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### **Chemistry - Wikipedia**

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dissolved in water. Sucrose is a nonelectrolyte, which means that the solution contains whole C<sub>12</sub>H<sub>22</sub>O<sub>11</sub> molecules. In predicting the expected freezing point of a solution, one must consider not only the number of formula units present, but also the number of ions that result from each formula unit, in the case of ionic compounds.

### **Colligative Properties - Chemistry Encyclopedia - water ...**

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### **Metalloid - Wikipedia**

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