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Modern Chemistry Chapter 12. STUDY. PLAY. soluble. Capable of being dissolved. solution. A homogeneous mixture of two or more substances in a single phase. solvent. In a solution, the substance in which the solute dissolves. suspension. Mixture in which a substance will not dissolve in another and quickly separates out if left to stand.

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Modern Chemistry • CHAPTER 12 HOMEWORK 12-3 (pp. 273–375) ... MODERN CHEMISTRY SECTION 12-3 REVIEW 103 HRW material copyrighted under notice appearing earlier in this work. Name Date Class . SECTION 12-3 continued b. Explain the difference between an open system and a closed system. 5.

HOMEWORK 12-3

Modern Chemistry 1 Solutions CHAPTER 12 REVIEW Solutions Teacher Notes and Answers Chapter 12 SECTION 1 SHORT ANSWER 1. c 2. a 3. b 2. a. alcohol b. water c. the gels 3. The mixture is a colloid. The properties are consistent with those reported in Table 3 on page 404 of the text. The particle size is small, but not too small, and the mixture

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Modern Chemistry 88 Chapter Test Chapter: States of Matter PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

_____ 1. According to the kinetic-molecular theory, particles of an ideal gas a. attract each other but do not collide. b. repel each other and collide.

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Modern Chemistry 71 Chapter Test Name Class Date Chapter Test B, continued 17. A substance combines with oxygen, releasing a large amount of energy as heat ... 12. 3.4 13. SiO 2 14. Cl 4 15. Sn(CrO 4) 2 16. Ba(OH) 2 17. phosphorus(III) iodide 18. dinitrogen tetroxide 19. iron(II) nitrite 20. carbon tetrachloride

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CHAPTER 1 REVIEW Matter and Change MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Classify each of the following as a homogeneous or heterogeneous substance. a. sugar d. plastic wrap b. iron filings e. cement sidewalk c. granola bar 2. For each type of investigation, select the most appropriate branch of chemistry from the following

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Modern Chemistry • CHAPTER 7 HOMEWORK 7-3 (pp. 211–213) VOCABULARY Write the prefix that corresponds to each number. Try not to use the book. Number Prefix Number Prefix 16 27 38 49 510 GRAPHIC ORGANIZER Complete the chart by supplying the missing information. Name of Compound Formula of Compound CO 2 Nitrogen dioxide S 2 Cl 2 Carbon ...

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