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This question is a little easier. We do it the same way as the first step of the previous problem and then we stop. To find out how many moles of salt are contained in 300. ml of a 0.40 M NaCl solution, we start with the volume in liters (0.300 L) and multiply it by the number of moles per liter of solution, which is 0.40 moles over 1.00 L.

Calculations Using Molarity - dl.clackamas.edu

Hello and welcome to the very first Magoosh lesson on the TOEFL. Today, we're going to talk about a very general overview of the test. We won't talk about any specific details.

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