

# Biomolecular potential energy functions

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# Experiment

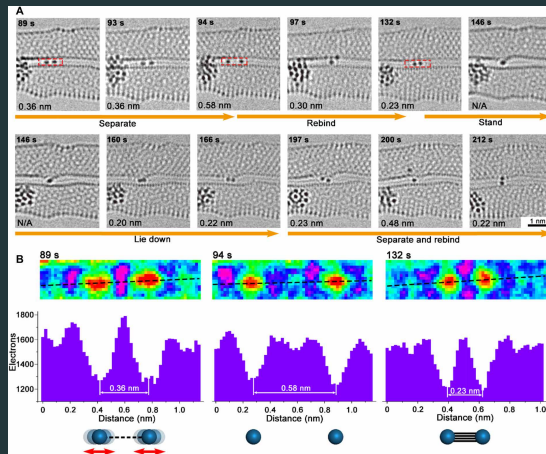
## I Transmission electron microscopy(TEM)

Cao *et al* imaged the structure and dynamics of  $\text{Re}_2$  molecules on the level of the single atom in real time

—— Cao *et al*, *Sci. Adv.* 2020; 6: eaay5849

- Re : the atomic number 75, weight: 186.21 u
- movie

# Experiment



A) Time-series AC-HRTEM images

B) (the first row) false-color images, (the second row) the intensity profiles, (the third row) the possible states of the two Re atoms

## Limit in Experiments

- They could not provide a dynamics of proteins, organic molecules, and water molecules.

## Contents

1. Potential Energy
2. QM
3. MM
4. QM/MM

# The Kinetic Theory of Ideal Gases

- There is no intermolecular interactions between the molecules of an ideal gas. In short, the potential energy  $E_{\text{pot}}$  of the system is zero:  $E_{\text{pot}} = 0$
- The total energy  $E_{\text{tot}}$  of the system becomes the kinetic energy  $E_{\text{kin}}$  of the system:  
 $E_{\text{tot}} = E_{\text{kin}} + E_{\text{pot}} = E_{\text{kin}}$ .

•

$$E_{\text{kin}} = \frac{3}{2} N k_B T = \sum_{A=1}^N \frac{1}{2} \frac{|\mathbf{p}_A|^2}{m_A}, \quad (1)$$

where  $N$  is the number of atoms in the system,  $k_B$  is the Boltzmann constant.  $T$  is the thermal temperature of the system.  $\mathbf{p}$  and  $m$  are the momentum and the mass.

- The momentum or the velocity is related to the thermal temperature  $T$ .

# Dynamics

- The motion of atoms is described by **Newton's laws of motion**
- The second law states:

The rate of change of the momentum of a body is directly proportional to the force applied, and this change in momentum takes place in the direction of the applied force

—— Newton

- $\mathbf{F}_A = -\nabla_A V$ , where  $\mathbf{F}$  is a force acting on  $A$ th atom.  $V$  is a potential energy.
- When we know the potential energy  $V$  of the system, the motion of atoms in the system is precisely predicted according to Newton's laws of motion:

$$\mathbf{R}_A(t + \delta t) = \mathbf{R}_A(t) + \mathbf{v}_A(t)\delta t + \frac{1}{2} \frac{\mathbf{F}_A(t)}{m_A} \delta t^2$$

# Potential Energy: The Born-Oppenheimer Approximation

- time-independent Schrödinger equation:

$$\begin{aligned} E &= \langle \Psi | \hat{H} | \Psi \rangle \\ \hat{H} &= - \sum_{i=1}^N \frac{1}{2} \nabla_i^2 - \sum_{A=1}^M \frac{1}{2M_A} \nabla_A^2 - \sum_{i=1}^N \sum_{A=1}^M \frac{Z_A}{|\mathbf{r}_i - \mathbf{R}_A|} \\ &\quad + \sum_{i=1}^N \sum_{j>i}^N \frac{1}{|\mathbf{r}_i - \mathbf{r}_j|} + \sum_{A=1}^M \sum_{B>A}^M \frac{1}{|\mathbf{R}_i - \mathbf{R}_j|} \end{aligned}$$

