Physics

SECTION - A

- A circular loop of radius r is carrying current I A. The ratio of magnetic field at the center of circular loop and at a distance r from the center of the loop on its axis is:
 - (1) $2\sqrt{2}$: 1
- (2) 1: $3\sqrt{2}$
- (3) $1:\sqrt{2}$

(4) $3\sqrt{2}$: 2

Sol. 1

Magnetic field at centre of coil $B_1 = \frac{\mu_0 I}{2r}$

on the axis at $x = r \Rightarrow B_2 = \frac{\mu_0 I r^2}{2(r^2 + x^2)^{\frac{3}{2}}}$

$$B_{2} = \frac{\mu_{0} I r^{2}}{2 \left(r^{2} + r^{2}\right)^{3/2}}$$

$$B_2 = \frac{\mu_0 I}{2(2\sqrt{2} r)}$$

$$\frac{B_1}{B_2} = 2\sqrt{2}$$

- The weight of a body at the surface of earth is 18 N. The weight of the body at an altitude of 3200 km above the earth's surface is (given, radius of earth $R_e = 6400$ km):
 - (1) 8 N
- (2) 4.9 N
- (3) 9.8 N

(4) 19.6 N

Sol. 1

Weight on earth surface W = mg = 18 N

Above earth surface $\Rightarrow W_2 = m \frac{GM}{(R+h)^2}$

h = 3200 km = R/2

$$W_2 = m \frac{GM}{\left(\frac{3R}{2}\right)^2} \Rightarrow W_2 = \frac{4}{9} mg$$

$$W_2 = \frac{4}{9} \times 18 \Longrightarrow W_2 = 8 N$$

- Two long straight wires P and Q carrying equal current 10 A each were kept parallel to each other at 5 cm distance. Magnitude of magnetic force experienced by 10 cm length of wire P is F_1 If distance between wires is halved and currents on them are doubled, force F_2 on 10 cm length of wire P will be:
 - $(1)\frac{F_1}{8}$
- (2) 8 F₁
- (3) 10 F₁

 $(4)\frac{F_1}{10}$

Sol.

$$F = \frac{\mu_0 I_1 I_2}{2\pi r} \Rightarrow F = \frac{\mu_0 I^2 \ell}{2\pi r}$$

$$\ell = 10 \text{ cm (Both)} \Rightarrow F \propto \frac{I^2}{r}$$

$$\frac{F_1}{F_2} = \left(\frac{I}{2I}\right)^2 \left(\frac{\frac{5}{2}}{5}\right) \Rightarrow \frac{F_1}{F_2} = \frac{1}{8} \Rightarrow F_2 = 8F_1$$

4. Given below are two statements :

Statement I : The temperature of a gas is -73° C. When the gas is heated to 527°C, the root mean square speed of the molecules is doubled.

Statement II: The product of pressure and volume of an ideal gas will be equal to translational kinetic energy of the molecules.

In the light of the above statements, choose the correct answer from the options given below:

- (1) Statement I is false but Statement II is true
- (2) Both Statement I and Statement II are false
- (3) Statement I is true but Statement II is false
- (4) Both Statement I and Statement II are true

Sol. 3

Statements-1

$$v_{rms} \propto \sqrt{T} \Rightarrow v_{rms_1} \propto \sqrt{273 - 73}$$

$$v_{rms_2} \propto \sqrt{273 + 527}$$

$$\frac{v_{rms_1}}{v_{rms_2}} = \sqrt{\frac{200}{800}} \Rightarrow v_{rms_2} = 2v_{rms_1}$$
(True)

Statements-2

Translation K.E. =
$$\frac{3}{2}$$
nRT = $\frac{3}{2}$ PV (False)

5. The maximum vertical height to which a man can throw a ball is 136 m. The maximum horizontal distance upto which he can throw the same ball is:

- (1) 272 m
- (2) 68 m
- (3) 192 m

(4) 136 m

Sol. 1

Max vertical height
$$H = \frac{v^2}{2g} = 136 \,\text{m}$$

Max horizontal distance
$$R = \frac{v^2}{g} \Rightarrow R = 2 \times 136 = 272 \text{ m}$$

6. Given below are two statements :

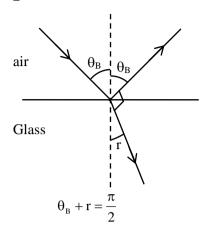
Statement I : If the Brewster's angle for the light propagating from air to glass is θ_B , then the Brewster's angle for the light propagating from glass to air is $\frac{\pi}{2} - \theta_B$

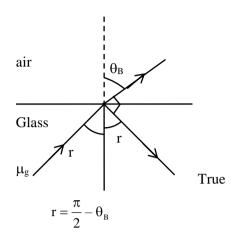
Statement II: The Brewster's angle for the light propagating from glass to air is $tan^{-1}(\mu_g)$ where μ_g is the refractive index of glass.

In the light of the above statements, choose the correct answer from the options given below:

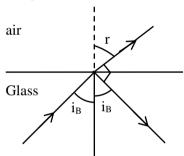
- (1) Both Statement I and Statement II are false
- (2) Statement I is true but Statement II is false
- (3) Statement I is false but Statement II is true
- (4) Both Statement I and Statement II are true

Sol. 2





For glass to air



$$\mu_g \sin i_B = 1 \cdot \sin r$$

$$r + i_B = \frac{\pi}{2}$$

$$\mu_{\rm g} \sin i_{\rm B} = \cos i_{\rm B} \Rightarrow \tan i_{\rm B} = \frac{1}{\mu_{\rm g}} \Rightarrow i_{\rm B} = \tan^{-1} \left(\frac{1}{\mu_{\rm g}} \right)$$

7. A 100 m long wire having cross-sectional area 6.25×10^{-4} m² and Young's modulus is 10^{10} Nm⁻² is subjected to a load of 250 N, then the elongation in the wire will be:

$$(1) 4 \times 10^{-3} \text{ m}$$

(2)
$$6.25 \times 10^{-3}$$
 m

$$(3) 6.25 \times 10^{-6} \text{ m}$$

$$(4) 4 \times 10^{-4} \text{ m}$$

Sol. 1

$$Stress = y \; strain \Rightarrow \frac{W}{A} = y \frac{\Delta \ell}{\ell}$$

$$\Delta \ell = \frac{W\ell}{yA} \Longrightarrow \Delta \ell = \frac{250 \times 100}{10^{10} \times 6.25 \times 10^{-4}}$$

$$\Delta \ell = 4 \times 10^{-3} \text{ m}$$

8. If two charges q_1 and q_2 are separated with distance 'd' and placed in a medium of dielectric constant K. What will be the equivalent distance between charges in air for the same electrostatic force?

(1)
$$2d\sqrt{k}$$

(2) 1.5 d
$$\sqrt{k}$$

(3)
$$d\sqrt{k}$$

For same force

$$\frac{q_1q_2}{4\pi\epsilon_0kd^2} = \frac{q_1q_2}{4\pi\epsilon_0r^2} \Longrightarrow r = d\sqrt{K}$$

9. Consider the following radioactive decay process

$${}^{218}_{84}A \xrightarrow{\alpha} A_1 \xrightarrow{\beta^-} A_2 \xrightarrow{\gamma} A_3 \xrightarrow{\alpha} A_4 \xrightarrow{\beta^+} A_5 \xrightarrow{\gamma} A_6$$

The mass number and the atomic number of A_6 are given by:

- (1) 210 and 84
- (2) 210 and 82
- (3) 211 and 80
- (4) 210 and 80

Sol.

$${}^{218}_{84}A \xrightarrow{\quad \alpha \quad} {}^{214}_{82}A_1 \xrightarrow{\quad \beta^\Theta \quad} {}^{214}_{83}A_2 \xrightarrow{\quad \gamma \quad} {}^{214}_{83}A_3 \xrightarrow{\quad \alpha \quad} {}^{210}_{81}A_4 \xrightarrow{\quad \beta^\Theta \quad} {}^{210}_{80}A_5 \xrightarrow{\quad \gamma \quad} {}^{210}_{80}A_6$$

10. From the photoelectric effect experiment, following observations are made. Identify which of these are correct.

A. The stopping potential depends only on the work function of the metal.

B. The saturation current increases as the intensity of incident light increases.

C. The maximum kinetic energy of a photo electron depends on the intensity of the incident light.

D. Photoelectric effect can be explained using wave theory of light.

Choose the correct answer from the options given below:

- (1) A, C, D only
- (2) B, C only
- (3) B only
- (4) A, B, D only

Sol. 3

$$v_{sp} = \frac{hv - \phi}{e}$$
 (v and ϕ both)

Intensity ↑ current ↑

$$kE_{max} = h\nu - \phi$$

Photoelectric effect is not explained by wave theory

11. Given below are two statements:

Statement I: An elevator can go up or down with uniform speed when its weight is balanced with the tension of its cable.

Statement II: Force exerted by the floor of an elevator on the foot of a person standing on it is more than his/her weight when the elevator goes down with increasing speed.

