#### FOAMtastic: Polystyrene Repurposing to Reduce Landfill Waste

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#### References:

- [1] J. Davis, "Styrofoam Facts Why You May Want To Bring Your Own Cup," *SEJ*, Apr. 10, 2019. https://www.sej.org/publications/backgrounders/styrofoam-facts-why-you-may-want-bring-your-own-cup
- [2] Baleen Group, "CIF 291 Town of Markham Polystyrene Densifier," May 2012.
- [3] NIST, "ThermoData Engine (TDE)," *trc.nist.gov*, 2009. https://trc.nist.gov/tde.html (accessed Feb. 28, 2025).
- [4] aspentech, "Aspen Plus | Leading Process Simulation Software | AspenTech," www.aspentech.com. https://www.aspentech.com/en/products/engineering/aspen-plus



# FOAMtastic

# Polystyrene Repurposing to Reduce Landfill Waste



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## Introduction

#### **Problem Statement**

Polystyrene (PS) is a durable and versatile plastic that can take over 500 years to degrade, which presents environmental risks due to its limited recyclability and potential to release harmful toxins [1]. With less than 10% being recycled, the majority accumulates in landfills. This project addresses these challenges by developing a pyrolysis process to convert PS waste into feedstock.

#### Context

PS is classified as #6 on the plastic resin identification scale, making it one of the most difficult plastics to recycle despite its common use in takeout containers and packaging.

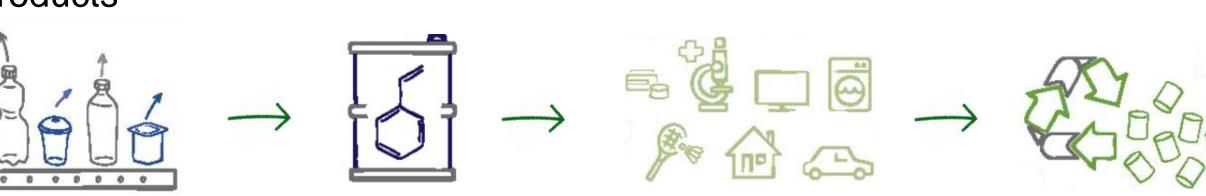
Using pyrolysis, a method that thermally degrades plastic anaerobically, this recycling process will:

Divert PS landfill waste from municipalities

Table 1: 2013 Town of Markham PS Densifier project [2]

Polystyrene Collected	Revenue (\$)	Profit/Loss (\$)
2100 tonnes/year	3,029.61	(53,036)

2. Develop a novel solution that can operationally repurpose waste into valuable products



### Constraints

Process capacity of 2100 tonnes/yr



Yield 98% purity of all products



Improve circularity



# Sustainable Development Goals



SDG 12: Responsible Consumption and Production

- Reducing virgin material use
- Addresses single use nature and low recyclability
- Improve waste management



SDG 13: Climate Action

- Diversion of waste from landfills and incineration
- Reduce greenhouse gas emissions
- Minimize energy use and emissions

