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Modern Chemistry 57 Chemical Bonding CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1.

Label each of the following statements as Tr	ue or False:	$\_$ a. If the formula mass of $\mathfrak c$	one
molecule is x u, the molar mass is x g/mol.	b.		

### **CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds**

CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Write formulas for the following compounds: CuCO 3 a. copper(II) carbonate Na 2SO 3 b. sodium sulfite (NH 4) 3PO 4 c. ammonium phosphate SnS 2 d. tin(IV) sulfide HNO 2 e. nitrous acid 2.

### 7 Chemical Formulas and Chemical Compounds

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Modern Chemistry • CHAPTER 7 HOMEWORK 7-3 (pp. 211–213) VOCABULARY Write the prefix that corresponds to each number. Try not to use the book. Number Prefix Number Prefix 16 27 38 49 510 GRAPHIC ORGANIZER Complete the chart by supplying the missing information. Name of Compound Formula of Compound CO 2 Nitrogen dioxide S 2 Cl 2 Carbon ...

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Modern Chemistry • 0	CHAPTER 7 HOMEWORK 7-1 (pp. 203-206) VOCABULARY Write t	rue or false	for
each statement. 1. A	chemical formula indicates the relative number of molecules of	each kind i	n a
chemical compound.	2. Monoatomic ions are ions formed from a single atom.	3.	

### **HOMEWORK 7-1 - Los Banos Unified School District**

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