In the light of the above statements, choose the correct answer from the options given below:

- (1) Both Statement I and Statement II are true
- (2) Statement I is false but Statement II is true
- (3) Statement I is true but Statement II is false
- (4) Both Statement I and Statement II are false

Sol. 3

Statement-1

When force balance it can move with uniform velocity (Uniform speed) True

Statement-2

Elevator going down with increasing speed means its acceleration is downwards

mg - N = ma (on person)

N = mg - ma (False)

- 1 g of a liquid is converted to vapour at 3×10^5 Pa pressure. If 10% of the heat supplied is used for increasing the volume by 1600 cm³ during this phase change, then the increase in internal energy in the process will be:
 - (1) 432000 J
- (2) 4320 J
- (3) 4800 J
- $(4) 4.32 \times 10^8 \text{ J}$

Sol. 2

10% of $\Delta Q = P\Delta V$ (W/D by gas)

$$\frac{\Delta Q}{10} = 3 \times 10^5 (1600 \times 10^{-6})$$

$$\Delta Q = 4800 \text{ J}$$

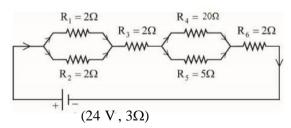
Using first law of the thermodynamics

$$\Delta Q = \Delta u + W$$

$$\Delta Q = \Delta u + \frac{\Delta Q}{10} \Longrightarrow \Delta u = \frac{9}{10} \Delta Q$$

$$\Delta u = \frac{9}{10} \times 4800 \Rightarrow \Delta u = 4320 J$$

As shown in the figure, a network of resistors is connected to a battery of 24 V with an internal resistance of 3Ω. The currents through the resistors R_4 and R_5 are I_4 and I_5 respectively. The values of I_4 and I_5 are:

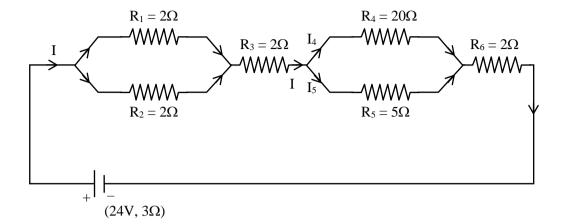


(1)
$$I_4 = \frac{2}{5} A$$
 and $I_5 = \frac{8}{5} A$

(2)
$$I_4 = \frac{24}{5}$$
 A and $I_5 = \frac{6}{5}$ A

(3)
$$I_4 = \frac{8}{5}$$
 A and $I_5 = \frac{2}{5}$ A

(4)
$$I_4 = \frac{6}{5}$$
 A and $I_5 = \frac{24}{5}$ A



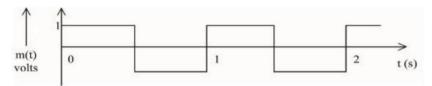
$$R_{eq} = 3 + 1 + 2 + \frac{20 \times 5}{25} + 2 \Rightarrow R_{eq} = 12\Omega$$

Current from battery $I = \frac{24}{12} \Rightarrow I = 2A$

$$I_4 + I_5 = 2A$$

$$I_4(20) = I_5(5) \Rightarrow I_5 = 4I_4 \Rightarrow I_4 = \frac{2}{5}A I_5 = \frac{8}{5}A$$

14. A modulating signal is a square wave, as shown in the figure.



If the carrier wave is given as $c(t) = 2\sin(8\pi t)$ volts, the modulation index is:

$$(1)^{\frac{1}{4}}$$

$$(2)^{\frac{1}{2}}$$

$$(4)\frac{1}{3}$$

Sol. 2

 $Modulation \ index \ \mu = \ \frac{A_{\scriptscriptstyle m}}{A_{\scriptscriptstyle c}}$

$$A_m = 1 \& A_c = 2$$

$$\mu = \frac{1}{2}$$

A conducting circular loop of radius $\frac{10}{\sqrt{\pi}}$ cm is placed perpendicular to a uniform magnetic field of 0.5 T. The magnetic field is decreased to zero in 0.5 s at a steady rate. The induced emf in the circular loop at 0.25 s is:

(1)
$$emf = 1mV$$

(2)
$$emf = 5mV$$

(3)
$$emf = 100 mV$$

(4)
$$emf = 10mV$$

$$emf = -\frac{d\phi}{dt} \Longrightarrow \epsilon = \frac{-d(BA)}{dt}$$

$$\epsilon = -A \frac{dB}{dt} \Longrightarrow \epsilon = -\pi R^2 \Biggl(\frac{0-B}{\Delta t} \Biggr)$$

$$\epsilon = \frac{\pi R^2 B}{\Delta t} \Rightarrow \epsilon = \frac{\pi \left(\frac{10}{\sqrt{\pi}} \times 10^{-2}\right)^2 \times 0.5}{0.5}$$

$$\varepsilon = 10^{-2} \text{ volt} = 10 \text{ m volt}$$

16. In \vec{E} and \vec{K} represent electric field and propagation vectors of the EM waves in vacuum, then magnetic field vector is given by :

(ω - angular frequency):

(1)
$$\omega(\bar{E} \times \bar{K})$$

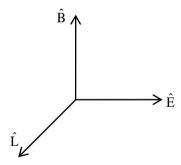
(2)
$$\omega(\bar{K} \times \bar{E})$$

(3)
$$\bar{K} \times \bar{E}$$

(4)
$$\frac{1}{\omega} (\bar{K} \times \bar{E})$$

Sol.

$$\vec{E} \& \vec{K} = \frac{W}{C}\hat{L}$$



$$\hat{B}=\hat{L}\!\times\!\hat{E}$$

$$\vec{B} = B\hat{B} \left\{ \frac{E}{B} = C \right\}$$

$$\vec{\mathbf{B}} = \frac{\mathbf{E}}{\mathbf{C}} (\hat{\mathbf{L}} \times \hat{\mathbf{E}})$$

$$\vec{B} = \frac{\omega}{C} \left(\frac{\hat{L} \times E\hat{E}}{\omega} \right) \Rightarrow \vec{B} = \frac{\vec{K} \times \vec{E}}{\omega}$$

Match List I with List II: 17.

| | LIST I | | LIST II | |
|----|-------------------------|------|--|--|
| A. | Planck's constant (h) | I. | $[M^1 L^2 T^{-2}]$ | |
| B. | Stopping potential (Vs) | II. | $[M^1 L^1 T^{-1}]$ | |
| C. | Work function (Ø) | III. | $[M^1 L^2 T^{-1}]$ | |
| D. | Momentum (p) | IV. | [M ¹ L ² T ⁻³ A ⁻¹] | |

Choose the correct answer from the options given below:

(1) A-I, B-III, C-IV, D-II

(2) A-III, B-I, C-II, D-IV

(3) A-II, B-IV, C-III, D-I

(4) A-III, B-IV, C-I, D-II

Sol. 4

(A) Planck's constant $h = \frac{E}{V}$

$$[h] = \frac{\left[M^{1}L^{2}T^{-2}\right]}{\left[T^{-1}\right]} \Rightarrow [h] = \left[M^{1}L^{2}T^{-1}\right]$$

(B) Stopping potential $V = \frac{W}{q}$

$$[v] = \frac{ML^2T^{-2}}{AT} \Rightarrow [v] = [ML^2T^{-3}A^{-1}]$$

- (C) Work function = $[ML^2T^{-2}]$
- (D) Momentum $[P] = [MLT^{-1}]$
- 18. A travelling wave is described by the equation

$$y(x,t) = [0.05\sin(8x - 4t)]m$$

The velocity of the wave is: [all the quantities are in SI unit]

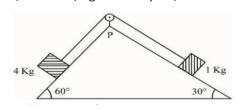
- $(1) 8 \text{ ms}^{-1}$
- $(2) 4 \text{ ms}^{-1}$
- $(3) 0.5 \text{ ms}^{-1}$
- $(4) 2 \text{ ms}^{-1}$

Sol. 3

$$y = 0.05 \sin(8x - 4t)$$

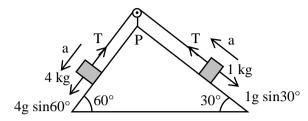
$$v = \frac{\omega}{k} \Rightarrow v = \frac{4}{8} \Rightarrow v = \frac{1}{2} \text{ m/s}$$

As per given figure, a weightless pulley P is attached on a double inclined frictionless surfaces. The 19. tension in the string (massless) will be (if $g = 10 \text{ m/s}^2$)



- (1) $(4\sqrt{3} + 1)N$ (2) $4(\sqrt{3} + 1)N$
- (3) $(4\sqrt{3}-1)N$
- $(4) 4(\sqrt{3}-1)N$

Sol. 2



$$4g\frac{\sqrt{3}}{2} - T = 4a$$
 ... (1)

$$T - \frac{g}{2} = 1a$$
 ... (2)

$$2\sqrt{3}g - T = 4\left(T - \frac{g}{2}\right) \Rightarrow 5T = (2\sqrt{3} + 2)g$$

$$T = \frac{10}{5} \left(2\sqrt{3} + 2 \right) \Longrightarrow T = 4 \left(\sqrt{3} + 1 \right) N$$

20. Given below are two statements: one is labelled as Assertion **A** and the other is labelled as Reason **R** Assertion **A**: Photodiodes are preferably operated in reverse bias condition for light intensity measurement.

Reason : The current in the forward bias is more than the current in the reverse bias for a p-n junction diode.

In the light of the above statements, choose the correct answer from the options given below:

- (1) A is true but R is false
- (2) A is false but R is true
- (3) Both A and R are true and R is the correct explanation of A
- (4) Both A and R are true but R is NOT the correct explanation of A

Sol. 4

Photodiode works in reverse bias and its is used as a intensity detector . (True)

Forward bias current is more as compaired to reverse bias current (True)

- **21.** Vectors $a\hat{i} + b\hat{j} + \hat{k}$ and $2\hat{i} 3\hat{j} + 4\hat{k}$ are perpendicular to each other when 3a + 2b = 7, the ratio of a to b is $\frac{x}{2}$. The value of x is
- Sol.

