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**Mole Ratios Worksheets - Printable Worksheets**

HS Chemistry POGIL Activity Topic: ... show how you calculated your answer.  $21.0 \text{ mol A} \times 5 \text{ mol C} = 52.5 \text{ mol C}$ . ... mole ratio between moles of given to moles of  $\text{H}_2$  definition moles explained reacting mole ratio molar  $\text{H}_2$

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The molecular weight of  $\text{O}_2$  is 32.00 u, so a mole of  $\text{O}_2$  would have a mass of 32.00 g and would contain  $6.022 \times 10^{23}$   $\text{O}_2$  molecules. Each  $\text{O}_2$  molecule is composed of two oxygen atoms, so one mole of  $\text{O}_2$  contains two moles of oxygen atoms. 3. The mole is sometimes described as “the chemist’s dozen”.

**Moles & Stoichiometry Answers Key Questions & Exercises**

substance in the reaction is called the mole ratio. For example, the mole ratio of sodium hydroxide to sodium sulfate is 2:1. or 2/1. The mole ratio can be used to “convert” one substance in an equation to another using dimensional analysis (sometimes called “factor-label” – see the POGIL entitled Basic Skills Supplement –

**POGIL 04 - Stoichiometry 4 - Reactants, Products, and the ...**

MODEL 1: Mole-to-Mole Stoichiometry (1-step) Example: mole-to-mole Question: Given the following equation:  $\text{N}_2(\text{g}) + 3 \text{H}_2(\text{g}) \rightarrow 2 \text{NH}_3(\text{g})$  How many moles of  $\text{H}_2$  are needed to produce 34.8 moles of  $\text{NH}_3$ ? Mole ratio Answer:  $34.8 \text{ mol NH}_3 \times \frac{3 \text{ mol H}_2}{2 \text{ mol NH}_3} = 52.2 \text{ mol H}_2$  OR  $34.8 \text{ mol NH}_3 \times 3 \text{ mol H}_2 = 52.2 \text{ mol H}_2$  2 mol  $\text{NH}_3$  2 mol  $\text{NH}_3$  Key Questions: 1.

### **POGIL- Stoichiometry**

Mole Ratio Worksheet Pogil Worksheets have been used as an effective medium to improve the imagination, fine motor skills and understanding of your kids. Worksheets today are designed to cater to every aspect of your child's learning and they are given a lot of importance by parents and teachers alike.

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