

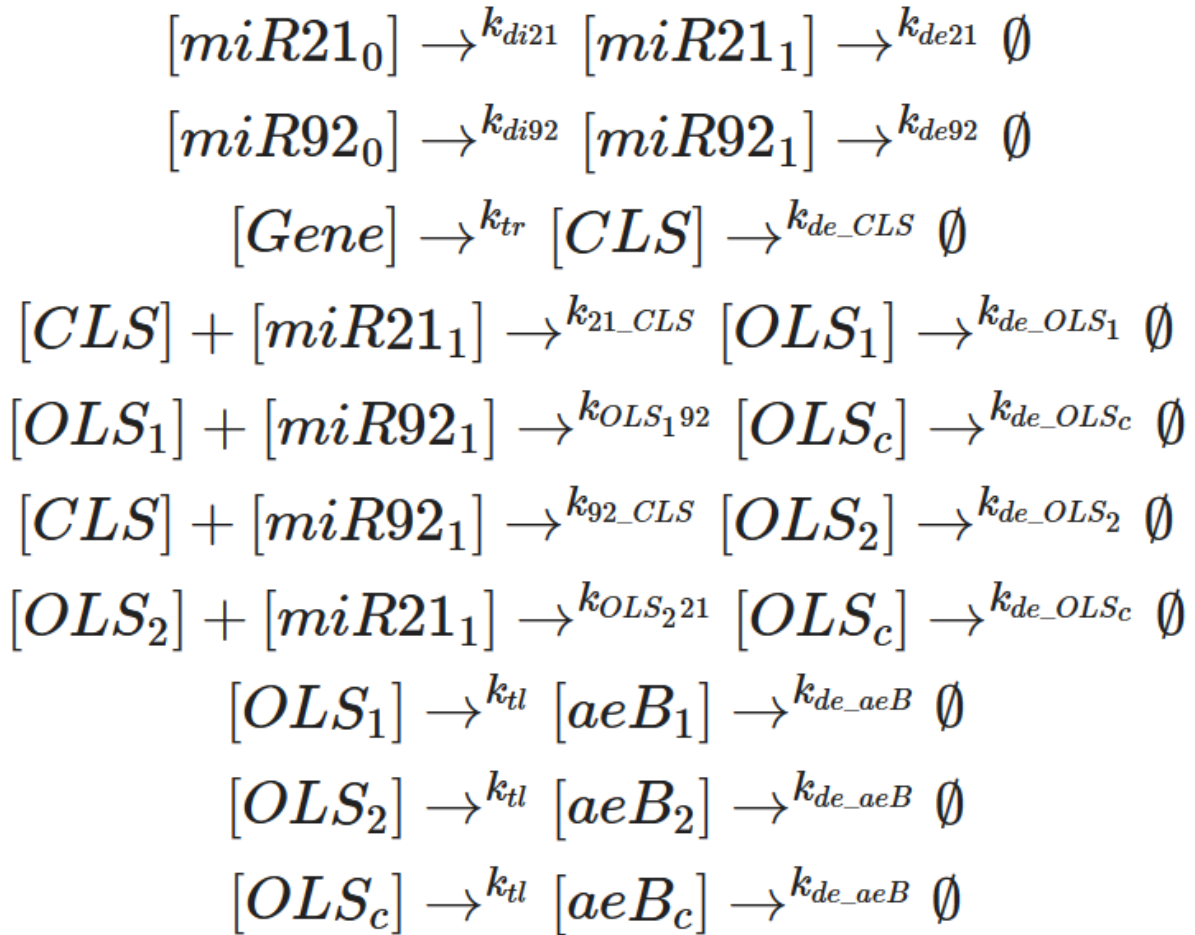
## Kinetic Modeling Equation for LIRA

### Abbreviations

Abbreviation	Description
$k_{di21}$	miR21 diffusion rate constant
$k_{di92}$	miR92a diffusion rate constant
$k_{de\_aeB}$	aeBlue degradation rate constant
$k_{de21}$	miR21 degradation rate constant
$k_{de92}$	miR92a degradation rate constant
$k_{de\_OLS}$	Open LIRA Switch degradation rate constant
$k_{de\_CLS}$	Close LIRA Switch degradation rate constant
$k_{tr}$	Transcription Rate Constant
$k_{tl}$	Translation Rate Constant
$k_{21\_CLS}$	miR21-CLS Complex Formation Constant
$k_{92\_CLS}$	miR92a-CLS Complex Formation Constant
$k_{OLS192}$	OLS1-miR92 Complex Formation Constant or OLSc Formation
$k_{OLS221}$	OLS2-miR21 Complex Formation Constant or OLSc Formation
$miR21_0$	miR21 extracellular
$miR92_0$	miR92a extracellular
$miR21_1$	miR21 intracellular
$miR92_1$	miR92a intracellular
Gene	Gene Coding for LIRA and Reporter
CLS	Close LIRA Switch
OLS	Open LIRA Switch
$OLS_1$	Open LIRA Switch formed by CLS-miR21 complex formation

