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Chapter 12 - Chemical Kinetics . 12.1 Reaction Rates . A. Chemical kinetics 1. Study of the speed with which reactants are converted to products B. Reaction Rate 1. The change in concentration of a reactant or product per unit of time [ ] t A t t concentration of A at time t concentration of A at time t Rate  $\Delta A = - \frac{\Delta A}{\Delta t} = \frac{1}{2} \frac{\Delta A}{\Delta t}$  1. a ...

**Chapter 12 - Chemical Kinetics - ScienceGeek.net**

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AP Chemistry Test (Chapter 12) Multiple Choice (40%) 1) Which of the following is a kinetic quantity? A) Enthalpy B) Internal Energy C) Gibbs free energy D) Entropy E) Rate of reaction 2) Of the following questions, which ones are thermodynamic, rather than kinetic concepts? I) Can substances react when we put them together?

**AP Chemistry Test (Chapter 12) Multiple Choice (40%)**

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