

## *Binary Molecular Nomenclature Worksheet Answers*

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**Binary Molecular Nomenclature Worksheet Answers**

Binary Molecular Compounds Compounds between two nonmetals First element in the formula is named first. Keeps its element name Gets a prefix if there is a subscript on it Second element is named second Use the root of the element name plus the -ide suffix Always use a prefix on the second element

**Naming Binary Molecular Compounds - SCCPSS**

Nomenclature Worksheet 6: Binary Covalent Compounds Please complete the following table: Name of Covalent Compound 1. carbon dioxide 2. phosphorus triiodide 3. sulfur dichloride 4. nitrogen trifluoride 5. dioxygen difluoride Formula of Covalent Compound 16.  $N_2F_4$  7.  $SCl_2$  8.  $ClF_3$  9.  $SiO_2$  10.  $P_4O_{10}$

**Nomenclature Worksheet 5: Ionic Compounds Summary**

Naming Binary Molecular Compounds Naming Binary Molecular Compounds. OUTLINE. 1. Naming Binary Molecular Compounds. 2. Binary Molecular Compounds. 3. List of Prefixes. 4. Naming Binary Compounds. 5. Practice - Write the Formula ... Answers - Write the Formula Answers - Write the Formula ...

**Binary Molecular Nomenclature - ScienceGeek.net**

Formulas and Nomenclature Binary Molecular Worksheet Name the following compounds. 1.  $N_2O_4$  2.  $N_2O_3$  3.  $P_2O_5$  4.  $Cl_2O_7$  5.  $CO_2$  6.  $OF_2$  Write formulas. 1. phosphorus pentachloride 2. carbon monoxide 3. carbon tetrachloride 4. nitrogen trifluoride 5. sulfur hexafluoride 6. dinitrogen trioxide

**Formulas and Nomenclature Binary Ionic - svsd.net**

Binary Molecular Nomenclature and Molar Mass Rules for Binary Molecular Compounds 1. The naming system is for compounds composed of two nonmetallic elements. 2. The first element keeps its name a. The first element gets a prefix if it has a subscript in the formula 3. The second element gets the -ide suffix (ending) a.

**Binary Molecular Compound Practice - ScienceGeek.net**

Nomenclature Worksheet Part 2-Binary Molecular Compounds. When two different elements combine, they form a binary compound. We have seen some ionic binary compounds such as sodium chloride,  $NaCl$ , or copper(II) oxide,  $CuO$ . But, non-metals can also form binary compounds, the most famous of which is water,  $H_2O$ . We have learned how to name binary ...

**Nomenclature Worksheet Part 2 - laney.edu**

Images in curved mirrors worksheet answers binary compounds worksheet answers 5.10 molecular compounds names and formulas wkst key trustee: trustee:. Answer key for naming compounds worksheet, binary ionic and covalent names. 1# Naming Ionic Compounds - Answer Key Give the name of the following ionic compounds: Name 1)  $Na$  Naming Compounds ...

**Naming Ionic Compounds Worksheet And Answers**

Binary Ionic Compounds ANSWER KEY. Directions: First quickly scan the worksheet and circle any metals (as a symbol or as a name) that is a transition metal. Then either give the same or chemical formula as necessary for each problem below. Remember that transition metals can have multiple oxidation states, so you are required

**Binary Ionic Compounds - achs.roselleschools.org**

Naming Covalent Compounds - Key Write the formulas for the following covalent compounds: 1) Nitrogen tribromide  $NBr_3$  2) hexaboron silicide  $B_6Si$  3) chlorine dioxide  $ClO_2$  4) hydrogen iodide  $HI$  5) iodine pentafluoride  $IF_5$  6) dinitrogen trioxide  $N_2O_3$  7) ammonia (nitrogen trihydride)  $NH_3$  8) phosphorus triiodide  $PI_3$

**Naming Covalent Compounds Worksheet**

Covalent Nomenclature Worksheet -Answer Key. Write the correct name for: 1) As<sub>4</sub>O<sub>10</sub>. tetrarsenic decoxide. 2) BrO<sub>3</sub>. bromine trioxide. 3) BN. boron mononitride. 4) N<sub>2</sub>O<sub>3</sub>. dinitrogen trioxide. 5) NI<sub>3</sub>. nitrogen triiodide. 6) SF<sub>6</sub>. sulfur hexafluoride. ... Covalent Nomenclature Worksheet ...

**Covalent Nomenclature Worksheet - Brooklyn High School**

Covalent Compound Naming Worksheet 1 covalentname.sxw. Naming Covalent Compounds Solutions Write the formulas for the following covalent compounds: 1) antimony tribromide SbBr<sub>3</sub> 2) hexaboron silicide B<sub>6</sub>Si 3) chlorine dioxide ClO<sub>2</sub> 4) hydrogen iodide HI 5) iodine pentafluoride IF<sub>5</sub>

**Covalent Compound Naming Worksheet - Prairie Science**

Naming Covalent Compounds(Type III) Worksheet. Write the formulas for the following covalent compounds: 1) nitrogen tribromide \_\_\_\_\_ ... Naming Covalent Compounds - Key. Write the formulas for the following covalent compounds: 1) Nitrogen tribromide . NBr<sub>3</sub>. 2) hexaboron silicide . B<sub>6</sub>Si. 3) chlorine dioxide . ClO<sub>2</sub>.

**Naming Covalent Compounds Worksheet**

Naming Covalent Compounds Worksheet. ... Naming Covalent Compounds Solutions. Write the formulas for the following covalent compounds: 1) antimony tribromide SbBr<sub>3</sub>. ... Naming Ionic Compounds - Answer Key. Give the name and molar mass of the following ionic compounds: Name . 1) 2)

**Naming Covalent Compounds Worksheet**

Nomenclature Packet Worksheet I: Binary Ionic Compounds (representative metals) – metals from groups 1A, 2A, and 3A (1, 2, and 13) have constant charges as ions and do NOT get Roman Numerals in their names 1. Name the following ionic compounds: a. Al<sub>2</sub>O<sub>3</sub> Aluminum oxide b. Cs<sub>2</sub>O Cesium oxide c. Rb<sub>3</sub>N Rubidium nitride d. Ca<sub>3</sub>N<sub>2</sub> Calcium ...

**Nomenclature Packet - Mr. Midtgard's Chemistry Class**

Nomenclature Worksheet 3 – Covalent (Molecular) Compounds Part A: Name the following covalent compounds. 1. CO Carbon monoxide 2. CO<sub>2</sub> Carbon dioxide 3. N<sub>2</sub>O<sub>3</sub> Dinitrogen trioxide 4. N<sub>2</sub>omit 5. NP Nitrogen monophosphide 6. SCl<sub>2</sub> Sulfur dichloride 7. P<sub>2</sub>O<sub>5</sub> Diphosphorous pentoxide 8. NBr<sub>3</sub> Nitrogen tribromide 9. Cl<sub>4</sub> Carbon tetraiodide 10 ...

## Binary Molecular Nomenclature Worksheet Answers

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