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Molarity Worksheets - Printable Worksheets

Molarity Problems Worksheet Use M or mol/L as unit for molarity. Remember that 1 Liter = 1000 mL. Do not confuse M, L, and mL! Some problems ask for volume – by algebra, V = n/M. Some problems ask for number of moles – n = V M. 1. What is the molarity of a 0.30 liter solution containing 0.50 moles of NaCl? 2.

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3. What is the molarity of a solution of HNO 3 that contains 12.6 grams HNO 3 in 1.0 L of solution? ? mol HNO 3 = 12.6 g HNO 3 \times 1 mol HNO 3 63.0 g HNO 3 = 0.200 mol HNO 3 M = 0.200 mol HNO 3 1.0 L = 0.200 M 4. How many grams of potassium nitrate are required to prepare 0.250 L of a 0.700 M solution? 0.700 M = moles of solute 0.250 L moles of ...

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Molarity Worksheet W 331 Everett Community College Student Support Services Program What is the molarity of the following solutions given that: 1) 1.0 moles of potassium fluoride is dissolved to make 0.10 L of solution.

Molarity Worksheet W 331 - Everett Community College

Molarity Worksheet # 2 identifiera _____ What does molarity mean? Number of moles of solute. 1 liter solution. What is the molarity of a solution that contains 4.53 moles of lithium nitrate in 2.85 liters of solution? 4.53 mol LiNO 3 = 1.59 M LiNO 3.2.85 L soln

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