# Molar Mass Answer Key

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# Molar Mass Worksheet - nclark.net

Molar Mass Practice Worksheet . Find the molar masses of the following compounds: 1) NaBr Na 1 (22.989770)

#### **Molar Mass Practice Worksheet - Gravity Simulator**

Directions: Find the formula mass of the following compounds. Round atomic masses to the tenth of a decimal place. Place your final answer in the FORMULA MASS COLUMN. The molar ratio is an important comcept in solving stoichiometry problems. The sources for these ratios are the coefficients of a balanced equation.

#### **CHEMISTRY COMPUTING FORMULA MASS WORKSHEET**

The molar mass of a substance is the mass of one mole of the substance. This collection of ten chemistry test questions deals with calculating and using molar masses. The answers appear after

the final question. A periodic table is necessary to complete the questions. Tetra Images/Getty Images. Calculate the molar mass of CuSO4.

#### Molar Mass - Chemistry Test Questions - ThoughtCo

Relative Mass and the Mole answer key Created Date: 20171005134609Z ...

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Molar Mass Worksheet Key. Water (H. 20) Element Number MM H 2 1.01 g/mole O 1 16.00 g/mole Total MM = 18.02 g/mole. carbon dioxide, 2CO. 2 Element Number MM O 2 16.00 g/mole C 1 12.01 g/mole Total MM = 44.01 g/mole. sodium chloride, NaCl. Element Number MM Cl 1 35.45 g/mole Na 1 22.99 g/mole Total MM = 58.44 g/mole.

#### ws molar mass - Saddleback College

View Homework Help - molar volume:molar mass #1 answer key.doc from ORGO 205 at Johns Hopkins University. Name: Period: Date: Worksheet: Molar Volume/Molar Mass #1 1. How many moles of molecules are

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1 mole = molar mass (could be atomic mass from periodic table or molecular mass) 1 mole = 22.4 L of a gas at STP ... Mole Calculation Worksheet - Answer Key What are the molecular weights of the following compounds? 1) NaOH 22.99 + 16.00 + 1.01 = 40.00 grams/mol 2) H 3 PO 4

# Mole Calculation Worksheet - sheffieldschools.org

MOLAR MASS Determine the molar mass (the mass of one mole) of each compound below. Remember to include units (either amu or g/mol). Hint: for hydrated compounds such as CuSO  $4 \cdot 5H$  2O, add the molar mass of the salt (CuSO 4) to the mass of the number of water molecules indicated by the coefficient (5H 2O = 5 water molecules).

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