

## ***Molar Mass Worksheet Answer Key***

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### **Molar Mass Worksheet Answer Key**

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Molar Mass Worksheet – Answer Key Calculate the molar masses of the following chemicals: 1)  $\text{Cl}_2$  71 g/mol 2)  $\text{KOH}$  56.1 g/mol 3)  $\text{BeCl}_2$  80 g/mol 4)  $\text{FeCl}_3$  162.3 g/mol 5)  $\text{BF}_3$  67.8 g/mol 6)  $\text{CCl}_2\text{F}_2$  121 g/mol 7)  $\text{Mg}(\text{OH})_2$  58.3 g/mol 8)  $\text{UF}_6$  352 g/mol 9)  $\text{SO}_2$  64.1 g/mol 10)  $\text{H}_3\text{PO}_4$  98 g/mol 11)  $(\text{NH}_4)_2\text{SO}_4$  132.1 g/mol 12)  $\text{CH}_3\text{COOH}$  60 g/mol 13)  $\text{Pb}(\text{NO}_3)_2$  331.2 g/mol 14)  $\text{Ga}_2(\text{SO}_3)_3$  379.7 g/mol Avogadro's Number and the Mole 1) How many moles of water does  $6.02 \times 10^{23}$  molecules represent?

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Molar Mass Worksheet – Answer Key Calculate the molar masses of the following chemicals: 1)  $\text{Cl}_2$  71 g/mol 2)  $\text{KOH}$  56.1 g/mol 3)  $\text{BeCl}_2$  80 g/mol 4)  $\text{FeCl}_3$  162.3 g/mol 5)  $\text{BF}_3$  67.8 g/mol 6)  $\text{CCl}_2\text{F}_2$  121 g/mol 7)  $\text{Mg}(\text{OH})_2$  58.3 g/mol 8)  $\text{UF}_6$  352 g/mol 9)  $\text{SO}_2$  64.1 g/mol 10)  $\text{H}_3\text{PO}_4$  98 g/mol 11)  $(\text{NH}_4)_2\text{SO}_4$  132.1 g/mol 12)  $\text{CH}_3\text{COOH}$  60 g/mol 13)  $\text{Pb}(\text{NO}_3)_2$  331.2 g/mol 14)  $\text{Ga}_2(\text{SO}_3)_3$  379.7 g/mol

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Molar Mass Practice Worksheet . Find the molar masses of the following compounds: 1)  $\text{NaBr}$  Na 1 (22.989770)

### **Molar Mass Practice Worksheet - Gravity Simulator**

Molar Mass Worksheet and Key For the following compounds, write the chemical formula and determine the molar mass. Write the units! water carbon dioxide sodium chloride calcium hydroxide potassium permanganate ( $\text{KMnO}_4$ ) sodium carbonate barium nitrate hydrogen monochloride sulfuric acid ( $\text{H}_2\text{SO}_4$ ) acetic acid ( $\text{C}_2\text{H}_4\text{O}_2$ ) SEE NEXT PAGE FOR KEY

### **ws molar mass - Saddleback College**

Find the formula mass of the following compounds. Round atomic masses to the tenth of a decimal place. Place your final answer in the FORMULA MASS COLUMN. CHEMISTRY COMPUTING FORMULA

MASS WORKSHEET Problem Set-up example: Find the formula mass of  $\text{Ca}(\text{NO}_3)_2$  Ca:  $1 \times 40.1 = 40.1$  N:  $2 \times 14.0 = 28.0$  O:  $6 \times 16.0 = 96.0$  \_\_\_\_

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1 mole = molar mass (could be atomic mass from periodic table or molecular mass) 1 mole = 22.4 L of a gas at STP ... Mole Calculation Worksheet - Answer Key What are the molecular weights of the following compounds? 1) NaOH  $22.99 + 16.00 + 1.01 = 40.00$  grams/mol 2)  $\text{H}_3\text{PO}_4$

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What is its molar mass? 7. Find the molar mass of acetylene  $\text{C}_2\text{H}_2$ , the gas used in welding torches. 8. Vinegar contains acetic acid,  $\text{HC}_2\text{H}_3\text{O}_2$ . What is the molar mass of this substance? 9. What is the molar mass of magnesium sulfate,  $\text{MgSO}_4$ ? 10. Find the molar mass of phosphorus pentachloride,  $\text{PCl}_5$ . 11. Bromine is diatomic. Find the molar ...

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