

## *Naming Worksheet 6 Binary Covalent Compounds Answer Key*

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### **Naming Worksheet 6 Binary Covalent**

Naming Covalent Compounds Solutions. Write the formulas for the following covalent compounds:

1) antimony tribromide  $\text{SbBr}_3$ . 2) hexaboron silicide  $\text{B}_6\text{Si}$  3) chlorine dioxide  $\text{ClO}_2$ . 4) hydrogen iodide  $\text{HI}$  5) iodine pentafluoride  $\text{IF}_5$ . 6) dinitrogen trioxide  $\text{N}_2\text{O}_3$ . 7) ammonia  $\text{NH}_3$ .

### **Covalent Compound Naming Worksheet**

Naming Covalent Compounds. Showing top 8 worksheets in the category - Naming Covalent Compounds. Some of the worksheets displayed are Covalent compound naming work, Binary covalent ionic only, Naming covalent compounds work, Chemistry naming covalent compounds work name write, How to name covalent compounds, Naming covalent compounds work, Naming ions and chemical compounds, Covalent.

### **Naming Covalent Compounds - Printable Worksheets**

Naming Covalent Compounds - Key. Write the formulas for the following covalent compounds: 1) Nitrogen tribromide  $\text{NBr}_3$ . 2) hexaboron silicide  $\text{B}_6\text{Si}$ . 3) chlorine dioxide  $\text{ClO}_2$ .

### **Naming Covalent Compounds Worksheet**

Naming Covalent Compounds - Key. Write the formulas for the following covalent compounds: 1) Nitrogen tribromide  $\text{NBr}_3$ . 2) hexaboron silicide  $\text{B}_6\text{Si}$ . 3) chlorine dioxide  $\text{ClO}_2$ . 4) hydrogen iodide  $\text{HI}$ . 5) iodine pentafluoride  $\text{IF}_5$ . 6) dinitrogen trioxide  $\text{N}_2\text{O}_3$ . 7) ammonia (nitrogen trihydride)  $\text{NH}_3$ . phosphorus triiodide  $\text{PI}_3$ . 9) dihydrogen ...

### **Naming Covalent Compounds Worksheet - My Chemistry Class**

Nomenclature Worksheet 6: Binary Covalent Compounds Please complete the following table: Name of Covalent Compound 1. carbon dioxide 2. phosphorus triiodide 3. sulfur dichloride 4. nitrogen trifluoride 5. dioxygen difluoride Formula of Covalent Compound 6.  $\text{N}_2\text{F}_4$  7.  $\text{SCl}_2$  8.  $\text{ClF}_3$  9.  $\text{SiO}_2$  10.  $\text{P}_4\text{O}_{10}$

### **Nomenclature Worksheet 5: Ionic Compounds Summary**

Nomenclature Worksheet 6: Binary Covalent Compounds Please complete the following table: 6. 8, 9. Formula of Covalent Compound 6.  $\text{N}_2\text{F}_4$  8.  $\text{SCl}_2$  9.  $\text{ClF}_3$

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Naming Covalent Compounds Solutions. Write the formulas for the following covalent compounds: 1) antimony tribromide  $\text{SbBr}_3$ . 2) hexanitrogen silicide  $\text{N}_6\text{Si}$ . 3) chlorine dioxide  $\text{ClO}_2$ . 5) iodine pentafluoride  $\text{IF}_5$ . 6) dinitrogen trioxide  $\text{N}_2\text{O}_3$ . 7) ammonia  $\text{NH}_3$ . ... Naming Covalent Compounds Worksheet

### **Naming Covalent Compounds Worksheet**

Binary Molecular Nomenclature and Molar Mass Rules for Binary Molecular Compounds 1. The naming system is for compounds composed of two nonmetallic elements. 2. The first element keeps its name. 3. The first element gets a prefix if it has a subscript in the formula. 4. The second element gets the -ide suffix (ending).

### **Binary Molecular Compound Practice - ScienceGeek.net**

If a binary compound is composed of two nonmetals, it is a covalent/molecular compound. Use the appropriate prefixes to name these simple compounds. Mono- is only used as a prefix on the second element in the formula. The ending for the second element is always removed and replaced with -ide. 1 = mono...

