

Mixed Mole Problems Chemistry If8766 Answers

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Mixed Mole Problems Chemistry If8766

MIXED MOLE PROBLEMS - KEY 1. a) How many grams are there in 1.5×10^{25} molecules of CO₂? 1.110g 1 mol 44.0 g 6.0210molecules 1.510molecules CO1mol 3 23 2 25!=!!!! b) What volume would the CO

KEY - CP - Mixed Mole Problems

MIXED PROBLEMS Name What volume of NH₃ at STP is produced if 25.0 g of N₂ is reacted with an excess of ... Chemistry IF8766 Olnstructional Fair, Inc. STOICHIOMETRY: MOLE-MOLE PROBLEMS I. N₂ + 3H₂ Name How many moles of hydrogen are needed to completely react with two moles

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Mixed Mole Problems Directions: Solve the following problems. 1) How many grams are there in 1.5×10^{25} molecules of CO₂? 2) What volume would the CO₂ in Problem 1 occupy at STP? 3) A sample of NH₃ gas occupies 75.0 liters at STP. How many molecules is this? 4) What is the mass of the sample of NH₃ in Problem 3?

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Mixed Mole Problems Directions: Solve the following problems. 1) How many grams are there in 2.5×10^{25} molecules of CO₂? ANSWER: 1800 g 2) What volume would the CO₂ in Problem 1 occupy at STP? ANSWER: 930 L 3) A sample of NH₃ gas occupies 57.0 liters at STP. How many molecules is this? ANSWER: 1.53×10^{24} 4) What is the mass of the sample of NH₃

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Stoichiometry Mixed Problems Instructional Fair Answers

Stoichiometry: Mixed Problems (KEY) 1) N₂ + 3H₂ → 2NH₃ What volume of NH₃ at STP is produced if 25.0 of N₂ is reacted with an excess of H₂? 3 3 3 2 3 2 2 40.0L NH 1mol NH 22.4L NH 1mol N 2mol NH 28.0g N 25.0g N 1mol N × × × = 2) 2KClO₃ → 2KCl + 3O₂ If 5.0g of KClO₃ is decomposed, what volume of O₂ is produced at STP? 2

Stoichiometry: Mixed Problems (KEY)

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3. How many grams of ammonia are produced in the reaction in Problem 2? $4 \text{ O}_2 + \text{AgNO}_3 + 0 \text{ Ag} +$
How many grams of silver chloride are produced from 5.0 g of silver nitrate reacting with an excess
of barium chloride? 5J 143B 5. How much barium chloride is necessary to react with the silver
nitrate in Problem 4? ÛQCA vvtc Chemistry IF8766 64

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