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The Avogadro constant, named after scientist Amedeo Avogadro, is the number of constituent particles, usually molecules, atoms or ions that are contained in the amount of substance given by one mole. It is the proportionality factor that relates the molar mass of a substance to the mass of a sample, is designated with the symbol N A or L, and is defined to have the value of exactly 6.022 140 76 ...

Avogadro constant - Wikipedia

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Chemistry 101 - Introduction and Index of Topics

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History of chemistry - Wikipedia

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How can I calculate number of OH group in a Polyol?

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Tests for gases | Chemstuff

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Wiggler - Super Mario Wiki, the Mario encyclopedia

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