

## *Naming Compounds Handout Answers*

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### **Naming Compounds Handout Answers**

CHEMISTRY Naming Compounds Handout Key v 1-3 page 2 of 7 p. 4 Name each of the following polyatomic ions: CN<sup>-</sup> = cyanide ion CrO<sub>4</sub><sup>-2</sup> = chromate ion SO<sub>4</sub><sup>-2</sup> = sulfate ion NO<sub>3</sub><sup>-</sup> = nitrate ion OH<sup>-</sup> = hydroxide ion PO<sub>4</sub><sup>-3</sup> = phosphate ion NH<sub>4</sub><sup>+</sup> = ammonium ion C<sub>2</sub>H<sub>3</sub>O<sub>2</sub><sup>-</sup> = acetate ion Combine each pair of ions to get the formula of the compound they form:

### **Naming Cmpds Handout Key - Seattle Central College**

Worksheet Template Pizzaz Handout Pizzazz Math Worksheets Book E from Naming Ionic Compounds Practice Worksheet Answers , source: maveno.us Lesson 1 Intro to Chemical Bonding from Naming Ionic Compounds Practice Worksheet Answers

### **Naming Ionic Compounds Practice Worksheet Answers ...**

Worksheet 5.1 Writing and Naming Ionic Compounds with Polyatomic Ions and Transition Metals Section A Write the name of the ionic compounds containing polyatomic ions

### **Answer Key Worksheet 5.1 Naming and Writing Ionic ...**

Naming Compounds Handout IONIC COMPOUNDS versus MOLECULAR COMPOUNDS ionic compound: consist of cations (positive ions) and anions (negative ions) held together by electrostatic attraction - usually metal + nonmetal(s) - made of monatomic ions, polyatomic ions, and/or both

### **Naming Compounds Handout - Peninsula School District**

Answers - Naming Chemical Compounds . Name the following chemical compounds: 1) NaBr sodium bromide. 2) Ca(C<sub>2</sub>H<sub>3</sub>O<sub>2</sub>)<sub>2</sub> calcium acetate. 3) P<sub>2</sub>O<sub>5</sub> diphosphorus pentoxide. 4) Ti(SO<sub>4</sub>)<sub>2</sub> titanium(IV) sulfate. 5) FePO<sub>4</sub> iron(III) phosphate. 6) K<sub>3</sub>N potassium nitride. 7) SO<sub>2</sub> sulfur dioxide. 8) CuOH copper(I) hydroxide. 9) Zn(NO

### **Answers - Naming Chemical Compounds**

Covalent Compound Naming Worksheet 1 covalentname.sxw. Naming Covalent Compounds Solutions Write the formulas for the following covalent compounds: 1) antimony tribromide SbBr<sub>3</sub> 2) hexaboron silicide B<sub>6</sub>Si 3) chlorine dioxide ClO<sub>2</sub> 4) hydrogen iodide HI 5) iodine pentafluoride IF<sub>5</sub>

### **Covalent Compound Naming Worksheet**

Worksheet Generator; Plans & Pricing; Blog; Print Instructions NOTE: Only your test content will print. To preview this answer key, click on the File menu and select Print Preview. Click here to print this answer key! Click here to save or print this answer key as a PDF! ... Naming Compounds Answer Key. 1. What is the chemical name for [math ...

### **Naming Compounds Answer Key - HelpTeaching.com**

Naming Compounds Worksheet 1. The formulas and common names for several substances are given below. Give the systematic names for these substances. Common Name Chemical Formula Systematic Name a. Sugar of lead Pb(CH<sub>3</sub>COO)<sub>2</sub> Note: CH<sub>3</sub>COO is a polyatomic ion. See your polyatomic ion table for name and charge. b. Blue vitrol CuSO<sub>4</sub> c. Epsom salts ...

