

# Investigation of a novel concept for carbothermic reduction of alumina

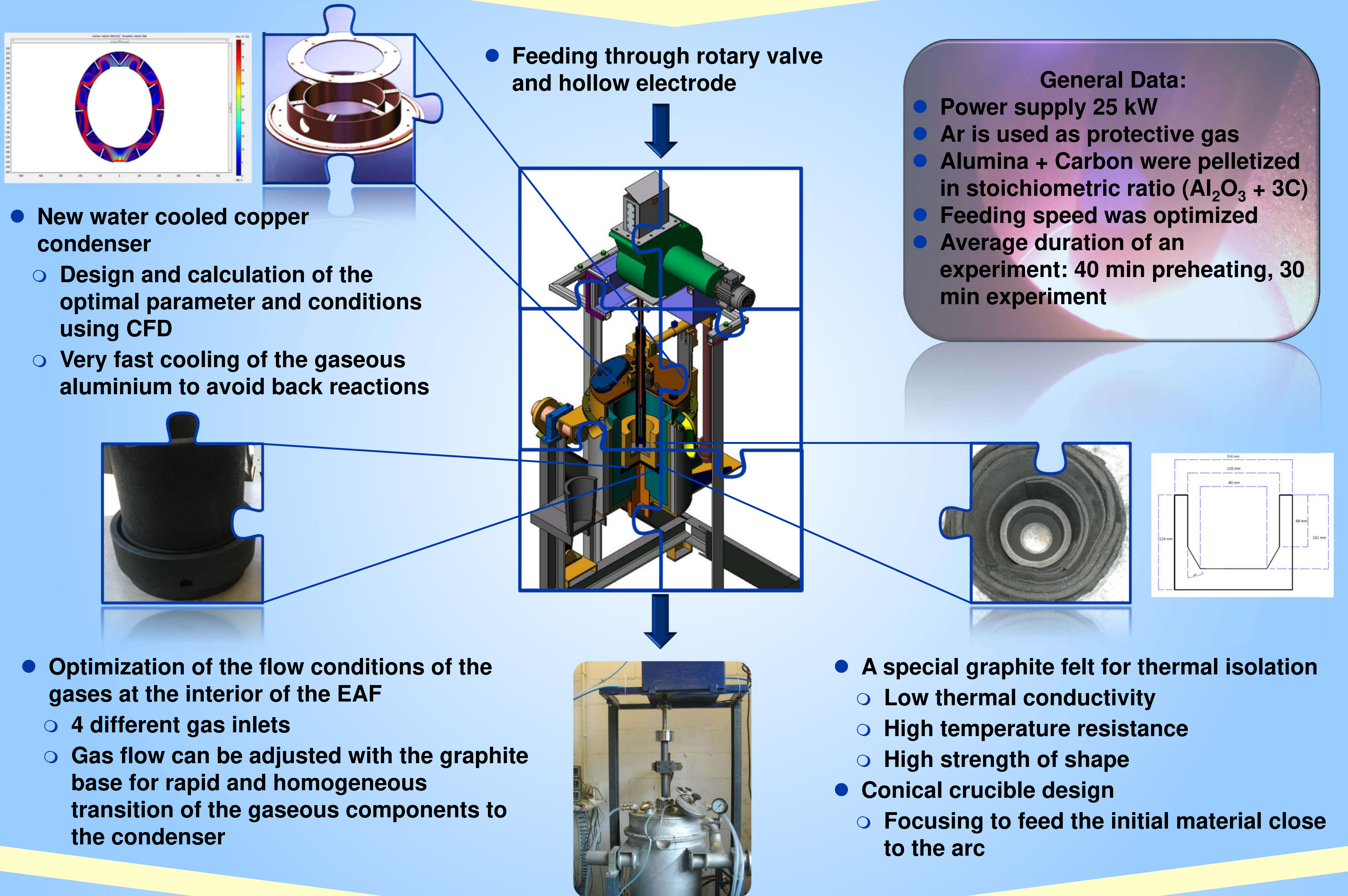
## Motivation

- Conventional Hall-Héroult process is one of the most energy and CO<sub>2</sub> intensive processes
  - No alternative aluminium production process was able to prevail in industrial scale
  - The most promising alternative is the carbothermic reduction of alumina in an EAF, which is characterized by high flexibility and power density as well as providing the required process temperatures
  - A thermodynamic calculation indicates that the aluminium comproportionation reaction is preventing the complete alumina reduction in the liquid phase and leading to vapour losses  
( $T = 2250^{\circ}\text{C}$ ;  $\text{Al}_2\text{O}_{3(l)} + 5\text{Al}_{(l)} = 3\text{Al}_2\text{O}_{(g)} + \text{Al}_{(g)}$ )
- Suppression of liquid metal formation → Suppression of  $\text{Al}_2\text{O}_{(g)}$  formation → higher alumina reduction yields → processing above  $2500^{\circ}\text{C}$  → decreasing oxygen content of the system