$$a\hat{i} + b\hat{j} + \hat{k}$$
 is \perp to $(2\hat{i} - 3\hat{j} + 4\hat{k})$

$$\vec{A} \cdot \vec{B} = 0 \Rightarrow 2a - 3b - 4 = 0$$

 $2a - 3b = -4$

Given
$$3a + 2b = 7$$

$$\frac{2\left(\frac{a}{b}\right) - 3}{3\left(\frac{a}{b}\right) + 2} = \frac{-4}{7} \Rightarrow 14\frac{a}{b} - 21 = -12\frac{a}{b} - 8$$

$$26\frac{a}{b} = 13 \Rightarrow \frac{a}{b} = \frac{1}{2} = \frac{x}{2}$$

$$x = 1$$

- Assume that protons and neutrons have equal masses. Mass of a nucleon is 1.6×10^{-27} kg and radius of nucleus is 1.5×10^{-15} A^{1/3} m. The approximate ratio of the nuclear density and water density is $n \times 10^{13}$. The value of n is
- Sol. 1

$$\rho_{\text{Nucleus}} = \frac{A(m)}{\frac{4}{3}\pi R^3} \Longrightarrow$$

$$\rho_{\rm N} = \frac{3}{4\pi} \frac{Am}{\left(1.5 \times 10^{-15} \; A^{\frac{1}{3}}\right)^3}$$

$$\frac{\rho_{\rm N}}{\rho_{\rm W}} = \frac{3}{4\pi} \frac{(1.6) \times 10^{-27}}{(1.5)^3 \times 10^{-45} \times 10^3}$$

$$\frac{\rho_{\rm N}}{\rho_{\rm w}} = 11 \times 10^{13}$$

- 23. A hollow cylindrical conductor has length of 3.14 m, while its inner and outer diameters are 4 mm and 8 mm respectively. The resistance of the conductor is $n \times 10^{-3}\Omega$. If the resistivity of the material is $2.4 \times 10^{-8}\Omega$ m. The value of n is
- Sol. 2

$$R = \frac{\rho \ell}{A} \Rightarrow R = \frac{\rho \ell}{\pi \left(r_2^2 - r_1^2\right)}$$

$$R = \frac{2.4 \times 10^{-8} \times 3.14}{\pi (4^2 - 2^2) \times 10^{-6}}$$

$$R = 2 \times 10^{-3} \,\Omega$$

- A stream of a positively charged particles having $\frac{q}{m} = 2 \times 10^{11} \frac{C}{kg}$ and velocity $\vec{v}_0 = 3 \times 10^7 \text{îm/s}$ is deflected by an electric field $1.8\hat{j}$ kV/m. The electric field exists in a region of 10 cm along x direction. Due to the electric field, the deflection of the charge particles in the y direction is _____ mm
- Sol. 2

$$y = \frac{1}{2}at^2$$

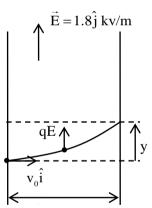
$$y = \frac{1}{2} \frac{qE}{m} t^2$$

$$\ell = v_0 t$$

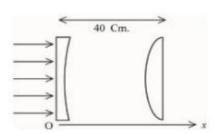
$$y = \frac{1}{2} \frac{qE}{m} \left(\frac{\ell}{v_0} \right)^2$$

$$y = \frac{1}{2} (2 \times 10^{11}) (1.8 \times 10^3) \left(\frac{0.1}{3 \times 10^7} \right)^2$$

$$y = 2 \text{ mm}$$

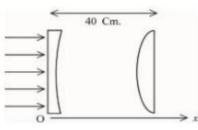


25. As shown in the figure, a combination of a thin plano concave lens and a thin plano convex lens is used to image an object placed at infinity. The radius of curvature of both the lenses is 30 cm and refraction index of the material for both the lenses is 1.75. Both the lenses are placed at distance of 40 cm from each other. Due to the combination, the image of the object is formed at distance = ___cm, from concave lens.



Magnitude of focal length of both lens

$$f = \frac{R}{\mu - 1} \Rightarrow f = \frac{30}{1.75 - 1} \Rightarrow f = 40 \text{ cm}$$



$$f = -40 \text{ cm}$$
 $f = +40 \text{ cm}$

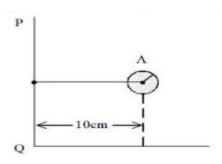
Concave lens will form image at its focus for convex lens $\frac{1}{v} - \frac{1}{u} = \frac{1}{f} \Rightarrow \frac{1}{v} - \frac{1}{-80} = \frac{1}{+40}$

V = +80 cm

From concave lens distance of image of d = 80 + 40

d = 120 cm

26. Solid sphere A is rotating about an axis PQ. If the radius of the sphere is 5 cm then its radius of gyration about PQ will be \sqrt{x} cm. The value of x is ______



Sol. 110

$$I_{PO} = I_{cm} + md^2$$

$$mk^{2} = \frac{2}{5}mR^{2} + md^{2} \Rightarrow k = \sqrt{\frac{2}{5}(5)^{2} + (10)^{2}}$$

$$k = \sqrt{110}$$
 cm

A block of a mass 2 kg is attached with two identical springs of spring constant 20 N/m each. The block is placed on a frictionless surface and the ends of the springs are attached to rigid supports (see figure). When the mass is displaced from its equilibrium position, it executes a simple harmonic motion. The time period of oscillation is $\frac{\pi}{\sqrt{x}}$ in SI unit. The value of x is ______

Sol.

$$T = 2\pi \sqrt{\frac{m}{k_{eq}}}$$

$$T = 2\pi \sqrt{\frac{2}{2k}} \Rightarrow T = 2\pi \sqrt{\frac{1}{20}}$$

$$T = \frac{\pi}{\sqrt{5}}$$

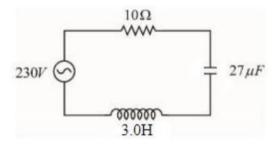
- **28.** A hole is drilled in a metal sheet. At 27°C, the diameter of hole is 5 cm. When the sheet is heated to 177°C, the change in the diameter of hole is $d \times 10^{-3}$ cm. The value of d will be _____ if coefficient of linear expansion of the metal is 1.6×10^{-5} /°C.
- **Sol.** 12

$$\Delta D = D \propto \Delta T$$

$$\Delta D = 5 \times 1.6 \times 10^{-5} \times (177 - 27)$$

$$\Delta D = 12 \times 10^{-3} \text{ cm}$$

29. In the circuit shown in the figure, the ratio of the quality factor and the band width is ______ S.



Sol. 10

$$Q = \frac{1}{R} \sqrt{\frac{L}{C}}$$
 & bandwidth = $\frac{R}{L}$

$$\frac{Q}{Bandwidth} = \frac{L}{R^2} \sqrt{\frac{L}{C}}$$

$$= \frac{3}{100} \times \sqrt{\frac{3}{27 \times 10^{-6}}}$$

- **30.** A spherical body of mass 2 kg starting from rest acquires a kinetic energy of 10000 J at the end of 5th second. The force acted on the body is _______N.
- **Sol.** 40

Impulse =
$$\Delta P$$

$$F\Delta T = P - O \Rightarrow F\Delta T = \sqrt{2mk}$$

$$F(5) = \sqrt{2 \times 2 \times 10000}$$

$$F = 40 \text{ N}$$

Chemistry

SECTION - A

31. 'A' and 'B' formed in the following set of reactions are:

$$\begin{array}{c|c}
OH & \text{HBr} \\
\hline
\Delta & A
\end{array}$$

Sol. 2

- 32. Decreasing order of the hydrogen bonding in following forms of water is correctly represented by A.Liquid water
 - B. Ice
 - C. Impure water

Choose the correct answer from the options given below:

- (1) B > A > C
- (2) A > B > C
- (3) A = B > C
- (4) C > B > A

Sol.

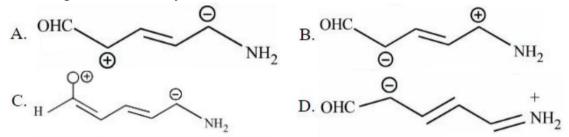
ice > liquid water > impure water

OR

ice $> H_2O$ liq. > impure H_2O

Hydrogen bond $\propto \frac{1}{\text{Temp}}$

33. Increasing order of stability of the resonance structures is:



Choose the correct answer from the options given below:

(1) D, C, A, B

(2) D, C, B, A

(3) C, D, A, B

(4) 4. C, D, B, A

Sol. Bonus

$$\begin{array}{c} O^+\\ \\ H \\ \hline \\ (C) \\ \\ (+\text{ve on O}) \end{array} \leftarrow \begin{array}{c} OHC \\ \\ \hline \\ (B) \\ \\ (Back \ bonding) \end{array} \leftarrow \begin{array}{c} OHC \\ \\ \hline \\ NH_2 \end{array} \leftarrow \begin{array}{c} OHC \\ \\ \hline \\ (A) \end{array}$$

Final correct order C<B<A<D

34. 'R' formed in the following sequence of reactions is:

$$(1) \underbrace{\begin{array}{c} NaCN \\ NaCN \\ HOAc \end{array}} \stackrel{`P'}{\longrightarrow} \underbrace{\begin{array}{c} EtOH \\ H^+ \end{array}} \stackrel{`Q'}{\longrightarrow} \underbrace{\begin{array}{c} (i) \ 2MeMgBr \\ (ii) \ H_3O^+ \end{array}} \stackrel{`R'}{\longrightarrow} \underbrace{\begin{array}{c} R' \\ major \ product \end{array}}$$

$$(2) \underbrace{\begin{array}{c} OH \ Me \\ CI \end{array}} \stackrel{Me}{\longrightarrow} \underbrace{\begin{array}{c} OH \ Me \\ OH \ Me \end{array}}$$

$$(3) \underbrace{\begin{array}{c} OH \ Me \ Me \end{array}} \stackrel{(4) \underbrace{\begin{array}{c} OH \ Me \ Me \ Me \end{array}} \stackrel{(4) \underbrace{\begin{array}{c} OH \ Me \ Me \ Me \ Me \end{array}}$$

$$CI \xrightarrow{O} CH_3 \xrightarrow{NaCN} CI \xrightarrow{O^-} CH_3 \xrightarrow{COOH} CH_3 \xrightarrow{COOH$$

- The primary and secondary valencies of cobalt respectively in [Co(NH₃)₅Cl^{Cl}Cl₂ are: 35.
 - (1) 3 and 6
- (2) 2 and 6
- (3) 3 and 5
- (4) 2 and 8

- Sol. 1 [CO(NH₃)₅Cl]Cl₂
- 36. An ammoniacal metal salt solution gives a brilliant red precipitate on addition of dimethylglyoxime. The metal ion is:
 - $(1) \text{ Co}^{2+}$
- $(2) Ni^{2+}$
- $(3) \text{ Fe}^{2+}$
- $(4) Cu^{2+}$

 $NiCl_2+NH_4OH+dmg \rightarrow Rosy red ppt$

 $[Ni(dmg)_2]$

- 37. Reaction of BeO with ammonia and hydrogen fluoride gives A which on thermal decomposition gives BeF2 and NH₄ F. What is 'A'?
 - $(1) (NH_4)_2 BeF_4$
- $(2) H_3 NBeF_3$
- $(3) (NH_4)Be_2 F_5$
- (4) (NH₄)BeF₃

Sol.