# Process Development & Tools

### **Process Development**

Literature review on PS/plastic pyrolysis

Develop process diagram

**Preliminary** Analysis

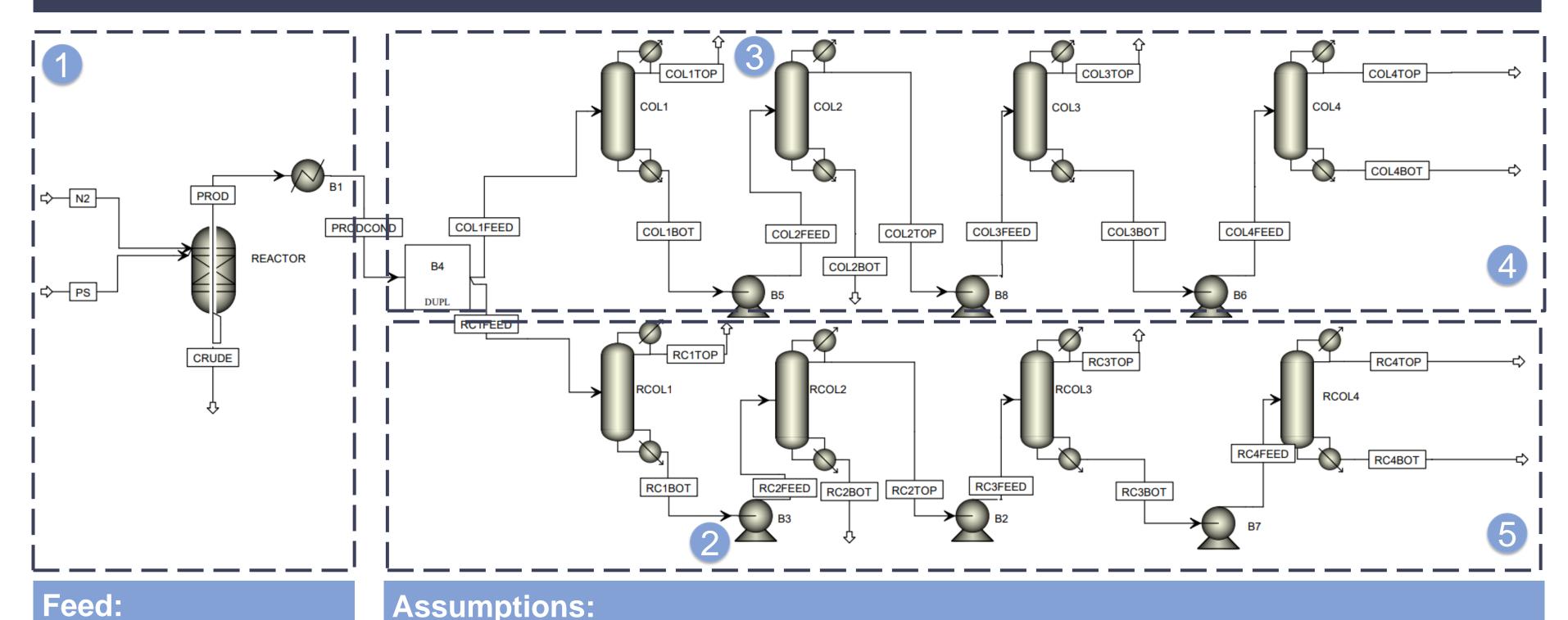
Simulate in Aspen

Validate model with literature

## **Engineering Tools**

- Aspen Plus v14 for process simulation of shortcut and waspentech rigorous design of the process
- Microsoft Excel for cost analysis and general calculation
- Python for model validation using Matplotlib, NumPy, SciPy, and pandas

# Process Flow Diagram & Analysis



Reactor

100 kg/hr N2

257 kg/hr PS

N2 and PS are the feed of the reactor for the pyrolysis process. B1 (heat exchanger) ensures the input of the distillation columns are at the set temperature.

**Product** 

**Production Location** 

Mass (tonne/year)

**Product Purity** 

Uses

## 2 Pumps

## Distillation Columns

Fixed conversion reactor

of the process. Separations in each column: 1. Toluene

Table 2: Key Analysis Results based on shortcut method design

Toluene

COL 1

158

99.71%

Paint solvent,

rubber

- - 2. Alpha methyl styrene 3. Ethylbenzene
  - 4. Cumene and Styrene

Sieve tray columns used to separate the products

Alpha Methyl Styrene

COL 2

281

99.97%

Paint, resins, coatings,

synthetic rubbers

Feed contamination is negligible

## 4 Shortcut Method

Modelling of the process using "simple" distillation columns. There are fewer parameters that can be changed:

- Specifications
- Calculation Options Convergence

Ethylbenzene

COL 3

387

99.61%

5 Rigorous Method

NRTL thermodynamic model

Steady-state

The shortcut method provides inputs for the rigorous distillation. More parameters can be changed:

- Configuration
- Streams
- Pressure
- Condenser
- Styrene Cumene COL 4 COL 4 1343 98.02% 99.97%

