## FINAL JEE-MAIN EXAMINATION - JULY, 2022

(Held On Thursday 28th July, 2022)

# TIME: 9:00 AM to 12:00 NOON

### **PHYSICS**

#### **SECTION-A**

1. The dimensions of  $\left(\frac{B^2}{\mu_0}\right)$  will be :

(if  $\mu_0$ : permeability of free space and

B: magnetic field)

- (A)  $[M L^2 T^{-2}]$
- (B)  $[M L T^{-2}]$
- (C)  $[M L^{-1} T^{-2}]$
- (D)  $[M L^2 T^{-2} A^{-1}]$

Official Ans. by NTA (C)

**Sol.** 
$$u = \frac{B^2}{2\mu_0}$$

u → Energy per unit volume

$$\left[\frac{\mathbf{B}^2}{\mu_0}\right] = \left[\mathbf{u}\right] = \frac{\left[\mathbf{M}\mathbf{L}^2\mathbf{T}^{-2}\right]}{\left[\mathbf{L}^3\right]} = \left[\mathbf{M}\mathbf{L}^{-1}\mathbf{T}^{-2}\right]$$

2. A NCC parade is going at a uniform speed of 9 km/h under a mango tree on which a monkey is sitting at a height of 19.6 m. At any particular instant, the monkey drops a mango. A cadet will receive the mango whose distance from the tree at time of drop is:

(Given  $g = 9.8 \text{ m/s}^2$ )

- (A) 5 m
- (B) 10 m
- (C) 19.8 m
- (D) 24.5 m

Official Ans. by NTA (A)

Sol. Monkey

$$\begin{array}{c}
\downarrow \\
t = 0 \xrightarrow{Vt}
\end{array}$$

Time taken by mango =  $\sqrt{\frac{2n}{g}}$ 

$$=\sqrt{\frac{2\times19.6}{9.8}}=2 \text{ second}$$

Distance = vt

$$=9\times\frac{5}{18}\times2=5m$$

### **TEST PAPER WITH SOLUTON**

- 3. In two different experiments, an object of mass 5 kg moving with a speed of 25 ms<sup>-1</sup> hits two different walls and comes to rest within
  - (i) 3 second, (ii) 5 seconds, respectively. Choose the correct option out of the following :
  - (A) Impulse and average force acting on the object will be same for both the cases.
  - (B) Impulse will be same for both the cases but the average force will be different.
  - (C) Average force will be same for both the cases but the impulse will be different.
  - (D) Average force and impulse will be different for both the cases.

Official Ans. by NTA (B)

**Sol.** Impulse = change in momentum

$$I = \Delta P$$

$$F_{aug} = \frac{\Delta P}{\Delta t}$$

$$\Delta t_1 = 3$$
  $\Delta t_2 = 5$ 

$$\Delta P_1 = \Delta P_2$$

$$I_1 = I_2$$

F<sub>avg</sub> in case (i) is more than (ii)

- 4. A balloon has mass of 10 g in air. The air escapes from the balloon at a uniform rate with velocity 4.5 cm/s. If the balloon shrinks in 5 s completely. Then, the average force acting on that balloon will be (in dyne).
  - (A)3
- (B)9
- (C) 12
- (D) 18

Official Ans. by NTA (B)

Sol. 
$$F = \frac{dm}{dt}v$$

$$=\frac{10g}{5s}\left(4.5\frac{cm}{s}\right)=9\frac{gcm}{s^2}=9 \text{ dyne}$$

- If the radius of earth shrinks by 2% while its mass 5. remains same. The acceleration due to gravity on the earth's surface will approximately:
  - (A) decrease by 2%
- (B) decrease by 4%
- (C) increase by 2%
- (D) increase by 4%

Official Ans. by NTA (D)

$$\textbf{Sol.} \quad g = \frac{GM}{R^2}$$

$$M = constant g < \frac{1}{R^2}$$

$$100 \ \frac{\Delta g}{g} = -2 \frac{\Delta R}{R} 100$$

% change = -2(-2)

% change in g = 4%

increase by 4%

The force required to stretch a wire of cross-6. section 1 cm<sup>2</sup> to double its length will be:

(Given Yong's modulus of the wire =  $2 \times 10^{11} \text{ N/m}^2$ )

- (A)  $1 \times 10^7 \,\text{N}$
- (B)  $1.5 \times 10^7 \text{ N}$
- (C)  $2 \times 10^7 \text{ N}$
- (D)  $2.5 \times 10^7 \text{ N}$

Official Ans. by NTA (C)

$$Sol. \quad F = \gamma A \frac{\Delta \ell}{\ell}$$

$$=2\times10^{11}\times10^{-4}\left(\frac{2\ell-\ell}{\ell}\right)$$

$$=2\times10^7 \text{ N}$$

- A Carnot engine has efficiency of 50%. If the 7. temperature of sink is reduced by 40°C, its efficiency increases by 30%. The temperature of the source will be:
  - (A) 166.7 K
- (B) 255.1 K
- (C) 266.7 K
- (D) 367.7 K

Official Ans. by NTA (C)

Sol. 
$$\eta = 1 - \frac{T_L}{T_W}$$

$$\frac{1}{2} = 1 - \frac{T_{L}}{T_{H}}$$

$$\frac{1}{2}(1\cdot3) = 1 - \left(\frac{T_L - 40}{T_H}\right)$$

$$\frac{1}{2}(1\cdot3) = \frac{1}{2} + \frac{40}{T_{\text{H}}}$$
  $T_{\text{H}} = 266.7 \text{ K}$ 

$$T_{\rm H} = 266.7 \, \rm K$$

8. Given below are two statements:

> Statement I: The average momentum of a molecule in a sample of an ideal gas depends on temperature.

> **Statement II:** The rms speed of oxygen molecules in a gas is v. If the temperature is doubled and the oxygen molecules dissociate into oxygen atoms, the rms speed will become 2v.

> In the light of the above statements, choose the correct answer from the options given below:

- (A) Both Statement I and Statement II are true
- (B) Both Statement I and Statement II are false
- (C) Statement I is true but Statement II is false
- (D) Statement I is false but Statement II is true

Official Ans. by NTA (D)

**Sol.**  $[P_{avg} = 0]$  (due to random motion)

$$v_{rms} = \sqrt{\frac{3RT}{M}}$$

$$T_{\text{new}} = 2T$$

$$M_{new} = \frac{M}{2}$$

$$\frac{v_{\rm new}}{v} = \frac{\sqrt{\frac{2T}{M/2}}}{\sqrt{\frac{T}{M}}}$$

$$v_{\text{new}} = 2v$$

**9.** In the wave equation

$$y = 0.5 \sin \frac{2\pi}{\lambda} (400 t - x) m$$

the velocity of the wave will be:

- (A) 200 m/s
- (B)  $200\sqrt{2}$  m/s
- (C) 400 m/s
- (D)  $400\sqrt{2} \text{ m/s}$

Official Ans. by NTA (C)

**Sol.**  $y = 0.5 \sin\left(\frac{2\pi}{\lambda} 400t - \frac{2\pi}{\lambda}x\right)$ 

$$\omega = \frac{2\pi}{\lambda} 400$$

$$K = \frac{2\pi}{\lambda}$$

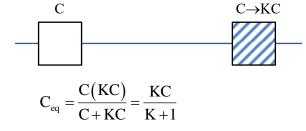
$$v = \frac{\omega}{k}$$

$$[v = 400 \text{ m/s}]$$

- 10. Two capacitors, each having capacitance 40  $\mu F$  are connected in series. The space between one of the capacitors is filled with dielectric material of dielectric constant K such that the equivalence capacitance of the system became 24  $\mu F$ . The value of K will be:
  - (A) 1.5
- (B) 2.5
- (C) 1.2
- (D) 3

Official Ans. by NTA (A)

Sal



$$24 = \frac{K40}{K+1}$$

$$[K=1\cdot 5]$$

- 11. A wire of resistance  $R_1$  is drawn out so that its length is increased by twice of its original length.

  The ratio of new resistance to original resistance is:
  - (A)9:1
- (B) 1:9
- (C) 4:1
- (D) 3:1

Official Ans. by NTA (A)

Sol.  $R_1 = \rho \frac{L_1}{A_1}$ 

$$R_2 = \rho \left( \frac{3L_1}{A_1/3} \right) = 9\rho \frac{L_1}{A_1}$$

$$\therefore \frac{R_2}{R_1} = 9$$

- **12.** The current sensitivity of a galvanometer can be increased by :
  - (A) decreasing the number of turns
  - (B) increasing the magnetic field
  - (C) decreasing the area of the coil
  - (D) decreasing the torsional constant of the spring

    Choose the most appropriate answer from the

    options given below:
  - (A) (B) and (C) only
- (B) (C) and (D) only
- (C)(A) and (C) only
- (D) (B) and (D) only

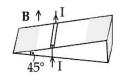
Official Ans. by NTA (D)

**Sol.** 
$$i = \left(\frac{K}{NAB}\right)\theta$$

$$\therefore \frac{d\theta}{di} = \frac{NAB}{K}$$

density 0.45 kg m<sup>-1</sup> is lying horizontally on a smooth incline plane which makes an angle of 45° with the horizontal. The minimum current flowing in the rod required to keep it stationary, when 0.15 T magnetic field is acting on it in the vertical upward direction, will be:

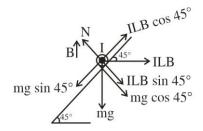
 ${Use g = 10 m/s^2}$ 



- (A)30A
- (B) 15 A
- (C) 10 A
- (D) 3 A

Official Ans. by NTA (A)

Sol.