- Assumption: the electrons in a molecule are moving in the field of fixed nuclei. (from Szabo and Ostlund, **Modern Quantum Chemistry**)

$$\begin{aligned} \hat{H}_{\text{elec}} &= - \sum_{i=1}^N \frac{1}{2} \nabla_i^2 - \sum_{i=1}^N \sum_{A=1}^M \frac{Z_A}{|\mathbf{r}_i - \mathbf{R}_A|} + \sum_{i=1}^N \sum_{j>i}^N \frac{1}{|\mathbf{r}_i - \mathbf{r}_j|} \\ E_{\text{elec}} &= \langle \Psi_{\text{elec}} | \hat{H}_{\text{elec}} | \Psi_{\text{elec}} \rangle \\ E &= E_{\text{elec}} + \sum_{A=1}^M \sum_{B>A}^M \frac{1}{|\mathbf{R}_i - \mathbf{R}_j|} \end{aligned}$$

## Movie

- Born-Oppenheimer Molecular Dynamics (BOMD)
- Path Integral Molecular Dynamics (PIMD)



## Potential Energy

- Intra-molecular potential energy of the  $I$ th monomer:

$$E_I - E_I^{\min} = E_{\text{bond}} + E_{\text{angle}} + E_{\text{torsion}},$$

where  $E_I = \langle \Psi_I | \hat{H}_I | \Psi_I \rangle$  represents the potential energy of  $I$ th monomer.

- Inter-molecular potential energy:

$$E_{IJ} - E_I - E_J = E_{\text{LJ}} + E_{\text{Coul}},$$

where  $E_{IJ} = \langle \Psi_{IJ} | \hat{H}_{IJ} | \Psi_{IJ} \rangle$  represents the potential energy of the dimer consisting of  $I$ th and  $J$ th monomers.

•

$$E_{\text{LJ}} = \sum_{A>B} 4\epsilon_{AB} \left[ \left( \frac{\sigma_{AB}}{R_{AB}} \right)^{12} - \left( \frac{\sigma_{AB}}{R_{AB}} \right)^6 \right], \quad E_{\text{Coul}} = \sum_{A>B} \frac{q_A q_B}{R_{AB}}$$



## Polarization Energy

$$\begin{aligned}
 E_I^{\text{pol}} &= \langle \Psi_{I:Q_I} | \hat{H}_{I:Q_I} | \Psi_{I:Q_I} \rangle - \langle \Psi_I | \hat{H}_{I:Q_I} | \Psi_I \rangle \\
 E^{\text{pol}} &= 0.5 \sum_I E_I^{\text{pol}}
 \end{aligned}$$

# Quantum Chemistry Software

- Plane-wave:

$$\Psi(\mathbf{r}) = \frac{1}{\sqrt{\Omega}} \exp[i \mathbf{G} \cdot \mathbf{r}],$$

where  $\Omega$  is the volume of the box,  $\mathbf{G}$ : Reciprocal lattice vector

- ABINIT, CP2K, CPMD, Quantum ESPRESSO6, VASP
- Density Functional Theory
- PBC (solid or liquid)

- With Gaussian-type orbitals:

$$\Psi_k(\mathbf{r}) = \exp[-\alpha_k |\mathbf{r} - \mathbf{R}|^2]$$

- Gaussian, GAMESS(US), MPQC, NWChem, PSI, PySCF, Q-Chem
- Hartree-Fock Theory, Kohn-Sham Density Functional Theory, Møller-Plesset perturbation theory (MP), Coupled-Cluster method
- molecules

## Prerequisites: Install the dependencies

- python 3.7
- (MM) Install OpenMM

```
conda install -c conda-forge openmm  
or  
conda install -c omnia openmm
```

- (QM) Install pyscf

```
pip install pyscf
```

- (GeomOpt) Install pyberny

```
pip install pyberny
```

## Prerequisites: Exercise

- <https://github.com/swillow/pdb2amber>
- <https://github.com/swillow/modelingworkshop>

## pyscf: Python-based Simulations of Chemistry Framework

- Free
- Python
- Easy to install:

```
$ pip install pyscf
```

## drawback

- we have to know how to use a python.

