Current Situation

- Potential cytotoxic behavior of vanadium when using alloy Ti-6Al-4V as implant material for bio applications
- Comparatively low strength and wear properties of CP titanium in contrast to alloy Ti-6Al-4V





Artificial hip joint

Artificial knee joint

Background

- Titanium shows an extremely high solubility of oxygen
- Oxygen is dissolved interstitially and promotes the formation of hcp α -structure
- → Strength and hardness of Ti are increased
- → Simultaneously ductility is decreased

Motivation

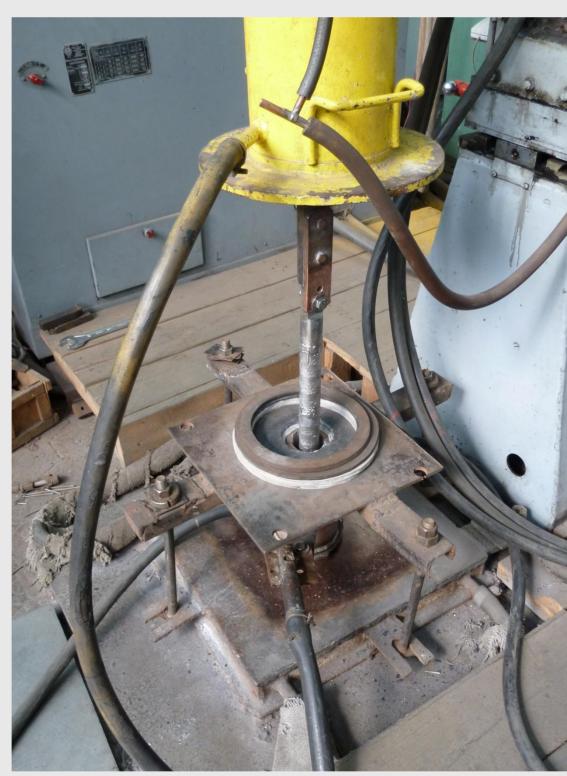
- Intentional addition of oxygen to increase the strength of pure titanium
- Further improving of ductility by a currently investigated method

Experimental Procedure

- Remelting of Ti-sponge electrodes under a technically pure CaF₂-slag
- Inserting of oxygen during remelting by different methods of TiO₂ addition:
 - constitution of the powder (conventional powder vs. nano-powder)
 - type of introduction (via electrode vs. via slag)
 - → amount of powder
- Oxygen pick up by dissolution of TiO₂
 (TiO) _ (Til + 2 (O)

 $(TiO_2)_{slag} = [Ti]_{Ti} + 2 [O]_{Ti}$

 Homogenous solidification structure provides uniform distribution of oxygen in the as-cast material







Remelted ingot showing a smooth surface

Process Characteristics

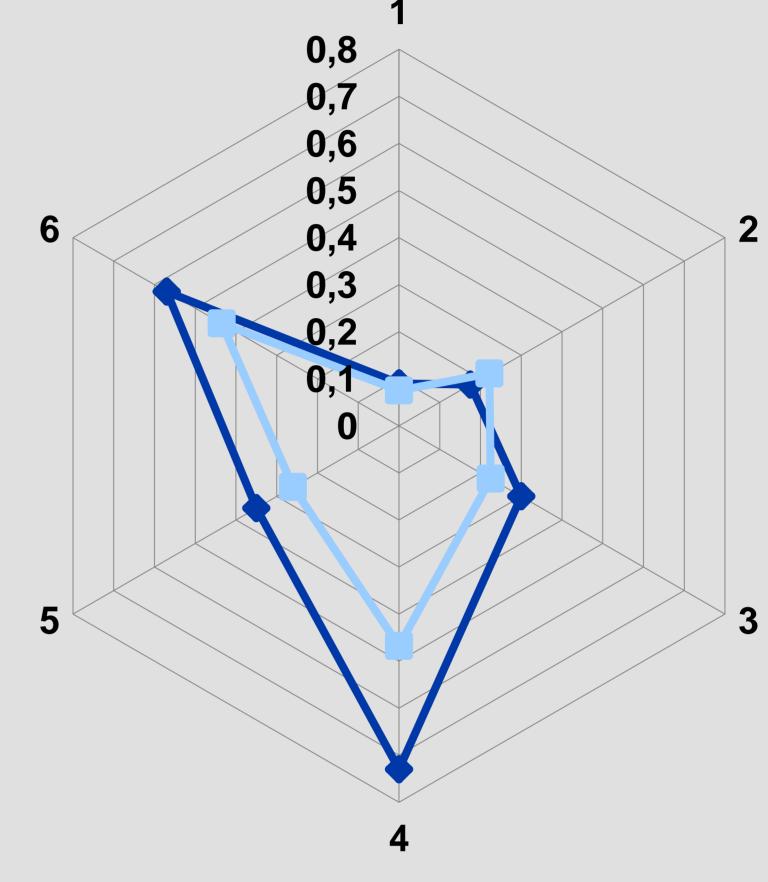
- Power supply: 724 kW
- Electrode weight: 2.5 kg
- Utilized amount of CaF₂ slag: 400 g
- Protective argon atmosphere

