniques
3. Identify various batteries and recognize its commercial applications
4. Classify the different types of Energy and its future scope
5. Apply the knowledge of environmental pollution and degradation to solve related problems

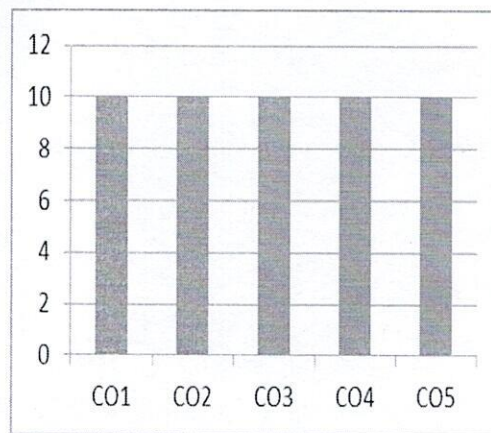
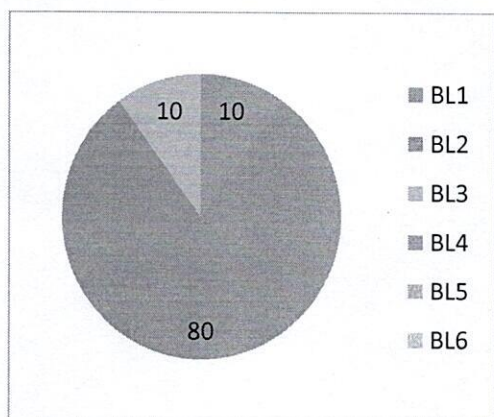
Instructions to the candidates:

- 1) (CO1/CO2/CO....)at the beginning of question/sub question indicates the course outcome related to the question.
- 2) All questions are compulsory.
- 3) Neat diagrams must be drawn wherever necessary.
- 4) Figures to the right indicate full marks.
- 5) Assume suitable data, if necessary.
- 6) Other Instructions, if any.

CO	Sub Question		Marks	BL
CO1	a)	Explain boiler corrosion, state its causes and preventative measures.	[5]	L2
	b)	On the basis of your understanding, explain the Zeolite process with a suitable diagram for water softening and give its advantages and disadvantages.	[5]	L2
OR				
	c)	Applying your knowledge to explain hardness of water, also explain it with chemical reaction of soap in soft water as well as in hard water.	[5]	L3
CO2	a)	What is E-waste? Explain collection, segregation, transportation and its disposal.	[5]	L2
	b)	Applying your knowledge to explain solid waste management with suitable technique.	[5]	L3

P.T.O.

CO3	a)	Explain Lithium –Ion battery with suitable diagram and chemical reaction, also state any two advantages of it.	[5]	L2
	b)	Explain Fuel Cell with suitable diagram. Explain its advantages and disadvantages.	[5]	L2
CO4	a)	Remember and write any six characteristics of good fuel.	[5]	L1
	b)	As per your understanding explain the working mechanism of Solar Cooker with diagram.	[5]	L2
OR				
	c)	A Chemist was burnt 0.75 g of coal completely in a Bomb Calorimeter; he observed the increase in the temperature of 2600 g of water was 1.75 ⁰ C. If the water equivalent of the calorimeter is 165 g, calculate the gross calorific value in KJ/Kg. Specific heat of water = 4.187 KJ/ kg ⁰ C	[5]	L3
CO5	a)	What are the sources of air pollution in your area and explain its controlling methods?	[5]	L2
	b)	Explain water quality index as per WHO. Explain various sources of water pollution.	[5]	L2



- BL – Bloom's Taxonomy Levels (1- Remembering, 2- Understanding, 3 – Applying, 4 – Analysing, 5 –Evaluating, 6 - Creating).