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(4) A coffee-cup calorimeter having a heat capacity of 472 J/OC is used to measure the heat evolved when the following aqueous solutions, both initially at 22.6 OC, are mixed: 100 g of a solution containing 6.62 g of lead (II) nitrate, and 100 g of solution containing 6.00 g

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About This Quiz & Worksheet. If you're confused about the format of the AP Chemistry test--or just need some tips on how to prepare--then this quiz/worksheet combo is perfect for you.

Quiz & Worksheet - Multiple Choice Questions on the AP ...

6. A 50.0L reaction vessel contains 1.00mol N 2, 3.00mol H 2, and 0.500mol NH 3.Will more ammonia, NH 3, be formed or will it dissociate when the mixture goes to equilibrium at 400.oC? The equation is: $N2(g) + 3H 2(g) \leftrightarrow 2NH 3(g)$ Kc is 0.500 at 400. oC. 7.

AP Chemistry Equilibrium Worksheet

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on how those properties interact with each other in various contexts.

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