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atoms on each side of an equation must balance, therefore, the mass (number of grams) must balance, but not the number of molecules nor the volume of gases. The subscripts in a correct formula tell the number of atoms in one molecule.

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Balance the equation: $\text{KMnO}_4 + \text{HCl} = \text{KCl} \dots$ To have the same amount of each element on each side of the eq... $2\text{KMnO}_4 + 16\text{HCl} = 2\text{KCl} + 2\text{MnCl}_2 + 8\text{H}_2\text{O} + 5\text{Cl}_2$ What is the purpose of balancing chemic... To have the same amount of each element on each side of the eq... What law does balancing chemical equati... The law of conservation...

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Balancing Chemical Equations Answer Key

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