

Modern Chemistry Reaction Energy Review Answers

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Modern Chemistry 137 Reaction Kinetics CHAPTER 17 REVIEW Reaction Kinetics SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. Refer to the energy diagram below to answer the following questions. ____ a. Which letter represents the energy of the activated complex? (a) A (c) C (b) B (d) D ____ b.

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CHAPTER 17 REVIEW Reaction Kinetics MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. The reaction for the decomposition of hydrogen peroxide is $2\text{H}_2\text{O}_2(\text{aq}) \rightarrow 2\text{H}_2\text{O}(\text{l}) + \text{O}_2(\text{g})$. List three ways to speed up the rate of decomposition. For each one, briefly explain why it is effective, based on collision theory.

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