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### Atkins 9th Edition Physical Chemistry

□□□□. D15.2 See Figures 15.8 and 15.10, the Brief Illustration in Section 15.3(a), and Self-test 15.6 for details. If the levels are non-degenerate, then both quantities increase from zero to a limiting value in a roughly sigmoidal shape.

### □□□□ 9□ □□□ (Atkins Physical Chemistry 9th Edition, P.W ...

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### Chemistry - Wikipedia

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In physical and organic chemistry, the dispersity is a measure of the heterogeneity of sizes of molecules or particles in a mixture. A collection of objects is called uniform if the objects have the same size, shape, or mass. A sample of objects that have an inconsistent size, shape and mass distribution is called non-uniform. The objects can be in any form of chemical dispersion, such as ...

### Dispersity - Wikipedia

dissolved in water. Sucrose is a nonelectrolyte, which means that the solution contains whole C<sub>12</sub>H<sub>22</sub>O<sub>11</sub> molecules. In predicting the expected freezing point of a solution, one must consider not only the number of formula units present, but also the number of ions that result from each formula unit, in the case of ionic compounds.

### Colligative Properties - Chemistry Encyclopedia - water ...

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