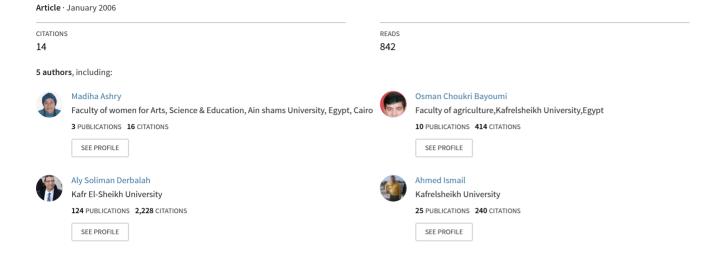
Monitoring and removal of pesticide residues in drinking water collected from Kafr-El-Shiekh governorate



MONITORING AND REMOVAL OF PESTICIDE RESIDUES IN DRINKING WATER COLLECTED FROM KAFR EL-SHEIKH GOVERNORATE, EGYPT.

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ABSTRACT

Drinking water contamination by agrochemicals particularly pesticides is an important environmental and health issue in different part of the world. Monitoring of pesticide residues was conducted at different sources of water purification plants in Kafr El-Sheikh governorate (Kafr El-Sheikh, Fowa, El-Hamoul, Mehalet Abuali and Ebshan). Furthermore, the removal of pesticide residues in drinking water by using powdered activated carbon was evaluated at both the intake and outlet of each water purification plant. The results showed that, aldrin, lindane, DDE, DDD, dimethoate, methyl parathion and malathion were the detected pesticides. Organochlorin pesticides (OCPs) detected with higher concentration organophosphates (OPPs) at all sampling sites. Aldrin and lindane were detected with higher frequency and concentration level than other compounds. The concentration levels of detected pesticides were much lower in outlets than at the intake points of water purification plants. water treatment itself may reduce or remove the pesticide residues from water. The concentration level of detected pesticides in Kafr El-Sheikh governorate still lower than the maximum residue limits (MRLs). The powdered activated carbon effectively reduces pesticide residues in drinking water, which promise as water purification technique.

Key words: Drinking water, Pesticide residues, powdered activated carbon, Organochlorine, organophosphates pesticides, removal pesticides.

INTRODUCTION

Pesticide contamination of surface water and ground water from agricultural use has been a concern for a long time. Attention is usually focused on contamination by organochlorine pesticides (OCPs) due to their toxicity and persistence in environment; and contamination by common pesticides, such as organophosphorous pesticides (OCPs) due to misuse and runoffs. Organochlorine pesticides (OCPs) are one of the most persistent organic micropollutants present in water (Gangesh *et. al.*, ۲۰۰٦). They are a possible risk to environment because of their toxicity and ability to bioaccumulation. Because of their highly persistent properties and potential

threat to human health, OCPs has prohibited for producing and using in most developed countries and OPPs are used as a substitute for OCPs in many countries nowadays because they can degrade more easily in the environment. Although OPPs as a whole are not the most toxic pollutants, they can be traced in a wide range of surface water. (Chunzhou, *et. al.*, Y···°; Sankararamakrishnan, *et. al.*, Y···° and Van Dijk-Looijaard and Van Gendren, Y···)

Pesticide residues reach the aquatic environment through direct runoff, leaching, careless disposal of empty containers, equipment washings, etc. (Miliadis, 1995). Surface water contamination may have ecotoxicological effects for aquatic flora and fauna as well as for human health if used for public consumption (Forney and Davis, 1941; Leonard, 1944; Miyamoto et. al., 1999; Mulla and Mian, 1941). Contamination of ground water resources by pesticides has brought increased environmental concern (Foster et al., 1991; Schiavon et al., 1990; Guzzella et al., 1991; Soutter and Pannatier, 1991; Papadopoulou-Mourkidou et al., 1991; The problem has become more prominent in countries where ground water aquifers constitute the main drinking water resources for rural and adjacent urban areas (Tuxen et al., 1991).

