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Unit 8 3 Molar Concentration Answer Key

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Determine what the volume of gas would be at STP. 3. Determine the mass, then the moles of Mg used. Modeling Chemistry TN Modeling Curriculum Committee Pope John Paul II High School 307. 4. Write a balanced equation to describe the reaction between the Mg and HCl. Determine the number of moles of H2 gas in your sample.

Unit 9, Worksheet 1— Dalton's Law of Partial Pressures

Modeling Chemistry 1 U2 ws 3 v2.0 . Name . Date Pd . Unit 2 Worksheet 2 - Measuring Pressure . Problem 1 and 2 Calculate the pressure using the barometer . 1. 720 mmHg = 0.95 atm. 2. 760 mmHg = 1 atm. Problems 3 and 4. Calculate the pressure of the gas in the flask connected to the manometer. 3. a. 127-84= 43mmHg 730 + 43 = 773 mmHg

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