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Problem Example 1. The Normal Saline solution used in medicine for nasal irrigation, wound cleaning and intravenous drips is a 0.91% (w/v) solution of sodium chloride in water. How would you prepare 1.5 L of this solution? Solution: The solution will contain 0.91 g of NaCl in 100 mL of water, or 9.1 g in 1 L. Thus you will add $(1.5 \times 9.1\text{g}) = 13.6\text{ g}$ of NaCl to 1.5 L of water.

Solutions and Concentrations - Chem1

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$K = ^\circ\text{C} + 273$ $F = ^\circ\text{C} \times 1.8 + 32$ Pressure, simple mercury barometer. Pressure is the force exerted over an area: $P = F/A$ Due to gravity, the atmosphere exerts a pressure of 101 kPa at sea level.

Phases and Phase Equilibria - MCAT Review

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Activity coefficient - Wikipedia

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Undergraduate-level courses, in alphabetical order - NMHU

AFR 101 - The Foundations of the U.S. Air Force I This is a survey course designed to introduce students to the United States Air Force and provides an overview of the basic characteristics, missions, and organization of the Air Force.

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