Pollution of surface and ground water is a real risk to human health because of the potential health hazards of their contents of inorganic and organic compounds. Pesticides are group of hazardous compounds that may pollute water due to their extensive application in agriculture as insecticides, fungicides, herbicides, plant growth regulators and sterilants. Although much benefit is obtained from their uses, they have some undesirable side effects such as toxicity, carcinogenity and mutagenity (Becker and Wilson, 19A.; Kouras et al., 199A). Adsorption is one of the well-known methods used in removal of such hazardous compounds from polluted water. Organochlorine pesticides remain stable in water for many years after their use Shukla *et al.*, (۲۰۰٦). Due to the long resistance time of organochlorine pesticides in the environment, their removal by different degradations processes not sufficient and high cost. However, their removal by adsorption on activated carbon may be the effective way to overcome their high persistence. Activated carbon is the most widely used adsorbent material for this purpose due to its efficiency and economic feasibility (Yoshida et al., ۱۹۹۳; Zhao et al., ۱۹۹۸).

In Egypt, drinking water limit and maximum permissible level (MPLs) are lacking adequate documentation and continuous revision and they appear to be copied from other existing list lack of logical consistency in limit set for drinking water probably due to the lack of communication

between health experts and decision-making authorities. Thus, the main objectives of this study monitor the presences of pesticide residues in surface water in Kafr El-Sheikh governorate, determination their concentration levels, to study the impact of water treatment process on the presence of pesticides in water, to evaluate the activated carbon as technique for pesticide residues removal from drinking water.

MATERIALS AND METHODS

Chemicals

Selected pesticides, organochlorine (aldrin, Lindane (γ -HCH), DDE (γ , γ -(dichloro γ , γ bis (ξ -clorophenyl) ethane), DDD (dichloro-diphenyl-dichloroethane)) and organophosphorus (dimethoate, malathion, methyl parathion, chlorpyriphos and diazinon,) were obtained from the Environmental Protection Agency (EPA), USA.

Petroleum ether, dichloromethane and n-hexane as well as anhydrous sodium sulfate were obtained from Merck Co., Germany. The chemicals used were analytical grade for pesticide residues analysis. The used powdered activated carbon having specific surface area ($^{\wedge} \cdot ^{\circ} m^{\gamma}/g$, particle size $^{\circ} \cdot ^{\circ} \cdot ^{\circ} \cdot ^{\circ}$ higher than $^{\circ} \cdot ^{\circ} \cdot ^{\circ} \cdot ^{\circ}$ and iodine number $^{\wedge} \cdot ^{\circ} \cdot ^{\circ} \cdot ^{\circ} \cdot ^{\circ}$ obtained from Merck Co., Germany.

Sample Preparation

T L water samples in cleaned, sterilized and solvent washed glass bottles from each place were collected for a year (one time every three months) from five water purification plants in Kafr El-Sheikh governorate (Kafr El-Sheikh, Fowa, El-Hamoul, Mehalet Abuali and Ebshan). Samples were collected from the intake point of each water purification plant and its outlet (finished treated drinking water). The samples were transferred to the laboratory in ice container. Samples had filtered through fiberglass filter to remove turbidity and debris; and stored at ξ °C prior to extraction.

Rosetta branch feeds the main municipal water purification plants at Fowa and Mehalet Abuali. Demeatta branch divided into Meat yazed canal (which feed Kafr El-Sheikh area) and Terra Sea canal which feed El-Hamoul and Ebshan areas. These sampling sites were selected according to their proximity to residential areas and agricultural activities.

