

Balancing Act Practice Answers Key

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Balancing Act Answer Key: Page 1 Problems $2\text{Ca} + \text{O}_2 \rightarrow 2\text{CaO}$ $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$ $2\text{Cu}_2\text{O} + \text{C} \rightarrow 4\text{Cu} + \text{CO}_2$ $2\text{H}_2\text{O} \rightarrow 2\text{H}_2 + \text{O}_2$ Hint: Add the O atoms on the product side together when doing the counts. Page 2 Practice Problems 1. $2\text{Na} + \text{MgF}_2 \rightarrow 2\text{NaF} + \text{Mg}$ 2. $\text{Mg} + 2\text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2$ 3. $\text{Cl}_2 + 2\text{KI} \rightarrow 2\text{KCl} + \text{I}_2$ 4.

Name Atoms are not or during a chemical reaction. Mg + O

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Balance each equation. Be sure to show work. Remember you cannot add subscripts of place coefficients in the middle of a chemical formula.

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Balancing Act Practice Name Balance each equation. Be sure to show your lists! Remember you cannot add subscripts or place coefficients in the middle of a chemical formula.

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Balancing Equations Practice ANSWER KEY Part A: Identify the following parts of each chemical formula by circling the subscripts and drawing a square around the coefficients. H_2 2 HCl 4 O $2CH_4$ 3 CO 3 2 NaOH Part B: List the symbols for the atoms in each formula and give the number of each. C_2H_6 2MgO 4P 4O₁₀

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