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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Label each of the following statements as True or False: True a. If the formula mass of one molecule is  $x$  amu, the molar mass is  $x$  g/mol. False b. Samples of equal numbers of moles of two different chemicals

### 7 Chemical Formulas and Chemical Compounds

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3(s) is strongly heated, it produces solid  $\text{MgO}$  as gaseous  $\text{CO}_2$  is driven off. a. What is the percentage loss in mass as this reaction occurs? b. Assign the oxidation number to each atom in  $\text{MgCO}_3$ . c. Does the oxidation number of carbon change upon the formation of  $\text{CO}_2$ ? 64 CHEMICAL FORMULAS AND CHEMICAL COMPOUNDS MODERN CHEMISTRY

### 7 Chemical Formulas and Chemical Compounds

see figure 3-3 page 67 . C: Modern Atomic Theory. Some of Dalton's postulates have been proven to

be incorrect. His third postulate indicated that atoms cannot be subdivided. His second postulate indicates atoms of the same element have identical mass. Isotopes. His theory has been modified -- not discarded. Homework: Section 3-1

### Chapter 3 Modern Chemistry - Prepchem

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