Extraction procedure

and then 'ml mixture of petroleum ether and dichloromethane ('''' v/v) was added. The separatory funnel was shacken vigorously for about four

minutes with periodic venting to release excess pressure. The organic layer was allowed to separate for $\ '$ min. and was collected into a $\ '$ ml flask. A second $\ '$ ml of the organic solvent was added and extraction procedure was repeated twice. The combined extract was percolated through an anhydrous sodium sulphate column. The dried extract was evaporated using rotary evaporator adjusted at $\ '$ o C until the volume reached $\ '$ ml. The final extract had transferred quantitatively by rinsing with $\ '$ ml aliquots of the organic solvent into a concentrator tube and then evaporated to dryness. The residue was dissolved in n-hexane and One microliter was injected into the GC in the split-less mode. This method was according to **Tahboub**, et al, $(\ '$ \cdot \cdot \circ).

Gas chromatograph (HRGC-ECD and NPD)

Residues of the monitored pesticides had analyzed by using a gas chromatograph (GC) model HP-one equipped with a tritium electron capture detector (H-ECD) for detection of chlorinated pesticides. A nitrogen phosphorus detector (NPD) was employed to determine the organophosphorus. Megabore column Hp- 7. A part No. 19.900-177; (7.m. x ·, ο τ mm id, ·, το μm film thickness) was used. Split injection (·, Λ min. hold) was applied with temperature at YY.°C. For H-ECD the carrier gas used was helium at a flow rate of Y,o ml/min, the make up gas was nitrogen at To ml/min, anode purge, nitrogen at & ml/min., temperature was YA.°C The initial oven temperature was $\wedge \cdot \circ C$ ($\wedge \min$) $\rightarrow (\nabla \cdot \circ C/\min) \vee \vee \cdot \circ C \rightarrow (\nabla \cdot \circ C/\min) \vee \vee \circ C \rightarrow (\nabla \cdot \circ C/\min) \vee \vee \circ C \rightarrow (\nabla \cdot \circ C/\min) \vee \vee \circ C \rightarrow (\nabla \cdot \circ C/\min) \vee \circ C \rightarrow (\nabla \cdot \circ C/\min) \vee \vee \circ C \rightarrow (\nabla \cdot \circ C/\min) \vee \circ C \rightarrow (\nabla \circ C/\min) \vee \circ (\nabla \circ C/\min) \vee ($ (\'\circ'C/min) \'\cdot\'\circ'C hold \'\cdot\' min. For the NPD, the carrier gas used was hydrogen at a flow rate of \(\xi\) ml/min. and the make up gas was helium at \(\tau\). ml/min with temperature at YYo C. Initial oven temperature was ^.°C () \min) \rightarrow ($^{\circ}$ C/ \min) $^{\circ}$ C \rightarrow ($^{\circ}$ C/ \min) $^{\circ}$ C \rightarrow ($^{\circ}$ C/ \min) $^{\circ}$ A \cdot hold $^{\circ}$ C min. The percentage recovery of organochlorine and organophosphorus pesticides used were > 90% and $\wedge \wedge / \cdot$, respectively for all water samples.

Application of powdered activated carbon

Collected water samples from selected sites in winter season were used to evaluate the powdered activated carbon (PAC) for pesticides removal. Accurately weighted carbon doses (Y·±·, Ymg) were added to YL of water samples from each site (intake and outlet points). The starting time of the experiment was the time of PAC addition and kept until one hour according to Kouras, et al., (Yqqh). After the addition of PAC, stirring was kept at Y·rpm in order to keep PAC particles sufficiently suspended. Samples were filtered through ·, 50 mm membrane filter then extracted and analyzed.

RESULTS AND DISCUSSION

The results of the analysis of the water samples from Kafr El-sheikh governorate have shown the presence of both organochlorine and organophosphate pesticide residues. The compounds detected were aldrrin, lindane, DDE, DDD, Dimethoate, methyl parathion and malathion in water samples at all sampling sites either at the intake or outlet point of each water purification plant but chlorpyriphos and diazinon were not detected in all samples. Results of the sample analysis were summarized in Tables (\(\frac{1}{2}\)). The concentrations of organochlorine and organophosphorus pesticides ranged at the intake points from \(\frac{1}{2}\), \(\frac{1}{2}\) and \(\frac{1}{2}\).

