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Modern Chemistry 33 Chapter Test Name Class Date Chapter Test B, continued 15. The energy state of an atom is called its ground state. 16. The number of waves that pass a point in one second is called. 17. When an electron drops from a higher-energy state to a lower-energy state, a(n) spectrum is produced. 18.

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CHAPTER 4 PRE-TEST Arrangement of Electrons in Atoms In the space provided, write the letter of the term that best completes each ... Chapter 4 PRE TEST, continued \_\_\_\_ 8. What is the frequency of a photon whose energy is  $3.4 \times 10^{-19} \text{ J}$ ? ... Modern Chemistry 4 Electron Arrangement Chapter 4 PRE-TEST, continued

### **CHAPTER 4 PRE-TEST Arrangement of Electrons in Atoms**

Holt McDougal Modern Chemistry 1 Chapter Test Assessment Chapter Test B Chapter: The Periodic Law PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. \_\_\_\_ 1. In his periodic table, Mendeleev did not list all of the elements in order

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Holt McDougal Modern Chemistry 1 Chapter Test Assessment Chapter Test A Chapter: The Periodic Law Use the periodic table below to answer the questions in this Chapter Test. In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. \_\_\_\_ 1.

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Modern Chemistry 84 Chapter Test Chapter: States of Matter In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. \_\_\_\_ 1. The kinetic-molecular theory of gases assumes that the particles of an ideal gas are separated by great distances. This implies that the gas

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CHAPTER 1 REVIEW Matter and Change MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Classify each of the following as a homogeneous or heterogeneous substance. a. sugar d. plastic wrap b. iron filings e. cement sidewalk c. granola bar 2. For each type of investigation, select the most appropriate branch of chemistry from the following

### **mc06se cFMsr i-vi**

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Modern Chemistry 92 Chapter Test Name Class Date Chapter Test B, continued 31. Calculate the energy released by freezing 13.8 g of a liquid. Its molar mass is 82.9 g/mol, and its molar enthalpy of fusion is 4.60 kJ/mol. PART VI Define the term equilibrium vapor pressure. Use it to explain the following two phenomena.

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MODERN CHEMISTRY The atomic number of nickel-60 is 28. How many neutrons does this isotope have? Carbon-14 has 8 neutrons. What is the atomic number of carbon-14? 11 CHAPTER 3 TEST HW material copyrighted under notice appearing earlier in this work.

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