$OLS_2$	Open LIRA Switch formed by CLS-miR92 complex formation
$OLS_c$	Open LIRA Switch formed by CLS-miR21-miR92a complex formation
$aeB_1$	aeBlue chromprotein translated from $OLS_1$
$aeB_2$	aeBlue chromprotein translated from $OLS_2$
$aeB_c$	aeBlue chromprotein translated from $OLS_c$
$aeB_t$	aeBlue chromprotein total = $aeB_1 + aeB_2 + aeB_c$
$\emptyset$	Degradated substance

### Chemical Equation for two miRNAs



## ODE for two miRNAs

$$\frac{d[miR21_0]}{dt} = -k_{di21}[miR21_0]$$

$$\frac{d[miR92_0]}{dt} = -k_{di92}[miR92_0]$$

$$\frac{d[miR21_1]}{dt} = k_{di21}[miR21_0] - k_{de21}[miR21_1] - k_{21\_CLS}[CLS][miR21_1] - k_{OLS_221}[OLS_2][miR21_1]$$

$$\frac{d[miR92_1]}{dt} = k_{di92}[miR92_0] - k_{de92}[miR92_1] - k_{92\_CLS}[CLS][miR92_1] - k_{OLS_192}[OLS_1][miR92_1]$$

$$\frac{d[CLS]}{dt} = k_{tr} - k_{de\_CLS}[CLS] - k_{21\_CLS}[CLS][miR21_1] - k_{92\_CLS}[CLS][miR92_1]$$

$$\frac{d[OLS_1]}{dt} = k_{21\_CLS}[CLS][miR21_1] - k_{OLS_192}[OLS_1][miR92_1] - k_{de\_OLS_1}[OLS_1]$$

$$\frac{d[OLS_2]}{dt} = k_{92\_CLS}[CLS][miR92_1] - k_{OLS_221}[OLS_2][miR21_1] - k_{de\_OLS_2}[OLS_2]$$

$$\frac{d[OLS_c]}{dt} = k_{OLS_192}[OLS_1][miR92_1] + k_{OLS_221}[OLS_2][miR21_1] - k_{de\_OLS_c}[OLS_c]$$

$$\frac{d[aeB_1]}{dt} = k_{tl}[OLS_1] - k_{de\_aeB}[aeB_1]$$

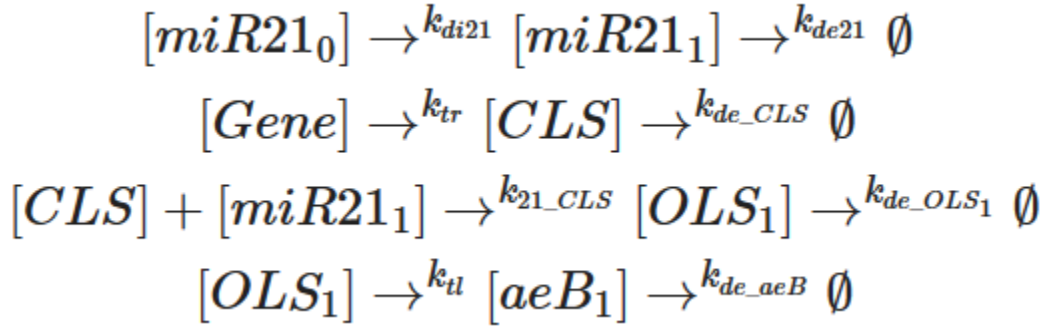
$$\frac{d[aeB_2]}{dt} = k_{tl}[OLS_2] - k_{de\_aeB}[aeB_2]$$

$$\frac{d[aeB_c]}{dt} = k_{tl}[OLS_c] - k_{de\_aeB}[aeB_c]$$

$$\frac{d[aeB_t]}{dt} = \frac{d[aeB_1]}{dt} + \frac{d[aeB_2]}{dt} + \frac{d[aeB_c]}{dt}$$

## Chemical Equation for one miRNA

To simulate 1 miRNA, we take it as miR21, because the parameter values we use in this simulation for both miR21 and miR92a are the same.



