

## *Modern Chemistry Chapter 4 Test Answers*

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Modern Chemistry 33 Chapter Test Name Class Date Chapter Test B, continued 15. The energy state of an atom is called its ground state. 16. The number of waves that pass a point in one second is called. 17. When an electron drops from a higher-energy state to a lower-energy state, a(n) spectrum is produced. 18.

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CHAPTER 4 PRE-TEST Arrangement of Electrons in Atoms In the space provided, write the letter of the term that best completes each ... Chapter 4 PRE TEST, continued \_\_\_\_ 8. What is the frequency of a photon whose energy is  $3.4 \times 10^{-19} \text{ J}$ ? ... Modern Chemistry 4 Electron Arrangement Chapter 4 PRE-TEST, continued

### **CHAPTER 4 PRE-TEST Arrangement of Electrons in Atoms**

Taken from the book Modern Chemistry by Holt, Rinehart, and Winston on Chapters 4 and 5, which deals with electrons and the periodic table. Includes the chapter vocabulary and a few other useful things.

### **Modern Chemistry (Holt, Rinehart, and Winston): Chapter 4 ...**

Modern Chemistry 92 Chapter Test Name Class Date Chapter Test B, continued 31. Calculate the energy released by freezing 13.8 g of a liquid. Its molar mass is 82.9 g/mol, and its molar enthalpy of fusion is 4.60 kJ/mol. PART VI Define the term equilibrium vapor pressure. Use it to explain the following two phenomena.

### **Assessment Chapter Test B - clarkchargers.org**

Holt McDougal Modern Chemistry 1 Chapter Test Assessment Chapter Test B Chapter: The Periodic Law PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. \_\_\_\_ 1. In his periodic table, Mendeleev did not list all of the elements in order

### **Assessment Chapter Test B - Wag & Paws**

E.C. The approximate composition of naturally occurring magnesium is as follows: 79% magnesium-24, 10% magnesium-25, and 11% magnesium-26. Calculate the atomic mass of magnesium.

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