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Most atoms are less likely to react with other atoms when they have valence electrons. 13. When an atom loses an electron, a(an) ion forms. 14. When two atoms share two pairs of electrons, a(an) forms. 15. A mixture made of two or more elements that has the properties of metal is a(an) .
double bond eight elements ionic chemical bond alloy ...

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Atoms and Bonding Multiple Choice Write the letter of the correct answer on the line at the left. ____
1. A(n) ____ is an atom or group of atoms that has an ... Element E: exists as a nonpolar molecule of two identical atoms with a double bond Element G: forms an ion with a 3+ charge when reacting with nitrogen

Atoms and Bonding Chapter Test A Atoms and Bonding

After reading Lesson 7.3, answer the following questions. Metallic Bonds and Metallic Properties 1. Is the following sentence true or false? Metals are made up of cations and valence electrons, not neutral atoms. 2. What are metallic bonds? 3. Name three properties of metals that can be explained by metallic bonding. a. b. c. 4.

BONDING AND INTERACTIONS

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125-129, Atoms, Bonding, and the Periodic Table 130-137, Ionic Bonds 132, Figure 3 - Formation of an ... information about fuel cells when they answer the questions in "Figure 6: How Can Chemical ... Pearson Interactive Science, 12 Module Series, ©2011 ...

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Atoms and Bonding Chapter Test Atoms and Bonding Multiple Choice Write the letter of the correct answer on the line at the left. ____ 1. A(n) ____ is an atom or group of atoms that has an electric charge. a. compound b. ion c. crystal d. neutron ____ 2. A covalent bond is called ____ if the atoms equally

Atoms and Bonding Chapter Test Atoms and Bonding

fluoride consist of hydrogen bonding and permanent dipole-dipole attractions, in addition to the weak van der Waals' attractive forces that occur between the molecules of the other hydrogen halides.

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Chapter 4 Lesson 1 Atoms, Bonding, and the Periodic Table Ms. Schreurs 8th Grade Earth Science Introduction to Chemistry . My Planet Diary Elemental Effects Page 124 . What Determines an Element's ... -Atoms form bonds so they have 8 valence electrons (hydrogen needs 2)

Chapter 4 Lesson 1 Atoms, Bonding, and the Periodic Table

Chemical Bonding II: molecular shapes, Valence Bond Theory, and molecular orbital Theory When your body metabolizes a mole of sucrose, it obtains 5644 kJ of energy. Some artificial sweeteners, such as saccharin, for example, are not metabolized at all—they just pass through the body unchanged—and therefore have no caloric value.

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