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The vapor pressure of the NaCl solution forming is lower than the vapor pressure of pure water at 0°C. The lower vapor pressure of the NaCl solution results in a lower freezing point. Name Date Class MODERN CHEMISTRY CHAPTER 14 MIXED REVIEW 121 HRW material copyrighted under notice appearing earlier in this work.

14 Ions in Aqueous Solutions and Colligative Properties

Modern Chemistry 1 Solutions CHAPTER 12 REVIEW Solutions Teacher Notes and Answers Chapter 12 SECTION 1 SHORT ANSWER 1. c 2. a 3. b 2. a. alcohol b. water c. the gels 3. The mixture is a colloid. The properties are consistent with those reported in Table 3 on page 404 of the text. The particle size is small, but not too small, and the mixture

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Modern Chemistry 113 Ions In Aqueous Solutions And Colligative Properties CHAPTER 13 REVIEW

Ions in Aqueous Solutions and Colligative Properties SECTION 2 PROBLEMS Write the answer on the line to the left. Show all your work in the space provided. 1. _____ a. Predict the boiling point of a 0.200 m solution of glucose in water.

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HOMEWORK 13-1 - Los Banos Unified School District

Modern Chemistry 2 Stoichiometry CHAPTER 9 REVIEW Stoichiometry MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Given the following equation: C ... Modern Chemistry 3 Stoichiometry MIXED REVIEW continued c. If 0.1 mol of N₂ combine with H₂, what must be true about the quantity of H

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CHAPTER 1 REVIEW Matter and Change MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Classify each of the following as a homogeneous or heterogeneous substance. a. sugar d. plastic wrap b. iron filings e. cement sidewalk c. granola bar 2. For each type of investigation, select the most appropriate branch of chemistry from the following

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