 $(NH_4)_2 BeF_4 \xrightarrow{\Delta} BeF_2 + NH_4F$

38. Match List I with List II

| LIST I | | LIST II | |
|--------|-----------------------|---------|----------|
| A. | Reverberatory furnace | I. | Pig Iron |
| B. | Electrolytic cell | II. | Aluminum |
| C. | Blast furnace | III. | Silicon |
| D. | Zone Refining furnace | IV. | Copper |

Choose the correct answer from the options given below:

(1) A-IV, B-II, C-I, D-III

(2) A-I, B-III, C-II, D-IV

(3) A-III, B-IV, C-I, D-II

(4) A-I, B-IV, C-II, D-III

Sol. 1

Reverberatory furnance \rightarrow Cr

Electrolysis cell → Ar

Blast furnance → Pig iron

Zone refining furnance \rightarrow silicon

39. Match List I with List II

| LIST I | | LIST II | |
|--------|----------------------------|---------|---------------------------------|
| A. | Chlorophyll | I. | Na ₂ CO ₃ |
| В. | Soda ash | II. | CaSO ₄ |
| C. | Dentistry, Ornamental work | III. | Mg ²⁺ |
| D. | Used in white washing | IV. | Ca(OH) ₂ |

Choose the correct answer from the options given below:

- (1) A-II, B-I, C-III, D-IV
- (2) A-III, B-I, C-II, D-IV
- (3) A-II, B-III, C-IV, D-I
- (4) A-III, B-IV, C-I, D-II

Sol. 2

Chlrophyl \rightarrow Mg²⁺

Sodaash \rightarrow Na₂CO₃

Destistry &

ornamental work \rightarrow CaSO₄

White washing \rightarrow Ca(OH)₂

40. In the following given reaction, 'A' is

$$CH_{3}$$

$$C = CH_{2}$$

$$+ HBr \longrightarrow \text{major product}$$

$$CH_{3} \longrightarrow CH_{3}$$

$$CH_{4} \longrightarrow CH_{3}$$

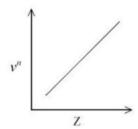
$$CH_{4} \longrightarrow CH_{4}$$

$$CH_{4} \longrightarrow CH_{4}$$

$$CH_{4} \longrightarrow CH_{4}$$

$$\begin{array}{c} CH_3 \\ C=CH_2 \\ H^+ \end{array} \xrightarrow{\qquad \qquad } \begin{array}{c} CH_3 \\ Rearrangment \\ ring\ expansion \end{array}$$

41. It is observed that characteristic X-ray spectra of elements show regularity. When frequency to the power " n " i.e. v^n of X-rays emitted is plotted against atomic number " Z ", following graph is obtained.



The value of " n " is

- (1) 3
- (2) 2
- (3) 1
- $(4)\frac{1}{2}$

Sol.

$$\sqrt{\nu} \propto z$$

$$v^n \propto z$$

$$n = 1/2$$

42. Given below are two statements:

Statement I : Noradrenaline is a neurotransmitter.

Statement II: Low level of noradrenaline is not the cause of depression in human.

In the light of the above statements, choose the correct answer from the options given below

- (1) Statement I is correct but Statement II is incorrect
- (2) Both Statement I and Statement II are correct
- (3) Both Statement I and Statement II are incorrect
- (4) Statement I is incorrect but Statement II is correct
- Sol. 1

Fact

- 43. Which of the Phosphorus oxoacid can create silver mirror from AgNO₃ solution?
 - $(1) (HPO_3)_n$
- $(2) H_4 P_2 O_6$
- $(3) H_4 P_2 O_5$
- $(4) H_4 P_2 O_7$

Sol.

Silver mirror test can gives by p⁺³, p⁺¹ ox acid

$$H_4 \stackrel{+3}{P_2} O_5 + Ag_2O \rightarrow Ag$$

Silver mirror

44. Compound (X) undergoes following sequence of reactions to give the Lactone (Y).

Compound (X)
$$(ii)$$
 HCHO, KOH (iii) KCN(ale) (iii) H₃O⁺ (iii) H₃O⁺ Lactone (Y)

Compound (X) is

$$\begin{array}{c} \operatorname{CH}_3 \\ | \\ (1) \operatorname{HOH}_2 \operatorname{C} - \operatorname{C} - \operatorname{CHO} \\ | \\ \operatorname{CH}_3 \end{array}$$

$$\begin{array}{c} \text{CH}_{3} \\ \text{(3)} \\ \text{H}_{2} \\ \text{C} \\ \text{CH}_{3} \\ \end{array}$$

Sol.

$$\begin{array}{c} \text{CH}_{2}\text{OH} \\ \text{CH}_{3} \\ \text{CH}_{3} \end{array} \begin{array}{c} \text{CH}_{2}\text{OH} \\ \text{CH}_{3} \\ \text{CH}_{3} \end{array} \begin{array}{c} \text{CH}_{2}\text{OH} \\ \text{CH}_{3} \\ \text{CH}_{3} \end{array} \begin{array}{c} \text{CH}_{2}\text{OH} \\ \text{CH}_{3} \\ \text{CH}_{3} \\ \text{CH}_{3} \end{array} \begin{array}{c} \text{CH}_{2}\text{OH} \\ \text{CH}_{3} \\ \text{CH}_{4} \\ \text{CH}_{3} \\ \text{CH}_{4} \\ \text{CH}_{4} \\ \text{CH}_{5} \\ \text{$$

$$\begin{array}{c|cccc} CH_2OH & CH_2OH \\ CH_3 & C-CH & CH_3O^+ \\ CH_3 & C-CH & CH_3 & C-CH \\ CH_3 & C-CH & CN \\ \end{array}$$

cyano hydrine

$$OH$$
 H_3C
 CH_3

Lactone

45. Order of Covalent bond;

A. KF > KI; LiF > KF

 $C. SnCl_4 > SnCl_2; CuCl > NaCl$

E. KF < KI; CuCl > NaCl

B. KF < KI; LiF > KF

D. LiF > KF; CuCl < NaCl

Choose the correct answer from the options given below:

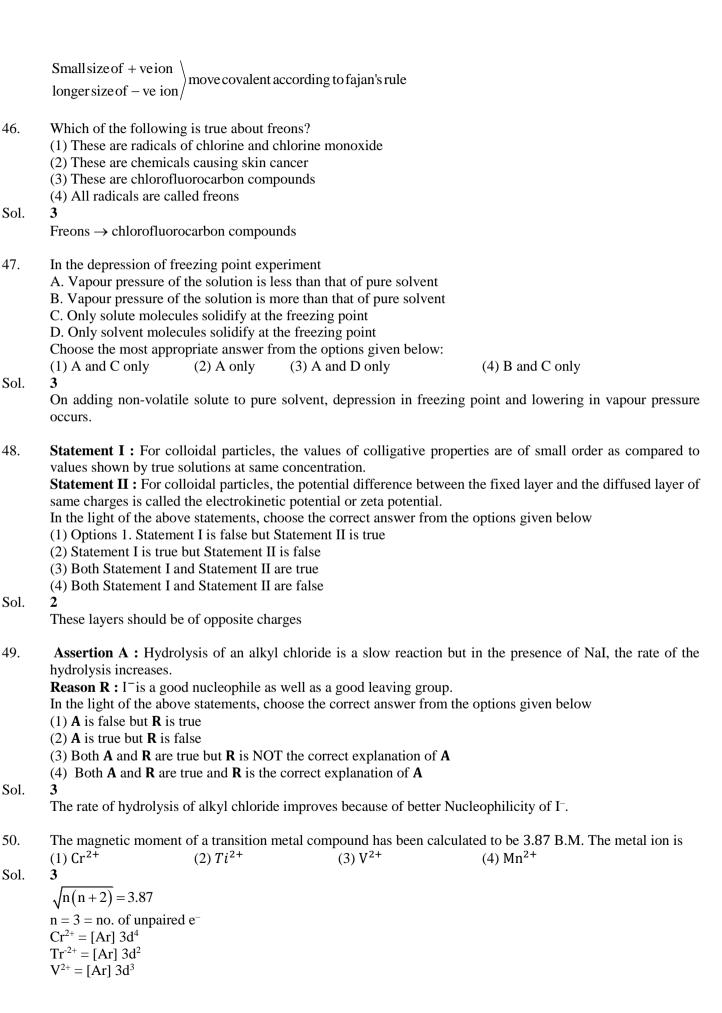
(1) C, E only

(2) B, C, E only (3) A, B only

(4) B, C only

Sol.

2



$$Mn^{2+} = [Ar] 3d^5$$

SECTION - B

- When Fe_{0.93}O is heated in presence of oxygen, it converts to Fe₂O₃. The number of correct statement/s from 51. the following is
 - A. The equivalent weight of $Fe_{0.93}O$ is $\frac{Molecular weight}{0.79}$ B. The number of moles of Fe^{2+} and Fe^{3+} in 1 mole of $Fe_{0.93}O$ is 0.79 and 0.14 respectively

 - C. Fe_{0.93}O is metal deficient with lattice comprising of cubic closed packed arrangement of O^{2-} ions
 - D. The % composition of Fe^{2+} and Fe^{3+} in $Fe_{0.93}O$ is 85% and 15% respectively
- Sol.

$$2x+(0.93-x)3=2$$

$$-x+3\times0.93=2$$

$$x = 0.79$$

$$0.79 = \text{no. of Fe}^{2+} \text{ ion}$$

$$0.14 = \text{no. of Fe}^{3+} \text{ ion}$$

$$nf = 0.79$$

Equivalent wt =
$$\frac{\text{Molecular weight}}{0.79}$$

Due to presence of Fe³⁺ in FeO lattice, Metal deficiency occurs.

% Composition :-
$$Fe^{2+}$$
 ions = $\frac{0.79}{0.93} \times 100$

$$Fe^{3+} ion = \frac{0.14}{0.93} \times 100$$
15%

- The number of correct statement/s from the following is 52.
 - A. Larger the activation energy, smaller is the value of the rate constant.
 - B. The higher is the activation energy, higher is the value of the temperature coefficient.
 - C. At lower temperatures, increase in temperature causes more change in the value of k than at higher temperature
 - D. A plot of $\ln k \nu v \frac{1}{T}$ is a straight line with slope equal to $-\frac{E_a}{R}$
- Sol.

$$K = A e^{-Ea/RT}$$

$$lnK = lnA - Ea/RT$$

slope of
$$\ln K$$
 vs $1/T = -Ea/R$

The higher is the activation energy, higher is the value of the temperature coefficient.

