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The molecular weight of O2 is 32 Pogil the mole answer key. 00 u, so a mole of O2 would have a mass of 32. 00 g and would contain $6.022 \times 1023 \text{ O}$. 2 molecules. Each O2 molecule is composed of two oxygen atoms, so one mole of O2 contains two moles of oxygen atoms Pogil the mole answer key. 3. The mole is sometimes described as "the chemist's dozen".

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POGIL- Stoichiometry

Chem 115 POGIL Worksheet - Week 4 Moles & Stoichiometry Answers Key Questions & Exercises 1. The atomic weight of carbon is 12.0107 u, so a mole of carbon has a mass of 12.0107 g.

Moles & Stoichiometry Answers Key Questions & Exercises

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