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Chapter 11 Small-Scale Lab
Practice Problems (Chapter 5): Stoichiometry CHEM 30A Part I: Using the conversion factors in your tool box g A mol A mol A 1. How many moles CH_3OH are in 14.8 g CH_3OH? 2. What is the mass in grams of 1.5 x 1016 atoms S? 3. How many molecules of CO_2 are in 12.0 g CO_2? 2 4. What is the mass in grams of 1 atom of Au? KEY Tool Box: To ...

Practice Problems (Chapter 5): Stoichiometry
Textbook pages: Chapter 12 Key Terms: stoichiometry. mole-mole problems. mass-mass problems. mass-volume problems. volume-volume problems. Directions: Use this information as a general reference tool to guide you through this unit. By the conclusion of this unit, you should know the following: Quantitative relationships exist in all chemical ...

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12.2 Chemical Calculations 12 - Evaluation 2016

e. 12 g 7. How many grams of nitric acid, HNO 3, can be prepared ... Practice Test Ch3 Stoichiometry (page 3 of 3) 1. d It might be easiest to balance the equation with mostly whole numbers: 2 NH 3 + ⁷⁄₂O ... 7. c First you must realize this is a limiting reactant problem. You can tell this since you are given quantities for both reactants.

Practice Test Ch 3 Stoichiometry Name Per

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