

## *Modern Chemistry Chapter 7 Homework Answers*

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### Modern Chemistry Chapter 7 Homework

Modern Chemistry • CHAPTER 7 HOMEWORK 7-3 (pp. 211–213) VOCABULARY Write the prefix that corresponds to each number. Try not to use the book. Number Prefix Number Prefix 16 27 38 49 510 GRAPHIC ORGANIZER Complete the chart by supplying the missing information. Name of Compound Formula of Compound CO<sub>2</sub> Nitrogen dioxide S<sub>2</sub> Cl<sub>2</sub> Carbon ...

### HOMEWORK 7-3 - Kettering City School District

Modern Chemistry • CHAPTER 7 HOMEWORK 7-2 (pp. 206–211) VOCABULARY Match the symbol with the name that describes it. 1. Br ... CHAPTER 7 • HOMEWORK 7-1 VOCABULARY 1. false 2. true 3. false 4. false 5. true SKILL BUILDER AgCl—In the compound, there is one atom of silver for each atom of chlorine.

### HOMEWORK 7-2 - Kettering City School District

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Modern Chemistry • CHAPTER 7 HOMEWORK 7-5 VOCABULARY Complete each sentence. 1. Oxidation numbers are assigned to the atoms composing a compound or ion in order to \_\_\_\_\_. 2. The atoms in a pure element have an oxidation number of \_\_\_\_\_. 3. The more electronegative element in a binary molecular compound is assigned the number that ...

### HOMEWORK 7-5 - apcscience.com

Modern Chemistry • CHAPTER 7 HOMEWORK 7-1 (pp. 203–206) VOCABULARY Write true or false for each statement. 1. A chemical formula indicates the relative number of molecules of each kind in a chemical compound. \_\_\_\_ 2. Monoatomic ions are ions formed from a single atom. \_\_\_\_ 3.

### HOMEWORK 7-1 - Los Banos Unified School District

Modern Chemistry • CHAPTER 6 HOMEWORK 6-7 VOCABULARY Write true or false for each statement. 1. A polyatomic ion is a group of covalently bonded atoms. \_\_\_\_ 2. NaCl is a molecular compound. \_\_\_\_ 3. Lattice energy is released when the ions in a crystal lattice are separated from each other. \_\_\_\_ 4. Ionic compounds are electrical conductors ...

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Modern Chemistry • CHAPTER 22 HOMEWORK 22-2 (pp. 702–704) VOCABULARY Write true or false.

1. The band of stability represents the stable nuclei cluster over a range of electron-proton ratios. \_\_\_\_ 2. The stability of a nucleus is greatest when the nucleons are in a 1:1 ratio. \_\_\_\_ 3. 2, 20, 50, and 126 are all magic numbers. \_\_\_\_ 4.

### Modern Chemistry CHAPTER 22 HOMEWORK 22-1

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