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Stoichiometry Practice Worksheet Solve the following stoichiometry grams-grams problems: 1) Using the following equation: 2 NaOH + H 2SO 4 2 H 2O + Na 2SO 4 How many grams of sodium sulfate will be formed if you start with 200.0

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Practice Problems: Stoichiometry (Answer Key) Balance the following chemical reactions: a. 2 CO \pm O 2 2 CO 2 b. 2 KNO 3 2 KNO 2 \pm O 2 c. 2 O 3 3 O 2 d. NH 4 NO 3 N 2 O \pm 2 H 2 O e. 4 CH 3 NH 2 \pm 9 O 2 4 CO 2 \pm 10 H 2 O \pm 2 N 2 f. Cr(OH) 3 \pm 3 HClO 4 Cr(ClO 4) 3 \pm 3 H 2 O Write the balanced chemical equations of each reaction:

Practice Problems: Stoichiometry (Answer Key)

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Chapter 12 Stoichiometry 12.1 The Arithmetic of Equations 12.2 Chemical Calculations 12.3 Limiting Reagent and Percent Yield excess of 1800 is a logical answer. • The unit of the known (FSW 3 HP 2) cancels. • The answer has the correct unit (W). 3 Evaluate Does the result make sense?

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