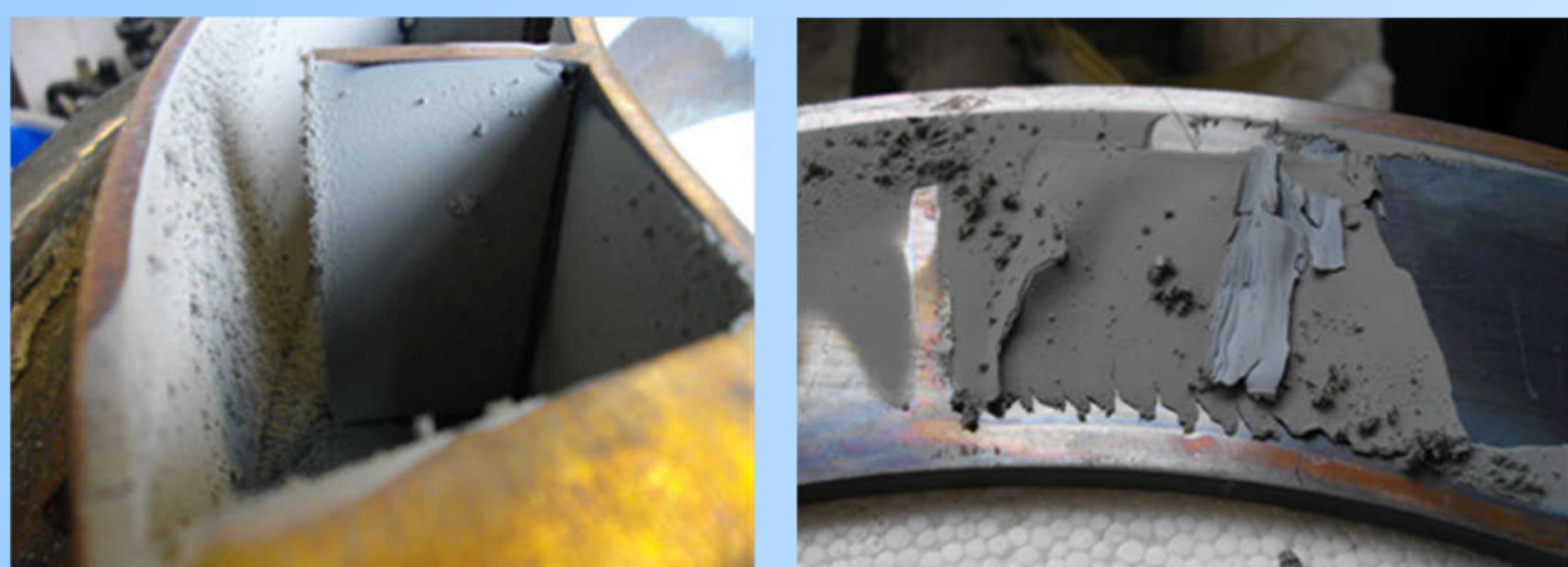
→ **Al generation over the gaseous phase under protective gas**

## Process development



## Results

### Condensed Aluminium

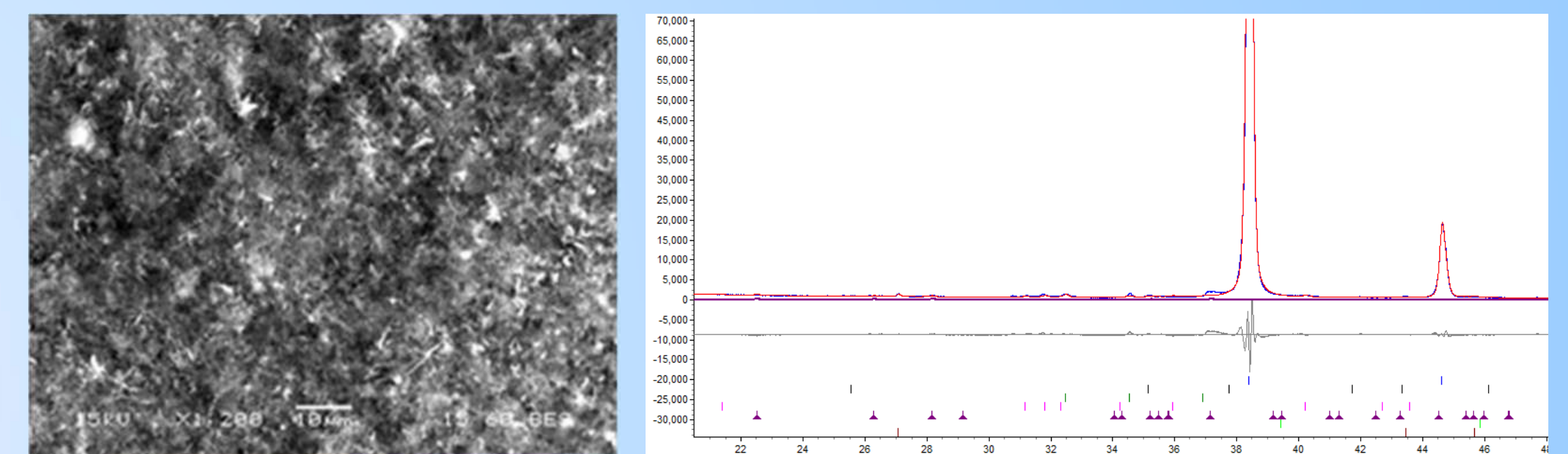


- Fast and clean physical vapour deposition
- Successive optimization of the specific parts led to increasing Al contents in the condensed material

### Progress in metal content

5 wt.-% Al  
15 wt.-% Al  
40 wt.-% Al  
60 wt.-% Al  
>90 wt.-% Al

### SEM Analysis



- After realization of the aforementioned processing and optimization steps: high amount of Al, no corundum, no graphite, no cubic  $\text{Al}_2\text{O}_3$ , minimal amounts of  $\text{Al}_2\text{OC}$  and  $\text{Al}_4\text{C}_3$ , traces of  $\text{Al}_4\text{O}_4\text{C}$

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**Successful generation of Al in a 25 kW lab-scale EAF ✓**