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An aldol condensation is a condensation reaction in organic chemistry in which an enol or an enolate ion reacts with a carbonyl compound to form a β -hydroxyaldehyde or β -hydroxyketone (an aldol reaction), followed by dehydration to give a conjugated enone.

Aldol condensation - Wikipedia

A free-radical reaction is any chemical reaction involving free radicals. This reaction type is abundant in organic reactions. Two pioneering studies into free radical reactions have been the discovery of the triphenylmethyl radical by Moses Gomberg (1900) and the lead-mirror experiment described by Friedrich Paneth in 1927. In this last experiment tetramethyllead is decomposed at elevated ...

Free-radical reaction - Wikipedia

ABSTRACT. Based on the reaction products of ethylenediaminetetraacetic acid (EDTA) with 1,2-phenylenediamine (o-PDA), a novel thermal decomposition pathway of EDTA is proposed. The strong acidic medium and the presence of o-PDA facilitate the decomposition of EDTA as evidenced by the relatively lower reaction temperature. In addition to the steps described in literatures, rearrangement process ...

Thermal decomposition of ethylenediaminetetraacetic acid ...

However, with steric hindered Grignard reagents, the reaction may proceed by single-electron transfer.. Grignard reactions will not work if water is present; water causes the reagent to rapidly decompose. So, most Grignard reactions occur in solvents such as anhydrous diethyl ether or tetrahydrofuran (THF), because the oxygen in these solvents stabilizes the magnesium reagent.

Grignard reaction - Simple English Wikipedia, the free ...

A detailed investigation was performed on benzyloxybenzene by reacting it with sodium borohydride and different metal salts in methanol at 0 °C (). Salts of transition metals such as Fe, Co, Ni, Pd and Cu were evaluated (). $\text{NiCl}_2 \cdot 6\text{H}_2\text{O}$ provided maximum yields of phenol in 5 min, while other metal salts gave inferior yields. An examination of the effect of the solvents on the outcome of the ...

$\text{NiCl}_2 \cdot 6\text{H}_2\text{O}/\text{NaBH}_4$ in methanol: a mild and efficient ...

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The use of polar aprotic solvents in acid-catalyzed biomass conversion reactions can lead to improved reaction rates and selectivities. We show that further increases in catalyst performance in ...

Effects of chloride ions in acid-catalyzed biomass ...

Figure 1. Mediator and BRD4 Co-occupy Promoters of Active Genes in Multiple Myeloma (A) Gene tracks of MED1, BRD4, H3K27Ac, and H3K4Me3 ChIP-seq occupancy at the enhancer (left) and promoter (right) of SMARCA4 in MM1.S MM cells. The x axis shows genomic position, and enhancer-containing regions are depicted with a white box.

Selective Inhibition of Tumor Oncogenes by Disruption of ...

A polar aprotic solvent does not hydrogen bond to nucleophiles to a significant extent, meaning that the nucleophiles have greater freedom in solution. Under these conditions, nucleophilicity

correlates well with basicity – and fluoride ion, being the most unstable of the halide ions, reacts fastest with electrophiles.

What Makes A Good Nucleophile? - Master Organic Chemistry

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Clarithromycin Monograph for Professionals - Drugs.com

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