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Any alkali added to the solution is consumed by the hydronium ions. These ions are mostly regenerated as the equilibrium moves to the right and some of the acid dissociates into hydronium ions and the conjugate base. If a strong acid is added, the conjugate base is protonated, and the pH is almost ...

Buffer solution - New World Encyclopedia

In chemistry, a base is thought of as a substance which can accept protons or any chemical compound that yields hydroxide ions (OH⁻) in solution. It is also commonly referred to as any substance that can react with an acid to decrease or neutralize its acidic properties, change the color of indicators (e.g. turn red litmus paper blue), feel slippery to the touch when in solution, taste bitter ...

Base (chemistry) - New World Encyclopedia

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PM2.5 in China: Measurements, sources, visibility and ...

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Keratin: Structure, mechanical properties, occurrence in ...

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