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1)Temperature change 2)Precipitate formation 3)Color change 4)Release of gas 1 answ. 2. see explanantion 1 answ. 3.

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CHAPTER 8 REVIEW Class S E-cī7b IV hemical Equations and Reactions MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. b d A balanced chemical equation represents all the following except (a) experimentally established facts. (b) the mechanism by which reactants combine to form products.

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CHAPTER 1 REVIEW Matter and Change MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Classify each of the following as a homogeneous or heterogeneous substance. a. sugar d. plastic wrap b. iron filings e. cement sidewalk c. granola bar 2. For each type of investigation, select the most appropriate branch of chemistry from the following

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CHAPTER 6 REVIEW Chemical Bonding MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. a.

6 Chemical Bonding - Effingham County Schools / Overview

Modern Chemistry 47 Chemical Bonding CHAPTER 6 REVIEW Chemical Bonding SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. ____ In metals, the valence electrons are considered to be (a) attached to particular positive ions. (c) immobile. (b) shared by all surrounding atoms.(d) involved in covalent bonds. 2.

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