

## ***Mole Ratios Pogil Answer Key***

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The molecular weight of O<sub>2</sub> is 32 Pogil the mole answer key. 00 u, so a mole of O<sub>2</sub> would have a mass of 32. 00 g and would contain 6. 022 x 10<sup>23</sup> O. 2 molecules. Each O<sub>2</sub> molecule is composed of two oxygen atoms, so one mole of O<sub>2</sub> contains two moles of oxygen atoms Pogil the mole answer key. 3. The mole is sometimes described as "the chemist's dozen".

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Chem 115 POGIL Worksheet - Week 4 Moles & Stoichiometry Answers Key Questions & Exercises 1. The atomic weight of carbon is 12.0107 u, so a mole of carbon has a mass of 12.0107 g.

#### **Moles & Stoichiometry Answers Key Questions & Exercises**

This is called Stoichiometry- (stoi-key-ah-meh-tree) Another way of looking at it is using the mole ratio from the balanced equation and information about one compound in the reaction to determine information about another compound in the equation. A mole ratio is a conversion factor derived from the coefficients of a balanced chemical equation

#### **POGIL- Stoichiometry**

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HS Chemistry POGIL Activity Topic: ... show how you calculated your answer.  $21.0 \text{ mol A} * 5 \text{ mol C} = 52.5 \text{ mol C}$ . ... mole ratio between moles of given to moles of  $\text{H}_2\text{O}$  definition moles explained reacting mole ratio molar  $\text{H}_2\text{O}$

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