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Modern Chemistry 69 Chapter Test Chapter: Chemical Equations and Reactions PART I In the space provided, write the letter of the term or phrase that best com-pletes each statement or best answers each question. _____ 1. The production of a slightly soluble solid compound in a double-displacement reaction results in the formation of a a. gas. b ...

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Modern Chemistry 35 Chapter Test Name Class Date Chapter Test B, continued Write the orbital notation for the following elements in the space provided. 37. lithium, atomic number 3 38. carbon, atomic number 6 39. neon, atomic number 10 PART VI Write the answers to the questions on the line to the left, and show your work in the space provided.

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SHORT ANSWER Answer the following questions in the space provided. 1. Use the concept of potential energy to describe how a covalent bond forms between two atoms. As the atoms involved in the formation of a covalent bond approach each other, the electron-proton attraction is stronger than the electron-electron and proton-proton repulsions.

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CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

10 States of Matter - Ms. Agostine's Chemistry Page

SHORT ANSWER Answer the following questions in the space provided. 1. b A balanced chemical equation represents all the following except (a) experimentally established facts.

8 Chemical Equations and Reactions - David Brearley High ...

Modern Chemistry 143 Chapter Test Name Class Date Chapter Test B, continued PART V Write the answers to the following problems on the line to the left, and show your work in the space provided.

____39. What is the hydronium ion concentration of an aqueous solution that has a pH of 5.0?

____40.

Assessment Chapter Test B - Baumapedia

SHORT ANSWER Answer the following questions in the space provided. 1. Given the following equation: C 3H 4(g) + xO 2(g) \rightarrow 3CO 2(g) + 2H 2O(g) 4 a. What is the value of the coefficient x in this equation? 40.07 g/mol b. What is the molar mass of C 3H 4? 2 mol O 2:1 mol H 2O c. What is the mole ratio of O 2 to H 2O in the above equation? 0.20 ...

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Modern Chemistry 46 Chapter Test Chapter: Chemical Bonding In the space provided, write the
letter of the term or phrase that best completes each statement or best answers each question.
1. The charge on an ion is a. always positive. b. always negative. c. either positive or negative
d. zero. 2.

Assessment Chapter Test A - Kettering City School District

PROBLEMS Write the answers to the questions on the line to the left, and show your work in the space rovided. 25. 26. MODERN CHEMISTRY The atomic number of nickel-60 is 28. How many neutrons does this iso- tope have? Carbon-14 has 8 neutrons. What is the atomic number of carbon-14? 11 CHAPTER 3 TEST

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