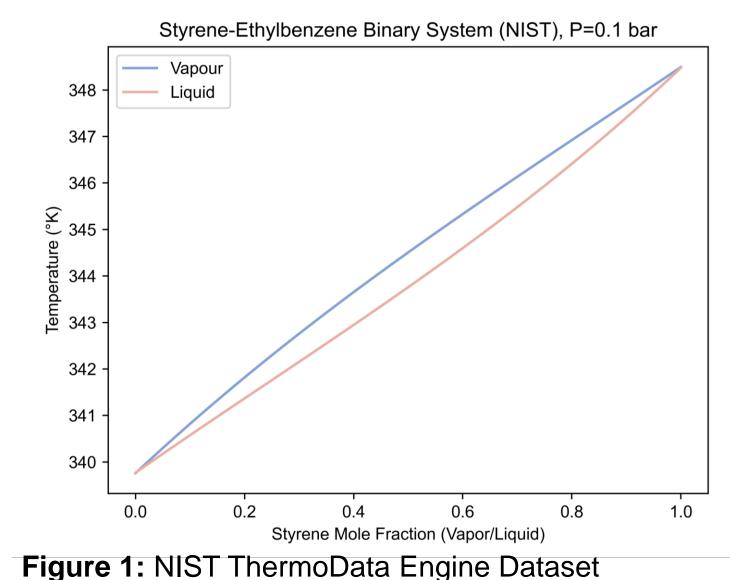
#### New PS, Phenol, Inks, dyes, rubbers perfume, new PS acetone

# Thermodynamic Model Verification

## NRTL (Non-Random Two-Liquid)

Thermodynamic model chosen for modelling vapor-liquid equilibrium:

- Effective at modelling non-ideal interactions between hydrocarbons (i.e., styrene/toluene)
- VLE data obtained from NIST ThermoData Engine [3] and NRTL Aspen Plus Simulation [4]
- Simulated temperatures closely reflect literature



Styrene-Ethylbenzene Binary System (NRTL), P=0.1 bar Figure 2: NRTL Thermodynamic Model Dataset

