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Cathode Ray Experiments Teaching Transparency

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Teaching Transparency Worksheets Chemistry: Matter and Change • Chapter 4 87 1. What is a cathode ray? 2. What do the experiments in A, B, and C have in common? 3. Examine the cathode ray experiment in A. Describe the path of the cathode ray from its origin to its termination. 4. Compare the experimental setup in B with the setup in C.

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set up experiment to measure the ratio of the charge of an electron to its mass. He found this ratio to be constant and did not depend on the kind of gas in the cathode ray tube or the type of metal used for the electrodes.

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JJ Thompson did an experiment using a cathode ray tube and found that electrons passed from the surface of the cathode and accelerate toward the positively charged anode. The anode absorbed many electrons, but is the anode has slits some electrons will pass through. He is credited with the first discovery of the electron.

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J.J. Thomson's cathode ray tube experiments led to a very important scientific discovery, the electron. In this lesson learn what a cathode ray tube is and how J.J. Thomson made his discovery.

J.J. Thomson's Cathode Ray Tube (CRT): Definition ...

Thomson's First Cathode Ray Experiment. Thomson had an inkling that the 'rays' emitted from the electron gun were inseparable from the latent charge, and decided to try and prove this by using a magnetic field. His first experiment was to build a cathode ray tube with a metal cylinder on the end.

Cathode Ray Experiment by J. J. Thomson - Explorable.com

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Electrons as Particles Thomson's Cathode Ray Tube Experiment Worksheet Key Background Joseph John Thomson was an Englishman. While studying mathematics at Trinity College, he applied and received a scholarship to Cambridge University in London, England. Thomson spent his entire career with the university where his interests in math led him to

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