 $mg \sin 45^{\circ} = ILB \cos 45^{\circ}$ 

$$\therefore \mathbf{I} = \left(\frac{\mathbf{m}}{\mathbf{L}}\right) \frac{\mathbf{g}}{\mathbf{B}}$$

$$=\frac{(0.45)(10)}{0.15}=30\,A$$

14. The equation of current in a purely inductive circuit is  $5\sin(49\pi t - 30^{\circ})$ . If the inductance is 30 mH then the equation for the voltage across the inductor, will be:

$$\left\{ \text{Let } \pi = \frac{22}{7} \right\}$$

- (A)  $1.47\sin(49\pi t 30^{\circ})$  (B)  $1.47\sin(49\pi t + 60^{\circ})$
- (C)  $23.1\sin(49\pi t 30^\circ)$  (D)  $23.1\sin(49\pi t + 60^\circ)$

Official Ans. by NTA (D)

$$Sol. \quad \mathbf{v}_0 = \mathbf{i}_0 \mathbf{x}_{\mathrm{L}}$$

$$=i_0(wL)$$

$$=(5)(49\pi)(30\times10^{-3})$$

$$= 23.1$$

Voltage will lead current by 90°.

$$\therefore$$
 V = 23.1 sin (49  $\pi$ t + 60°)

15. As shown in the figure, after passing through the medium 1. The speed of light  $v_2$  in medium 2 will be:

(Given  $c = 3 \times 10^8 \text{ ms}^{-1}$ )

$$\begin{array}{c|cccc} \text{Air} & \text{Medium 1} & \text{Medium 2} \\ & \mu_r = 1 & \mu_r = 1 \\ & \epsilon_r = 4 & \epsilon_r = 9 \\ \hline \hline C & v_1 & v_2 \\ \end{array}$$

- (A)  $1.0 \times 10^8 \, \text{ms}^{-1}$
- (B)  $0.5 \times 10^8 \, \text{ms}^{-1}$
- (C)  $1.5 \times 10^8 \,\mathrm{ms}^{-1}$
- (D)  $3.0 \times 10^8 \,\mathrm{ms}^{-1}$

Official Ans. by NTA (A)

Sol. 
$$\frac{\mu_2}{\mu_{air}} = \frac{C}{v_2}$$

$$\therefore \frac{\sqrt{\mu_{r_2} \varepsilon_{r_2}}}{(1)} = \frac{C}{v_2}$$

$$\therefore \sqrt{(1)(9)} = \frac{C}{v_2}$$

$$\therefore v_2 = \frac{C}{3}$$

- 16. In normal adjustment, for a refracting telescope, the distance between objective and eye piece is 30 cm. The focal length of the objective, when the angular magnification of the telescope is 2, will be:
  - (A) 20 cm
- (B) 30 cm
- (C) 10 cm
- (D) 15 cm

Official Ans. by NTA (A)

**Sol.** 
$$f_0 + f_e = 30$$

$$m = \frac{f_0}{f_e}$$

$$2 = \frac{f_0}{f_e} \Longrightarrow f_0 = 2f_e$$

So 
$$f_0 + \frac{f_0}{2} = 30$$

$$f_0 = 20 \text{ cm}$$

17. The equation  $\lambda = \frac{1.227}{x}$  nm can be used to find the de-Brogli wavelength of an electron. In this equation x stands for:

Where,

m = mass of electron

P = momentum of electron

K = Kinetic energy of electron

V = Accelerating potential in volts for electron

(A) 
$$\sqrt{mK}$$

(B) 
$$\sqrt{P}$$

(D) 
$$\sqrt{V}$$

### Official Ans. by NTA (D)

**Sol.**  $\lambda = \frac{h}{m\nu}$  (de-Broglie's wavelength)

$$\lambda \, \frac{h}{\sqrt{2m \big(K \cdot E\big)}}$$

$$h = \frac{h}{\sqrt{2mqV}}$$

Putting the values of m; q

We get 
$$\lambda = \frac{1 \cdot 22}{\sqrt{V}}$$
 nm

- 18. The half life period of a radioactive substance is 60 days. The time taken for  $\frac{7}{8}$ th of its original mass to disintegrate will be:
  - (A) 120 days
- (B) 130 days
- (C) 180 days
- (D) 20 days

### Official Ans. by NTA (C)

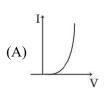
**Sol.** 7/8 disintegrates means 1/8 remains

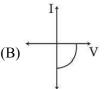
Or 
$$\left(\frac{1}{2}\right)^3$$

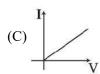
∴ 3 half lives

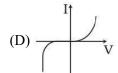
= 180 days

**19.** Identify the solar cell characteristics from the following options:









### Official Ans. by NTA (B)

- **Sol.** Conceptual / theory
- **20.** In the case of amplitude modulation to avoid distortion the modulation index  $(\mu)$  should be:
  - (A)  $\mu \leq 1$
- (B) μ≥1
- (C)  $\mu = 2$
- (D)  $\mu = 0$

## Official Ans. by NTA (A)

**Sol.** 
$$\mu = \frac{A_m}{A_c}$$

 $\mu \le 1$  to avoid distortion

because  $\mu > 1$  will result in interference between career frequency & message frequency.

#### **SECTION-B**

1. If the projection of  $2\hat{i} + 4\hat{j} - 2\hat{k}$  on  $\hat{i} + 2\hat{j} + \alpha\hat{k}$  is zero. Then, the value of  $\alpha$  will be

Official Ans. by NTA (5)

**Sol.** 
$$\vec{a} \cdot \vec{b} = 0$$

$$\vec{a} \cdot \vec{b} = 0$$

$$\therefore 2 \times 1 + 4 \times 2 - 2 \times \alpha = 0$$

$$\alpha = 5$$

2. A freshly prepared radioactive source of half life 2 hours 30 minutes emits radiation which is 64 times the permissible safe level. The minimum time, after which it would be possible to work safely with source, will be \_\_\_\_\_\_ hours.

Official Ans. by NTA (15)

**Sol.** 
$$A = A_0 \times 2^{-t/T}$$

$$\frac{A_0}{64} = A_0 \times 2^{-t/T}$$

$$\therefore$$
 t = 6T = 6 \times 2 \cdot 5 = 15 hours

3. In a Young's double slit experiment, a laser light of 560 nm produces an interference pattern with consecutive bright fringes' separation of 7.2 mm. Now another light is used to produce an interference pattern with consecutive bright fringes' separation of 8.1 mm. The wavelength of second light is \_\_\_\_\_\_ nm.

Official Ans. by NTA (630)

Sol. 
$$\beta \propto \lambda$$

$$\lambda_2 = \frac{9}{8} \, \lambda_1$$

$$\beta_2 = \frac{9}{8}\beta_1 = \frac{9}{8} \times 560 = \boxed{630} \text{ nm}.$$

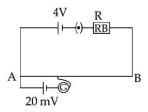
4. The frequencies at which the current amplitude in an LCR series circuit becomes  $\frac{1}{\sqrt{2}}$  times its maximum value, are 212 rad s<sup>-1</sup> and 232 rad s<sup>-1</sup>. The value of resistance in the circuit is  $R = 5\Omega$ . The self inductance in the circuit is \_\_\_\_\_ mH.

Official Ans. by NTA (250)

**Sol.** Band width = 
$$232 - 212 = \frac{R}{L}$$

$$\therefore L = \frac{5}{20} = 250 \text{ mH}$$

5. As shown in the figure, a potentiometer wire of resistance  $20\Omega$  and length 300 cm is connected with resistance box (R.B.) and a standard cell of emf 4 V. For a resistance 'R' of resistance box introduced into the circuit, the null point for a cell of 20 mV is found to be 60 cm. The value of 'R' is  $\Omega$ .



Official Ans. by NTA (780)

Sol. 
$$E = \frac{AC}{AB}(V_A - V_B)$$

$$\therefore 20 \times 10^{-3} = \frac{60}{300} \times \frac{4 \times 20}{R + 20}$$

$$\therefore R = \boxed{780} \Lambda$$

6. Two electric dipoles of dipole moments  $1.2 \times 10^{-30}$  cm and  $2.4 \times 10^{-30}$  cm are placed in two difference uniform electric fields of strengths  $5 \times 10^4$  NC<sup>-1</sup> and  $15 \times 10^4$  NC<sup>-1</sup> respectively. The ratio of maximum torque experienced by the electric dipoles will be  $\frac{1}{x}$ . The value of x is

Official Ans. by NTA (6)

**Sol.** 
$$\left|\tau\right|_{max} = PE$$

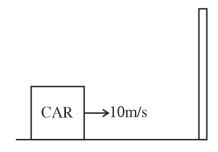
$$\frac{\tau_1}{\tau_2} = \frac{P_1 E_1}{P_2 E_2} = \frac{1 \cdot 2 \times 10^{-30} \times 5 \times 10^4}{2 \cdot 4 \times 10^{-30} \times 15 \times 10^4} = \frac{1}{6}$$

Hence x = 6

7. The frequency of echo will be \_\_\_\_\_\_ Hz if the train blowing a whistle of frequency 320 Hz is moving with a velocity of 36 km/h towards a hill from which an echo is heard by the train driver. Velocity of sound in air is 330 m/s.

Official Ans. by NTA (340)

**Sol.** The hill will be a secondary source.



 $f_1$  = frequency of the car w.r.t. the hill

$$f_1 = \left(\frac{v}{v - v_s}\right) f = \left(\frac{330}{320}\right) \times 320 = 330 \text{ Hz}$$

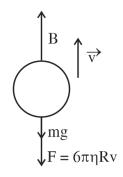
 $f_2$  = Frequency of the sound reflected by hill w.r.t. the car (echo)

$$f_2 = \left(\frac{v + v_0}{v}\right) f_1 = \frac{\left(330 + 10\right)}{330} \times 330 = 340 \text{ Hz}$$

2 mm, rises steadily through a solution of density 1750 kg m<sup>-3</sup> at the rate of 0.35 cms<sup>-1</sup>. The coefficient of viscosity of the solution is \_\_\_\_\_ poise (in nearest integer). (the density of air is negligible).

