

## ***Bond Energy Pogil Answers Key***

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As a final assessment I skip question 17 and the extension questions, and give them with a take-home quiz (here is the answer key) that summarizes all four models. The following day we grade it as a class so they can they receive immediate feedback.

### **answer key - BetterLesson**

POGIL on CHEMICAL BONDING. ... key idea: lattice energy - the energy released when an ionic solid is formed from its gaseous ions. This is the most favorable energy term in the process. ... Using bond energy data calculate H for the following reaction. Compare this to the H calculated using thermodynamic values.  $\text{CH}_4 + 4\text{F}_2 \rightarrow \text{CF}_4 + 4\text{HF}$ .

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Bond energy is defined as the amount of energy required to break a bond. These values are positive, indicating that bond breaking is endothermic. Bond energies are reported in kilojoules per mole (kJ/mol). The energy for breaking a hydrogen-hydrogen bond is 436 kJ/mol so when a hydrogen-hydrogen bond is formed the process releases 436 kJ/mol.

**Bond Energy Name Chem Worksheet 16-2**

Chem 115 POGIL Worksheet - Week 11 - Solutions Chemical Bonds Key Questions 1. Using the electronegativity values in the table above, calculate  $\delta\chi$  for the pairs of elements forming bonds in each of the following species, and describe the type of bonding as either pure covalent, polar covalent, or ionic: H<sub>2</sub>, HCl, PbBr<sub>2</sub>, NaCl, NCl<sub>3</sub>.

**Chem 115 POGIL Worksheet - Week 11 - Solutions Chemical Bonds**

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- POGIL (Process Oriented Guided Inquiry Learning) ... reads the POGIL out loud. The instructor will answer questions from the leader only.
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