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Modern Chemistry Test B Answer

Modern Chemistry 88 Chapter Test Chapter: States of Matter PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

____ 1. According to the kinetic-molecular theory, particles of an ideal gas a. attract each other but do not collide. b. repel each other and collide.

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Modern Chemistry 33 Chapter Test Name Class Date Chapter Test B, continued 15. The energy state of an atom is called its ground state. 16. The number of waves that pass a point in one second is called. 17. When an electron drops from a higher-energy state to a lower-energy state, a(n) spectrum is produced. 18.

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Holt McDougal Modern Chemistry 3 Chapter Test Chapter Test B, continued 16. The measure of the ability of an atom in a chemical compound to attract electrons from another atom in the compound is called _____. 17. The energy required to remove one electron from an atom is called its _____. 18.

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Modern Chemistry 57 Chapter Test Chapter Test B, continued ____ 7. In a reaction, the ions of two compounds exchange places in aqueous solution to form two new compounds. This reaction is called a a. synthesis reaction. b. decomposition reaction. c. single-displacement reaction. d. double-displacement reaction. ____ 8.

Assessment Chapter Test B - Hatboro

Modern Chemistry 139 Chapter Test Chapter: Acid-Base Titration and pH PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. ____ 1. The pH scale generally ranges from a. 0 to 1. b. 1 to 1. c. 0 to 7. d. 0 to 14. ____ 2.

Assessment Chapter Test B - Baumapedia

Holt McDougal Modern Chemistry 1 Chapter Test Assessment Chapter Test A Chapter: The Periodic Law Use the periodic table below to answer the questions in this Chapter Test. In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. ____ 1.

Assessment Chapter Test A - mrbijlani.weebly.com

Modern Chemistry 87 Chapter Test Name Class Date Chapter Test A, continued ____ 19. In general, ionic compounds have a. high boiling points and low melting points. b. low boiling points and high melting points. c. high boiling points and high melting points. d. low boiling points and low melting points. ____ 20.

Assessment Chapter Test A - Kettering City School District

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Modern Chemistry 46 Chapter Test Chapter: Chemical Bonding In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

_____ 1. The charge on an ion is a. always positive. b. always negative. c. either positive or negative. d. zero. _____ 2.

Assessment Chapter Test A - Kettering City School District

CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

10 States of Matter - Ms. Agostine's Chemistry Page

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Chemical Bonding SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. b In metals, the valence electrons are considered to be (a) attached to particular positive ions. (c) immobile. (b) shared by all surrounding atoms. (d) involved in covalent bonds. 2. a The fact that metals are malleable and ionic crystals are ...

6 Chemical Bonding - Effingham County Schools / Overview

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CHAPTER 1 REVIEW Matter and Change MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Classify each of the following as a homogeneous or heterogeneous substance. a. sugar d. plastic wrap b. iron filings e. cement sidewalk c. granola bar 2. For each type of investigation, select the most appropriate branch of chemistry from the following

mc06se cFMsr i-vi

MODERN CHEMISTRY HR W material copyrighted under notice in this work. Name CHAPTER 3 TEST continued Date Class SHORT ANSWER Write the answers to the following questions in the space

provided. 22, Give three of the main concepts in Dalton's atomic theory. 23. What is molar mass? How is it related to atomic mass?

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