Official Ans. by NTA (11)

**Sol.** As the bubble is rising steadily the net force acting on it will be zero



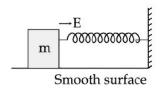
(Because of density of air the value of mg can be neglected)

So 
$$B = F \Rightarrow \frac{4\pi}{3} R^3 \rho g = 6\pi \eta R v$$

Putting  $R = 1 \text{mm} = 10^{-3} \text{m}$   $\rho = 1.75 \times 10^{3} \text{kg/m}^{3}$   $g = 10 \text{ m/s}^{2}$  $v = 0.35 \times 10^{-2} \text{ m/s}$ 

$$\eta = \frac{10}{9} \simeq 1.11 \text{ SI unit} = 11 \text{ poise} (CGS)$$

9. A block of mass 'm' (as shown in figure) moving with kinetic energy E compresses a spring through a distance 25 cm when, its speed is halved. The value of spring constant of used spring will be nE Nm<sup>-1</sup> for n =



Official Ans. by NTA (24)

**Sol.** Using work – energy theorem

$$W_{net} = (K_f - K_i)$$

$$\Rightarrow -\frac{1}{2}Kx^2 = \frac{1}{2}m\left(\frac{v}{2}\right)^2 - \frac{1}{2}mv^2 = \frac{E}{4} - E$$

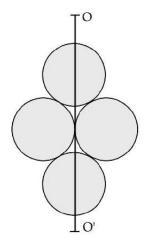
$$\Rightarrow \frac{1}{2}Kx^2 = \frac{3E}{4} \Rightarrow K = \frac{3E}{2x^2}$$

$$\Rightarrow K = \frac{3E}{2\times\left(\frac{1}{4}\right)^2} = 24E$$

$$n = 24$$

10. Four identical discs each of mass 'M' and diameter 'a' are arranged in a small plane as shown in figure. If the moment of inertia of the system about OO' is  $\frac{x}{4}$ Ma<sup>2</sup>. Then, the value of x will be

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Official Ans. by NTA (3)

Sol. 
$$I_1 = I_3 = \frac{MR^2}{4}$$

$$I_2 = \frac{MR^2}{4} + MR^2 = \frac{5}{4}MR^2 = I_4$$

So 
$$I = I_1 + I_2 + I_3 + I_4$$
  
 $= \frac{MR^2}{2} + \frac{5}{2}MR^2$   
 $= 3MR^2$ , Putting  $R = \frac{a}{2}$   
 $I = \frac{3Ma^2}{4}$ , So  $x = 3$ 

## FINAL JEE-MAIN EXAMINATION - JULY, 2022

(Held On Thursday 28th July, 2022)

### TIME: 9:00 AM to 12:00 NOON

#### **CHEMISTRY**

#### **SECTION-A**

- Identify the incorrect statement from the following.
   (A) A circular path around the nucleus in which an electron moves is proposed as Bohr's orbit.
  - (B) An orbital is the one electron wave function  $(\Psi)$  in an atom.
  - (C) The existence of Bohr's orbits is supported by hydrogen spectrum.
  - (D) Atomic orbital is characterised by the quantum numbers n and l only

Official Ans. by NTA (D)

- **Sol.** Atomic orbital is characterised by n, *l*, m.
- **2.** Which of the following relation is not correct?

(A) 
$$\Delta H = \Delta U - P\Delta V$$

(B) 
$$\Delta U = q + W$$

(C) 
$$\Delta S_{svs} + \Delta S_{surr} \ge 0$$

(D) 
$$\Delta G = \Delta H - T\Delta S$$

Official Ans. by NTA (A)

**Sol.** If U + Pv (By definition)

$$\Delta 14 = \Delta U + \Delta (Pr)$$
 at constant pressure

$$\Delta H = \Delta U + P \Delta V$$

3. Match List-I with List-II.

	List-I		List-II	
(A)	$Cd(s) + 2Ni(OH)_3(s) \rightarrow$	(I)	Primary	
	$CdO(s) + 2Ni(OH)_2(s) +$		battery	
	$H_2O(l)$			
(B)	$Zn(Hg) + HgO(s) \rightarrow$	(II)	Discharging of	
	ZnO(s) + Hg(l)		secondary	
			battery	
(C)	$2\text{PbSO}_4(s) + 2\text{H}_2\text{O}(l) \rightarrow$	(III)	Fuel cell	
	$Pb(s) + PbO_2(s) +$			
	$2H_2SO_4(aq)$			
(D)	$2H_2(g) + O_2(g) \rightarrow$	(IV)	Charging of	
	$2H_2O(l)$		secondary	
			battery	

Choose the correct answer from the options given below:

$$(A)(A) - (I), (B) - (II), (C) - (III), (D) - (IV)$$

$$(B)(A) - (IV), (B) - (I), (C) - (II), (D) - (III)$$

$$(C)(A) - (II), (B) - (I), (C) - (IV), (D) - (III)$$

$$(D)(A) - (II), (B) - (I), (C) - (III), (D) - (IV)$$

Official Ans. by NTA (C)

#### **TEST PAPER WITH SOLUTION**

Sol. (a) 
$$Cd(s) + 2Ni(OH)_3(s) \rightarrow CdO(s) + 2Ni(OH)_2(s) + H_2O(l)$$

Discharge of secondary Battery

(b) 
$$Zn(Hg) + HgO(s) \rightarrow ZnO(s) + Hg(l)$$

(Primary Battery Mercury cell)

(c) 
$$2PbSO_4(s) + 2H_2O(l) \rightarrow Pb(s) + PbO_2(s) + 2H_2SO_4(aq)$$

Charging of secondary Battery

(d) 
$$2H_2(g) + O_2(g) \rightarrow 2H_2O(l)$$
 – Fuel cell

**4.** Match List-I with List-II.

	List-I		List-II
	Reaction		Catalyst
(A)	$4NH_3(g) + 5O_2(g) \rightarrow$	(I)	NO(g)
	$4NO(g) + 6H_2O(g)$		
(B)	$N_2(g) + 3H_2(g) \rightarrow$	(II)	$H_2SO_4(l)$
	2NH <sub>3</sub> (g)		
(C)	$C_{12}H_{22}O_{11}(aq) + H_2O(l)$	(III)	Pt(s)
	$\rightarrow$ C <sub>6</sub> H <sub>12</sub> O <sub>6</sub> (Glucose) +		
	$C_6H_{12}O_6$ (Fructose)		
(D)	$2SO_2(g) + O_2(g) \rightarrow$	(IV)	Fe(s)
	$2SO_3(g)$		

Choose the correct answer from the options given below:

$$(A)(A) - (II), (B) - (III), (C) - (I), (D) - (IV)$$

$$(B)(A) - (III), (B) - (II), (C) - (I), (D) - (IV)$$

$$(C)(A) - (III), (B) - (IV), (C) - (II), (D) - (I)$$

$$(D)(A) - (III), (B) - (II), (C) - (IV), (D) - (I)$$

Official Ans. by NTA (C)

Sol.

(a) 
$${}^{4}\text{NH}_{3}(g) + 5\text{O}_{2}(g) \xrightarrow{\text{Pt(s)}} {}^{4}\text{NO}(g) + 6\text{H}_{2}\text{O}(g)$$
  
Ostwald process 500 K

(b) 
$$N_2 + 3H_2 \xrightarrow{Fe(s)} 2NH_3(g)$$

Haber's process

$$\text{(c)} \ \ C_{12} H_{22} O_{11}(\text{aq.}) + H_2 O(\ell) \xrightarrow{\ \ H^+ \ \ } C_6 H_{12} O_6 + C_6 H_{12} O_6 \\ \text{(fluctose)}$$

Inversion of sugar cane

(d) 
$$2SO_2(g) + O_2(g) \xrightarrow{NO(g)} 2SO_3(g)$$

- 5. In which of the following pairs, electron gain enthalpies of constituent elements are nearly the same or identical?
  - (A) Rb and Cs
- (B) Na and K
- (C) Ar and Kr
- (D) I and At

Choose the correct answer from the options given below:

- (A) (A) and (B) only
- (B) (B) and (C) only
- (C) (A) and (C) only
- (D) (C) and (D) only

Official Ans. by NTA (C)

**Sol.** Rb & Cs have nearly same electron gain enthalpy electron gain enthalpy = -46 kj/ml

Ar & Kr have same  $\Delta H_{eq}$ . Value is + 96 kj/ml

- **6.** Which of the reaction is suitable for concentrating ore by leaching process ?
  - (A)  $2Cu_2S + 3O_2 \rightarrow 2Cu_2O + 2SO_2$
  - (B)  $Fe_3O_4 + CO \rightarrow 3FeO + CO_7$
  - (C)  $Al_2O_3 + 2NaOH + 3H_2O \rightarrow 2Na[Al(OH)_4]$
  - (D)  $Al_2O_3 + 6MgO + 4Al$

Official Ans. by NTA (C)

- Sol.  $Al_2O_3 + 2NaOH + 3H_2O \rightarrow 2Na, [Al(OH)_4]$ Leaching.
- 7. The metal salts formed during softening of hardwater using Clark's method are:
  - (A) Ca(OH)<sub>2</sub> and Mg(OH)<sub>2</sub>
  - (B) CaCO<sub>3</sub> and Mg(OH)<sub>2</sub>
  - (C) Ca(OH)<sub>2</sub> and MgCO<sub>3</sub>
  - (D) CaCO<sub>3</sub> and MgCO<sub>3</sub>

Official Ans. by NTA (B)

- Sol. Clark's Method Reaction  $Ca(HCO_3)_2 + Ca(OH)_2 \rightarrow 2CaCO_3 + 2H_2O$ 
  - $Mg(HCO_3)_2 + 2Ca(OH)_2 \rightarrow 2CaCO_3 + Mg(OH)_2 + 2H_2O$

- **8.** Which of the following statement is incorrect? (A) Low solubility of LiF in water is due to its small hydration enthalpy.
  - (B) KO<sub>2</sub> is paramagnetic.
  - (C) Solution of sodium in liquid ammonia is conducting in nature.
  - (D) Sodium metal has higher density than potassium metal

Official Ans. by NTA (A)

- **Sol.** Low solubility of LiF in water is due to high lattice enthalpy
- **9.** Match List-I with List-II, match the gas evolved during each reaction.

	List-I		List-II
(A)	$(NH_4)_2 Cr_2 O_7 \xrightarrow{\Delta}$	(I)	H <sub>2</sub>
(B)	$KMnO_4 + HCl \rightarrow$	(II)	N <sub>2</sub>
(C)	$Al + NaOH + H_2O \rightarrow$	(III)	$O_2$
(D)	$NaNO_3 \xrightarrow{\Delta}$	(IV)	Cl <sub>2</sub>

Choose the correct answer from the options given below:

$$(A)(A) - (II), (B) - (III), (C) - (I), (D) - (IV)$$

$$(B)(A) - (III), (B) - (I), (C) - (IV), (D) - (II)$$

$$(C)(A) - (II), (B) - (IV), (C) - (I), (D) - (III)$$

$$(D)(A) - (III), (B) - (IV), (C) - (I), (D) - (II)$$

Official Ans. by NTA (C)

Sol. 
$$(NH_4)_2 Cr_2O_7 \xrightarrow{\Delta} N_2 + Cr_2O_3 + 4H_2O$$
  
 $KMnO_4 + HCl \rightarrow MnCl_2 + KCl + Cl_2 + H_2O$   
 $Al + NaOH + H_2O \rightarrow H_2 + Na[Al(OH)_4]$   
 $NaNO_3 \longrightarrow NaNO_2 + O_2$ 

- 10. Which of the following has least tendency to liberate  $H_2$  from mineral acids?
  - (A) Cu
- (B) Mn
- (C) Ni
- (D) Zn

Official Ans. by NTA (A)

**Sol.** Copper is least electropositive among the given metals and it lies below H in reactivity series

#### **11.** Given below are two statements:

**Statement I :** In polluted water values of both dissolved oxygen and BOD are very low.

**Statement II:** Eutrophication results in decrease in the amount of dissolved oxygen.

In the light of the above statements, choose the most appropriate answer from the options given below:

- (A) Both Statement I and Statement II are true
- (B) Both Statement I and Statement II are false
- (C) Statement I is true but Statement II is false
- (D) Statement I is false but Statement II is true

### Official Ans. by NTA (D)

- **Sol.** Since eutrophication is result of excessive growth of weed in water bodies, which consume dissolved oxygen of water bodies.
  - :. Eutrophication decreases amount of dissolved oxygen in water bodies.