Results

Analyzed and calculated oxygen content in the remelted material / wt%

Ca deoxidized during remleting

24 g conventional TiO₂ powder pressed with Ti electrode



10.82 g TiO₂ nano-powder addition to CaF₂-slag

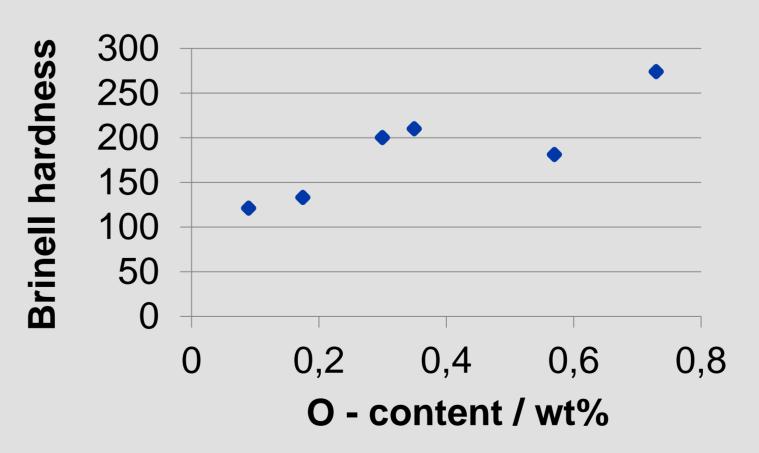
13.66 g conventional TiO₂ ₅ powder filled into an axial drilling in the electrode

10.82 g TiO₂ nano-powder filled into an axial drilling in the electrode

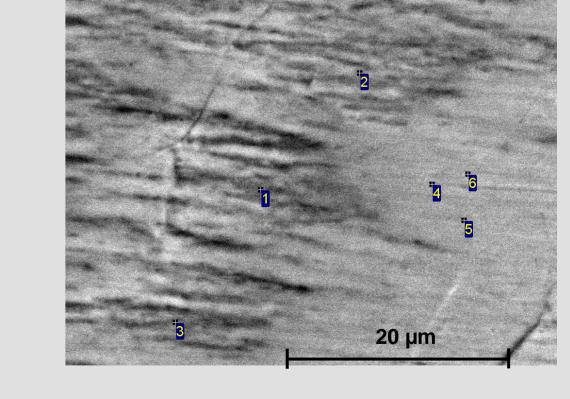
Analyzed O contentCalculated O content

23.17 g TiO₂ nano-powder filled into an axial drilling in the electrode

- → Increasing oxygen content with rising TiO₂ addition
- → Higher oxygen pick up by TiO₂ introduction via the electrode compared to previous mixing with the CaF₂-slag
- → No significant difference between conventional and nano-powder addition
- → Deviation between analyzed and calculated oxygen content due to minor CaF₂ impurities and residual atmospheric oxygen



Increased hardness with higher oxygen contents



Complete dissolution of TiO₂ particles in the ingot material (no TiO₂ inclusions detected)



Dipl.-Ing. S. Radwitz
IME Process Metallurgy and Metal Recycling
RWTH Aachen University, Germany

Prof. A.D. Ryabtsev, Prof. A.A. Troyansky
Department of Electrometallurgy
Donetsk National Technical University, Ukraine