The concentration range of detected pesticides in Kafr El-Sheikh area was for aldrin (*, *\(\tau_-\,\\\\), lindane (*, \(\tau_-\,\\\\\)), DDE (N.D.-\,\\\\)), DDD $(\cdot,\cdot,\cdot,\cdot,\cdot,\cdot)$, Dimethoate (N.D.- \cdot,\cdot,\cdot), methyl parathion (N.D.- \circ,\circ,\cdot) and malathion (N.D.-, 91). In Fowa area, the concentration range was for aldrin $(\cdot, 77-7, 79 \text{ ng/l})$, lindane $(\cdot, 19-77, \circ)$, DDE $(\cdot, \cdot \xi-\cdot, 17)$, DDD (N.D), Dimethoate (N.D.), methyl parathion $(\cdot, \cdot, \cdot, \cdot, \cdot, \cdot, \cdot, \cdot)$ and malathion (N.D.-, \(\frac{1}{2}\)). In El-Hamoul area showed that, the concentration range of detected pesticides was for aldrin (*, *\lambda-\\Y,\T\lambda\ ng/l), lindane (\xi, 90-\\Y,\Y\), DDE (',')-','), DDD (','o-','o), Dimethoate (N.D), methyl parathion (·,·)-), and malathion (N.D.-·, 7). The concentration range of detected pesticides in Mehalet Abuali area was for aldrin (*, \ -\ \,\ ng/l), lindane $(\Lambda, \Lambda 9-19,77)$, DDE $(\cdot, \cdot, \cdot, -\cdot, \pi \xi)$, DDD (N.D.), Dimethoate (N.D.), methyl parathion (Y, A5-Y, YT) and malathion (N.D.). Finally, the concentration range of detected pesticides in Ebshan area was for aldrin (*, ۲۲-۸, 05 ng/l), lindane (٦,٣٢-١٢,٧١), DDE (٠,٠٥-٠,٠٧), DDD (N.D.), Dimethoate (N.D.-1,1), methyl parathion (\cdot, ξ) - \vee ,90) and malathion $(N.D.-\xi, \neg 0)$.

Both the detection frequency and concentration level of organochlorine pesticides such as aldrin and lindane were higher than organophosphorus pesticides. This is due to the high persistence of organochlorine pesticides (OCs) compare to organophosphorus compounds (OPs).

The presence of DDT metabolites (DDD and DDE) instead of DDT due to itself most of the applied pesticides undergo to many transformation processes (Rand *et al.*, 1990). In addition, many pesticides eventually end up in ground water and their transformation products may remain for years (Belfroid *et al.*, 1994).

With the concerning the sampling sites, temporal analysis over a year period showed that, the organochlorine pesticides were abundant in the

following order Fowa > Mehalet Abuali > Ebshan > El-Hamoul > Kafr El-Sheikh. However, for organophosphates the abundant were as follow Kafr El-Sheikh > Fowa > Ebshan > El-Hamoul > Mehalet Abuali.

In Kafr El-Sheikh area (Table ¹), the level of pesticides residues (ng/l) in descending order was methyl parathion > DDD > dimethoate > lindane > malathion > aldrin > DDE. For fowa (Table ˁ), the level was lindane > aldrin > methyl parathion. However, in El-Hamoul area (Table ˁ), the level of residue was aldrin > lindane > methyl parathion > DDD > malathion > DDE. The ranking in Mehalet Abuali (Table ٤) depending the residue level was aldrin > lindane > methyl parathion > DDE > DDD > dimethoate > malathion. The concentration level in the last site Ebshan (Table °) was as follow lindane > aldrin > methyl parathion > malathion > DDE > DDD > dimethoate. Methyl parathion Lindane and aldrine had detected with higher detection frequency and concentration level than other compounds at all sampling area.

Table (1): Mean concentration of detected pesticide residues (ng/L) at intake and outlet of water in Kafr El-Sheikh.

