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CHAPTER 11 REVIEW Molecular Composition of Gases MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. The average speed of a gas molecule is most directly related to the . (a) polarity of the molecule (b) pressure of the gas (c) temperature of the gas (d) number of moles in the sample 2.

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CHAPTER 11 Molecular Composition of Gases ... the ideal gas law, as you will see in Section 11-2. MOLECULAR COMPOSITION OF GASES 335 Hydrogen gas 1 Volume 1 Molecule Chlorine gas 1 Volume 1 Molecule ... SECTION REVIEW 1. Explain Gay-Lussac's law of combining volumes. 2. State Avogadro's law and explain its significance.

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CHAPTER 11 REVIEW Gases SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. c The molar mass of a gas at STP is the density of that gas (a) multiplied by the mass of 1 mol. (c) multiplied by 22.4 L. (b) divided by the mass of 1 mol. (d) divided by 22.4 L.

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Chapter 11 - Molecular Composition of Gases 11-1 Volume-Mass Relationships of Gases I. Measuring and Comparing the Volumes of Reacting Gases A. Observations of Gay-Lussac 1. 2 liters H₂ + 1 liter O₂ → 2 liters H₂O vapor 2. 2 volumes H₂ + 1 volume O₂ → 2 volumes H₂O vapor 3. 1 volume H₂ + 1 volume Cl₂ → 2 volumes HCl 4.

Chapter 11 - Molecular Composition of Gases

CHAPTER 11 REVIEW Gases SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. State whether the pressure of a fixed mass of gas will increase, decrease, or stay the same in the following circumstances: ____ a. temperature increases, volume stays the same

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