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___ 28. The type of reaction that takes place when one element reacts with a compound to form a new compound and a different element is a _____. a. combination reaction c. single-replacement reaction b. decomposition reaction d. double-replacement reaction ___ 29. In a double-replacement reaction, the _____.

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Modern Chemistry 71 Chapter Test Name Class Date Chapter Test B, continued 17. A substance combines with oxygen, releasing a large amount of energy as heat and light, in a(n) . 18. The decomposition of a substance by an electric current is called . 19. A(n) orders the elements by the ease with which they undergo certain chemical reactions. 20.

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Distinguish between the single- and double-replacement reactions indicated below by completing each, identifying each as to type, and explaining, in words, what has occurred in each.

05 CTR ch11 7/9/04 3:33 PM Page 281 CHEMICAL REACTIONS 11

Chapter 11 - Chemical Reactions - Standardized Test Prep - Page 381: 6 Answer To write the net ionic equation for the formation of the gas, react potassium carbonate with hydrochloric acid.

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Chemical Reactions Review IDENTIFY THE TYPE OF REACTION AND BALANCE THE EQUATION: 1. Sb + I 2 Sbl 3 2. Li + H ... 11. Solid potassium nitrate yields solid potassium nitrite and oxygen gas. 12. Calcium metal reacts with chlorine gas to produce solid calcium chloride. ... Chemistry Test Review Sheet - Chapters 10 & 11 Author:

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