Polluted water has low value of dissolved oxygen, but high value of BOD (Biological oxygen demand), since chemical and organic matter requires dissolved oxygen to get decompose.

#### **12.** Match List-I with List-II.

	List-I		List-II
(A)		(I)	Spiro compound
(B)	$\rightarrow$	(II)	Aromatic compound
(C)	$\searrow$	(III)	Non-planar Heterocyclic compound
(D)		(IV)	Bicyclo compound

Choose the correct answer from the options given below:

$$(A)(A) - (II), (B) - (I), (C) - (IV), (D) - (III)$$

$$(B)(A) - (IV), (B) - (III), (C) - (I), (D) - (II)$$

$$(C)(A) - (III), (B) - (IV), (C) - (I), (D) - (II)$$

$$(D)(A) - (IV), (B) - (III), (C) - (II), (D) - (I)$$

Official Ans. by NTA (C)

Sol.



: Non-planar heterocyclic Compound

: Bicyclo Compound

 $\triangleright$ 

: Spiro Compound



: Aromatic Compound

**13.** Choose the correct option for the following reactions.

$$B \xleftarrow{\text{(BH_3)}_2}_{\text{H}_2\text{O}_2/\text{OH}^{\Theta}} \text{H}_3\text{C} - \overset{\text{CH}_3}{\text{C}} - \overset{\text{Hg(OAc)}_2, \text{H}_2\text{O}}{\text{NaBH}_4} \rightarrow \text{A}$$

- (A) 'A' and 'B' are both Markovnikov addition products.
- (B) 'A' is Markovnikov product and 'B' is anti-Markovnikov product.
- (C) 'A' and 'B' are both anti-Markovnikov products.
- (D) 'B' is Markovnikov and 'A' is anti-Markovnikov product.

Official Ans. by NTA (B)

Sol.

$$CH_{3}$$

$$CH_{3}$$

$$CH_{3}$$

$$CH_{3}$$

$$CH_{2}$$

$$CH_{3}$$

$$C$$

3

14. Among the following marked proton of which compound shows lowest  $pK_a$  value?

$$\begin{array}{c} H & O \\ \hline \parallel & \parallel \\ (B) & H_2C-C-CH_3 \end{array}$$

$$(C) \qquad \qquad \stackrel{\text{H}}{\longrightarrow} CH_3$$

Official Ans. by NTA (C)

Sol.

(A)
$$\begin{array}{ccc}
H & O \\
CH_2-C-OH \longrightarrow CH_2-C-OH(+R) \\
\parallel & Less stable \\
O & (Cross resonance)
\end{array}$$

(B)
$$\begin{array}{c}
H \\
CH_2-C-CH_3 \longrightarrow CH_2-C-CH_3 (+I \text{ effect}) \\
0 \qquad O
\end{array}$$

(C) 
$$H$$
  $\hookrightarrow$   $CH_3$   $\hookrightarrow$   $CH-C-CH_3(+I)$   $O$   $O$  Resonance through benzene ring acidic) (most stable)

So it has least pK<sub>a</sub> value.

**15.** Identify the major product A and B for the below given reaction sequence.

$$(1) CH_3CHCH_3, AlCl_3 
(2) O_2 
(3) H, H_2O$$

$$P \xrightarrow{Br_2 in CS_2}$$

$$(B)$$

$$(B) \begin{picture}(60,0) \put(0,0){\line(1,0){100}} \put(0,0){\line(1,0)$$

$$(C) \begin{picture}(60,0) \put(0,0){\line(1,0){100}} \put(0,0){\line(1,0)$$

$$(D) \begin{picture}(100,0){\line(1,0){100}} \put(100,0){\line(1,0){100}} \put$$

Official Ans. by NTA (B)

Sol.

$$\begin{array}{c}
CI \\
CH_3 \\
CH-CH_3
\end{array}$$

$$\begin{array}{c}
CH_3 \\
CH-CH_3
\end{array}$$

$$\begin{array}{c}
O_2 \\
H^+
\end{array}$$

$$\begin{array}{c}
O_2 \\
(P)
\end{array}$$

$$\begin{array}{c}
OH \\
\hline
OH \\
\hline
CS_2
\end{array}$$

$$\begin{array}{c}
OH \\
\hline
OH \\
Br
\end{array}$$

$$\begin{array}{c}
(B)
\end{array}$$

$$\begin{array}{c}
OH \\
\hline
Na_2Cr_2O_7 \\
\hline
/H_2SO_4
\end{array}$$

$$\begin{array}{c}
O \\
O \\
O
\end{array}$$

$$O$$

**16.** Identify the correct statement for the below given transformation.

$$CH_3 - CH_2 - CH_2 - CH_2 - CH_3 \xrightarrow{C_2H_5ONa} A + B$$

$$\oplus N(CH_3)_3 \xrightarrow{C_2H_5OH} (Major) + (Minor)$$

(A) A - 
$$CH_3CH_2CH = CH-CH_3$$
,  
B -  $CH_3CH_2CH_2CH = CH_2$ ,  
Saytzeff products

(B) A - 
$$CH_3CH_2CH = CH-CH_3$$
,  
B -  $CH_3CH_2CH_2CH = CH_2$ ,  
Hafmann products

(C) A - 
$$CH_3CH_2CH_2CH = CH_2$$
,  
B -  $CH_3CH_2CH = CHCH_3$ ,  
Hofmann products

(D) A - 
$$CH_3CH_2CH_2CH = CH_2$$
,  
B -  $CH_3CH_2CH = CHCH_3$ ,  
Saytzeff products

### Official Ans. by NTA (C)

Sol.

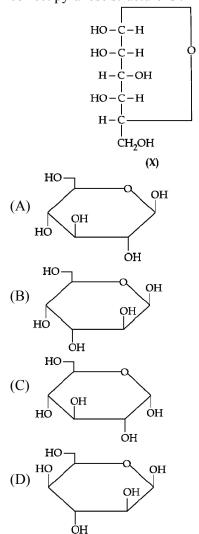
- **17.** Terylene polymer is obtained by condensation of :
  - (A) Ethane-1, 2-diol and Benzene-1, 3 dicarboxylic acid
  - (B) Propane-1, 2-diol and Benzene-1, 4 dicarboxylic acid
  - (C) Ethane-1, 2-diol and Benzene-1, 4 dicarboxylic acid
  - (D) Ethane-1, 2-diol and Benzene-1, 2 dicarboxylic acid

Official Ans. by NTA (C)

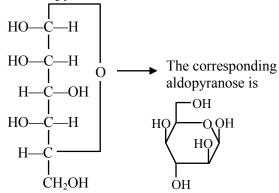
Sol.

Ethane 1,2 diol + 
$$CO_2H$$
 $CH_2OH$ 
 $CO_2H$ 
 $CH_2OH$ 
 $CH_2OH$ 
 $CO_2H$ 
 $CO_2H$ 

**18.** For the below given cyclic hemiacetal (X), the correct pyranose structure is :



### Sol. Correct pyranose structure is



X(Hemiacetal)

- **19.** Statements about Enzyme Inhibitor Drugs are given below:
  - (A) There are Competitive and Non-competitive inhibitor drugs.
  - (B) These can bind at the active sites and allosteric sites.
  - (C) Competitive Drugs are allosteric site blocking drugs.
  - (D) Non-competitive Drugs are active site blocking drugs.

Choose the correct answer from the options given below:

(A)(A), (D) only

(B) (A), (C) only

(C)(A),(B) only

(D) (A), (B), (C) only

#### Official Ans. by NTA (C)

- **Sol.** Enzyme inhibitors can be competitive inhibitors (inhibit the attachment of substrate on active site of enzyme) and non-competitive inhibitor (changes the active site of enzyme after binding at allosteric site.)
- **20.** For kinetic study of the reaction of iodide ion with  $H_2O_2$  at room temperature :
  - (A) Always use freshly prepared starch solution.
  - (B) Always keep the concentration of sodium thiosulphate solution less than that of KI solution.
  - (C) Record the time immediately after the appearance of blue colour.
  - (D) Record the time immediately before the appearance of blue colour.
  - (E) Always keep the concentration of sodium thiosulphate solution more than that of KI solution. Choose the correct answer from the options given below:

(A)(A), (B), (C) only

(B) (A), (D), (E) only

(C) (D), (E) only

(D)(A), (B), (E) only

Official Ans. by NTA (A)

**Sol.** The is recorded immediately after the blue colour appears.

Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> is kept in limited amount.

#### **SECTION-B**

1. In the given reaction,

$$X + Y + 3Z \rightleftharpoons XYZ$$

if one mole of each of X and Y with 0.05 mol of Z gives compound XYZ<sub>3</sub>. (Given : Atomic masses of X, Y and Z are 10, 20 and 30 amu, respectively). The yield of XYZ<sub>3</sub> is g.

(Nearest integer)

Official Ans. by NTA (2)

Sol. 
$$X + Y + 3Z \longrightarrow XYZ_3$$

Z is L.R.

$$\frac{0.05}{3} = 1 \text{ mole of } XYZ_3$$

Mass of XYZ<sub>3</sub> = 
$$\frac{0.05}{3} \times (10 + 20 + 30 \times 3)$$
  
= 2g

2. An element M crystallises in a body centred cubic unit cell with a cell edge of 300 pm. The density of the element is 6.0 g cm<sup>-3</sup>. The number of atoms present in 180 g of the element is \_\_\_\_\_ × 10<sup>23</sup>.

