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Modern Chemistry • CHAPTER 22 HOMEWORK 22-2 (pp. 702-704) VOCABULARY Write true or false.

1. The band of stability represents the stable nuclei cluster over a range of electron-proton ratios. \_\_\_\_ 2. The stability of a nucleus is greatest when the nucleons are in a 1:1 ratio. \_\_\_\_ 3. 2, 20, 50, and 126 are all magic numbers. \_\_\_\_ 4.

**Modern Chemistry CHAPTER 22 HOMEWORK 22-1**

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**mc06se cFMsr i-vi**

Modern Chemistry 47 Chemical Bonding CHAPTER 6 REVIEW Chemical Bonding SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. \_\_\_\_ In metals, the valence electrons are considered to be (a) attached to particular positive ions. (c) immobile. (b) shared by all surrounding atoms. (d) involved in covalent bonds. 2.

**CHAPTER 6 REVIEW Chemical Bonding**

Algebra: A Combined Approach (4th Edition) Martin-Gay, Elayn Publisher Pearson ISBN 978-0-32172-639-1 Algebra and Trigonometry 10th Edition Larson, Ron Publisher Cengage Learning

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Modern Chemistry 88 Chapter Test Chapter: States of Matter PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

\_\_\_\_ 1. According to the kinetic-molecular theory, particles of an ideal gas a. attract each other but do not collide. b. repel each other and collide.

**Assessment Chapter Test B - clarkchargers.org**

Holt McDougal Modern Chemistry Student Edition. Publisher: Holt McDougal. ISBN-13: 9780547586632. The Holt Modern Chemistry Student Edition (Hardcover) stand-alone textbook is available for an additional student.. Modern Chemistry presents a balanced approach to conceptual and problem-solving instruction. It was designed to make chemistry accessible to more students than ever.

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CHAPTER 7 • HOMEWORK 7-1 VOCABULARY 1. false 2. true 3. false 4. false 5. true SKILL BUILDER AgCl—In the compound, there is one atom of silver for each atom of chlorine. Ba(OH) 2 —The subscript 2 refers to everything inside the parentheses, so there are two hydroxide ions for

**HOMEWORK 7-2**

and 0 kg 0 g 0 0 m × × × × × × × × ×

**APPENDIX D Problem Bank - St. Charles Parish**

CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

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