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Modern Chemistry 91 Chapter Test Name Class Date Chapter Test B, continued 27. Distinguish between ionic crystals and metallic crystals. PART V Write the answers to the following questions on the line to the left, and show your work in the space provided. The molar enthalpy of fusion for water is 6.008 kJ/mol. 28.

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Modern Chemistry 35 Chapter Test Name Class Date Chapter Test B, continued Write the orbital notation for the following elements in the space provided. 37. lithium, atomic number 3 38. carbon, atomic number 6 39. neon, atomic number 10 PART VI Write the answers to the questions on the line to the left, and show your work in the space provided.

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MODERN CHEMISTRY STATES OF MATTER 81 ... Name Date Class CHAPTER 10 REVIEW States of Matter SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. Identify whether the descriptions below describe an ideal gas or a real gas. ... States of Matter Answer . STATES OF MATTER MODERN CHEMISTRY ...

10 States of Matter - Ms. Agostine's Chemistry Page

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Modern Chemistry 84 Chapter Test Chapter: States of Matter In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. ____ 1. The kinetic-molecular theory of gases assumes that the particles of an ideal gas are separated by great distances. This implies that the gas

Assessment Chapter Test A - Kettering City School District

Modern Chemistry • CHAPTER 9 HOMEWORK 9-1 (pp. 275-279) VOCABULARY ... Modern Chemistry • CHAPTER 9 HOMEWORK 9-2 (pp. 280-282) VOCABULARY ... Circle the letter of the best answer. 1. For the general chemical equation $2A + 3B \rightarrow 4C + D$, how many moles of A are required to produce

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