Time of	Concentration (ng/L) at intake point									
samples	Aldrin	Lindane	DDE	DDD	Di.	Me	Ma.			
Spring	۰,۳۱	1,.9	٠,٢	1,17	1,10	۲,۳۳	٠,٩١			
Summer	٠,٤٨	٠,١٦	N.D	N.D	N.D	N.D	N.D			
Autumn	٠,٠٩	٠,١٢	N.D	N.D	N.D	N.D	N.D			
Winter	٠,٠٨	٠,١٢	N.D	N.D	N.D	0,07	N.D			
		Concentration (ng/L) at outlet point								
Spring	٠,٠٦	٠,٠٣	N.D	٠,٠٧	N.D	N.D	N.D			
Summer	٠,١٨	٠,٠٧	N.D	N.D	N.D	N.D	N.D			
Autumn	٠,٠٣	N.D	N.D	N.D	N.D	N.D	N.D			
Winter	٠,٠٧	N.D	N.D	N.D	N.D	١,٠	N.D			

N.D = Not detected, Di=Dimethoate, Me. = Methyl parathion, Ma= Malathion

Table ($^{\gamma}$): Mean concentration of detected pesticide residues (ng/L) at intake and outlet of water in Fowa.

Time of		Concentration (ng/L) at intake point								
samples	Aldrin	Lindane	DDE	DDD	Di.	Me	Ma.			
Spring	٧,٣٢	17,71	٠,٠٦	N.D	N.D	٠,٣٠	٠,٧٤			
Summer	٣٠,٣٩	٣٢,٥	N.D	N.D	N.D	9, ٤9	N.D			
Autumn	٤,٥٢	٩,٧٤	٠,١٣	N.D	N.D	٠,١	N.D			
Winter	٠,٤٤	٩,٢	N.D	N.D	N.D	N.D	N.D			
		Concentration (ng/L) at outlet point								
Spring	٠,٧١	٠,٢٣	N.D	N.D	N.D	٠,٢٤	N.D			
Summer	٠,٢٢	٧,٩٩	٠,٠٤	N.D	N.D	٠,٦٧	N.D			
Autumn	1,41	N.D	N.D	N.D	N.D	N.D	N.D			
Winter	٠,٢٦	٠,١٩	٠,٠٤	N.D	N.D	N.D	N.D			

N.D = Not detected, Di=Dimethoate, Me. = Methyl parathion, Ma= Malathion

Table ($^{\circ}$): Mean concentration of detected pesticide residues (ng/L) at intake and outlet of water in El-Hamoul.

Time of	Concentration (ng/L) at intake point									
samples	Aldrin	Lindane	DDE	DDD	Di.	Me	Ma.			
Spring	17,08	N.D	N.D	1,.0	N.D	١,٧	N.D			
Summer	۱۷,۳۸	17,77	٠,١	N.D	N.D	٠,٠١	٠,٦١			
Autumn	9,08	N.D	N.D	N.D	N.D	N.D	N.D			
Winter	٠,١٢	N.D	۰,۰۳	N.D	N.D	N.D	N.D			
		Concentration (ng/L) at outlet point								
Spring	۲,۰۹	N.D	N.D	N.D	N.D	N.D	N.D			
Summer	٠,٧٨	٤,٩٥	٠,٠١	N.D	N.D	N.D	N.D			
Autumn	N.D	N.D	N.D	۲,٠٥	N.D	N.D	N.D			
Winter	٠,٠٨	N.D	N.D	N.D	N.D	N.D	N.D			

N.D = Not detected, Di=Dimethoate, Me. = Methyl parathion, Ma= Malathion

The highest detected concentration during the study period at all sampling area for each detected pesticides was as follow aldrin (r), $^{\xi}$ ng/l) at Mehalet Abuali, lindane (r), $^{\circ}$ ng/l) at fowa area, DDE ($^{\circ}$, r ng/l) Mehalet Abuali, DDD ($^{\circ}$, $^{\circ}$ ng/l) at Kafr El-Sheikh, methyl parathion ($^{\circ}$, $^{\xi}$ ng/l) at Fowa area and malathion ($^{\xi}$, r o ng/l) at Ebshan area.

