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use these activities to yourself study the vocabulary and major concepts presented in this chapter

Quia - Chapter 15 "Water and Aqueous Systems"

WATER AND AQUEOUS SYSTEMS chapter ... Crystals of copper sulfate pentahydrate always contain five molecules of water for each copper and sulfate ion pair.

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Chapter 15 - Water and Aqueous Systems - 15.2 Homogeneous Aqueous Systems - 15.2 Lesson Check - Page 501: 17 Answer CH₄ doesn't dissolve in water because it is a molecular compound with no net dipole KCl is soluble because of its ionic bonds.

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CHAPTER 15,Water and Aqueous Systems(continued) 6. Circle the letter next to each sentence that describes a result of the surface tension of water. a. In a full glass of water, the water surface seems to bulge over the rim of the glass. b. Water beads up into small, nearly spherical drops on a paper towel.

SECTION 15.1 WATER AND ITS PROPERTIES (pages 445-449)

The hydrogen bonding in water cause the high surface tension and low vapor pressure of water. After you claim an answer you'll have 24 hours to send in a draft. An editor will review the submission and either publish your submission or provide feedback. Next Answer Chapter 15 - Water and Aqueous ...

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