53. For independent processes at 300 K

| Process | $\Delta H/kJ \text{ mol}^{-1}$ | $\Delta S/J K^{-1}$ |
|---------|--------------------------------|---------------------|
| A | -25 | -80 |
| В | -22 | 40 |
| С | 25 | -50 |
| D | 22 | 20 |

The number of non-spontaneous processes from the following is

Sol.

For process A

$$\Delta G = -25 \times 10^3 - 300(-80)$$

$$=-25000+24000$$

$$=-1000 \Rightarrow \Delta G < 0$$
 spontaneous

For process B

$$\Delta G = -22 \times 10^3 - 300(40)$$

=
$$-22000-12000 \Rightarrow \Delta G < 0$$
 spontaneous

For process C

$$\Delta G = 25 \times 10^3 - 300 (-50)$$

$$\Delta G > 0 \Rightarrow$$
 Non-spontaneous

For process D

$$\Delta G = 22 \times 10^3 - 300(20)$$

 $\Delta G > 0 \Rightarrow$ Non-spontaneous.

54. 5 g of NaOH was dissolved in deionized water to prepare a 450 mL stock solution. What volume (in mL) of this solution would be required to prepare 500 mL of 0.1M solution?

Given: Molar Mass of Na, O and H is 23,16 and 1 g mol⁻¹ respectively

Sol. 180

Molarity of stock solution

$$=\frac{5/40}{450}\times1000$$

$$=\frac{50}{4\times45}=\frac{10}{36}M$$

$$M_1V_1=M_2V_2$$

$$\frac{10}{36}$$
 × V = 0.1×500

$$V = \frac{50 \times 36}{10} = 180 \text{ ml}$$

- 55. If wavelength of the first line of the Paschen series of hydrogen atom is 720 nm, then the wavelength of the second line of this series is nm. (Nearest integer)
- Sol. 492

Paschen series :-

$$z = 1$$

Ist line:
$$-4 \rightarrow 3$$

$$\frac{1}{\lambda} = \mathbf{R} \times (1)^2 \left(\frac{1}{3^2} - \frac{1}{4^2} \right)$$

$$\frac{1}{720} = R\left(\frac{7}{144}\right) \tag{1}$$

IInd line
$$\rightarrow 5 \rightarrow 3$$

$$\frac{1}{\lambda} = \mathbf{R} \times (1) \left(\frac{1}{3^2} - \frac{1}{5^2} \right)$$

$$\frac{1}{\lambda} = R\left(\frac{16}{225}\right) \tag{2}$$

Equation $(2) \div \text{equation } (1)$

$$\frac{\lambda}{720} = \frac{7}{144} \times \frac{225}{16}$$

$$\lambda = 492.18$$

56. Uracil is a base present in RNA with the following structure. % of N in uracil is

Sol. 25

Moleculer formula of uracil

 $=C_4N_2H_4O_2$

% of N =
$$\frac{28}{112} \times 100 = 25\%$$

- 57. The dissociation constant of acetic acid is $x \times 10^{-5}$. When 25 mL of 0.2MCH₃COONa solution is mixed with 25 mL of 0.02MCH₃COOH solution, the pH of the resultant solution is found to be equal to 5. The value of x is
- Sol. **10**

$$K_a = x \times 10^{-5}$$

 $CH_3COOH \rightarrow 0.02 \text{ M } \& 25 \text{ ml}$

 $CH_3COONa \rightarrow 0.2 M \text{ and } 25 \text{ ml}$

$$pH = \, p^{K_a} + log \frac{\left[salt\right]}{\left[acid\right]}$$

$$5 = \, p^{K_a} \, + log \, \frac{0.2 \! \times \! 25}{0.02 \! \times \! 25} \! = \, p^{K_a} \, + log \, \, 10$$

$$p^{K_a}=4$$

$$K_a = 10^{-4} = 10{\times}10^{-5}$$

Hence x = 10

58. Number of moles of AgCl formed in the following reaction is

$$Cl \longrightarrow Cl \longrightarrow AgNO_3 \longrightarrow (A) + X AgCl \downarrow$$

Sol. 2

After treat with AgNO₃, Cl⁻ remove and possible carbocation will form

Position 1: Vinyllic carbocation forms, unstable, So not possible

Position 2 : sp² hybridised carbocation is unstable, so not possible

Position 3: Forms 2° carbocation which will be in conjugation with ring

Position 4: 3° stable carbocation will form.

- The d-electronic configuration of $[CoCl_4]^{2-}$ in tetrahedral crystal field is $e^m t_2^n$. Sum of "m" and "number of 59. unpaired electrons" is
- Sol.

$$CO^{+3} \rightarrow 3d^7$$

At 298 K, a 1 litre solution containing 10mmol of $Cr_2O_7^{2-}$ and 100mmol of Cr^{3+} shows a pH of 3.0. Given: $Cr_2O_7^{2-} \to Cr^{3+}$; $E^\circ = 1.330 \text{ V}$ and $\frac{2.303\text{RT}}{\text{F}} = 0.059 \text{ V}$ The potential for the half cell reaction is $x \times 10^{-3} \text{ V}$. The value of x is 60.

Given:
$$Cr_2O_7^{2-} \to Cr^{3+}$$
; $E^{\circ} = 1.330 \text{ V}$ and $\frac{2.303\text{RT}}{\text{F}} = 0.059 \text{ V}$

Sol. 917

$$14H^{+}+Cr_{2}O_{7}^{2-}+6e^{-} \rightarrow 2Cr^{3+}+7H_{2}O$$

$$E = E^{o} - \frac{2.303RT}{6F} log \frac{\left[Cr^{3+}\right]^{2}}{\left[Cr_{2}O_{7}^{2-}\right]\left[H^{+}\right]^{14}}$$

$$pH = 3$$

$$[H^+]=10^{-3}$$

$$E = 1.330 - \frac{0.059}{6} \log \frac{10^{-2}}{10^{-2} (10^{-42})}$$

$$E = 0.917$$

$$=917\times10^{-3}$$

$$x = 917$$

Mathematics

SECTION - A

- **61.** Let $\vec{u} = \hat{\imath} \hat{\jmath} 2\hat{k}$, $\vec{v} = 2\hat{\imath} + \hat{\jmath} \hat{k}$, $\vec{v} \cdot \vec{w} = 2$ and $\vec{v} \times \vec{w} = \vec{u} + \lambda \vec{v}$. Then $\vec{u} \cdot \vec{w}$ is equal to
 - (1) 2
- $(2)\frac{3}{2}$
- (3)1

 $(4)-\frac{2}{3}$

Sol. (3)

$$\overline{\mathbf{v}} \cdot \overline{\mathbf{w}} = \overline{\mathbf{u}} + \lambda \overline{\mathbf{v}}$$

$$\overrightarrow{v} \cdot \overrightarrow{w} \cdot \overrightarrow{v} = \overrightarrow{u} \cdot \overrightarrow{v} + \lambda \overrightarrow{v} \cdot \overrightarrow{v}$$

$$0 = 2 - 1 + 2 + \lambda + 4 + 1 + 1$$

$$\lambda = \frac{-3}{6} \Rightarrow \lambda = \frac{-1}{2}$$

Now

$$\vec{v} \times \vec{w} = \vec{u} + \lambda \vec{v}$$

$$\overrightarrow{\mathbf{v}} \times \overrightarrow{\mathbf{w}} \cdot \overrightarrow{\mathbf{w}} = \overrightarrow{\mathbf{u}} \cdot \overrightarrow{\mathbf{w}} + \lambda \overrightarrow{\mathbf{v}} \cdot \overrightarrow{\mathbf{w}}$$

$$0 = \overline{u} \cdot \overline{w} + \lambda 2$$

$$\overline{\mathbf{u}} \cdot \overline{\mathbf{w}} = -2\lambda = 1$$

- **62.** $\lim_{t\to 0} \left(1^{\frac{1}{\sin^2 t}} + 2^{\frac{1}{\sin^2 t}} + \dots + n^{\frac{1}{\sin^2 t}}\right)^{\sin^2 t}$ is equal to
 - (1) n²
- $(2)^{\frac{n(n+1)}{2}}$
- (3) n
- $(4) n^2 + n$

Sol. (3)

$$\underset{t\to 0}{1} \left(1^{\frac{1}{\sin^2 t}} + 2^{\frac{1}{\sin^2 t}} + + n^{\frac{1}{\sin^2 t}}\right)^{\sin^2 t}$$

$$= \lim_{t \to 0} n \left(\left(\frac{1}{n} \right)^{\frac{1}{\sin^2 t}} + \left(\frac{2}{n} \right)^{\frac{1}{\sin^2 t}} + \dots + \left(\frac{n-1}{n} \right)^{\frac{1}{\sin^2 t}} + 1 \right)^{\sin^2 t}$$

$$= n \cdot \big[0 + 0 + \ldots + 1\big]^0$$

$$=$$
 \boxed{n}

63. Let α be a root of the equation $(a-c)x^2 + (b-a)x + (c-b) = 0$

where a, b, c are distinct real numbers such that the matrix $\begin{bmatrix} \alpha^2 & \alpha & 1 \\ 1 & 1 & 1 \\ a & b & c \end{bmatrix}$

is singular. Then, the value of $\frac{(a-c)^2}{(b-a)(c-b)} + \frac{(b-a)^2}{(a-c)(c-b)} + \frac{(c-b)^2}{(a-c)(b-a)}$ is

- (1)12
- (2)9
- (2)

(4)6

$$(a-c) x^2 + (b-a) x + (c-b) = 0$$
 $(a \ne c)$

$$\boxed{x=1}$$
 is one root & other root is $\boxed{\frac{c-b}{a-c}}$ (1)

$$now \begin{vmatrix} \alpha^2 & \alpha & 1 \\ 1 & 1 & 1 \\ a & b & c \end{vmatrix}$$
 is singular

$$\Rightarrow \begin{vmatrix} \alpha^2 & \alpha & 1 \\ 1 & 1 & 1 \\ a & b & c \end{vmatrix} = \mathbf{O} \Rightarrow \alpha^2(c - b) - \alpha (c - a) + (b - a) = 0$$

$$\Rightarrow \alpha^2 (c-b) + \alpha (a-c) + (b-a) = 0$$

satisfied by
$$\alpha = 1$$
 or $\alpha = \frac{b-a}{c-b}$ (2)