## Code

---

```
1 from pyscf import gto, scf
2 mol = gto.M(atom='H 0 0 0; H 0 0 1.2', basis='ccpvdz')
3 mf = scf.RHF(mol)
4 mf.kernel()
```

---

python run\_qmmm.py -i input.json

## input.json

---

```
1 {  
2     "theory": "qm",  
3     "job": "ener", # gopt  
4     "qm": {  
5         "method": "scf",  
6         "basis": "6-31gs",  
7         "charge": 0,  
8         "fname_geom": "qm_step4.xyz",  
9         "esp": true,  
10        "esp_opts": {"resp": true, "resp_hfree": true}  
11    }  
12  
13 }
```

---



# Molecular Mechanics Force Field

- The **force field** refers to the *functional form* and *parameter* sets used to calculate the potential energy of a system.
- The parameters for a chosen energy function are derived from experiments, calculations in quantum mechanics, or both.

- 

$$E = E_{\text{intra}} + E_{\text{inter}}$$

$$E_{\text{intra}} = E_{\text{bond}} + E_{\text{angle}} + E_{\text{torsion}}$$

$$E_{\text{inter}} = E_{\text{vdW}} + E_{\text{Coul}}$$

$$E_{\text{bond}} = \frac{k}{2}(r - r_0)^2$$

- The **force constant**  $k$  is determined from the experimental *Infrared spectrum*, *Raman spectrum*, or high-level *quantum mechanical calculations*. This value is related to the vibrational frequencies.

## Molecular Mechanics Force Field Parameter Sets

- AMBER
- CHARMM
- MMFF
- OPLS
- AMOEBA

### Topology

- lists of chemical bonds, angles, torsional angles, *et al*

## OpenMM: A High Performance Molecular Dynamics Library

- <http://openmm.org/>
- Free
- Python
- Easy to install:

```
| $ conda install -c conda-forge openmm
```

## drawback

- we have to know how to use a python.

## Code

---

```
1 from simtk.openmm.app import *
2 from simtk.openmm import *
3 from simtk.unit import *
4 from sys import stdout
5
6 prmtop = AmberPrmtopFile('input.prmtop')
7 inpcrd = AmberInpcrdFile('input.inpcrd')
8 system = prmtop.createSystem(nonbondedMethod=PME, nonbondedCutoff=1*nanometer, \
9                             constraints=HBonds)
10 integrator = LangevinIntegrator(300*kelvin, 1/picosecond, 0.002*picoseconds)
11 simulation = Simulation(prmtop.topology, system, integrator)
12 simulation.context.setPositions(inpcrd.positions)
13 if inpcrd.boxVectors is not None:
14     simulation.context.setPeriodicBoxVectors(*inpcrd.boxVectors)
15 simulation.minimizeEnergy()
16 simulation.reporters.append(PDBReporter('output.pdb', 1000))
17 simulation.reporters.append(StateDataReporter(stdout, 1000, step=True, \
18     potentialEnergy=True, temperature=True))
19 simulation.step(10000)
```

---

## Main protein from SARS-CoV-2

- Jin, Z. *et al* Nature, Vol 582, page 289 (2020)
  - PDB: 6LU7
    - 6LU7 contains coordinates of heavy atoms (C, N, O, S) of a monomer.
1. separate a substrate (or inhibitor) from a protein
  2. add H atoms
    - (option) build a dimer from a monomer
  3. generate an AMBER parameter and topology file (prmtop) of a protein (to get the charge of the protein)
  4. build a system consisting of a protein, water, and salt.
  5. generate an AMBER parameter and topology file (prmtop) of the whole system.
  6. Minimization, NVT MD simulations, and NPT MD simulations

## Step 1. Separate a ligand from a protein

■ `python separate_complex.py -i input_step1.json`

### input\_step1.json

```
1 {  
2     "fname_xtal": "./MPRO/6lu7.pdb",  
3     "fname_protein": "./MPRO/protein.pdb",  
4     "fname_ligand": "./MPRO/subtrate.pdb"  
5 }
```

## Step 2. Add H atoms

- using AmberTools: `conda install -c conda-forge ambertools`

```
| reduce abc.pdb > abc_H.pdb
```

- using pdb2pqr: `pip install pdb2pqr`

```
| pdb2pqr30 --ff AMBER --with-ph=7 --nodebump  
--ffout=AMBER abc.pdb abc.pqr
```

- using OpenMM:

