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Intravenous (IV) saline solutions are often administered to patients in the hospital. Normal saline solution contains 0.90g NaCl in exactly 100. ml of solution. Calculate the molarity of this solution. © POGIL 2005, 2006 3/5 Revised by: Lorraine Giloni, ««GreetingLine»» and Kelly Levy Edited by Linda Padwa and David Hanson, Stony Brook University

Molarity - Pogil - Yumpu

Molarity Worksheet # 2 identifica ____ What does molarity mean? Number of moles of solute. 1 liter solution. What is the molarity of a solution that contains 4.53 moles of lithium nitrate in 2.85 liters of solution? $4.53 \text{ mol LiNO}_3 = 1.59 \text{ M LiNO}_3$ 2.85 L soln

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