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States that equal volumes of gases at the same temperature and... the pressure of a gas is directly proportional to the Kelvin t... the rate of effusion of a gas is inversely proportional to the... The amount of force exerted per unit area of a surface The SI unit for force; the force that will increase the speed... The SI unit...

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Typically, hydrogen gas is collected by water displacement. Recalculate the volume of the dry gas taking into account the partial pressure of water vapor at the reaction temperature. HRW material copyrighted under notice appearing earlier in this work. SHORT ANSWER Answer the following questions in the space provided.

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given volume of a given gas at STP.You can also use the molar volume of a gas to find the volume, at STP, of a known number of moles or a known mass of a gas. These types of problems can also be solved using the ideal gas law, as you will see in Section 11-2. MOLECULAR COMPOSITION OF GASES 335 Hydrogen gas 1 Volume 1 Molecule Chlorine gas 1 ...

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Concept Review j j. Section: What is Ch€iSti'Y Cornpiete each statement below by underlining the correct word or phrase in brackets. 1. A chemical is any substance that has [definit,e, indefinite] composition. Changes in chemicals, or chemical reactions, take place [only in test tubes, all. 2.

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B. - Charles's Law: Volume-Temperature Relationship 1. The experiments of Jacques Charles showed that all gases _____ to the same extent when

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