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By raising the pH using something basic, you'll strip the grease molecules of their protons giving them a polar functional group. Grease compounds are very unlikely to ever become soluble, but with polar functional groups they will be able to emulsify.

### **Mark Rosengarten**

Unit 12: Electrochemistry-Key Regents Chemistry '14-'15 Mr. Murdoch Page 7 of 61 Key Electricity: Why study electricity in chemistry? Isn't that a physics topic? Well, yes it is, as I have taught

Regents physics as well. But to understand what you can DO with electricity in physics, you need to understand

### **Unit 12: Electrochemistry-Key Regents Chemistry '14 Mr ...**

All work must be shown. including numerical setup. units and answers must be properly rounded. A) Calculate the molarities of the following solutions: 1) 2 moles of NaCl in 4 L of solution 2) 34 grams of  $\text{CaSO}_4$  in 2 L of solution 3) 28 grams of KOH in 100. ml- of solution 4) 3,65 g HCl dissolved in 0.500 L of solution

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Check your answers at the end of the activity. Alien Juice Bar - A simple interactive that reviews acids, bases and the pH scale. It allows you to practice identifying acids, bases and neutral solutions. A stress-free way to review. ... YouTube Video by Mark Rosengarten. pH Scale. pH and pOH (Parts 1 and 2)

### **Unit 12 Acids and Bases - Ms. Kinsella - Google Sites**

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