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Investigate how calorimetry can be used to find relative specific heat values when different substances are mixed with water. Modify initial mass and temperature values to see effects on the system. One or any combination of the substances can be mixed with water. A dynamic graph (temperature vs. time) shows temperatures of the individual substances after mixing.

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Gizmo Warm-up A calorimeter is an insulated container filled with a liquid, usually water. When a hot object is placed in the calorimeter, heat energy is transferred from the object to the water and the water heats up. Calorimeters can be used to find a substance's specific heat capacity. You will use the Calorimetry Lab Gizmo™ to

Student Exploration: Calorimetry Lab

When a hot object is placed in the calorimeter, heat energy is transferred from the object to the water and the water heats up. Calorimeters can be used to find a substance's specific heat capacity. You will use the Calorimetry Lab Gizmo™ to determine the specific heat capacities of

various substances. 1. On the SIMULATION pane, select Copper.

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When a hot object is placed in the calorimeter, heat energy is transferred from the object to the water and the water heats up. Calorimeters can be used to find a substance's specific heat capacity. You will use the Calorimetry Lab Gizmo™ to determine the specific heat capacities of various substances. 1. On the SIMULATION pane, select Copper.

CalorimetryLabSERyanV - Name Ryan Villani Date Student ...

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Chemistry is fun. I will answer the question so that there is no Way To Fail the lab. Here are my Words To Follow Step One: The formula we will be using today is $W = T(\Delta F)$ where W is the work done, T is the temp change, and F represents mass. Plugging in our values of -13.11 and 35.5 to get our change in mass, we gain an answer of 22.1 moles.

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