

## ***Chapter 10 Physical Characteristics Of Gases Answer Key***

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306 CHAPTER 10. Diffusion is a process by which particles of a gas spread out spontaneously and mix with other gases. In contrast, effusion is a process by which gas particles pass through a tiny opening. The rates of effusion of different gases are directly proportional to the velocities of their particles.

## **CHAPTER 10 Physical Characteristics of Gases**

ideal gas an imaginary gas that perfectly fits all the assumptions of the kinetic molecular theory  
ideal gas based on the kinetic molecular theory  
real gas don't follow the rules  
elastic collisions  
collisions where there is no net loss of kinetic energy  
attraction or repulsion there are no forces of \_\_\_\_\_ or \_\_\_\_\_ between gas [...]

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\ Chapter 10: Physical Characteristics of Gases. Chapter 10: Physical Characteristics of Gases. The kinetic-molecular theory of matter can be used to explain the properties of solids, liquids, and gases in terms of the \_\_\_\_\_ of the particles and the \_\_\_\_\_ that act between them.

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### **10-1 The Kinetic-Molecular Theory of Matter A. - Intro**

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Chapter 10 – Physical Characteristics of Gases 1. ChemistryS. Martinez – Spring 2009 1 2. KMT – based on idea that particles of matter are always in motion. 2 3. 1. Gases consist of large numbers of tiny particles that are far apart relative to their size.2.

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CHAPTER 1. PHYSICAL CHARACTERISTICS Wild ruminants have many physical characteristics that are important adaptations for survival and production in their natural environments. Weights vary as fat reserves are accumulated and used in relation to seasonal variations in the availability of food resources.

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