The concentration levels of detected pesticides were much lower at outlet points than intake points and others pesticides had not detected in outlet points. This may due to water treatment processes inside purification plants that remove or reduce the residues of pesticides in water. Spite of, some pesticides still present in drinking water after treatment, their concentration level were much lower than the maximum residue limits

(MRLs) according to Egyptian Standards, 1940. Our results concerning the level of OCs and OPs residues in purified water are in agreement with those reported by Abd-Allah and Hesham, (7... and El-Sebae and Abu Elamayem (1944).

Table (٤): Mean concentration of detected pesticide residues (ng/L) at intake and outlet of water in Mehalet Abuali.

Time of	Concentration (ng/L) at intake point									
samples	aldrin	Lindane	DDE	DDD	Di.	Me	Ma.			
Spring	١٠,٧٤	17,97	٠,١٤	N.D	N.D	N.D	N.D			
Summer	٣١,٤	19,77	٠,٣٢	N.D	N.D	٣,٧٣	N.D			
Autumn	٨, ٤	9,77	٠,١٢	N.D	N.D	N.D	N.D			
Winter	٠,٤٩	N.D	٠,٣٤	N.D	N.D	N.D	N.D			
		Concentration (ng/L) at outlet point								
Spring	N.D	N.D	N.D	N.D	N.D	N.D	N.D			
Summer	۲,٤٩	۸,۸۹	N.D	N.D	N.D	۲,۸٤	N.D			
Autumn	N.D	N.D	N.D	N.D	N.D	N.D	N.D			
Winter	٠,١	N.D	٠,٠٩	N.D	N.D	N.D	N.D			

N.D = Not detected, Di=Dimethoate, Me. = Methyl parathion, Ma= Malathion

Table (°): Mean concentration of detected pesticide residues (ng/L) at intake and outlet of water in Ebshan.

and outlet of water in Boshan.										
Time of	Concentration (ng/L) at intake point									
samples	aldrin	Lindane	DDE	DDD	Di.	Me	Ma.			
Spring	٤,٤٥	٦,٣٢	N.D	N.D	N.D	N.D	N.D			
Summer	٨,٥٤	17,71	N.D	N.D	N.D	٧,٩٥	٤,٣٥			
Autumn	٠,٨٠	N.D	N.D	N.D	N.D	N.D	N.D			
Winter	٠,١١	N.D	٠,٠٧	N.D	N.D	N.D	N.D			
		Concentration (ng/L) at outlet point								
Spring	١,٨١	N.D	N.D	N.D	١,١	N.D	N.D			
Summer	۲,۱۳	N.D	N.D	N.D	N.D	٠,٤١	N.D			
Autumn	٠,٢٢	N.D	٠,٠٥١	N.D	N.D	N.D	N.D			
Winter	N.D	N.D	٠,٠٥	N.D	N.D	N.D	N.D			

N.D = Not detected, Di=Dimethoate, Me. = Methyl parathion, Ma= Malathion

The results indicated that DDD and DDE were detected at the outlets during autumn, spite of it not detected in intake points of El-Hamoul and Ebshan areas (table r and $^{\circ}$) respectively. These results agree with those reported by Abd-Allah and Hesham (r . r).

With the concern to sampling time, it was obvious that, the concentration of identified OCPs and OPPs residues were higher in spring

and summer seasons which coincides with the extensine agricultural activities during this period Abd-Allah and Hesham, Y. . . T).

El-Sebae and Abu Elamayem (1944) recorded detectable concentrations of some organochlorine pesticides (e.g., HCB, lindane, heptachlor, p,p-DDT, o,p-DDT) in freshwater samples (raw water, treated water, tap water and wastewater) from Alexandria City, Egypt. Generally, tap water contained pesticide residues higher than those found in the water taken from the plant after treatment.