(Nearest integer)

Official Ans. by NTA (22)

**Sol.** M is body certred cubic  $\therefore Z = 2$ 

Let mass of 1 atom of M is A

Edge length = 300 pm

Density =  $6g/cm^3$ 

$$\therefore 6g/cm^3 = \frac{Z \times A}{\left(300 \times 10^{-10}\right)^3} = \frac{2 \times A}{27 \times 10^{-24}}$$

$$A = 81 \times 10^{-24} g$$

 $\therefore$  Atomic mass = 48.6g

... Mole in 
$$180g = \frac{180}{48.6} = 3.7$$
 moles

Atoms of M =  $3.7 \times 6 \times 10^{23}$ 

 $= 22.22 \times 10^{23}$  atoms

**3.** The number of paramagnetic species among the following is \_\_\_\_\_\_.

 $B_2$ ,  $Li_2$ ,  $C_2$ ,  $C_2^-$ ,  $O_2^{2-}$ ,  $O_2^+$  and  $He_2^+$ 

Official Ans. by NTA (4)

- **Sol.** Paramagnetic  $B_2, C_2^-, O_2^+, He_2^+$
- 4. 150 g of acetic acid was contaminated with 10.2 g ascorbic acid ( $C_6H_8O_6$ ) to lower down its freezing point by (x × 10<sup>-1</sup>)°C. The value of x is \_\_\_\_\_. (Nearest integer) [Given  $K_f = 3.9 \text{ K kg mol}^{-1}$ ; Molar mass of ascorbic acid = 176 g mol<sup>-1</sup>]

Official Ans. by NTA (15)

**Sol.**  $150g CH_3COOH$   $10.2g ascorbic acid <math>\Rightarrow 0.058 \text{ moles}$ 

$$\Delta T_{\rm f} = (x \times 10^{-1})^{\circ} C$$

$$\Delta T_f = k_f \cdot \text{molality}$$

$$=3.9\times\frac{0.058}{150}\times1000$$

$$=1.5^{\circ}C$$

$$=15\times10^{-1}$$
°C

5.  $K_a$  for butyric acid (C<sub>3</sub>H<sub>7</sub>COOH) is  $2 \times 10^{-5}$ . The pH of 0.2 M solution of butyric acid is \_\_\_\_ ×  $10^{-1}$ . (Nearest integer) [Given log 2 = 0.30]

Official Ans. by NTA (27)

**Sol.**  $K_a$  of Butyric acid  $\Rightarrow 2 \times 10^{-5}$  PKa = 4.7 pH of 0.2 M solution

$$pH = \frac{1}{2}pK_a - \frac{1}{2}\log C$$

$$=\frac{1}{2}(4\cdot7)\frac{1}{2}\log(0.2)$$

$$= 2.35 + 0.35 = 2.7$$

$$pH = 27 \times 10^{-1}$$

**6.** For the given first order reaction

 $A \rightarrow B$ 

the half life of the reaction is 0.3010 min. The ratio of the initial concentration of reactant to the concentration of reactant at time 2.0 min will be equal to \_\_\_\_\_\_\_. (Nearest integer)

Official Ans. by NTA (100)

**Sol.** A  $\rightarrow$  B  $t_{1/2} = 0.3010 \text{ min}$ A<sub>0</sub>/A<sub>1</sub> at time 2 min = ?

$$K = \frac{2.303}{t} \log \left[ \frac{A_0}{A_t} \right]$$

$$\Rightarrow \frac{0.693}{t_{\frac{1}{2}}} = \frac{2.303}{2} \log \left(\frac{A_0}{A_t}\right)$$

Or 
$$\frac{2.303 \times 0.3010}{0.3010} = \frac{2.303}{2} \log \frac{A_0}{A_t}$$

$$\log \frac{A_0}{A_t} = 2$$

$$\therefore \frac{A_0}{A_1} = 10^2 = 100$$

7. The number of interhalogens from the following having square pyramidal structure is:

ClF<sub>3</sub>, IF<sub>7</sub>, BrF<sub>5</sub>, BrF<sub>3</sub>, I<sub>2</sub>Cl<sub>6</sub>, IF<sub>5</sub>, ClF, ClF<sub>5</sub>

Official Ans. by NTA (3)

- **Sol.** Square pyramidal structures are BrF<sub>5</sub>, IF<sub>5</sub> and ClF<sub>5</sub>.
- 8. The disproportionation of  $MnO_4^{2-}$  in acidic medium resulted in the formation of two manganese compounds A and B. If the oxidation state of Mn in B is smaller than that of A, then the spin-only magnetic moment ( $\mu$ ) value of B in BM is . (Nearest integer)

Official Ans. by NTA (4)

**Sol.** 
$$MnO_4^{2-} \xrightarrow{H^+} MnO_4^- + MnO_2$$

No. of unpaired  $\overline{e} = 3$ 

$$\therefore \mu = \sqrt{15} = 3.877$$

Nearest Integer = 4

9. Total number of relatively more stable isomer(s) possible for octahedral complex  $[Cu(en)_2(SCN)_2]$  will be

Official Ans. by NTA (3)

**Sol.**  $[Cu(en)_2(SCN)_2]$ 

10. On complete combustion of 0.492 g of an organic compound containing C, H and O, 0.7938 g of  $CO_2$  and 0.4428 g of  $H_2O$  was produced. The % composition of oxygen in the compound is \_\_\_\_\_.

Official Ans. by NTA (46)

Sol. 0.492g of  $C_xH_yO_z$ Gives 0.7938 g  $CO_2$  = 0.018 moles 0.4428g  $H_2O$  = 0.0246 moles So moles of C = 0.018 ⇒ 0.216 g Moles of H = 0.049 ⇒ 0.049g ∴ wt. of Oxygen = 0.492 – 0.216 – 0.049

=0.227g

% of Oxygen = 
$$\frac{0.227}{0.492} \times 100$$
 46 (approx.)

# FINAL JEE-MAIN EXAMINATION - JULY, 2022

(Held On Thursday 28th July, 2022)

## TIME: 9:00 AM to 12:00 NOON

#### **MATHEMATICS**

#### **SECTION-A**

- Let the solution curve of the differential equation 1.  $xdy = (\sqrt{x^2 + y^2} + y)dx$ , x > 0, intersect the line x = 1 at y = 0 and the line x = 2 at  $y = \alpha$ . Then the value of  $\alpha$  is:
- (A)  $\frac{1}{2}$  (B)  $\frac{3}{2}$  (C)  $-\frac{3}{2}$  (D)  $\frac{5}{2}$

Official Ans. by NTA (B)

$$Sol. \quad xdy = \left(\sqrt{x^2 + y^2} + y\right)dx$$

$$xdy - ydx = \sqrt{x^2 + y^2}dx$$

$$\frac{xdy - ydx}{x^2} = \sqrt{1 + \frac{y^2}{x^2}} \cdot \frac{dx}{x}$$

$$\frac{d(y/x)}{\sqrt{1+\left(\frac{y}{x}\right)^2}} = \frac{dx}{x}$$

$$\ln\left(\frac{y}{x} + \sqrt{\left(\frac{y}{x}\right)^2 + 1}\right) = \ln x + R$$

$$\frac{y + \sqrt{y^2 + x^2}}{x} = cx$$

$$y + \sqrt{y^2 + x^2} = cx^2$$

$$x = 1, y = 0 \Rightarrow 0 + 1 = C \Rightarrow C = 1$$

Curve is 
$$y + \sqrt{x^2 + y^2} = x^2$$

$$x = 2, y = \alpha$$

$$2 + \sqrt{4 + \alpha^2} = 4$$

$$4 + \alpha^2 = 16 + \alpha^2 = 8\alpha$$

$$\alpha = \frac{3}{2}$$

#### TEST PAPER WITH SOLUTION

- 2. Considering only the principal values of the inverse trigonometric functions, the domain of the function  $f(x) = \cos^{-1}\left(\frac{x^2 - 4x + 2}{x^2 + 3}\right)$  is:

  - $(A)\left(-\infty,\frac{1}{4}\right] \qquad \qquad (B)\left[-\frac{1}{4},\infty\right)$
  - (C)  $\left(-\frac{1}{3}, \infty\right)$  (D)  $\left(-\infty, \frac{1}{3}\right]$

Official Ans. by NTA (B)

Sol.

$$\left|\frac{x^2 + 4x + 2}{x^2 + 3}\right| \le 1$$

$$\iff \left(x^2 - 4x + 2\right)^2 \le \left(x^2 + 3\right)^2$$

$$\Leftrightarrow (x^2 - 4x + 2)^2 - (x^2 + 3)^2 \le 0$$

$$\Leftrightarrow \left(2x^2 - 4x + 5\right)\left(-4x - 1\right) \le 0$$

$$\Leftrightarrow -4x - 1 \le 0 \rightarrow x \ge -\frac{1}{4}$$

- vectors  $\vec{a} = (1+t)\hat{i} + (1-t)\hat{j} + \hat{k}$ ,  $\vec{b} = (1-t)\hat{i} + (1+t)\hat{i} + 2\hat{k}$  and  $\vec{c} = t\hat{i} - t\hat{i} + \hat{k}$ ,  $t \in R$ be such that for  $\alpha$ ,  $\beta$ ,  $\gamma \in R$ ,  $\alpha \vec{a} + \beta \vec{b} + \gamma \vec{c} = \vec{0}$  $\Rightarrow \alpha = \beta = \gamma = 0$ . Then, the set of all values of t is:
  - (A) a non-empty finite set
  - (B) equal to N
  - (C) equal to  $R \{0\}$
  - (D) equal to R

Official Ans. by NTA (C)

### **Sol.** By its given condition

 $: \vec{a}, \vec{b}, \vec{c}$  are linearly independent vectors

$$\Rightarrow \left[\overline{a}\,\overline{b}\,\overline{c}\right] \neq 0 \qquad ..(i)$$

Now, 
$$\lceil \overline{a} \ \overline{b} \ \overline{c} \rceil$$

$$= \begin{vmatrix} 1+t & 1-t & 1\\ 1-t & 1+t & 2\\ t & -t & 1 \end{vmatrix}$$

$$C_2 \rightarrow C_1 + C_2$$

$$\begin{vmatrix} 1+t & 2 & 1 \\ 1-t & 2 & 2 \\ t & 0 & 1 \end{vmatrix}$$

$$= 2 \begin{vmatrix} 1+t & 1 & 1 \\ 1-t & 1 & 2 \\ t & 0 & 1 \end{vmatrix}$$

$$=2\Big[\Big(1+t\Big)-\Big(1-t\Big)+t\Big]$$

$$= 2[3t] = 6t$$

$$\begin{bmatrix} \overline{a} \ \overline{b} \ \overline{c} \end{bmatrix} \neq 0 \Rightarrow t \neq 0$$