## Code

```
1 ...  
2 pdb = PDBFile('input.pdb')  
3 modeller = Modeller(pdb.topology, pdb.positions)  
4 modeller.addHydrogens(forcefield)  
5 ...
```

## Step 3 and 5. Build an AMBER prmtop file

- download a zip file from <https://github.com/swillow/pdb2amber>

```
python pdb2amber.py -i input.json
```

### input.json

```
1 {  
2     "fname_pdb": "dimer.pdb",  
3     "fname_prmtop": "dimer.prmtop",  
4     "fname_ff": [  
5         "./data/protein.ff14SB.xml",  
6         "./data/wat_opc3.xml"  
7     ]  
8 }
```



## Build an AMBER prmtop file: advanced

### input.json

---

```
1 {
2     "fname_pdb": "dimer.pdb",
3     "fname_prmtop": "dimer.prmtop",
4     "fname_ff": [
5         "./data/protein.ff14SB.xml",
6         "./data/wat_opc3.xml"
7     ],
8     "linked_residues": [
9         ["FE1 FES", "SG CYF"],
10        ["FE2 FES", "SG CYF"],
11        ["P FMN", "OG1 THO"],
12        ["FE FE", "SG CYG"]
13    ]
14 }
```

---

## Step 4. Build a system containing a protein, (membrane) water, and salt

python build\_system.py -i input\_build.json

### input\_build.json

```
1  {
2      "fname_protein": "dimer.pdb",
3      "protein_charge": -8,
4      "fname_system": "dimer_wat.pdb",
5      "box": [90,120,120],
6      "membrane": {
7          "fname_unit": "dppe_box15.pdb",
8          "bool": false
9      },
10     "solvent": {
11         "fname_unit": "water_box15.xyz",
12         "bool": true,
13         "NaCl(Molarity)": 0.1
14     }
15 }
```

## OpenMM Script Builder

■ <http://builder.openmm.org/>

## Minimization

python AmberOpenMM.py -i input\_min.json

### input\_min.json

```
1  {
2      "job": "min",
3      "min": {
4          "pdb": "input.pdb",
5          "prmtop": "input.prmtop",
6          "save_pdb": "minimized.pdb",
7          "fc_pos": 500.0,
8          "Platform": "OpenCL",
9          "maxInteraction": 5000,
10         "tolerance": 100.0
11     }
12 }
```

## NVT MD simulation

python AmberOpenMM.py -i input\_nvt.json

### input\_step1\_nvt.json

```
1 {
2     "job": "nvt",
3     "nvt": {
4         "pdb": "minimized.pdb",
5         "prmtop": "input.prmtop",
6         "save_pdb": "step1.pdb",
7         "save_state": "step1.rst",
8         "stdout": "bomd_nvt_step1.dat",
9         "dcd": "traj_step1.dcd",
10        "fc_pos": 300.0,
11        "dt_fs": 2.0,
12        "Temperature": 300.0,
13        "Platform": "OpenCL",
14        "totalSteps": 125000,
15        "nstdout": 1000,
16        "ndcd": 5000
17    }
18 }
```

### input\_step2\_nvt.json

```
1 {
2     "job": "nvt",
3     "nvt": {
4         "pdb": "step1.pdb",
5         "prmtop": "input.prmtop",
6         "save_pdb": "step2.pdb",
7         "save_state": "step2.rst",
8         "stdout": "bomd_nvt_step2.dat",
9         "dcd": "traj_step2.dcd",
10        "dt_fs": 2.0,
11        "Temperature": 300.0,
12        "Platform": "OpenCL",
13        "totalSteps": 125000,
14        "nstdout": 1000,
15        "ndcd": 5000
16    }
17 }
```

## input\_step3\_npt.json

```
1 {
2   "job": "npt",
3   "npt": {
4     "pdb": "step2.pdb",
5     "prmtop": "input.prmtop",
6     "save_pdb": "step3.pdb",
7     "load_state": "step2.rst",
8     "save_state": "step3.rst",
9     "stdout": "bomd_npt_step3.dat",
10    "dcd": "traj_step3.dcd",
11    "dt_fs": 2.0,
12    "Temperature": 300.0,
13    "Barostat" : "MonteCarloMembraneBarostat",
14    "Pressure" : 1.0,
15    "Platform": "OpenCL",
16    "nstep_cycle": 125000,
17    "ncycle": 20,
18    "nstdout": 1000,
19    "ndcd": 5000
20  }
21 }
```

