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rate of effusion. (Assume all gases are at the same temperature and pressure.) (a) He (b) Xe (c) HCl (d) Cl<sub>2</sub>.

### **CHAPTER 11 REVIEW Gases - Manasquan Public Schools**

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### **Holt McDougal Modern Chemistry Chapter 9: Stoichiometry ...**

(d) number of valence electrons involved in a reaction. 2. d Which of the following would not be studied within the topic of stoichiometry? (a) the mole ratio of Al to Cl in the compound aluminum chloride (b) the mass of carbon produced when a known mass of sucrose decomposes

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Chapter 11 - Gases Chapter 11 focuses on gas behavior and the gas laws. In Chapter 10, students were given an overview of the kinetic-molecular theory of matter and discussed how this theory explains the chemistry of particles in the solid, liquid, and gas phases.

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