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Investigate how calorimetry can be used to find relative specific heat values when different substances are mixed with water. Modify initial mass and temperature values to see effects on the system. One or any combination of the substances can be mixed with water. A dynamic graph (temperature vs. time) shows temperatures of the individual substances after mixing.

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Student Exploration: Calorimetry Lab Vocabulary: calorie, calorimeter, joule, specific heat capacity Prior Knowledge Questions (Do these BEFORE using the Gizmo.) 1. The Latin word calor means "heat," and meter comes from the Greek word meaning "to measure." What do you think a calorimeter does? Measures heat 2. Where have you heard the word calorie before?

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Gizmo Warm-up A calorimeter is an insulated container filled with a liquid, usually water. When a hot object is placed in the calorimeter, heat energy is transferred from the object to the water and the water heats up. Calorimeters can be used to find a substance's specific heat capacity. You will use the Calorimetry Lab Gizmo™ to

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Calorimetry Virtual Lab - BetterLesson

A metal of mass 13.11g and at 81 degrees Celsius is transferred to a calorimeter containing 25.0mL of deionized water at 25 degrees celsius. The final equilibrium temperature is 30.0 degrees celsius. What is the specific heat capacity of the metal? The specific heat capacity of H2O is 4.18 J/(g*degrees celsius). Assume the density of water to 1.00g/mL.

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