It is the worth to mention that, most of these organochlorines had virtually phased out many years ago and their presence in water residues from past application and this is due to firstly, the persistent nature of these compounds. Spite of, the organophosphate pesticides in analyzed water, they were not detected in all seasons due to the low stability of these pesticides compare to organochlorine pesticides. Based on our spatiotemporal monitoring, the detected pesticides in drinking water samples collected from different water resources exhibited fluctuations due to residental and agricultural activities along sampling points.

Secondly, the Nile water originates from the African plateau and crosses eight countries before reaching Egyptian territory (e.g., Sudan, Ethiopia, Uganda, Tanzania, Kenya, Zaire, Rwanda, and Burundi). While flowing through these countries, the Nile River is loaded with various types of pesticides and many other contaminants. Thus, it arrives in Egypt after already being contaminated with different pollutants, including the persistent chlorinated pesticides (El-Sebae *et al.*, 1997).

Thirdly, combustion of domestic wastes is a potential source of PTS in the Egyptian environment with a decreasing abundance in the order PAHs>PCBs> DDTs>HCBs>chlordane>HCHs> endosulfan (Barakat, Y., Y). Finally, the misuse of these pesticides by concerned individuals in addition to the lack of or week national control plants are behind the presence of these pesticides in water. The occurrence of such pesticide residues in drinking water represents an environmental and health hazard. Frequent monitoring program had urgently needed in order to assess health risks associated with such contaminates especially with chronic exposure or a life-long intake of contaminated drinking water.

The results in Table 7 and 7 showed that, the powdered activated carbon (PAC) significantly reduce pesticide residues in winter season samples in both intake and outlet point of each water purification plant. PAC quantitatively reduced the concentration level of detected pesticides from 7° to 7...% (such as aldrin fig 1) of their initial concentration level (before PAC addition) at all sampling areas.

Table (7): Mean concentration of detected pesticides before and after activated carbon addition to winter season water samples at intake points

detivated carbon addition to writer season water samples at make points									
Place of	Before treatment (ng/L)								
samples	Aldrin	Lindane	DDE	DDD	Di.	Me	Ma.		
Kafr El-Sheikh	٠,٠٨	٠,١٢	N.D	N.D	N.D	0,07	N.D		
Fowa	٠,٤٤	٩,٢	N.D	N.D	N.D	N.D	N.D		
El-Hamoul	٠,١٢	N.D	۰,۰۳	N.D	N.D	N.D	N.D		
M. Abuali	٠,٤٩	N.D	٠,٣٤	N.D	N.D	N.D	N.D		
Ebshan	٠,١١	N.D	٠,٠٧	N.D	N.D	N.D	N.D		
	After treatment (ng/L)								
Kafr El-Sheikh	٠,٠١	٠,٠٢	N.D	N.D	N.D	٠,٢٠	N.D		
Fowa	٠,٠٦	٠,١	N.D	N.D	N.D	N.D	N.D		
El-Hamoul	٠,٠٢	N.D	N.D	N.D	N.D	N.D	N.D		
M. Abuali	٠,٠٢	N.D	N.D	N.D	N.D	N.D	N.D		
Ebshan	N.D	N.D	٠,٠٢	N.D	N.D	N.D	N.D		

M. Abuali = Mehalet Abuali, N.D = Not detected, Di=Dimethoate, Me. = Methyl parathion, Ma= Malathion

Table (Y): Mean concentration of detected pesticides before and after activated carbon addition to winter season water samples at outlet points

