

## *Naming Molecular Compounds Worksheet With Answers*

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### **Naming Covalent Compounds - Printable Worksheets**

Naming Prefixes 1 mono- 2 di- 3 tri- 4 tetra- 5 penta- 6 hexa- 7 hepta- 8 octa- 9 nona- 10 deca- A molecular compound is a group of atoms held together by a covalent bond. Compounds made entirely of non-metals are generally molecular compounds. Carbon tetrachloride, CCl<sub>4</sub>, is an example of a molecular compound.

### **Naming Molecular Compounds Name Chem Worksheet 9-2**

Title: Microsoft Word - 6-20-Naming Molecular Compounds Wkst.doc Author: Brent White Created Date: 6/21/2005 10:00:59 PM

### **CHEMISTRY: A Study of Matter - Georgia Public Broadcasting**

Covalent Compound Naming Worksheet 1 covalentname.sxw. Naming Covalent Compounds Solutions Write the formulas for the following covalent compounds: 1) antimony tribromide SbBr<sub>3</sub> 2) hexaboron silicide B<sub>6</sub>Si 3) chlorine dioxide ClO<sub>2</sub> 4) hydrogen iodide HI 5) iodine pentafluoride IF<sub>5</sub>

### **Covalent Compound Naming Worksheet**

Naming Binary Molecular Compounds. Recall that a molecular formula shows the number of atoms of each element that a molecule contains. A molecule of water contains two hydrogen atoms and one oxygen atom, so its formula is H<sub>2</sub>O. A molecule of octane, which is a component of gasoline, contains 8 atoms of carbon and 18 atoms of hydrogen.

### **5.8: Naming Molecular Compounds - Chemistry LibreTexts**

Nomenclature Worksheet 3 - Covalent (Molecular) Compounds Author: hwdsb Last modified by: Jason Getz Created Date: 10/6/2013 3:51:00 PM Company: HWDSB Other titles: Nomenclature Worksheet 3 - Covalent (Molecular) Compounds

### **Nomenclature Worksheet 3 - Covalent (Molecular) Compounds**

Naming Covalent Compounds - Key Write the formulas for the following covalent compounds: 1) Nitrogen tribromide NBr<sub>3</sub> 2) hexaboron silicide B<sub>6</sub>Si 3) chlorine dioxide ClO<sub>2</sub> 4) hydrogen iodide HI 5) iodine pentafluoride IF<sub>5</sub> 6) dinitrogen trioxide N<sub>2</sub>O<sub>3</sub> 7) ammonia (nitrogen trihydride) NH<sub>3</sub> 8) phosphorus triiodide PI<sub>3</sub>

### **Naming Covalent Compounds Worksheet**

Naming Ionic Compounds Practice Worksheet Name the following ionic compounds: 1) NH<sub>4</sub>Cl \_\_\_\_\_ 2) Fe(NO<sub>3</sub>)<sub>3</sub> ... Mixed Ionic/Covalent Compound Naming For each of the following questions, determine whether the compound is ionic or covalent and name it appropriately. 1) Na<sub>2</sub>CO

### **Naming Ionic Compounds Practice Worksheet**

Naming Covalent Compounds Solutions. Write the formulas for the following covalent compounds: 1) antimony tribromide SbBr<sub>3</sub> 2) hexanitrogen silicide N<sub>6</sub>Si 3) chlorine dioxide ClO<sub>2</sub> 4) iodine pentafluoride IF<sub>5</sub> 5) dinitrogen trioxide N<sub>2</sub>O<sub>3</sub> 6) ammonia NH<sub>3</sub> ... Naming Covalent Compounds

Worksheet Author:

### **Naming Covalent Compounds Worksheet**

1 Naming Compounds Tutorial and Worksheet Since we use different methods in naming binary covalent (molecular) compounds and ionic compounds, the first step in naming or writing the formula of a compound is to determine which of the 2 compound classes it belongs to. This

### **Binary Covalent Ionic only - Saddleback College**

Covalent compound naming: Covalent Compound Naming Practice: For those of you who love covalent compounds, you're in for a real treat!; More Covalent Compound Naming Practice: And if you still love covalent compounds even after the last worksheet, here's another for your viewing pleasure.; Naming Covalent Compounds (dd-ch): Let's name our little covalent friends.

### **Naming Worksheets | The Cavalcade o' Chemistry**

Acids are compounds containing an ionizable proton ( $H^+$ ), since an acid is a proton contributor (a hydrogen atom which has lost its electron). The polyatomic anions derived from acids are named by dropping the -ic (or -ous) suffix from the acid name and adding the -ate (or -ite) suffix, respectively. Compounds containing polyatomic anions are named using the Type I or Type II naming systems ...

### **Ionic Compound Naming Worksheet ~ Ionic Compounds**

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### **CHEMISTRY: A Study of Matter - Alex Anguiano**

NAMING MOLECULAR COMPOUNDS Indicate number of atoms of each element with Greek prefix before element name: # of atoms Greek Prefix # of atoms Greek Prefix 1 mono (usually omitted) 6 hexa 2 di 7 hepta 3 tri 8 octa 4 tetra 9 nona 5 penta 10 deca. CHEMISTRY Naming Compounds Handout page 10 of 12

### **Naming Compounds Handout - faculty.seattlecentral.edu**

Propose a definition for "binary molecular compounds." Binary molecular compounds contain two different elements that are covalently bonded 15. Collaborate with your group members to write a list of rules for recognizing and naming binary molecular compounds from their chemical formulas. 1.

### **(DOC) Pogil ans naming molec compds-s | Gul Awkum ...**

Can you pick the correct name for these molecular compounds? Test your knowledge on this science quiz to see how you do and compare your score to others. Quiz by CMHSMacomber

### **Naming Molecular Compounds Quiz - By CMHSMacomber**

Binary Molecular Compounds Compounds between two nonmetals First element in the formula is named first. Keeps its element name Gets a prefix if there is a subscript on it Second element is named second Use the root of the element name plus the -ide suffix Always use a prefix on the second element

### **Naming Binary Molecular Compounds - SCCPSS**

The most widely accepted format for nomenclature has been established by IUPAC. Molecular compounds are made when two or more elements share electrons in a covalent bond to connect the elements. Typically, non-metals tend to share electrons, make covalent bonds, and thus, form molecular compounds. Rules for Naming Molecular Compounds:

### **Naming Molecular Compounds | Introduction to Chemistry**

Naming Mixed Ionic and Covalent - Answers Name the following compounds. Remember, they may be either ionic or covalent compounds, so make sure you use the right naming method! 1) NaF

sodium fluoride 2) NF 3 nitrogen trifluoride 3) Li 2O lithium oxide 4) Al 2S 3 aluminum sulfide 5) MgSO 4 magnesium sulfate 6) SiH 4 silicon tetrahydride 7) KNO 3 ...

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