Considering the principal 4. values of the inverse trigonometric functions, the sum of solutions  $\cos^{-1}(x) - 2\sin^{-1}(x) = \cos^{-1}(2x)$  is equal to :

(C) 
$$\frac{1}{2}$$

(C) 
$$\frac{1}{2}$$
 (D)  $-\frac{1}{2}$ 

#### Official Ans. by NTA (A)

**Sol.** 
$$\cos^{-1} x = 2\sin^{-1} x = \cos^{-1} 2x$$

$$\cos^{-1} x - 2 \left( \frac{\pi}{2} - \cos^{-1} x \right) = \cos^{-1} 2x$$

$$\cos^{-1} x - \pi + 2\cos^{-1} x = \cos^{-1} 2x$$

$$3\cos^2 x = \pi + \cos^{-1} 2x$$
 ...(1

$$\cos\left(3\cos^{-1}x\right) = \cos\left(\pi + \cos^{-1}2x\right)$$

$$4x^3 - 3x = -2x$$

$$4x^3 = x \Rightarrow x = 0, \pm \frac{1}{2}$$

All satisfy the original equation

$$sum = -\frac{1}{2}to + \frac{1}{2} = 0$$

 $*, \bigcirc \in \{\land, \lor\}.$ 5. operations If Let the  $(p*q) \odot (p \odot \sim q)$  is a tautology, then the ordered pair (\*, ⊙) is:

$$(A)\;(\vee,\wedge)\quad \ (B)\;(\vee,\vee) \ \ \, (C)\;(\wedge,\wedge)\quad (D)\;(\wedge,\vee)$$

### Official Ans. by NTA (B)

Well check each option Sol.

For A 
$$\pi = v \text{ of } 0 = \Lambda$$

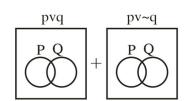
$$(pvq) \land (pv \sim q)$$

$$\equiv pv(q \land \sim q)$$

$$\equiv pv(c) \equiv p$$

For B: 
$$* = v$$
,  $O = v$ 

$$(pvq) \lor (pv \sim q) \equiv t$$
 using Venn Diagrams



6. Let a vector  $\vec{a}$  has a magnitude 9. Let a vector  $\vec{b}$ be such that for every  $(x,y) \in R \times R - \{(0,0)\}$ , the vector  $(x\vec{a} + y\vec{b})$  is perpendicular to the vector  $(6y\vec{a} - 18x\vec{b})$ . Then the value of  $|\vec{a} \times \vec{b}|$  is equal to:

(A) 
$$9\sqrt{3}$$
 (B) 2

(B) 
$$27\sqrt{3}$$
 (C) 9

### Official Ans. by NTA (B)

**Sol.** 
$$|\vec{a}| = 9 \& (x\vec{a} + y\vec{b}) \cdot (6y\vec{a} - 18x\vec{b}) = 0$$

$$\Rightarrow 6xy|\overline{a}|^2 - 18x^2(\overline{a} \cdot \overline{b}) + 6y^2(\overline{a} \cdot \overline{b}) - 18xy|\overline{b}|^2 = 0$$

$$\Rightarrow 6xy(|\overline{a}|^2 - 3|\overline{b}|^2) + (\overline{a} \cdot \overline{b})(y^2 - 3x^2) = 0$$

This should hold  $\forall x, y \in R \times R$ 

$$\therefore |\overline{a}|^2 = 3|\overline{b}|^2 \& (\overline{a} \cdot \overline{b}) = 0$$

Now 
$$\left| \overline{a} \times \overline{b} \right|^2 = \left| \overline{a} \right|^2 \left| \overline{b} \right|^2 - \left( \overline{a} \cdot \overline{b} \right)^2$$

$$= \left| \overline{a} \right|^2 \cdot \frac{\left| \overline{a} \right|^2}{3}$$

$$\therefore |\overline{a} \times \overline{b}| = \frac{|\overline{a}|^2}{\sqrt{3}} = \frac{81}{\sqrt{3}} = 27\sqrt{3}$$

- 7. For  $t \in (0, 2\pi)$ , if ABC is an equilateral triangle with vertices A(sint, -cost), B(cost, sint) and C(a, b) such that its orthocentre lies on a circle with centre  $\left(1, \frac{1}{3}\right)$ , then  $(a^2 b^2)$  is equal to:
  - (A)  $\frac{8}{3}$

- (B) 8
- (C)  $\frac{77}{9}$
- (D)  $\frac{80}{9}$

Official Ans. by NTA (B)

**Sol.**  $s = \sin t, c = \cos t$ 

Let orthocentre be (h,k)

Since it if an equilateral triangle hence orthocentre coincides with centroid.

- $\therefore$  a+s+c=3h, b+s-c=3k
- $\therefore$   $(3h-a)^2 + (3k-b)^2 = (s+c)^2 + (s-c)^2 = 2(s^2+c^2) = 2$
- $\therefore \left(h \frac{a}{3}\right)^2 + \left(K \frac{b}{3}\right)^2 = \frac{2}{9},$

circle centre at  $\left(\frac{a}{3}, \frac{b}{3}\right)$ 

Gives,  $\frac{a}{3} = 1, \frac{b}{3} = \frac{1}{3} \Rightarrow a = 3, b = 1$ 

- $\therefore a^2 b^2 = 8$
- 8. For  $\alpha \in N$ , consider a relation R on N given by  $R = \{(x,y): 3x + \alpha y \text{ is a multiple of } 7\}.$  The relation R is an equivalence relation if and only if :  $(A) \ \alpha = 14$ 
  - (B)  $\alpha$  is a multiple of 4
  - (C) 4 is the remainder when  $\alpha$  is divided by 10
  - (D) 4 is the remainder when  $\alpha$  is divided by 7

Official Ans. by NTA (D)

**Sol.** For R to be reflexive  $\Rightarrow$  x R x

$$\Rightarrow$$
 3x +  $\alpha$ x = 7x  $\Rightarrow$  (3+ $\alpha$ )x = 7K

$$\Rightarrow$$
 3+ $\alpha$  = 7 $\lambda$   $\Rightarrow$   $\alpha$  = 7 $\lambda$  - 3 = 7N + 4, K,  $\lambda$ , N  $\in$  I

 $\therefore$  when  $\alpha$  divided by 7, remainder is 4.

R to be symmetric  $xRy \Rightarrow yRx$ 

$$3x + \alpha y = 7N_1, 3y + \alpha x = 7N_2$$

$$\Rightarrow (3+\alpha)(x+y) = 7(N_1 + N_2) = 7N_3$$

Which holds when  $3 + \alpha$  is multiple of 7

 $\therefore \alpha = 7N + 4$  (as did earlier)

R to be transitive

 $xRy \& yRz \Rightarrow xRz$ .

$$3x + \alpha y = 7N_1 & 3y + \alpha z = 7N_2$$
 and

 $3x + \alpha z = 7N_3$ 

$$\therefore 3x + 7N_2 - 3y = 7N_3$$

$$\therefore 7N_1 - \alpha y + 7N_2 - 3y = 7N_3$$

$$\therefore 7(N_1 + N_2) - (3 + \alpha)y = 7N_3$$

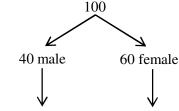
$$\therefore (3+\alpha)y = 7N$$

Which is true again when  $3 + \alpha$  divisible by 7, i.e. when  $\alpha$  divided by 7, remainder is 4.

- 9. Out of 60% female and 40% male candidates appearing in an exam, 60% candidates qualify it. The number of females qualifying the exam is twice the number of males qualifying it. A candidate is randomly chosen from the qualified candidates. The probability, that the chosen candidate is a female, is:
  - (A)  $\frac{3}{4}$
- (B)  $\frac{11}{16}$
- (C)  $\frac{23}{32}$
- (D)  $\frac{13}{16}$

Official Ans. by NTA (A)

Sol.



20 male qualified

40 female qualified

Probability that chosen candidate is female =  $\frac{40}{60} = \frac{2}{3}$ 

**10.** If y = y(x),  $x \in \left(0, \frac{\pi}{2}\right)$  be the solution curve of the

 $\left(\sin^2 2x\right)\frac{dy}{dx} + \left(8\sin^2 2x + 2\sin 4x\right)y =$ 

$$2e^{-4x}(2\sin 2x + \cos 2x)$$
, with  $y(\frac{\pi}{4}) = e^{-\pi}$ ,

then  $y\left(\frac{\pi}{6}\right)$  is equal to :

differential equation

- (A)  $\frac{2}{\sqrt{3}}e^{-2\pi/3}$
- (B)  $\frac{2}{\sqrt{3}}e^{2\pi/3}$
- (C)  $\frac{1}{\sqrt{3}}e^{-2\pi/3}$ 
  - (D)  $\frac{1}{\sqrt{3}}e^{2\pi/3}$

Official Ans. by NTA (A)

Sol. Given differential equation can be re-written as

 $\frac{dy}{dx} + (8 + 4\cot 2x)y = \frac{2e^{-4x}}{\sin^2 2x} (2\sin x + \cos 2x)$ which is a linear diff. equation.

