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Atomic Mass & Mass Number Atomic mass and mass number are not the same. Atomic mass is the
atom's actual mass • 1 atomic mass unit (amu) is equal to 1/12 of the mass of the carbon-12 atom
• a proton has a mass of 1.007 amu • a neutron has a mass of about 1.008 amu • an electron has a
very small mass of 0.00055 amu

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Dalton's Atomic Theory (page 101) 3. Is the following sentence true or false? John Dalton gathered evidence for the existence of atoms by measuring the masses of elements that reacted to form compounds. 4. What theory did Dalton propose to explain why the elements in a compound always join in the same way? 5.

Chapter 4 Atomic Structure Section 4.1 Studying Atoms

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Sample answer: Salt water is a mixture that is a solution. The salt and water are so well blended that they appear to be one substance. Cereal with milk is a mixture that is not a solution. You can easily see and separate the pieces of cereal from the milk. Cereal with milk is a heterogeneous mixture.

Chapter 1 Introduction to Physical a. Science b. c ...

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Slide 12-26 Atomic Mass and Atomic Mass Number •The atomic mass number A is the sum of the number of protons and the number of neutrons in an atom. $A = \text{number of protons} + \text{number of neutrons}$ •The atomic mass scale is established by defining the mass of ^{12}C to be exactly 12 u. • u is the symbol for the atomic mass unit: $1\text{ u} = 1.66 \times 10^{-27}\text{ kg}$ •Molecular mass is the sum of the atomic masses ...

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