Chapter 11 Review Gases Mixed Answers

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CHAPTER . 11 . REVIEW . Gases . SHORT ANSWER MIXED REVIEW continued . 4. 226 ml A 250. mL sample of gas is collected at 57° C. What volume will the gas sample occupy at 25° C? 5. 0_._7_l H 2 reacts according to the following equation representing the synthesis of

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CHAPTER 11 REVIEW Gases MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Consider the following data table: Approximate pressure (kPa) Altitude above sea level (km) 100 0 (sea level) 50 5.5 (peak of Mt. Kilimanjaro) 25 11 (jet cruising altitude) 0.1 22 (ozone layer) a.

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Temperature of gas depends on average kinetic energy of particles Ch 10 pg 329-330 Parameters for an ideal gas Kinetic molecular theory only applies to "ideal" gases. Most gases are only nearly ideal if pressure is not too high and temperature is not to low

Gases - Los Angeles County High School for the Arts

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CHAPTER 11 REVIEW Molecular Composition of Gases MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. The average speed of a gas molecule is most directly related to the . (a) polarity of the molecule (b) pressure of the gas (c) temperature of the gas (d) number of moles in the sample 2.

11 Molecular Composition of Gases - Madison Public Schools

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Review Previous Concepts Gases CHAPTER 11 1. What is a mole? What information does the molar mass of a ... 344 CHAPTER 11. Dalton's Law of Partial Pressures The pressure exerted by each gas in a mixture is called the ... of 0.28 atm when mixed with one or more gases. Dalton's law of partial pressures states that the total

CHAPTER 11 Gases - St. Charles Parish

CHAPTER 11 REVIEW Gases SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. State whether the pressure of a fixed mass of gas will increase, decrease, or stay the same in the following circumstances:

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Chapter 11 181 Chapter 11 - Gases Review Skills 11.1 Gases and Their Properties Ideal Gases Properties of Gases Discovering the Relationships Between Properties The Relationship Between Volume ... neon gas mixed with argon gas. (Objs 21 & 22) a. If the total pressure of the mixture of gases is 1.30 kPa and the partial pressure of neon

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