Now, if
$$\alpha = 1$$
 then $\forall a \neq b \neq c$

$$\sum \frac{(a-c)^2}{(b-a)(c-b)} = \frac{\sum (a-c)^3}{(a-b)(b-c)(c-a)}$$
$$= \frac{3(a-b)(b-c)(c-a)}{(a-b)(b-c)(c-a)}$$

[if
$$A+B+C=0$$
 \Rightarrow

$$A^3 + B^3 + C^3 = 3ABC$$

64. The area enclosed by the curves $y^2 + 4x = 4$ and y - 2x = 2 is :

$$(2)^{\frac{22}{3}}$$

&

$$(3)^{\frac{23}{3}}$$

$$(4)^{\frac{25}{2}}$$

$$y^2 + 4x = 4$$

$$y = 2 + 2x$$

P:
$$y^2 = 4(1 - x)$$

$$L: y = 2(1 + x)$$

Now

$$y^2 + 4\left(\frac{y}{2} - 1\right) = 4$$

$$y^2 + 2y - 8 = 0$$

$$(y+4)(y-2)=0$$

$$A = \int_{-4}^{2} \left[\left(\frac{4 - y^2}{4} \right) - \left(\frac{y - 2}{2} \right) \right] dy$$
$$A = \int_{-4}^{2} \left[2 - \frac{y^2}{4} - \frac{y}{2} \right] dy$$

$$= \left[2y - \frac{y^3}{12} - \frac{y^2}{4} \right]^2$$

$$=\left(4-\frac{8}{12}-1\right)-\left(-8+\frac{64}{12}-4\right)$$

$$A = 9$$

- Let p, $q \in \mathbb{R}$ and $(1 \sqrt{3}i)^{200} = 2^{199}(p + iq)$, $i = \sqrt{-1}$ Then $p + q + q^2$ and $p q + q^2$ are roots of the **65.** equation
 - $(1)x^2 4x 1 = 0$ $(2)x^2 4x + 1 = 0$ $(3)x^2 + 4x 1 = 0$ $(4)x^2 + 4x + 1 = 0$
- Sol.

$$(1-\sqrt{3}i)^{200} = 2^{199}(p+iq)$$

$$\Rightarrow 2^{200} \operatorname{cis} \left(\frac{-\pi}{3} \right)^{200} = 2^{199} (p + iq)$$

$$\Rightarrow 2^{200} \left(\operatorname{cis} \left(-\frac{200\pi}{3} \right) \right) = 2^{199} \left(p + iq \right)$$

$$\Rightarrow 2\left(\operatorname{cis}\left(-66\pi - \frac{2\pi}{3}\right)\right) = \left(p + iq\right)$$

$$\Rightarrow 2 \left[cis \left(\frac{-2\pi}{3} \right) \right] = (p + iq)$$

$$\Rightarrow 2\left\lceil \frac{-1}{2} - \frac{\sqrt{3}i}{2} \right\rceil = (p + iq)$$

$$\Rightarrow$$
 p = -1, q = $-\sqrt{3}$

Now

$$\alpha = p + q + q^2 = 2 - \sqrt{3}$$

$$\beta=p-q+q^2=2+\sqrt{3}$$

req. quad is $x^2 - 4x + 1 = 0$

Let N denote the number that turns up when a fair die is rolled. If the probability that the system of 66. equations

$$x + y + z = 1$$

$$2x + Ny + 2z = 2$$

$$3x + 3y + Nz = 3$$

has unique solution is $\frac{k}{6}$, then the sum of value of k and all possible values of N is

- (1)21
- (2)18
- (3)20
- (4)19

Sol. **(3)**

for unique solu.

$$\Delta \neq 0$$

$$\begin{vmatrix} 1 & 1 & 1 \\ 2 & N & 2 \\ 3 & 3 & N \end{vmatrix} \neq 0$$

$$\Rightarrow$$
 $(N^2 - 6) - (2N - 6) + (6 - 3N) \neq 0$

$$\Rightarrow$$
 N² – 5N + 6 \neq 0

$$\Rightarrow$$
 N \neq 3

$$N \neq 2$$

Hence N can be $\{1, 4, 5, 6\}$ Fav case : $\frac{4}{6} = \frac{k}{6} \Rightarrow \boxed{k=4}$

- For three positive integers p, q, r, $x^{pq^2} = y^{qr} = z^{p^2r}$ and r = pq + 1 such that $3,3\log_y x,3\log_z y$, 67. $7\log_x z$ are in A.P. with common difference $\frac{1}{2}$. Then r - p - q is equal to
 - (1) -6
- (2)12
- (3)6

- Sol. **(4)**
 - $x^{pq^2} = v^{qr} = z^{p^2r}$

- r = pq + 1
- 3, $3\log_x^x$, $3\log_z^y$, $7\log_x^z$ are in A.P.

Now

$$3\log_{y}^{x} = 3 + \frac{1}{2} = \frac{7}{2} \Rightarrow \log_{y}^{x} = \frac{7}{6}$$

$$x^6 = v^7$$

$$3\log_z^y = 3 + 1 = 4 \Rightarrow \log_z^y = \frac{4}{3}$$

$$v^3 = z^4$$

$$7\log_{x}^{z} = 3 + \frac{3}{2} = \frac{9}{2} \Rightarrow \log_{x}^{z} = \frac{19}{14}$$

$$z^{14} = x^9$$

Now

$$x^{pq^2} = x^{\frac{6}{7}qr} = x^{\frac{9p^2r}{14}}$$

$$pq^2 = \frac{6}{7}qr = \frac{9}{14}p^2r$$

$$pq = \frac{6}{7}r$$

$$pq = \frac{6}{7}r \qquad q^2 = \frac{9}{14}pr$$

$$r = pq + 1$$

$$\Rightarrow q^3 = \frac{9}{14} \frac{6}{7} r \cdot r$$

$$\Rightarrow r = \frac{6}{7}r + 1$$

$$\Rightarrow \boxed{r=7}$$

$$\Rightarrow q = 3$$

Now

$$r - p - q$$

$$=7-2-3$$

- **68.** The relation $R = \{(a, b) : \gcd(a, b) = 1, 2a \neq b, a, b \in Z\}$ is :
 - (1) reflexive but not symmetric
 - (2) transitive but not reflexive
 - (3) symmetric but not transitive
 - (4) neither symmetric nor transitive

 $gcd(a, b) = 1, 2a \neq b$

reflexive gcd (a, a)=a

Not possible

symmetric gcd (b, a) = $1 \& 2a \neq b$

Not possible

$$a=2,b=1$$

transitive

$$(a, b) = (2, 3)$$
 gcd $\{a, b\} = 1, 2a \neq b$

$$(b, c) = (3, 4)$$
 gcd

$$(b, c) = (3, 4)$$
 gcd $\{c, d\} = 1, 2a \neq c$

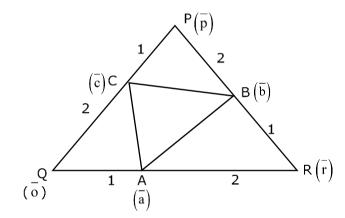
$$(a, c) = (2, 4)$$

$$(a, c) = (2, 4)$$
 gcd $\{2, 4\} = 2$, $2a = c$

Not possible

Let PQR be a triangle. The points A, B and C are on the sides QR, RP and PQ respectively such that 69. $\frac{QA}{AR} = \frac{RB}{BP} = \frac{PC}{CQ} = \frac{1}{2}$. Then $\frac{\text{Area} (\triangle PQR)}{\text{Area} (\triangle ABC)}$ is equal to

Sol. **(2)**



$$\overline{a} = \frac{r}{3}$$

$$\overline{b} = \frac{\overline{b} + 2\overline{r}}{3}$$

$$\bar{c} = \frac{2\bar{p}}{3}$$

$$\Delta PQR = \frac{1}{2} \left| \vec{r} \times \vec{p} \right|$$

$$\Delta PQR = \frac{1}{2} | \vec{a} \times \vec{b} + \vec{b} \times \vec{c} + \vec{c} \times \vec{a} |$$

$$=\frac{1}{2}\left|\frac{\bar{r}\times\bar{p}}{9}+\frac{4(\bar{r}\times\bar{p})}{9}+\frac{2}{9}\bar{p}\times\bar{r}\right|$$

$$= \frac{1}{18} \left| 3(\bar{r} \times \bar{p}) \right|$$

Hence
$$\frac{|\Delta PQR|}{|\Delta ABC|} = 3$$

- Let y = y(x) be the solution of the differential equation $x^3 dy + (xy 1) dx = 0, x > 0, y\left(\frac{1}{2}\right) = 3 e$. **70.** Then y(1) is equal to
 - (1)1
- (2) e
- (3)3
- (4) 2 e

$$x^3 dy + (xy - 1) dx = 0$$

$$x^3 \frac{dy}{dx} = 1 - xy$$

$$x^3 \frac{dy}{dx} + xy = 1$$

$$\left. \frac{\mathrm{d}y}{\mathrm{d}x} + \frac{y}{x^2} = \frac{1}{x^3} \right|_{\mathrm{LDE}}$$

$$IF = e^{\int \frac{1}{x^2} dx} = e^{-\frac{1}{x}}$$

$$y \cdot e^{\frac{-1}{x}} = \int e^{\frac{-1}{x}} \cdot \frac{1}{x^3} dx$$

$$\frac{-1}{x} = t$$

$$=-\int e^{t}t dt$$

$$y \cdot e^{\frac{-1}{x}} = -e^t(t-1) + k$$

$$y \cdot e^{\frac{-1}{x}} = -e^{\frac{-1}{x}} \left(\frac{-1}{x} - 1\right) + k$$

$$y = \left(\frac{1}{x} + 1\right) + k e^{\frac{1}{x}}$$

put
$$x = \frac{1}{2}$$
 \Rightarrow $3 - e = (2 + 1) + k e^2$

$$3-e=(2+1)+k e^2$$

$$k = -\frac{1}{e}$$

Now
$$y(1) = 2 + \frac{-1}{e}e$$

If A and B are two non-zero $n \times n$ matrics such that $A^2 + B = A^2 B$, then 71.

