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1. Calculate the number of moles of baking soda used in the lab. 2. Calculate the number of moles of sodium chloride produced in the lab. 3. What is the experimental mole ratio of baking soda (NaHCO_3) to sodium chloride (NaCl)? 4. Write a balanced equation for the reaction that took place in the experiment. 5.

Lab: Episode 801—Datasheet Name Can You Find the Mole Ratio?

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8-01,02,03 Note Taking Guide Ep 801 - Mole Cafe

Lab Episode 801 Datasheet Answers Ebook Lab Episode 801 Datasheet Answers currently available at spaldingchamber.co.uk for review only, if you need complete ebook Lab Episode 801 Datasheet Answers please fill out registration form to access in our databases. Summary : Delegation strategies for the nclex prioritization for the nclex infection

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Lab: Episode 801—Datasheet Name Can You Find the Mole Ratio? In this experiment, baking soda (NaHCO_3) reacts with hydrochloric acid (HCl) to produce sodium chloride, carbon dioxide, and water. You will determine the moles of reactant used and moles of product produced using both the lab data and stoichiometry and compare the two. Data Table:

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