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# Statistical Mechanics, Donald A. McQuarrie

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## Raoult's law - Wikipedia

In chemistry, colligative properties are properties of solutions that depend on the ratio of the number of solute particles to the number of solvent molecules in a solution, and not on the nature of the chemical species present. The number ratio can be related to the various units for concentration of solutions. The assumption that solution properties are independent of nature of solute ...

## Colligative properties - Wikipedia

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