(1)
$$A^2 = I$$
 or $B = I$

$$(2) A^2B = I$$

$$(3) AB = I$$

$$(4) A^2B = BA^2$$

$$A^2 + B = A^2B \qquad \dots (1)$$

$$A^2 - A^2B + B = 0$$

$$A^2 - A^2B - (I - B) = -I$$

$$(I - A^2) (I - B) = I$$

$$(I - A^2)$$
 & $(I - B)$ are inverses of each other

$$(I - B) (I - A^2) = I$$

$$I - B - A^2 + BA^2 = I$$

$$BA^2 = B + A^2$$
(2)

$$A^2B = BA^2$$

The equation $x^2 - 4x + [x] + 3 = x[x]$, where [x] denotes the greatest integer function, has: **72.**

- (1) a unique solution in $(-\infty, 1)$
- (2) no solution
- (3) exactly two solutions in $(-\infty, \infty)$
- (4) a unique solution in $(-\infty, \infty)$

Sol. **(4)**

$$x^2 - 4x + [x] + 3 = x [x]$$

$$x^2 - 4x + 3 = (x - 1)[x]$$

$$(x-1)(x-3) = (x-1)[x]$$

$$x = 1$$
 or

$$x = 1$$
 or $x - 3 = [x]$

$$x - [x] = 3$$

$${x} = 3$$

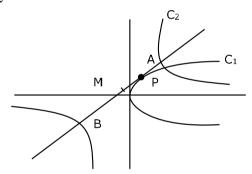
Let a tangent to the curve $y^2 = 24x$ meet the curve xy = 2 at the points A and B. Then the mid points 73. of such line segments AB lie on a parabola with the

- (1) Length of latus rectum $\frac{3}{2}$
- (2) directrix 4x = -3
- (3) length of latus rectum 2
- (4) directrix 4x = 3

Sol.

$$c_1: y^2 = 24 x$$
 & $c_2: xy = 2$

$$c_2 : xy = 2$$



AB: [Tangent to parabola at p(t)]

$$ty = x + 6t^2$$
(1)

[chord with given mid point of hyperbola]

 $T = S_1$

$$\frac{x}{h} + \frac{y}{k} = 2 \qquad \dots (2)$$

from (1) & (2)

$$\frac{-1}{\frac{1}{h}} = \frac{t}{\frac{1}{k}} = \frac{6t^2}{2}$$

$$-h = kt = 3t^2$$

$$h = -3t^2$$
 & $k = 3t$

$$c k = 3$$

$$h = -3\frac{k^2}{9} \implies \boxed{y^2 = -3x}$$

$$\ell$$
 (LR) = 3 & directrix is $x = \frac{3}{4}$

74. Let Ω be the sample space and $A \subseteq \Omega$ be an event.

Given below are two statements:

- (S1): If P(A) = 0, then $A = \emptyset$
- (S2): If P(A) = 1, then $A = \Omega$

Then

- (1) both (S1) and (S2) are true
- (2) only (S1) is true
- (3) only (S2) is true
- (4) both (S1) and (S2) are false
- **Sol.** (4)

Let
$$\Omega = [0, 1]$$

Let A \rightarrow selecting $\frac{1}{2}$

$$A = \left\{ \frac{1}{2} \right\}$$

then, P(A) = 0 but $A \neq \phi$

$$B = A^{c} = [0,1] - \left\{ \frac{1}{2} \right\}$$

$$P(B) = 1$$

but $B \neq \Omega$

Ans = 4

- **75.** The value of $\sum_{r=0}^{22} {}^{22}C_r {}^{23}C_r$ is
 - (1) $^{44}C_{23}$

(2) $^{45}C_{23}$

(3) ⁴⁴C₂₂

 $(4)^{45}C_{24}$

Sol. (2

$$\sum_{r=0}^{22} {}^{22}C_r \cdot {}^{23}C_r$$

$$= \sum_{r=0}^{22} {}^{22}C_r \cdot {}^{23}C_r$$

$$^{45}C_{22} = ^{45}C_{23}$$

76. The distance of the point (-1,9,-16) from the plane 2x + 3y - z = 5 measured parallel to the line

$$\frac{x+4}{3} = \frac{2-y}{4} = \frac{z-3}{12}$$
 is

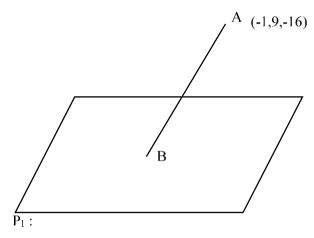
(1) 31

(2) $13\sqrt{2}$

(3) $20\sqrt{2}$

(4) 26

Sol. (4)



$$L_1$$
:

L_{AB}:
$$\frac{x+1}{3} = \frac{y-9}{-4} = \frac{z+16}{12} = t$$

B:
$$(3t-1, 9-4t, 12t-16)$$
 lies on plane
 $2(3t-1)+3(9-4t)-(12t-16)=5$
 $-18t-2+27+16=5$
 $-18t+25+16=5$
 $-18t=-36$ $\Rightarrow \boxed{t=2}$

B:
$$(5, 1, 8)$$
 & A: $(-1, 9, -16)$
 $1(AB) = \sqrt{36 + 64 + 576} = \sqrt{676} = \boxed{26}$

77.
$$\tan^{-1}\left(\frac{1+\sqrt{3}}{3+\sqrt{3}}\right) + \sec^{-1}\left(\sqrt{\frac{8+4\sqrt{3}}{6+3\sqrt{3}}}\right)$$
 is equal to:
 $(1)\frac{\pi}{3}$ $(2)\frac{\pi}{4}$ $(3)\frac{\pi}{6}$ $(4)\frac{\pi}{2}$

Sol. (1)

$$\tan^{-1}\left(\frac{1+\sqrt{3}}{\sqrt{3}(1+\sqrt{3})}\right) + \sec^{-1}\left(\sqrt{\frac{8+4\sqrt{3}}{6+3\sqrt{3}}}\right)$$

$$= \tan^{-1}\left(\frac{1}{\sqrt{3}}\right) + \sec^{-1}\sqrt{\frac{4(2+\sqrt{3})}{3(2+\sqrt{3})}}$$

$$= \tan^{-1}\left(\frac{1}{\sqrt{3}}\right) + \sec^{-1}\frac{2}{\sqrt{3}}$$

$$= \frac{\pi}{6} + \frac{\pi}{6} = \boxed{\frac{\pi}{3}}$$

78. Let
$$f(x) = \begin{cases} x^2 \sin\left(\frac{1}{x}\right) & , x \neq 0 \\ 0 & , x = 0 \end{cases}$$

Then at x = 0

(1) *f* is continuous but not differentiable

(2) f and f' both are continuous

(3) f' is continuous but not differentiable

(4) f is continuous but f' is not continuous

Sol. (4

$$f(x) = \begin{cases} x^2 \sin \frac{1}{x} & x \neq 0 \\ 0 & x = 0 \end{cases}$$

 \rightarrow cont. of f(x) at x = 0

LHL=
$$\lim_{h \to 0} h^2 \sin \frac{1}{(-h)} = 0$$

$$RHL = \lim_{h \to 0} h^2 \sin \frac{1}{(-h)} = 0$$

$$f(0) = 0$$
continuous at $x = 0$

 \rightarrow Diff. of f(x) at x = 0

$$RHD = dt_{h\to 0} \frac{h^2 \sin\left(\frac{1}{h}\right) - 0}{h} = dt_{h\to 0} h \sin\frac{1}{h} = 0$$

$$LHD = dt_{h\to 0} \frac{h^2 \sin\left(\frac{-1}{h}\right) - 0}{-h} = dt_{h\to 0} h \sin\frac{1}{h} = 0$$

Hence f(x) is diff. at x = 0

Now diff. f(s) at x = 0

$$f'(x) = \begin{bmatrix} 2x \sin\left(\frac{1}{x}\right) + x^2 \cos\left(\frac{1}{x}\right) \left(\frac{-1}{x^2}\right) & x \neq 0 \\ 0 & x = 0 \end{bmatrix}$$

$$f'(x) = \begin{bmatrix} 2x \sin\left(\frac{1}{x}\right) - \cos\left(\frac{1}{x}\right) & x \neq 0 \\ 0 & x = 0 \end{bmatrix}$$

hence f'(x) limit ossicilate at x = 0

hence f'(x) is D.C. at x = 0

79. The compound statement
$$(\sim (P \land Q)) \lor ((\sim P) \land Q) \Rightarrow ((\sim P) \land (\sim Q))$$
 is equivalent to

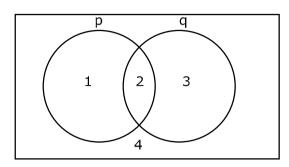
$$(1) (\sim Q) \vee P$$

$$(2)((\sim P)\vee Q)\wedge(\sim Q)$$

$$(3) (\sim P) \vee Q$$

$$(4) ((\sim P) \lor Q) \land ((\sim Q) \lor P)$$

$$(\sim (p \land q) \lor (\sim p \land q)) \rightarrow \sim p \land \sim q$$



$$(1+2+4)+2 \Rightarrow 4$$

$$(1 + 2 + 4) \implies 4 = \sim (1 + 2 + 4) + 4$$

$$= \sim (1 + 2 + 4) + 4$$

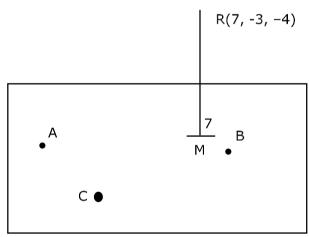
= 3 + 4

The distance of the point (7, -3, -4) from the plane passing through the points (2, -3, 1), (-1, 1, -2)80. and (3, -4, 2) is:

(3)
$$5\sqrt{2}$$

(4)
$$4\sqrt{2}$$

Sol. **(3)**



$$\overline{n}_{p} = \overline{AB} \times \overline{DC} = \begin{vmatrix} \hat{i} & \hat{j} & k \\ 3 & -4 & 3 \\ -1 & 1 & -1 \end{vmatrix} = <1, 0, -1>$$

