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Modern Chemistry 175 Nuclearchemistry CHAPTER 21 REVIEW Nuclear Chemistry SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. Match each of the following statements with the process(es) to which they apply, using one of the choices below: (1) fission only (3) both fission and fusion

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Chemistry Predict the mode of decay of (a) carbon-14, (b) (b) xenon-118. (b) Xenon has an atomic number of 54. Thus, xenon-118 has 54 protons and 118 - 54 = 64 neutrons, giving it a neutron-to-proton ratio of According to Figure 21.2, stable nuclei in this region of the belt

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Modern Chemistry • CHAPTER 22 HOMEWORK 22-2 (pp. 702-704) VOCABULARY Write true or false.

1. The band of stability represents the stable nuclei cluster over a range of electron-proton ratios.

2. The stability of a nucleus is greatest when the nucleons are in a 1:1 ratio. \_\_\_\_\_ 3. 2, 20, 50, and 126 are all magic numbers. \_\_\_\_\_ 4.

#### Modern Chemistry CHAPTER 22 HOMEWORK 22-1

Chapter 15:Acid--Base Titration and pH. pH Analysis pH Analysis: Quad Color Indicator. Chapter 17: Reaction Kinetics. Collision Theory. Chapter 21: Nuclear Chemistry. Nuclear Decay. Chapter 22: Organic Chemistry. Dehydration Synthesis. Chapter 23: Biological Chemistry. Cell Energy Cycle RNA and Protein Synthesis

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