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Molar Mass The molar mass of a substance is the mass of one mole of that substance. Because of the nature of the mole, the atomic mass of an element in atomic mass units is equal to the molar mass of that substance in grams. Worksheet - Moles, mass and molar mass Unit 1 and 2 VCE Chemistry 2014 Introduction - Transition Week 2013 ...

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Chemistry IF8766 . ssì KC)O so -150 caco S de phosphorus de . MOLES AND MASS Name Determine the number of moles in each of the quantities below. 1. 25 g of Naa 2. 125 g of 3. 100. g of 4 74 g of KCl z LB (03 a9/)3c 168 , 0991957/0/ m l.t9eoS D enstructional Fair, Inc.

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09/22/2017: 1. Warm-up: how do you determine the charge of an ion using the periodic table 2. Ion Charges Quiz 3. Molecular Model Lab *If absent, complete the following online activity Make-Up Models Lab . 1.

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MIXED MOLE PROBLEMS - KEY 1. a) How many grams are there in 1.5×10^{25} molecules of CO 2? 1.110g 1 mol 44.0 g 6.0210molecules 1.510molecules CO1mol 3 23 2 25!=!!!! b) What volume would the CO

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