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Dissociation. The separating of a molecule into simpler molecules, atoms, radicals, or ions. Net Ionic Equation. An equation that includes only those compounds and ions that undergo a chemical change in a reaction in an aqueous solution.

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Modern Chemistry Chapter 13 Ions in Aqueous Solutions and Colligative Properties Section 1- Compounds in Aqueous Solutions dissociation is the separation of ions that occurs when an ionic compound dissolves see sample problem A on page 436 do practice problem #1 on page 436 Solubility precipitation reactions occur when combinations of ions in a solution have an extremely low solubility and a ...

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figure 13-2a page 396 - filter the solution through filter paper and both solute and solvent will pass through - the solute particles are that small - 0.01 to 1 nm in diameter. figure 13-2b page 396. Types of Solution. different types list in table 13-1 page 396 - note that you must specify which is solute and which is solvent

F.J. DeHoratius - Chapter 13 - Modern Chemistry

Modern Chemistry 1 Solutions CHAPTER 12 REVIEW Solutions Teacher Notes and Answers Chapter 12 SECTION 1 SHORT ANSWER 1. c 2. a 3. b 2. a. alcohol b. water c. the gels 3. The mixture is a colloid. The properties are consistent with those reported in Table 3 on page 404 of the text. The particle size is small, but not too small, and the mixture

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Holt Modern Chemistry Chapter 13 Section 1 Review Answers ... Modern Chemistry Chapter 12-Solutions Section 1- Types of Mixtures Solutions are homogeneous mixtures of two or more substances in a single phase. Soluble describes a substance as capable of being dissolved. Solvent is the dissolving medium in a solution.

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