## ODE for one miRNA

$$\begin{aligned} \frac{d[miR21_0]}{dt} &= -k_{di21}[miR21_0] \\ \frac{d[miR21_1]}{dt} &= k_{di21}[miR21_0] - k_{de21}[miR21_1] - k_{21\_CLS}[CLS][miR21_1] \\ \frac{d[CLS]}{dt} &= k_{tr} - k_{de\_CLS}[CLS] - k_{21\_CLS}[CLS][miR21_1] \\ \frac{d[OLS_1]}{dt} &= k_{21\_CLS}[CLS][miR21_1] - k_{de\_OLS_1}[OLS_1] \\ \frac{d[aeB]}{dt} &= k_{tl}[OLS_1] - k_{de\_aeB}[aeB_1] \end{aligned}$$

## Modeling Parameters of LIRA

Symbol	Description	Value/Range	Unit	Reference(s)	Notes
$k_{di21}$	miR21 diffusion rate constant	$1 \times 10^{-2}$	$s^{-1}$	(Levine <i>et al.</i> , 2007)	General miRNA rate diffusion with size ~20 nucleotide
$k_{di92}$	miR92a diffusion rate constant	$1 \times 10^{-2}$	$s^{-1}$	(Levine <i>et al.</i> , 2007)	General miRNA rate diffusion with size ~20 nucleotide
$k_{de\_aeB}$	aeBlue degradation rate constant	$2.7 \times 10^{-4}$	$s^{-1}$	(Alon 2020)	General reporter degradation rate (1 hour)
$k_{de21}$	miR21 degradation rate constant	$3 \times 10^{-4}$	$s^{-1}$	(Baabu <i>et al.</i> , 2022)	General RNA degradation rate
$k_{de92}$	miR92a degradation rate constant	$3 \times 10^{-4}$	$s^{-1}$	(Baabu <i>et al.</i> , 2022)	General RNA degradation rate
$k_{de\_OLS}$	Open LIRA Switch degradation rate constant	$3 \times 10^{-4}$	$s^{-1}$	(Baabu <i>et al.</i> , 2022)	General RNA degradation rate
$k_{de\_CLS}$	Close LIRA Switch degradation rate constant	$3 \times 10^{-4}$	$s^{-1}$	(Baabu <i>et al.</i> , 2022)	General RNA degradation rate
$k_{tr}$	Trancription Rate Constant*	0.000779	$M.s^{-1}$	(Stögbauer <i>et al.</i> , 2012)	RNA Toehold and reporter transcription rate ~2.2 NtPs, this TU has around 1603 nucleotides
$k_{tl}$	Translation Rate Constant*	0.0344	$s^{-1}$	<a href="#">Average translation rate -</a>	Average translation rate in <i>E coli</i> ~8 amino acid per second

				<a href="#">Bacteria Escherichia coli - BNID 111689 (harvard.edu)</a>	and aeBlue chromoprotein has 231 amino acids
$k_{21\_CLS}$	miR21-CLS Complex Formation Constant	$1 \times 10^5$	$M^{-1}.s^{-1}$	(Baabu <i>et al.</i> , 2022)	miRNA-RNA toehold complex formation rate
$k_{92\_CLS}$	miR92a-CLS Complex Formation Constant	$1 \times 10^5$	$M^{-1}.s^{-1}$	(Baabu <i>et al.</i> , 2022)	miRNA-RNA toehold complex formation rate
$k_{OLS192}$	OLS1-miR92 Complex Formation Constant or OLS <sub>c</sub> Formation	$1 \times 10^5$	$M^{-1}.s^{-1}$	(Baabu <i>et al.</i> , 2022)	miRNA-RNA toehold complex formation rate
$k_{OLS221}$	OLS2-miR21 Complex Formation Constant or OLS <sub>c</sub> Formation	$1 \times 10^5$	$M^{-1}.s^{-1}$	(Baabu <i>et al.</i> , 2022)	miRNA-RNA toehold complex formation rate

\*assumption: OLS<sub>1</sub>, OLS<sub>2</sub>, OLS<sub>c</sub> have same degradation rate

Known that In E coli translation rate is 8 amino acids/second ([Average translation rate - Bacteria Escherichia coli - BNID 111689 \(harvard.edu\)](#)) and our aeBlue chromoprotein has 231 amino acids. So  $k_t$  (**Translation Rate Constant**) = 1/average time for translation

Average time for translation = Number of amino acids × Time for addition of one amino acid

Time for addition of one amino acid =  $\frac{1}{8}$  = 0.125 second per one amino acid

Average time for translation =  $231 \times 0.125$

Translation Rate Constant =  $\frac{1}{231 \times 0.125}$  = ~0.0344 / second

### Trancription Rate Constant

molar\_sec = 1603.62 nucleotide/sec / (330 grams/mol \*  $6.022 \times 10^{23}$  molecules/mol)

molar\_sec = 0.000779 molar/sec

## References

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