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Chem 3322 -- Physical Chemistry II Syllabus TA (Dineli) web page: TA (Dineli) web page NEW
LECTURE NOTES Lecture notes, part 1 Lecture notes, part 2

Chem 3322 -- Physical Chemistry II

Calculate a boiling point or pressure using the Antoine Equation: The Pressure-Temperature Nomograph is a graphical application of the Clausius-Clapeyron Equation, which assumes the heat of vaporization is a constant over a pressure range.

Pressure-Temperature Nomograph Interactive Tool | Sigma ...

Algebra: A Combined Approach (4th Edition) Martin-Gay, Elayn Publisher Pearson ISBN 978-0-32172-639-1 Algebra and Trigonometry 10th Edition Larson, Ron Publisher Cengage Learning

Textbook Answers | GradeSaver

In chemistry, an ionic compound is a chemical compound composed of ions held together by electrostatic forces termed ionic bonding. The compound is neutral overall, but consists of positively charged ions called cations and negatively charged ions called anions. These can be simple ions such as the sodium (Na^+) and chloride (Cl^-) in sodium chloride, or polyatomic species such as the ammonium ...

Ionic compound - Wikipedia

Physical Chemistry 3

CHEMISTRY OF BOOK

Raoult's law ($P_i = x_i P_i^0$ / law) is a law of thermodynamics established by French chemist François-Marie Raoult in 1887. It states that the partial pressure of each component of an ideal mixture of liquids is equal to the vapour pressure of the pure component multiplied by its mole fraction in the mixture. In consequence, the relative lowering of vapour pressure of a dilute solution ...

Raoult's law - Wikipedia

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En chimie, une force de van der Waals, interaction de van der Waals ou liaison de van der Waals est une interaction électrique de faible intensité entre deux atomes, molécules, ou entre une molécule et un cristal. Bien qu'il soit possible de décrire sommairement cette interaction en considérant les forces électriques qui sont présentes entre tous les couples de charges électriques qui ...

Force de van der Waals — Wikipédia

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