activated carbon addition to writer season water samples at outlet points										
Place of		Before treatment (ng/L)								
samples	Aldrin	Lindane	DDE	DDD	Di.	Me	Ma.			
Kafr El-Sheikh	٠,٠٧	N.D	N.D	N.D	N.D	١,٠	N.D			
Fowa	٠,٢٦	٠,١٩	٠,٠٤	N.D	N.D	N.D	N.D			
El-Hamoul	٠,٠٨	N.D	N.D	N.D	N.D	N.D	N.D			
M. Abuali	٠,١	N.D	٠,٠٩	N.D	N.D	N.D	N.D			
Ebshan	N.D	N.D	٠,٠٥	N.D	N.D	N.D	N.D			
		After treatment (ng/L)								
Kafr El-Sheikh	N.D	N.D	N.D	N.D	N.D	N.D	N.D			
Fowa	۰,۰۳	٠,٠٢	٠,٠٢	N.D	N.D	N.D	N.D			
El-Hamoul	N.D	N.D	N.D	N.D	N.D	N.D	N.D			
M. Abuali	N.D	N.D	N.D	N.D	N.D	N.D	N.D			
Ebshan	N.D	N.D	N.D	N.D	N.D	N.D	N.D			

M. Abuali = Mehalet Abuali, N.D = Not detected, Di=Dimethoate, Me. = Methyl parathion, Ma= Malathion

The reduced levels of detected pesticides were much lower than maximum admissible concentration of individual pesticides in drinking water according to Egyptian Standards, 1940. The carbon dose used in this study is directly comparable with those previously reported (1 / 9 mg/L) for the removal of pesticides (Robeck *et al.*, 1970 and Kouras, *et al.*, 199A). The results in this study agree with Erol and Numan (1 ··0) they found that

the pesticides such as ametryn, aldicarb, dinoseb and diuron could be removed to a certain extent from aqueous solutions by adsorption onto the high surface area of the activated carbon. It remains to say that, suitable concentration and adequate exposure times are limiting factors determining the efficiency at activated carbon in removal of pesticide residues.

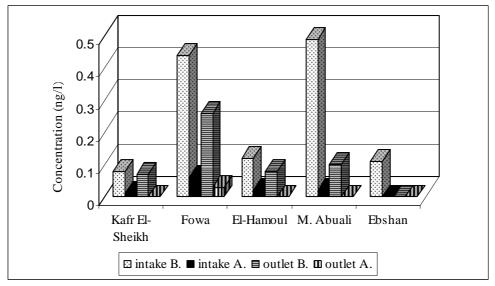


Fig (1): Mean concentration of aldrin after (A) and before (B) activated carbon addition to intake and outlet water.

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الملخص العربي

تتبع متبقيات المبيدات وازالتها في مياه الشرب التي تم تجميعها من محافظة كفر الشيخ بجمهورية مصر العربية

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يؤدى الاستخدام التزايد للكيماويات الزراعية وخاصة المبيدات الى تسربها ووصولها الى مصادر المياة السطحية والجوفية مما يشكل خطورة صحية وبيئية فى اجزاء مختلفة من العالم. فى هذه الدراسة تم اجراء رصد مكانى وزمانى للمبيدات فى مصادر مياة الشرب قبل وبعد معاملتها فى المحطات المختارة فى محافظة كفر الشيخ وايضا تم تقيم الفحم (الكربون النشط) كطريقة للتخلص من متبقيات المبيدات فى مياة الشرب سواء قبل او بعد معاملتها فى محطات التنقية. واوضحت نتائج تحليل العينات وجود متبقيات للمبيدات الكلورينية الالدرين واللندين والـ DDD و الـ DDD و الـ DDD و المركبات الفوسفورية الداى مثويت و الميثايل باراثيون والملاثيون. المركبات الكلورينية التى تم اكتشافها كانت اعلى فى التركيز من متبقيات المبيدات الفوسفورية العضوية. الالدرين واللندين تم اكتشافهم بتركزات عالية بالمقارنة بالمركبات الاخرى كما وجد ان عمليات المعالجة للمياة فى داخل المكتشفة فى محافظة كفر الشيخ يبقى اقل من الحد الاقصى للتركيز المسموح به فى مياه الشرب المكتشفة فى محافظة كفر الشيخ يبقى اقل من الحد الاقصى للتركيز المسموح به فى مياه الشرب الكربون النشط ادى الى انخفاض فى تركيزات المبيدات فى مياة الشرب بدرجة كبيرة مما يثبت انها المربقة فعالة فى تنقبة المياة