

## *Basic Atomic Structure Worksheet Answer Key*

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### **Basic Atomic Structure Worksheet Answer**

Basic Atomic Structure Worksheet H and the 1. 2. 4. 5. 6. 8. 9. The 3 particles of the atom are: protons (p), neutrons (n), electrons (e). Their respective charges are: The number of protons in ...

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Basic Atomic Structure Worksheet ANSWERS 1 a) protons b) neutrons c) electrons a) Positive b) Neutral c) negative 2 atomic number (or identity); charge 3 protons; electrons (in a neutral charge atom only!); same 4 average atomic weight; mass 5 mass number; nucleus 6 neutrons; protons (or atomic number); mass number 7 Lithium = Li = 3 Bromine = Br = ...

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3. The atomic number tells you the number of \_\_\_\_\_ in one atom of an element. It also tells you the number of \_\_\_\_\_ in a neutral atom of that element. The atomic number gives the “identity” of an element as well as its location on the periodic table. No two different elements will

### **Basic Atomic Structure Worksheet**

9 NOTE\*\*\*For these, I rounded the average atomic weight to the nearest whole number to get the mass number for the most abundant isotope. Mass number is the number of protons and neutrons; this number is not on the Periodic Table. Atomic weight is the average of all isotopes and this IS on the Periodic Table.

### **ww2.d155.org**

Law of Definite Proportions. Dalton's Atomic Theory was based in part on the work of the French scientist Joseph Louis Proust in 1799 who discovered what is now called the Law of Definite Proportions (also called the Law of Constant Composition):. A compound is always composed of the same elements in a fixed ratio by weight. Example: When 200.59 g of mercury reacts completely with 32.066 g ...

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Atomic Structure . Complete the separate worksheet on atomic structure. Use old materials from CPE or biology and/or Internet resources. It is essential that you are familiar with these concepts before beginning Honors Chemistry. Reading Assignment . Obtain your textbook prior to the first day of school. Read chapters one and two. As you read ...

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Atomic Basics Answer Key Part A: Atomic Structure 1. Draw five protons in the nucleus of the atom. Label them with their charge. 2. Draw six neutrons in the nucleus of the atom. 3. Draw two electrons in the first energy level and label them with their charge. 4. Draw three electrons in the second energy level and label them with their charge. 5.

### **Chemistry of Matter - sciencespot.net**

Basic Atomic Structure Worksheet and the 1. 2. 3. 4. 5. 6. 8. 9. The 3 particles of the atom are: Their respective charges are: The number of protons in one atom of ...

### **www.neshaminy.org**

Many students first learning about atomic structure and electricity notice a paradox with respect to the location of all the protons in an atom: despite their electrical charge, they are tightly bound together in a "core" called the nucleus. Based on what you know about electrical charges, explain why this is a paradox, and also what the ...

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