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Modern Chemistry: Chapter Tests with Answer Key, 2006 ...

Modern Chemistry 91 Chapter Test Name Class Date Chapter Test B, continued 27. Distinguish between ionic crystals and metallic crystals. PART V Write the answers to the following questions on the line to the left, and show your work in the space provided. The molar enthalpy of fusion for water is 6.008 kJ/mol. 28.

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Modern Chemistry 69 Chapter Test Chapter: Chemical Equations and Reactions PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. ____ 1. The production of a slightly soluble solid compound in a double-displacement reaction results in the formation of a a. gas. b. precipitate.

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a way to describe a chemical reaction using chemical formulas.... a solid that forms from a solution during a chemical reaction. A number in front of a chemical formula in an equation that in.... an equation in which the reactants and products in a chemical....

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SHORT ANSWER Answer the following questions in the space provided. 1. a A chemical bond between atoms results from the attraction between the valence electrons and of different atoms. (a) nuclei (c) isotopes (b) inner electrons (d) Lewis structures 2. b A covalent bond consists of (a) a shared electron. (c) two different ions. (b) a shared electron pair.

6 Chemical Bonding - Effingham County Schools / Overview

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CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

10 States of Matter - Ms. Agostine's Chemistry Page

Modern Chemistry 84 Chapter Test Chapter: States of Matter In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. ____ 1. The kinetic-molecular theory of gases assumes that the particles of an ideal gas are separated by great distances. This implies that the gas

Assessment Chapter Test A - Kettering City School District

Modern Chemistry 46 Chapter Test Chapter: Chemical Bonding In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. ____ 1. The charge on an ion is a. always positive. b. always negative. c. either positive or negative. d. zero. ____ 2.

Assessment Chapter Test A - Kettering City School District

SHORT ANSWER Answer the following questions in the space provided. 1. b The coefficients in a chemical equation represent the (a) masses in grams of all reactants and products.

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