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Mastering Chemistry-Solubility and Intermolecular Forces ...

To find the difference in pressure between methane and an ideal gas, you subtract the real pressure of methane from ideal pressure. $78.848 \text{ atm} - 67.3964 \text{ atm} = 11.45 \text{ atm}$ or -11.45 atm The official answer on Mastering Chemistry is the negative one, however, if you put in the positive 11.45, it still counts as correct.

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