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Chemical Equations and Reactions SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. List four metals that will not replace hydrogen in an acid. Choose from Cu, Ag, Au, Pt, Sb, Bi, and Hg. 2. Consider the metals iron and silver, both listed in Table 3 on page 286 of the text. Which one

8 Chemical Equations and Reactions - David Brearley High ...

Modern Chemistry 1 Solutions CHAPTER 12 REVIEW Solutions Teacher Notes and Answers Chapter 12 SECTION 1 SHORT ANSWER 1. c 2. a 3. b 2. a. alcohol b. water c. the gels 3. The mixture is a colloid. The properties are consistent with those reported in Table 3 on page 404 of the text. The particle size is small, but not too small, and the mixture

CHAPTER 12 REVIEW Solutions - Weebly

SHORT ANSWER Answer the following questions in the space provided. 1. Given the following equation: $C_3H_4(g) + xO_2(g) \rightarrow 3CO_2(g) + 2H_2O(g)$ 4 a. What is the value of the coefficient x in this equation? 40.07 g/mol b. What is the molar mass of C_3H_4 ? 2 mol O₂:1 mol H₂O c. What is the mole ratio of O₂ to H₂O in the above equation? 0.20 ...

mc06se cFMs i-vi

States of Matter SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. Identify whether the descriptions below describe an ideal gas or a real gas. ideal gas a. ... SECTION 5 MODERN CHEMISTRY STATES OF MATTER. STATES OF MATTER MODERN CHEMISTRY.

10 States of Matter - Ms. Agostine's Chemistry Page

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Modern Chemistry 47 Chemical Bonding CHAPTER 6 REVIEW Chemical Bonding SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. ____ In metals, the valence electrons are considered to be (a) attached to particular positive ions. (c) immobile. (b) shared by all surrounding atoms. (d) involved in covalent bonds. 2.

CHAPTER 6 REVIEW Chemical Bonding

CHAPTER 1 REVIEW Matter and Change MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Classify each of the following as a homogeneous or heterogeneous substance. a. sugar d. plastic wrap b. iron filings e. cement sidewalk c. granola bar 2. For each type of investigation, select the most appropriate branch of chemistry from the following

mc06se cFMsr i-vi

Chemical Bonding SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. b In metals, the valence electrons are considered to be (a) attached to particular positive ions. (c) immobile. (b) shared by all surrounding atoms. (d) involved in covalent bonds. 2. a The fact that metals are malleable and ionic crystals are ...

6 Chemical Bonding - Effingham County Schools / Overview

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Modern Chemistry 121 Acids and Bases CHAPTER 14 REVIEW Acids and Bases SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Answer the following questions according to the Brønsted-Lowry definitions of acids and bases: ____ a. What is the conjugate base of H_2SO_3 ? ____ b.

CHAPTER 14 REVIEW Acids and Bases - Manasquan Public Schools

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Modern Chemistry • CHAPTER 8 HOMEWORK 8-1 (pp. 241–245) VOCABULARY Define. ... Circle the letter of the best answer. 1. Which sentence is the correct translation for the equation $4\text{Fe(s)} + 3\text{O}_2\text{(g)} \rightarrow 2\text{Fe}_2\text{O}_3\text{(s)}$? a. Iron and dioxygen react to form bi-iron trioxide. b.

HOMEWORK 8-1 - Los Banos Unified School District

Section Q Iz: Determining Chemical Formula In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. For calculations that require the atomic mass of an element, round the atomic mass found on the periodic table on page 49 to two decimal places. 1.

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The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided.

1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered.

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