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Section Quiz: Using Chemical Formulas In the space provided, write the letter of the term or phrase that best completes ... Modern Chemistry . Name Section Quiz, continued Class Date 6. How many total moles of atoms are there in one mole of silver(I) phosphate, $\text{Ag}_3\text{P}_2\text{O}_7$? d. 11 7. What is the correct molar mass of $\text{Ag}_3\text{P}_2\text{O}_7$?

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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In a Stock system name such as iron(III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many sulfate ions can be attached to the iron atom. (c) the charge on ...

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Modern Chemistry 57 Chemical Bonding CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1.

Label each of the following statements as True or False: ____ a. If the formula mass of one molecule is x u, the molar mass is x g/mol. ____ b.

CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds

CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds MIXED REVIEW SHORT ANSWER

Answer the following questions in the space provided. 1. Write formulas for the following compounds: CuCO₃ 3 a. copper(II) carbonate Na₂SO₃ 3 b. sodium sulfite (NH₄)₃PO₄ 4 c. ammonium phosphate SnS₂ 2 d. tin(IV) sulfide HNO₂ 2 e. nitrous acid 2.

7 Chemical Formulas and Chemical Compounds

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Modern Chemistry • CHAPTER 7 HOMEWORK 7-3 (pp. 211-213) VOCABULARY Write the prefix that corresponds to each number. Try not to use the book. Number Prefix Number Prefix 16 27 38 49 510 GRAPHIC ORGANIZER Complete the chart by supplying the missing information. Name of Compound Formula of Compound CO₂ Nitrogen dioxide S₂ Cl₂ Carbon ...

HOMEWORK 7-3 - Kettering City School District

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Modern Chemistry • CHAPTER 7 HOMEWORK 7-1 (pp. 203-206) VOCABULARY Write true or false for each statement. 1. A chemical formula indicates the relative number of molecules of each kind in a chemical compound. ____ 2. Monoatomic ions are ions formed from a single atom. ____ 3.

HOMEWORK 7-1 - Los Banos Unified School District

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