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CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

### 10 States of Matter - Ms. Agostine's Chemistry Page

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CHEMISTRY NOTES - Chapter 10 States of Matter Goals : To gain an understanding of : 1.. The kinetic theory of matter. ... 2. 10% less dense than solids (average) (water is an exception - its liquid state is less dense than its solid state) ... Plasma is a state of matter in which gaseous atoms become excited to the point that electrons fly ...

### CHEMISTRY NOTES - Chapter 10 States of Matter

Modern Chemistry 88 Chapter Test Chapter: States of Matter PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

\_\_\_\_ 1. According to the kinetic-molecular theory, particles of an ideal gas a. attract each other but do not collide. b. repel each other and collide.

### Assessment Chapter Test B - clarkchargers.org

Modern Chemistry 84 Chapter Test Chapter: States of Matter In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. \_\_\_\_ 1.

The kinetic-molecular theory of gases assumes that the particles of an ideal gas are separated by great distances. This implies that the gas

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Modern Chemistry 84 Chapter Test Chapter: States of Matter In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. \_\_\_\_ 1. The kinetic-molecular theory of gases assumes that the particles of an ideal gas are separated by great distances. This implies that the gas

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Modern Chemistry 89 States Of Matter CHAPTER 10 REVIEW States of Matter SECTION 5 SHORT ANSWER Answer the following questions in the space provided. 1. Indicate whether each of the following is a physical or chemical property of water. \_\_\_\_ a. The density of ice is less than the density of liquid water. \_\_\_\_ b.

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