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Section Name Experiment 36 Advance Study Assignment: Qualitative Analysis of Group I Cations 1. At the start of the data page for this experiment, complete the flow diagram for the separation and identification of the ions in Group I 2. Write balanced net ionic equations for the following reactions a.

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experiment date(s): Page 1 of 1 Experiment 33 - Qualitative Analysis of Group I Cations Group I cations (a rather old method of classifying ions) refers to the three most easily identified metal ions: Pb2+, Hg 2 2+ (mercury I), and Ag+.

# **Experiment 33 - Qualitative Analysis of Group I Cations**

Experiment 36: Preparation and Reactions of Coordination Compounds: Oxalate Complexes Introduction: The purpose of this lab was to gain familiarity with coordination compounds by preparing a representative compound and to witness some typical reactions. The compound prepared was potassium dioxalatocuprate (II) dihyrdate.

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Chemistry 11 Lab (Sect 06) Fall 2002. Tuesdays, 1:00 (sharp!) - 4:15, Olin-Rice 343, Instructor: Keith Kuwata. Announcements: (12/9/02) Our final experiment (on 12/10) on pH and buffers is described on this handout; we will not be using Experiment 25 in the lab manual. Please make sure you understand all the problems in the ASA; here is a key.

# Chemistry 11 Lab (Sect 06) Fall 2002 - Macalester College

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Experiment 36- Qualitative Analysis of Group 1 Cations Abstract The purpose of this lab is to follow a classic analytical scheme to separate and identify the ions in a known mixture of Group 1 Cations, and then to then apply this scheme to identify the ions in an unknown mixture of Group 1 Cations. Experimental Procedure 1.

# **Experiment 36 Qualitative Analysis of Group ...**

The Guardian's Bim Adewunmi and Archie Bland get down to the business of taking the love test. Photograph: Teri Pengilley for the Guardian 3. Before making a telephone call, do you ever rehearse ...

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# Chemical Principles in the Laboratory 010, Emil Slowinski ...

Ch 16 27, 28, 31 (draw Lewis structure to rationalize the relative strengths of oxo acids), 34, 36, 42 (draw Lewis structures to justify answers), 49, 66, 68abc, 75, 100, 106 Chang Homework on thermodynamics to be announced Ch 18 33, 34, 41, 47 (must use Excel—be sure to turn in graphs), 49, 55, 58, 64 (again, you must use

# **Chemistry 20A: General Chemistry—Chemical Structure**

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# **Principles**

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