

Mastering Chemistry Answers Ch 13

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CH 8 HW 1 Item 13 Calculate the $[\text{OH}^-]$ of each aqueous solution with the following $[\text{H}_3\text{O}^+]$ Express the answer using 2 sig figures Part A: coffee, $1.2 \times 10^{-5} \text{ M}$ Part B: soap, $1.0 \times 10^{-8} \text{ M}$ Part C:...

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