

Chapter 10 The Mole Study Guide Answers

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Chapter 10 The Mole Study

The mole is large amounts of atoms at a time so that way the mass is numerically equal to the P.T. but the mass is in grams. Identify and use Avogadro's number Avogadro's number is 6.02×10^{23} .

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___ 3. the volume occupied by a mole of any gas at STP Match each item with the correct statement below. a. representative particle d. percent composition b. mole e. standard temperature and pressure c. Avogadro's number f. empirical formula ___ 4. the number of representative particles of a substance present in 1 mole of that substance ___ 5.

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Chapter 10 Study Guide The Mole Section 10 1 Measuring Matter

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Chapter 10 – Chemical Calculations and Chemical Equations 139 The section ends with a summary of equation stoichiometry problems and shows how the skills developed in Section 10.1 can be mixed with the new skills developed in this section. Section 13.3 completes our process of describing equation stoichiometry problems by

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