

## *Naming Acids Chemistry If8766 Answers*

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The name of  $\text{HClO}_3$  is chloric acid. 4)  $\text{HClO}_4$  has the  $\text{ClO}_4^-$  polyatomic ion and its name is perchlorate. Since the "ate" suffix is used, it gets changed to "ic.". The name of  $\text{HClO}_4$  is perchloric acid. In the ChemTeam's estimation there are two keys. You have to: recognize when a polyatomic is present and. know its name.

### **ChemTeam: Nomenclature - Acids**

Department of Chemistry and Physics: Answers: Acids Write the formula for each of the acids listed below: Nitric acid  $\text{HNO}_3$ : Hydrocyanic acid  $\text{HCN}$  (aq) Chloric acid  $\text{HClO}_3$ : Acetic acid  $\text{CH}_3\text{COOH}$ : Hydrobromic acid  $\text{HBr}$  (aq) ... Name each of the following acids:  $\text{HClO}_4$  perchloric acid:  $\text{HCOOH}$  formic acid:  $\text{H}_3\text{PO}_4$  phosphoric acid:  $\text{HCl}$  (aq) ...

### **Answers - Naming Acids - Arkansas State University**

Arkansas State University Department of Chemistry and Physics. Write the formula for each of the acids listed below: Nitric acid Hydrocyanic acid Chloric acid Acetic acid Hydrobromic acid Sulfurous acid Chlorous acid Boric acid Hydrochloric acid Phosphoric acid Nitrous acid Hydrofluoric acid Perchloric acid Hydroiodic acid Phosphorous acid Carbonic acid Sulfuric acid Formic acid.

### **Worksheet - Naming Acids**

Acids-For simplicity, the acids that we will be concerned with naming are really just a special class of ionic compounds where the cation is always  $\text{H}^+$ . So if the formula has hydrogen written first, then this usually indicates that the hydrogen is an  $\text{H}^+$  cation and that the compound is an acid.

### Naming Acids - AP Chemistry

Naming Ionic Compounds Directions: Name the following compounds using the Stock Naming System. 1.  $\text{CaCO}_3$  calcium carbonate\_\_\_\_\_ 2.  $\text{KCl}$  ... Chemistry IF8766 Page 45 Instructional Fair, Inc. Title: Name: Author:

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