### **Naming Compounds Worksheet - Saddleback College**

Naming Ionic Compounds Practice Worksheet Name the following ionic compounds: 1) NH<sub>4</sub>Cl \_\_\_\_\_ 2) Fe(NO<sub>3</sub>)<sub>3</sub> ... Answers - Naming Chemical Compounds 1) NaBr sodium bromide 2) Ca(C<sub>2</sub>H<sub>3</sub>O<sub>2</sub>)<sub>2</sub> calcium acetate 3) P<sub>2</sub>O<sub>5</sub> diphosphorus pentoxide 4) Ti(SO<sub>4</sub>)<sub>2</sub> titanium(IV) sulfate

### **Naming Ionic Compounds Practice Worksheet**

Naming Compounds What's it Made of? Metal and non-metal 2 non-metals 3 or more elements polyatomic compound covalent compound ionic compound USE GREEK PREFIXES Put prefixes in front of element names to tell how many atoms are there. Don't use "mono" for first name, but always for second name. Li<sub>2</sub>S Metal and non-metal— ionic

### **Naming Compounds - Mrs. McCutchen's Class**

Chem guide - answers NAMING ORGANIC COMPOUNDS: 1 1. Alkanes a) (i) (You shouldn't really have got this one wrong. Unless you made a careless mistake (like putting too many or too few hydrogens on the middle carbon), if you got it wrong, re-read the bit about naming alkanes and then look at these questions again.) (ii)

### **Chem guide - answers NAMING ORGANIC COMPOUNDS: 1**

Naming Ionic Compounds Worksheet W 306 Everett Community College Tutoring Center Student Support Services Program Name the following ionic compounds:

### **Naming Ionic Compounds Worksheet 1**

Mixed Naming Worksheet #2 W 303 Everett Community College Tutoring Center Student Support Services Program Write the names of the following chemical compounds:

### **Mixed Naming Worksheet #2 - Everett Community College**

Naming binary molecular compounds Worksheet#3. Binary molecular compounds are made from a combination of 2 different atoms, or in the case of diatomic molecules one kind of atom, ie. Br I N Cl H O F. When naming a binary molecular compound you need to use prefixes 1 = mono 6 = hexa 2 = di 7 = hepta 3 = tri 8 = octa 4 = tetra 9 = nona

### **Naming ions and chemical compounds**

CHEMISTRY Naming Compounds Handout page 8 of 12 Give the name for each compound given its chemical formula: Formula Individual Ions Name of Compound  $MgCl_2$  Mg +2  $Cl^-$  magnesium chloride  $LiOH$   $ZnCO_3$   $K_2S$   $FePO_4$   $SnO_2$   $CuBr_2$   $Ag_3N$   $Mn(CN)_2$   $AgC_2H_3O_2$  WRITING CHEMICAL FORMULAS GIVEN THE COMPOUND NAME Get the individual ions from the name, then combine them ...

### **Naming Compounds Handout - faculty.seattlecentral.edu**

1 Naming Compounds Tutorial and Worksheet Since we use different methods in naming binary covalent (molecular) compounds and ionic compounds, the first step in naming or writing the formula of a compound is to determine which of the 2 compound classes it belongs to. This

### **Binary Covalent Ionic only - Saddleback College**

Binary Ionic Compounds ANSWER KEY. Directions: First quickly scan the worksheet and circle any metals (as a symbol or as a name) that is a transition metal. Then either give the name or chemical formula as necessary for each problem below. Remember that transition metals can have multiple oxidation states, so you are required

### **Binary Ionic Compounds - Abraham Clark High School**

Naming Ionic Compounds - Answer Key Give the name of the following ionic compounds: Name 1)  $Na_2CO_3$  sodium carbonate 2)  $NaOH$  sodium hydroxide 3)  $MgBr_2$  magnesium bromide 4)  $KCl$  potassium chloride 5)  $FeCl_2$  iron (II) chloride 6)  $FeCl$

### **Naming Ionic Compounds - Answer Key - Mr. Siemianowski**

Naming Ionic Compounds. Give the name and molar mass of the following ionic compounds: ... Naming Ionic Compounds - Answer Key. Give the name and ...

### **Naming Ionic Compounds Practice Worksheet - MAFIADOC.COM**

Naming Covalent Compounds Worksheet. ... Naming Ionic Compounds . Give the name of the following ionic compounds: Name . 1) 2) 3)  $MgBr$  ... Naming Ionic Compounds - Answer Key. Give the name and molar mass of the following ionic compounds: Name . 1) 2) 3)  $MgBr_2$  magnesium bromide

## **Naming Compounds Handout Answers**

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