### **Naming Binary Covalent Compounds Flashcards | Quizlet**

Naming Ionic Compounds Practice Worksheet Name the following ionic compounds: 1)  $\text{NH}_4\text{Cl}$  2)  $\text{Fe}(\text{NO}_3)_3$  3) ... Mixed Ionic/Covalent Compound Naming For each of the following questions, determine whether the compound is ionic or covalent and name it appropriately. 1)  $\text{Na}_2\text{CO}_3$

### **Naming Ionic Compounds Practice Worksheet**

The electrons act as glue to keep the atoms together. Binary Compounds with Hydrogen When naming binary (two elements, here H + 1 other element) compounds containing hydrogen and another nonmetal, name the hydrogen first, and then name the nonmetal as if it were an anion. In other words, change its ending to "ide".

### **Answers to Worksheet 6 - University of Florida**

Yes No. Binary Covalent (Molecular) Compound: 1) Write the symbol of the first element in the compound's name, then the symbol of the second element in the compound's name. 2) Indicate how many atoms of each element the molecule contains using subscripts after the atomic symbol.

### **Binary Covalent Ionic only - Saddleback College**

Covalent Nomenclature Worksheet Author: Brooklyn City Board of Education Last modified by: Christopher Roegiers Created Date: 8/26/2008 2:17:00 AM Company: Brooklyn City Board of Education Other titles: Covalent Nomenclature Worksheet

### **Covalent Nomenclature Worksheet - Brooklyn High School**

Naming Ionic Compounds. Showing top 8 worksheets in the category - Naming Ionic Compounds. Some of the worksheets displayed are Naming ionic compounds work i, Binary covalent ionic only, Naming ionic compounds work 1, Naming ionic compounds practice work, Naming ionic compounds name chem work 8 2, Naming ions and chemical compounds, Ionic compound work, Covalent compound naming work.

### **Naming Ionic Compounds Worksheets - Printable Worksheets**

Start studying Binary Covalent Compounds (Worksheet 6). Learn vocabulary, terms, and more with flashcards, games, and other study tools.

### **Binary Covalent Compounds (Worksheet 6) Flashcards | Quizlet**

Nomenclature of Ionic and Covalent Compounds. Binary Ionic Compounds Containing a Metal and a Nonmetal. A binary compound is a compound formed from two different elements. There may or may not be more than one of each element. A diatomic compound (or diatomic molecule) contains two atoms, which may or may not be the same.

### **Formulas and Nomenclature of Ionic and Covalent Compounds**

As with ionic compounds, the system for naming covalent compounds enables chemists to write the molecular formula from the name and vice versa. This and the following section describe the rules for naming simple covalent compounds, beginning with inorganic compounds and then turning to simple organic compounds that contain only carbon and hydrogen.

### **3.6: Naming Covalent Compounds - Chemistry LibreTexts**

Take a look at the following examples to see how to use the prefixes when naming binary covalent compounds (the prefixes appear in bold). Note that chemists try to avoid putting an a and an o together with the oxide name, as in decaoxide, so they normally drop the a off the prefix. carbon dioxide:

### **How to Name Binary Covalent Compounds - dummies**

Mixed Naming Worksheet — Ionic, Covalent and Acids Directions: Be sure to determine what type of compound your substance is — ionic, binary covalent or an acid. Naming Compounds 1. N<sub>2</sub>O 2. NaBr 3. SiO<sub>2</sub> 4. CO<sub>2</sub> 5. C<sub>2</sub>H<sub>4</sub> 6. P<sub>2</sub>Br<sub>4</sub> 7. HNO<sub>3</sub> 8. Li<sub>2</sub>S 9. FeSO<sub>4</sub> 10. MgBr<sub>2</sub> 11. SeF<sub>2</sub> 12. AsCl<sub>3</sub> 13. AsCl<sub>3</sub>

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Name \_\_\_\_\_ Date \_\_\_\_\_ Period \_\_\_\_\_ Have you ever heard of sodium nitrate? It's a preservative used in foods like hot dogs. The chemical formula is NaNO<sub>3</sub>. How many types of atoms does this compound contain? There are three; sodium, nitrogen, and oxygen. How many atoms total are there in this compound?

There are five total. These five atoms act as one unit.

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