

## ***Molecular Formula Worksheet Answer Key***

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### **Molecular Formula Worksheet Answer Key**

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### **Chemical Formula Writing Worksheet Answer Key | Semesprit**

The mass of the empirical formula is 68 g/mole. Therefore, the molecular formula must be TWO times the empirical formula. The mass of the empirical formula is 126.9 g/mole. Therefore, the molecular formula must be TWO times the empirical formula. The mass of the empirical formula is 46 g/mole.

### **Empirical and Molecular Formula Worksheet**

Empirical and Molecular Formula Worksheet ANSWER KEY Write the empirical formula for the following compounds. 1) C<sub>6</sub>H<sub>6</sub> CH<sub>6</sub>) C<sub>8</sub>H<sub>18</sub> C<sub>4</sub>H<sub>9</sub> 7) WO<sub>2</sub> WO<sub>2</sub> 8) C<sub>2</sub>H<sub>6</sub>O<sub>2</sub> CH<sub>3</sub>O<sub>9</sub>) X<sub>39</sub>Y<sub>13</sub>X<sub>3</sub>Y<sub>6</sub>) A compound with an empirical formula of C<sub>2</sub>OH<sub>4</sub> and a molar mass of 88 grams per mole. What is the molecular formula of this compound?

### **Empirical\_Molecular\_Formula\_Worksheet\_answers - Empirical ...**

Molecular Formula Worksheet ANSWER KEY Write the molecular formulas of the following compounds: 1) A compound with an empirical formula of C<sub>2</sub>OH<sub>4</sub> and a molar mass of 88 grams per mole.

### **Percent Composition and Molecular Formula Worksheet**

EMPIRICAL AND MOLECULAR FORMULA WORKSHEET 1. An oxide of chromium is found to have the following % composition: 68.4 % Cr and 31.6 % O. Determine this compound's empirical formula. 2. The percent composition of a compound was found to be 63.5 % silver, 8.2 % nitrogen, and 28.3 % oxygen.

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Molecular Formula Worksheet ANSWER KEY Write the molecular formulas of the following compounds: 1) A compound with an empirical formula of C<sub>2</sub>OH<sub>4</sub> and a molar mass of 88 grams per mole.

### **Percent Composition and Molecular Formula Worksheet**

For chemistry help, visit <http://www.chemfiesta.com> © 2000 Cavalcade Publishing - All Rights Reserved 13. Calculate%the%mass%percent%of%carbon ...

### **Empirical and Molecular Formula Worksheet**

Chemical Formula Writing Worksheet Solutions Write chemical formulas for the compounds in each box. The names are found by finding the intersection between the cations and anions. Example: The first box is the intersection between the "zinc" cation and the "chloride" anion, so you should write "ZnCl<sub>2</sub>", as shown.

### **Chemical Formula Writing Worksheet Two - IMSA**

Printable Chemistry Worksheets - Chemical Names and Formulas. Here are printable chemistry worksheets in pdf format. The first worksheet in each set asks you to name the chemical formula based on the chemical name. The second worksheet asks for the chemical name based on the chemical formula. Answer keys are provided separately.

### Chemical Names and Formulas - Printable Worksheets

Place your final answer in the FORMULA MASS COLUMN. CHEMISTRY COMPUTING FORMULA MASS WORKSHEET Problem Set-up example: Find the formula mass of  $\text{Ca}(\text{NO}_3)_2$  ... A chemical engineer must be able to calculate the amounts of all reactants and products in order to determine if the

### CHEMISTRY COMPUTING FORMULA MASS WORKSHEET

Empirical and Molecular Formulas Worksheet . Objectives: • be able to calculate empirical and molecular formulas . Empirical Formula . 1) What is the empirical formula of a compound that contains 0.783g of Carbon, 0.196g of Hydrogen and 0.521g of Oxygen? 2) What is empirical formula of a compound which consists of 89.14% Au and 10.80% of O?

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