I.f. = 
$$e^{\int (8+4\cot 2x)dx}$$
 =  $e^{8x+2Cu(\sin 2x)}$   
=  $e^{8x} \cdot \sin^2 2x$ 

: solution is

 $y(e^{8x} \cdot \sin^2 2x) = \int 2e^{4x} (2\sin 2x + \cos 2x) dx + C$  $= e^{4x} \cdot \sin 2x + C$ 

Given 
$$y\left(\frac{\pi}{4}\right) = e^{-\pi} \Rightarrow C = 0$$

$$\therefore y = \frac{e^{-4x}}{\sin 2x}$$

$$\therefore y \left(\frac{\pi}{6}\right) = \frac{e^{-4 \cdot \frac{\pi}{6}}}{\sin\left(2 \cdot \frac{\pi}{6}\right)} = \frac{2}{\sqrt{3}} e^{-\frac{2\pi}{3}}$$

- 11. If the tangents drawn at the points P and Q on the parabola  $y^2 = 2x 3$  intersect at the point R(0, 1), then the orthocentre of the triangle PQR is:
  - (A)(0, 1)
- (B) (2, -1)
- (C)(6,3)
- (D)(2,1)

Official Ans. by NTA (B)

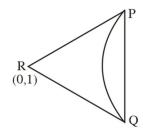
**Sol.** 
$$y^2 = 2x - 3$$
 ...(i)

Equation of chord of contact

$$PQ: r = 0$$

$$yx1 = (x+0) - 3$$

$$y = x - 3 \qquad \dots (2)$$



from (1) and (2)

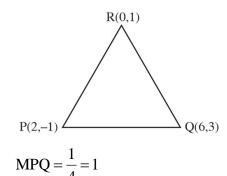
$$(\mathbf{x} \cdot 3)^2 = 2\mathbf{x} - 3$$

$$x^2 - 8x + 12 = 0$$

$$(x-2)(x-6)=0$$

$$x = 2 \text{ or } 6$$

$$y = -1 \text{ or } 3$$



$$MQR = \frac{2}{6} = \frac{1}{3}$$

$$MPR = \frac{2}{6} = \frac{1}{3}$$

$$MPR = \frac{2}{-2} = -1$$

$$MPQ \times MPR = - \Rightarrow PQ \perp PR$$

Orthocentre = 
$$P(2,-1)$$

Let C be the centre of the circle  $x^2 + y^2 - x + 2y =$ **12.** 

$$\frac{11}{4}$$
 and P be a point on the circle. A line passes

through the point C, makes an angle of  $\frac{\pi}{4}$  with the

line CP and intersects the circle at the points Q and R. Then the area of the triangle PQR (in unit<sup>2</sup>) is:

(A)2

- (C)  $8\sin\left(\frac{\pi}{8}\right)$  (D)  $8\cos\left(\frac{\pi}{8}\right)$

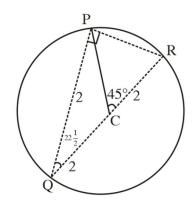
Official Ans. by NTA (B)

**Sol.** 
$$x^2 + y^2 - x + 2y = \frac{11}{4}$$

$$\left(x-\frac{1}{2}\right)^2+\left(y+1\right)^2=\left(2\right)^2$$

Or  $\Delta$  PQR

$$PR = QK \sin 2 \ge \frac{1}{3}$$



$$=4\cdot6\sin\frac{\pi}{8}$$

$$PQ = QR\cos 22\frac{1}{2}$$

$$=4\cos\frac{\pi}{8}$$

As 
$$\triangle PQR = \frac{1}{2}PR \times PQ$$
  
=  $\frac{1}{2} \left( 4^2 \sin \frac{\pi}{6} \right) \left( 4 \cos \frac{\pi}{8} \right)$ 

$$=4\sin\frac{\pi}{4} = \frac{4}{\sqrt{2}} = 2\sqrt{2}$$

- The remainder when  $7^{2022} + 3^{2022}$  is divided by 5 is: 13.
- (B) 2
- (C) 3
- (D) 4

Official Ans. by NTA (C)

**Sol.** 
$$7^{2022} + 3^{2022}$$

$$= (49)^{1011} + (9)^{1011}$$

$$= (50-1)^{1011} + (10-1)^{1011}$$

$$=5\lambda-1+5K-1$$

$$=5m-2$$

Remainder = 5 - 2 = 3

Let the matrix  $A = \begin{bmatrix} 0 & 1 & 0 \\ 0 & 0 & 1 \\ 1 & 0 & 0 \end{bmatrix}$  and the matrix

 $B_0 = A^{49} + 2A^{98}$ . If  $B_n = Adj(B_{n-1})$  for all  $n \ge 1$ ,

then  $det(B_4)$  is equal to:

- (A)  $3^{28}$
- (B)  $3^{30}$  (C)  $3^{32}$  (D)  $3^{36}$

Official Ans. by NTA (C)

**Sol.** 
$$A^2 = \begin{vmatrix} 0 & 1 & 0 \\ 0 & 0 & 1 \\ 1 & 0 & 0 \end{vmatrix} \begin{vmatrix} 0 & 1 & 0 \\ 0 & 0 & 1 \\ 1 & 0 & 0 \end{vmatrix}$$

$$= \begin{bmatrix} 0 & 0 & 1 \\ 1 & 0 & 0 \\ 0 & 1 & 0 \end{bmatrix}$$

 $a \leftrightarrow R_2$ 

$$-\begin{bmatrix} 1 & 0 & 0 \\ 0 & 0 & 1 \\ 0 & 1 & 0 \end{bmatrix}$$

 $R_2 \leftrightarrow R_3$ 

$$\begin{bmatrix} 1 & 0 & 0 \\ 0 & 1 & 0 \\ 0 & 0 & 1 \end{bmatrix} = I$$

$$B_0 = A^{49} + 2A^{98}$$
  
=  $A + 2I$   
 $B_n = Adj(B_n - 1)$ 

$$B_4 = Adj(Adj(Adj(AdjB_0)))$$
$$= |B_0|^{(n-1)^4}$$
$$= |B_0|^{16}$$

$$\mathbf{B}_0 = \begin{bmatrix} 0 & 1 & 0 \\ 0 & 0 & 1 \\ 1 & 0 & 0 \end{bmatrix} + \begin{bmatrix} 2 & 0 & 0 \\ 0 & 2 & 0 \\ 0 & 0 & 2 \end{bmatrix}$$

$$= \begin{bmatrix} 2 & 1 & 0 \\ 0 & 2 & 1 \\ 1 & 0 & 2 \end{bmatrix}$$

$$=2(4-0)-1(0-1)$$

=9

$$B_4(9)^{16} = (3)^{32}$$

**15.** Let 
$$S_1 = \left\{ z_1 \in \mathbb{C} : |z_1 - 3| = \frac{1}{2} \right\}$$
 and

$$S_2 = \{z_2 \in C : |z_2 - |z_2 + 1|| = |z_2 + |z_2 - 1||\}.$$
 Then,

for  $z_1 \in S_1$  and  $z_2 \in S_2$ , the least value of  $|z_2 - z_1|$ is:

(A) 0

(B) 
$$\frac{1}{2}$$

(C) 
$$\frac{3}{2}$$

(B) 
$$\frac{1}{2}$$
 (C)  $\frac{3}{2}$  (D)  $\frac{5}{2}$ 

Official Ans. by NTA (C)

Sol. 
$$|z_2 + |z_2 - 1||^2 = |z_2 - |z_2 + 1||^2$$
  

$$\Rightarrow |z_2 + |z_2 - 1||(\overline{z}_2 + |z_2 - 1|) = (z_2 - |z_2 + 1|)(\overline{z}_2 - (z_2 + 1))$$

$$\Rightarrow z_{2} |\overline{z}_{2} + 12_{2} - 1| - (\overline{z}_{2} - |z_{2} + 1|) + \overline{z}_{2} (|z_{2} - 1| + |z_{2} + 1|)$$

$$= |z_{2} + 1|^{2} = |z_{2} - 1|^{2}$$

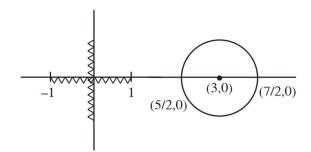
$$\Rightarrow \left[z_2 + \overline{z}_2\right) \left(\left|z_2 - 1\right|\right) + \left(z_2 + 1\right|\right) = 2\left(z_2 + \overline{z}_2\right)$$

$$\Rightarrow (z_2 + \overline{z}_2)(|z_2 - 1| + |z_2 + 1| - 2) = 0$$

$$\therefore z_2 + \overline{z}_2 = 0 \text{ or } |z_2 - 1| + |z_2 + 1| - 2 = 0$$

 $\therefore$  z<sub>2</sub> lie on imaginary axis. Or on real axis with in [-1,1]

Also  $|z_1 - 3| = \frac{1}{2}$  lie on circle having centre 3 and radius  $\frac{1}{2}$ .



Clearly 
$$|z_1 - z_2| \min = \frac{5}{2} - 1 = \frac{3}{2}$$

16. The foot of the perpendicular from a point on the circle  $x^{2} + y^{2} = 1$ , z = 0 to the plane 2x + 3y + z = 6lies on which one of the following curves?

(A) 
$$(6x + 5y - 12)^2 + 4(3x + 7y - 8)^2 = 1$$
,  
 $z = 6 - 2x - 3y$ 

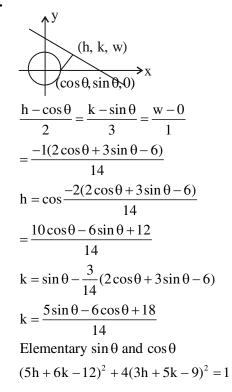
(B) 
$$(5x + 6y - 12)^2 + 4(3x + 5y - 9)^2 = 1$$
,  
 $z = 6 - 2x - 3y$ 

(C) 
$$(6x + 5y - 14)^2 + 9(3x + 5y - 7)^2 = 1$$
,  
 $z = 6 - 2x - 3y$ 

(D) 
$$(5x + 6y - 14)^2 + 9(3x + 7y - 8)^2 = 1$$
,  
 $z = 6 - 2x - 3y$ 

Official Ans. by NTA (B)

Sol.



