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□□□□. D15.2 See Figures 15.8 and 15.10, the Brief Illustration in Section 15.3(a), and Self-test 15.6 for details. If the levels are non-degenerate, then both quantities increase from zero to a limiting value in a roughly sigmoidal shape.

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Dispersity - Wikipedia

dissolved in water. Sucrose is a nonelectrolyte, which means that the solution contains whole C₁₂H₂₂O₁₁ molecules. In predicting the expected freezing point of a solution, one must consider not only the number of formula units present, but also the number of ions that result from each formula unit, in the case of ionic compounds.

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