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if two or more different compounds are composed of the same two elements, then the ratio of the masses of the second element combined with a certain mass of the first element is always a ratio of small whole numbers; textbook: if two or more different compounds are composed of the same two elements with a certain mass of the first element is always a ratio of small whole numbers

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see figure 3-2 page 67. Dalton's theory and Law of Definite Proportion --see figure 3-3 page 67 . C: Modern Atomic Theory. Some of Dalton's postulates have been proven to be incorrect. His third postulate indicated that atoms cannot be subdivided. His second postulate indicates atoms of the same element have identical mass. Isotopes

Chapter 3 Modern Chemistry - Prepchem

The Atoms: The Building Blocks of Matter chapter of this Holt McDougal Modern Chemistry Companion Course helps students learn the essential lessons associated with atoms.

Holt McDougal Modern Chemistry Chapter 3: Atoms: The ...

Modern Chemistry • CHAPTER 7 HOMEWORK 7-3 (pp. 211-213) VOCABULARY Write the prefix that corresponds to each number. Try not to use the book. Number Prefix Number Prefix 16 27 38 49 510 GRAPHIC ORGANIZER Complete the chart by supplying the missing information. Name of Compound Formula of Compound CO₂ Nitrogen dioxide S₂ Cl₂ Carbon ...

HOMEWORK 7-3 - Kettering City School District

CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Explain the difference between the mass number and the atomic number of a nuclide. Mass number is the total number of protons and neutrons in the nucleus of an isotope.

3 Atoms: The Building Blocks of Matter - Ms. Sheehan's ...

CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT ANSWER Answer the following questions in

the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

10 States of Matter - Ms. Agostine's Chemistry Page

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Modern Chemistry 1 Solutions CHAPTER 12 REVIEW Solutions Teacher Notes and Answers Chapter 12 SECTION 1 SHORT ANSWER 1. c 2. a 3. b 2. a. alcohol b. water c. the gels 3. The mixture is a colloid. The properties are consistent with those reported in Table 3 on page 404 of the text. The particle size is small, but not too small, and the mixture

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