Equation of plane

P:
$$1(x-2) + 0(y+3) - (z-1) = 0$$

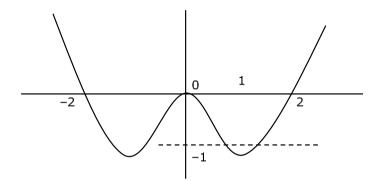
P:
$$x-z-1=0$$

$$d(RM) = \left| \frac{7+4-1}{\sqrt{2}} \right| = 5\sqrt{2}$$

SECTION - B

- **81.** Let $\lambda \in \mathbb{R}$ and let the equation E be $|x|^2 2|x| + |\lambda 3| = 0$. Then the largest element in the set S = $\{x + \lambda : x \text{ is an integer solution of E}\}$ is
- Sol.

$$|x|^2 - 2|x| + |\lambda - 3| = 0$$



$$|x|^2 - 2|x| = -|\lambda - 3|$$

LHS
$$=-1 \le |x|^2 - 2|x| < \infty$$

RHS $-|\lambda - 3| \le 0$

Now LHS = RHS only when

$$0 \le |\lambda - 3| \le 1$$
 &

$$x \in [-2,2]$$

$$-1 \le \lambda - 3 \le 1$$

$$2 \le \lambda \le 4$$

- **82.** Let a tangent to the curve $9x^2 + 16y^2 = 144$ intersect the coordinate axes at the points A and B. Then, the minimum length of the line segment AB is
- Sol.

$$\frac{x^2}{16} + \frac{y^2}{9} = 1$$

Elipse

Let $P = (4\cos\theta, 3\sin\theta)$

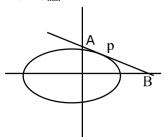
Tp:
$$\frac{x}{4}\cos\theta + \frac{y}{3}\sin\theta = 1$$

A: $(0, 3\csc\theta)$, B: $(4\sec\theta, 0)$

$$\ell(AB) = \sqrt{16\sec^2\theta + 9\cos^2\theta}$$

$$=\sqrt{16+9+\left(4\tan\theta-3\cot\theta\right)^2+24}$$

$$\ell(AB)_{\min} = 7$$



83. The shortest distance between the lines $\frac{x-2}{3} = \frac{y+1}{2} = \frac{z-6}{2}$ and $\frac{x-6}{3} = \frac{1-y}{2} = \frac{z+8}{0}$ is equal to

Sol. 1

$$L_1 : \overline{a} = <2, -1, 6> L_2 : \overline{b} = <6, 1, -8>$$
 $\overline{p} = <3, 2, 2> \overline{q} = <3, -2, 0>$

$$\begin{vmatrix} \hat{i} & \hat{j} & \hat{k} \\ = <4, 6, -12> \end{vmatrix}$$

$$\bar{p} \times \bar{q} = \begin{vmatrix} \hat{i} & \hat{j} & \hat{k} \\ 3 & 2 & 2 \\ 3 & -2 & 0 \end{vmatrix} = <4,6,-12 > \\ = <2,3,-6 >$$

$$b\Delta = \left| \frac{\left(\overline{b} - \overline{a}\right) . \left| \, \overline{p} \times \overline{q} \, \right|}{\mid \overline{p} \times \overline{q} \mid} \right|$$

$$= \left| \frac{\left(4\hat{i} + 2\hat{j} - 14\hat{k}\right) \cdot \left(2\hat{i} + 3\hat{j} - 6\hat{k}\right)}{\sqrt{4 + 9 + 36}} \right|$$

$$=\left|\frac{8+6+84}{\sqrt{49}}\right| = \left|\frac{98}{7}\right| = \boxed{14}$$

84. Suppose $\sum_{r=0}^{2023} r^{2} \, ^{2023} C_r = 2023 \times \alpha \times 2^{2022}$. Then the value of α is

Sol. (1012)

$$\sum_{r=0}^{n} r^{2} {}^{n}C_{r}$$

$$= \sum_{r=0}^{n} r^2 \cdot \frac{n}{r} \, {}^{n-1}C_{r-1}$$

$$= n \sum_{r=1}^{n} \left((r-1)^{n-1} C_{r-1} + {}^{n-1}C_{r-1} \right)$$

$$= n \sum_{r=2}^n \left(n-1\right)^{n-2} C_{r-2} + n \sum_{r=1}^n \, ^{n-1} C_{r-1}$$

$$= n \left(n-1\right) \left[2^{n-2}\right] + n \left[2^{n-1}\right]$$

$$= 2023 \cdot 2022 \cdot 2^{2021} + 2023 \cdot 2^{2022}$$

$$=2023 \cdot 2^{2021} [2022 + 2]$$

$$= 2023 \cdot 2^{2021} \cdot 2024$$

$$= 2023 \, \cdot \, 1012 \, \cdot \, 2^{2022} \, \Rightarrow \alpha \, = 1012$$

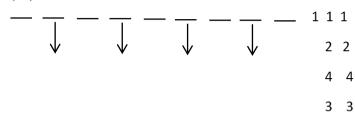
85. The value of
$$\frac{8}{\pi} \int_0^{\frac{\pi}{2}} \frac{(\cos x)^{2023}}{(\sin x)^{2023} + (\cos x)^{2023}} dx$$
 is

$$I = \frac{8}{\pi} \int_{0}^{\frac{\pi}{2}} \frac{\left(\cos x\right)^{2023}}{\left(\sin x\right)^{2023} + \left(\cos x\right)^{2023}} dx$$

$$2I = \frac{8}{\pi} \int_0^{\frac{\pi}{2}} 1 dx$$

$$2I = \frac{8}{\pi} \cdot \frac{\pi}{2} \Rightarrow I = 2$$

- **86.** The number of 9 digit numbers, that can be formed using all the digits of the number 123412341 so that the even digits occupy only even places, is
- Sol. (60)



4 even place can be occupied by 4 even digits

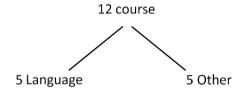
No of always =
$$\frac{4:}{2!2!} = 6$$

Odd place can be occupied by 5 odd digits

No of always =
$$\frac{5!}{3!2!} = 10$$

Total no.
$$= 6 \times 10 = 60$$

- **87.** A boy needs to select five courses from 12 available courses, out of which 5 courses are language courses. If he can choose at most two language courses, then the number of ways he can choose five courses is
- Sol. (546)



(0 language + 5 other) + (1 Language + 4 other) + (2 Language + 3 other)
=
$$5_{C_0} \cdot 7_{C_5} + 5_{C_1} \cdot 7_{C_4} + 5_{C_2} \cdot 7_{C_3}$$

$$= 21 + 175 + 350$$

 $= 546$

88. The 4th term of GP is 500 and its common ratio is
$$\frac{1}{m}$$
, $m \in \mathbb{N}$. Let S_n denote the sum of the first n terms of this GP. If $S_6 > S_5 + 1$ and $S_7 < S_6 + \frac{1}{2}$, then the number of possible values of m is

Sol. (12)

$$T_4 = 500 \implies a \left(\frac{1}{m}\right)^3 = 500 \implies a = 500 \text{ m}^3$$

Now
$$S_n - S_{n-1} = a \left(\frac{1 - r^n}{1 - r} \right) - a \left(\frac{1 - r^{n-1}}{1 - r} \right)$$

$$=\frac{a}{1-r}\Big[r^{n-1}(1-r)\Big]$$

$$= a r^{n-1}$$

$$= 500 \text{ m}^3 \left(\frac{1}{\text{m}}\right)^{n-1}$$

$$S_n - S_{n-1} = 500 \ m^{4-n}$$

Now
$$S_6 - S_5 > 1 \Rightarrow 500 \text{ m}^{-2} > 1 \dots (1)$$

$$\&~S_7 - S_6 < \frac{1}{2} \Rightarrow ~500~m^{-3} < \frac{1}{2} \ldots (2)$$

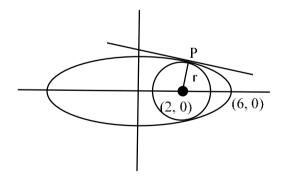
from (1)
$$m^2 < 500$$

from (2) $m^3 > 1000$ $10 < m \le 22$

Number of possible values of m is = 12

89. Let *C* be the largest circle centred at (2,0) and inscribed in the ellipse $\frac{x^2}{36} + \frac{y^2}{16} = 1$. If $(1, \alpha)$ lies on *C*, then $10\alpha^2$ is equal to

Sol. (118)



E:
$$\frac{x^2}{36} + \frac{y^2}{16} = 1 \& C$$
: $(x-2)^2 + y^2 = r^2$

For largest circle r is maximum

P (6cos θ , 4sin θ)

$$N_P$$
: 6 x sec θ – 4 y cosec θ = 20 pass (2, 0)

$$12 \sec\theta = 20 \Rightarrow \cos\theta = \frac{3}{5}$$

Now P:
$$\left(6 \times \frac{3}{5}, 4 \times \frac{4}{5}\right) \Rightarrow$$
 P: $\left(\frac{18}{5}, \frac{16}{5}\right)$

$$r = \sqrt{\left(2 - \frac{18}{5}\right)^2 + \left(\frac{16}{5}\right)^2}$$

$$r = \frac{\sqrt{64 + 256}}{5} = \frac{8\sqrt{5}}{5} = \frac{8}{\sqrt{5}}$$

C: =
$$(x-2)^2 + y^2 = \frac{64}{5}$$

Now $(1, \alpha)$ lies on C

$$\Rightarrow (1-2)^2 + \alpha^2 = \frac{64}{5}$$

$$\alpha^2 = \frac{64}{5} - 1$$

$$\alpha^2 = \frac{59}{5} \Rightarrow 10\alpha^2 = 118$$

90. The value of
$$12 \int_0^3 |x^2 - 3x + 2| dx$$
 is

Sol. (22

$$I = 12 \int_0^3 |x^2 - 3x + 2| dx$$

$$I = 12 \int_0^3 |(x-2)(x-1)| dx$$

$$=12\left[\int_{0}^{1}\left(x^{2}-3x+2\right)+\int_{1}^{2}-\left(x^{2}-3x+2\right)+\int_{2}^{3}\left(x^{2}-3x+2\right)\right]$$

$$=12\left[\left(\frac{1}{3}-\frac{3}{2}+2\right)-\left(\frac{7}{3}-\frac{9}{2}+2\right)+\left(\frac{19}{3}-\frac{15}{2}+2\right)\right]$$

$$= 12 \left[\frac{5}{6} + \frac{1}{6} + \frac{5}{6} \right]$$

$$=\frac{12\cdot11}{6}$$