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Calcium oxide reacts with water in a combination reaction to produce calcium hydroxide: CaO(s) + H 2 O(I) \rightarrow Ca(OH) 2 (s) In a particular experiment, a 3.00-g sample of CaO is reacted with excess water and 3.46 g of Ca(OH) 2 is recovered. What is the percent yield in this experiment?

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