## input\_step4\_npt.json

```
1 {
2   "job": "npt",
3   "npt": {
4     "pdb": "step3_20.pdb",
5     "prmtop": "input.prmtop",
6     "save_pdb": "step4.pdb",
7     "load_state": "step3_20.rst",
8     "save_state": "step4.rst",
9     "stdout": "bomd_npt_step4.dat",
10    "dcd": "traj_step4.dcd",
11    "dt_fs": 2.0,
12    "Temperature": 300.0,
13    "Barostat" : "MonteCarloBarostat",
14    "Pressure" : 1.0,
15    "Platform": "OpenCL",
16    "nstep_cycle": 125000,
17    "ncycle": 20,
18    "nstdout": 1000,
19    "ndcd": 5000
20  }
21 }
```

## Generate Ligand Force Fields

1. add hydrogen atoms

```
reduce ligand.pdb > ligand_H.pdb
```

2. check the molecular structure and identify the charge of the ligand

3. generate mol2 files (estimate the atomic point charges)

```
antechamber -fi pdb -fo mol2 -i ligand_H.pdb -o ligand_H.mol2 -c bcc -pf y -nc chg
```

4. generate frcmod

```
parmchk2 -i abc_H.mol2 -o abc_H.frcmod -f mol2
```

5. generate the parameter and topology file using a General Amber Force Field

```
tleap -s -f ligand.in
```

### **ligand.in**

```
source leaprc.gaff2
```

```
mol = loadmol2 ligand_H.mol2
```

```
loadamberparams ligand_H.frcmod
```

```
saveamberparm mol ligand.prmtop ligand.inpcrd
```

```
quit
```

### A minor problem of 'antechamber'

The sum of the atomic point charges does not become the total charge of the ligand.

### Build a force field parameter file for OpenMM from an AMBER prmtop file

```
python write_xml_pretty -i input_ff.json
```

```
1 {  
2     "fname_prmtop": "n3i.prmtop",  
3     "fname_xml": "n3i.ff.xml",  
4     "ff_prefix": "n3i"  
5 }
```



## Potential Energy of the system consisting of QM molecules and MM molecules

$$E = E_{\text{QM}} + E_{\text{MM}} + E_{\text{QM/MM}},$$

where  $E_{\text{QM/MM}}$  is the potential energy between QM and MM molecules.

- classical mechanics:

$$E_{\text{QM/MM}} = \sum_{A \in \text{QM}} \sum_{B \in \text{MM}} \left( 4\epsilon_{AB} \left[ \left( \frac{\sigma_{AB}}{R_{AB}} \right)^{12} - \left( \frac{\sigma_{AB}}{R_{AB}} \right)^6 \right] + \frac{q_A q_B}{R_{AB}} \right)$$

- embedded fragment method:

$$E_{\text{QM}} + E_{\text{QM/MM}} = \langle \Psi_{I:Q_I} | \hat{H}_{I:Q_I} | \Psi_{I:Q_I} \rangle + \sum_{A \in \text{QM}} \sum_{B \in \text{MM}} 4\epsilon_{AB} \left[ \left( \frac{\sigma_{AB}}{R_{AB}} \right)^{12} - \left( \frac{\sigma_{AB}}{R_{AB}} \right)^6 \right]$$

python run\_qmmm.py -i input.json

## input.json

---

```
1  {
2    "theory": "qmmm", # qm-pol
3    "job": "opt", # ener
4    "qm": {
5      "method": "scf",
6      "basis": "6-31gs",
7      "charge": 0,
8      "fname_geom": "qm_step4.xyz",
9      "fixed_atom_list": [1,16],
10     "constraints_list": [[23,38,500,1.85]],
11     "esp": true,
12     "esp_opts": {"resp": true,"resp_hfree": true}
13   },
14   "mm": {
15     "fname_prmtop": "mm_step4.prmtop",
16     "fname_geom": "mm_step4.pdb",
17     "fixed_atom_list": [2312,2314,2714,2716]
18   }
19 }
```