Fitted coefficients o VLE data used in temperature simulation

0.13%

Max. error

across

vapor/liquid

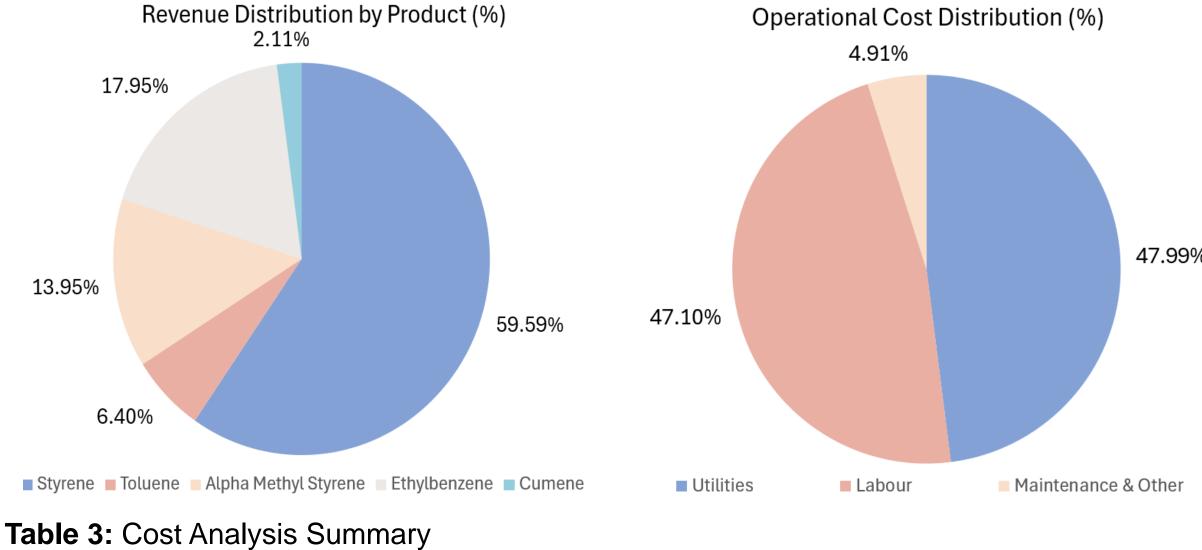
composition

for both

datasets

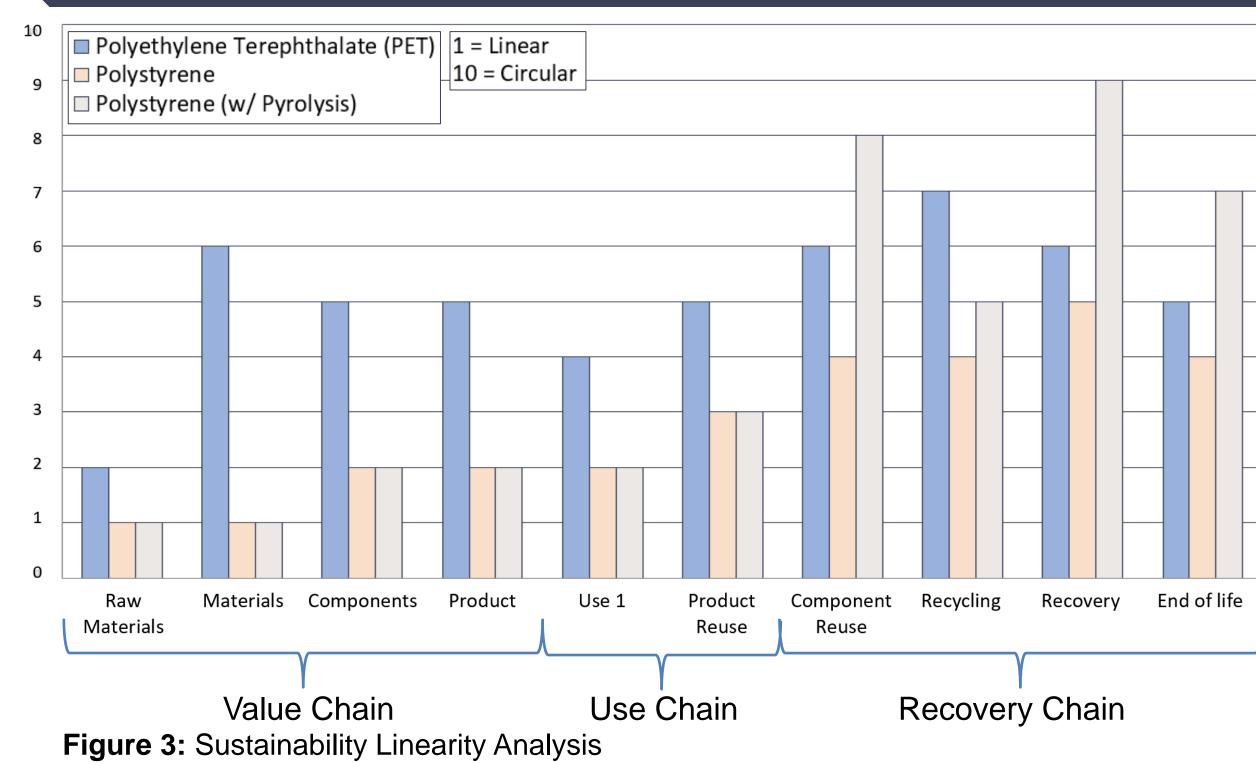
 $R^2 > = .995$ 

# Cost Analysis



Cost Analysis (\$/year) Total Equipment Cost (\$) 8,533,000 7,390,000 Revenue **Total Operational Cost** 2,113,000 6,420,000 **Profit** Payback Period: 4.1 years

# **Sustainability Linearity Analysis**



A linearity analysis was performed to evaluate a product's lifecycle from raw material to end of life. The developed pyrolysis process, improved all aspects of the recovery chain

# Conclusion

- The process effectively processes 100% of annual polystyrene waste collected by the Town of Markham.
- The process yields 98% purity of all products.
- The process improves the material circularity by improving its recyclability.

### **Future Direction**

Addition of Catalyst

 Using a decision matrix to compare the capital/operational costs, efficiency, and product quality found in literature, the recommended catalyst was found to be calcium oxide (CaO).

# Acknowledgments & References

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