17. If the minimum value of  $f(x) = \frac{5x^2}{2} + \frac{\alpha}{x^5}$ , x > 0, is

14, then the value of  $\alpha$  is equal to :

- (A)32
- (B) 64
- (C) 128
- (D) 256

Official Ans. by NTA (C)

**Sol.**  $\frac{x^2}{2} + \frac{x^2}{2} + \frac{x^2}{2} + \frac{x^2}{2} + \frac{x^2}{2} + \frac{\alpha}{2x^5} + \frac{\alpha}{2x^5}$ 

$$\geq 7 \left(\frac{\alpha^2}{2^7}\right)^{\frac{1}{7}}$$

$$\frac{7 \cdot (\alpha)^{2/7}}{2} = 14$$

$$(\alpha^2)^{1/7} = 2^2$$

$$\alpha = \left(2^2\right)^{7/2} = 2^7$$

- $\alpha = 128$
- **18.** Let  $\alpha$ ,  $\beta$  and  $\gamma$  be three positive real numbers. Let  $f(x) = \alpha x^5 + \beta x^3 + \gamma x, x \in R \text{ and } g: R \to R \text{ be}$

such that g(f(x)) = x for all  $x \in R$ . If  $a_1, a_2, a_3,..., a_n$  be in arithmetic progression with mean zero, then

the value of  $f\left(g\left(\frac{1}{n}\sum_{i=1}^{n}f(a_i)\right)\right)$  is equal to:

(A)0

(B) 3

- (C)9
- (D) 27

Official Ans. by NTA (A)

**Sol.** Consider a case when  $\alpha = \beta = 0$  then

$$f(x) = yx$$

$$g(x) = \frac{x}{y}$$

$$\frac{1}{n} \sum_{i=1}^{n} f(a_i) \Longrightarrow \frac{y}{n} (a_1 + a_2 + \dots + a_n)$$

$$=0$$

$$\Rightarrow f(g(0)) \Rightarrow f(0)$$

 $\Rightarrow 0$ 

**19.** Consider the sequence  $a_1$ ,  $a_2$ ,  $a_3$ , ..... such that

$$a_1 = 1$$
,  $a_2 = 2$  and  $a_{n+2} = \frac{2}{a_{n+1}} + a_n$  for  $n = 1, 2, 3, ...$ 

$$If \qquad \left(\frac{a_1 + \frac{1}{a_2}}{a_3}\right) \cdot \left(\frac{a_2 + \frac{1}{a_3}}{a_4}\right) \cdot \left(\frac{a_3 + \frac{1}{a_4}}{a_5}\right) \cdot \dots \cdot \left(\frac{a_{30} + \frac{1}{a_{31}}}{a_{32}}\right) = 2^{\alpha} \left({}^{61}C_{31}\right),$$

then  $\alpha$  is equal to:

- (A) 30
- (B) -31
- (C) -60
- (D) -61

Official Ans. by NTA (C)

**Sol.**  $a_{n+2} a_{n+1} - a_{n+1} a_n = 2$ 

Series will satisfy

$$a_1a_2$$
,  $a_2a_3$ ,  $a_3a_4$ ,  $a_4a_5$ ,

$$\frac{a_{n} + \frac{1}{a_{n+1}}}{a_{n+2}} = \frac{a_{n+2} - \frac{1}{a_{n+1}}}{a_{n+2}}$$

$$=1-\frac{1}{a_{n+1}a_{n+2}}$$

$$=1-\frac{1}{2(r+1)}$$

$$=\frac{2r+1}{2(r+1)}$$

Now proof is given by

$$=\prod_{r=1}^{30}\frac{(2r+1)}{2(r+1)}$$

$$=\frac{(1\cdot3\cdot5\cdot....\cdot61)}{2^{30}\cdot(2\cdot3\cdot\cdot\cdot31)}$$

$$\Rightarrow \frac{(1 \cdot 3 \cdot 5 \cdot \dots \cdot 61)}{|31 \cdot 2^{30}|} \times \frac{2^{30} \times |30|}{2^{30} \times |30|}$$

$$=\frac{61}{2^{60}|31\cdot|30}$$

$$\alpha = -60$$

**20.** The minimum value of the twice differentiable function  $f(x) = \int_{0}^{x} e^{x-t} f'(t) dt - (x^2 - x + 1)e^x$ ,  $x \in \mathbb{R}$ ,

is:

$$(A) - \frac{2}{\sqrt{e}}$$

(B) 
$$-2\sqrt{e}$$

(C) 
$$-\sqrt{e}$$

(D) 
$$\frac{2}{\sqrt{e}}$$

Official Ans. by NTA (A)

**Sol.** 
$$f(x) = e^x \cdot \int_0^x \frac{f'(t)}{e^t} dt$$

$$f'(x) = e^{x} \cdot \int_{0}^{x} \frac{f'(t)}{e^{t}} dt + e^{x} \cdot \frac{f'(x)}{e^{x}}$$
$$- \left[ (2x - 1) \cdot e^{x} + (x^{2} - x + 1) \cdot e^{x} \right]$$

$$\int_{0}^{x} \frac{f'(t)}{e^{t}} dt = x^{2} + x$$

$$\frac{f'(x)}{e^x} = 2x + 1$$

$$f'(x) = (2x+1) \cdot e^x$$

$$f'(x) = 0 \Rightarrow x = -\frac{1}{2}$$

$$f(x) = (2x + 1) \cdot e^x - 2e^x + C$$

$$f(0) = -1$$

$$-1 = 1 - 2 + C$$

$$C = 0$$

$$f(x) = e^x (2x - 1)$$

$$f\left(-\frac{1}{2}\right) = \frac{-2}{\sqrt{e}}$$

#### **SECTION-B**

Let S be the set of all passwords which are six to eight characters long, where each character is either an alphabet from {A, B, C, D, E} or a number from {1, 2, 3, 4, 5} with the repetition of characters allowed. If the number of passwords in S whose at least one character is a number from {1, 2, 3, 4, 5} is α×5<sup>6</sup>, then α is equal to \_\_\_\_\_.

Official Ans. by NTA (7073)

- **Sol.** Required no. = Total no character from  $\{1, 2, 3, 4, 5\}$ =  $(10^6 - 5^6) + (10^7 - 5^7) + (10^8 - 5^8)$ =  $10^6 (1 + 10 + 100) - 5^6 (1 + 5 + 25)$ =  $10^6 \times 111 - 5^6 \times 31$ =  $2^6 \times 5^6 \times 111 - 5^6 \times 31$ =  $5^6 (2^6 \times 111 - 31)$ =  $5^6 \times 7073$   $\alpha$  $\alpha = 7073$
- 2. Let P(-2, -1, 1) and  $Q\left(\frac{56}{17}, \frac{43}{17}, \frac{111}{17}\right)$  be the vertices of the rhombus PRQS. If the direction ratios of the diagonal RS are  $\alpha$ , -1,  $\beta$ , where both  $\alpha$  and  $\beta$  are integers of minimum absolute values, then  $\alpha^2 + \beta^2$  is equal to \_\_\_\_\_\_.

Official Ans. by NTA (450)

**Sol.** RS 
$$\equiv$$
  $(\alpha, -1, \beta)$ 

DR of PQ 
$$\equiv \left(\frac{56}{17} + 2, \frac{43}{17} + 1, \frac{111}{17} - 1\right)$$

$$\equiv \left(\frac{90}{17}, \frac{60}{17}, \frac{94}{17}\right)$$

$$\frac{90}{17}\alpha + \frac{60}{17}(-1) + \frac{94}{17}\beta = 0$$

$$90\alpha + 94\beta = 60$$

$$\beta = \frac{60 - 90\alpha}{94}$$

$$\beta = \frac{30(2-3\alpha)}{94}$$

$$\beta = -30 \frac{(3\alpha - 2)}{94}$$

$$\beta = \frac{-15}{47}(3\alpha - 2)$$

$$\Rightarrow \frac{\beta}{-15} = \frac{3\alpha - 2}{47}$$

$$\Rightarrow \beta = -15, \alpha = -15$$

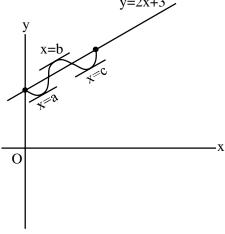
$$\alpha^2 + \beta^2 = 225 + 225$$

$$=450$$

3. Let  $f:[0,1] \to R$  be a twice differentiable function in (0, 1) such that f(0) = 3 and f(1) = 5. If the line y = 2x + 3 intersects the graph of f at only two distinct points in (0, 1), then the least number of points  $x \in (0, 1)$ , at which f''(x) = 0, is \_\_\_\_\_.

Official Ans. by NTA (2)

Sol.



$$f'(a) = f'(b) = f'(c) = 2$$
  

$$\Rightarrow f''(x) \text{ is zero}$$
for at least  $x_1 \in (a, b) \& x_2 \in (b, c)$ 

4. If  $\int_{0}^{\sqrt{3}} \frac{15x^3}{\sqrt{1+x^2+\sqrt{(1+x^2)^3}}} dx = \alpha\sqrt{2}+\beta\sqrt{3}$ , where

 $\alpha$ ,  $\beta$  are integers, then  $\alpha + \beta$  is equal to

#### Official Ans. by NTA (10)

Sol. Put 
$$1 + x^2 = t^2$$
  
 $2x dx = 2t dt$   
 $X dx = t dt$   

$$\therefore \int_{1}^{2} \frac{15(t^2 - 1)t dt}{\sqrt{t^2 + t^3}}$$

$$15 \int_{1}^{2} \frac{t(t^2 - 1)}{t\sqrt{1 + t}} dt$$
Put  $1 + t = u^2$   
 $dt = 2u du$   

$$15 \int_{\sqrt{2}}^{\sqrt{3}} \frac{(u^2 - 1)^2 - 1}{u} \times 2u du$$

$$30 \int_{\sqrt{2}}^{\sqrt{3}} (u^4 - 2u^2) du$$

$$30 \left(\frac{u^5}{5} - \frac{2u^3}{3}\right)_{-5}^{\sqrt{3}}$$

$$30\left[\frac{1}{5}\left(\sqrt{3}^5 - \sqrt{2}^5\right) - \frac{2}{3}\left(\sqrt{3}^3 - \sqrt{2}^3\right)\right]$$

$$30\left[\frac{1}{5}\left(9\sqrt{3} - 4\sqrt{2}\right) - \frac{2}{3}\left(3\sqrt{3} - 2\sqrt{2}\right)\right]$$

$$30\left[-\frac{1}{5} \times \sqrt{3} + \frac{8}{15}\sqrt{2}\right]$$

$$-6\sqrt{3} + 16\sqrt{2} = \alpha\sqrt{2} + \beta\sqrt{3}$$

$$\alpha = 16, \beta = -6$$

$$\therefore \alpha + \beta = 10$$

5. Let  $A = \begin{bmatrix} 1 & -1 \\ 2 & \alpha \end{bmatrix}$  and  $B = \begin{bmatrix} \beta & 1 \\ 1 & 0 \end{bmatrix}$ ,  $\alpha, \beta \in \mathbb{R}$ . Let  $\alpha_1$  be the value of  $\alpha$  which satisfies  $(A + B)^2 = A^2 + \begin{bmatrix} 2 & 2 \\ 2 & 2 \end{bmatrix}$  and  $\alpha_2$  be the value of  $\alpha$  which satisfies  $(A + B)^2 = B^2$ . Then  $|\alpha_1 - \alpha_2|$  is equal to \_\_\_\_\_.

Official Ans. by NTA (2)

Sol. 
$$A + B = \begin{bmatrix} \beta + 1 & 0 \\ 3 & \alpha \end{bmatrix}$$

$$(A + B)^{2} = \begin{bmatrix} \beta + 1 & 0 \\ 3 & \alpha \end{bmatrix} \begin{bmatrix} \beta + 1 & 0 \\ 3 & \alpha \end{bmatrix}$$

$$= \begin{bmatrix} (\beta + 1)^{2} & 0 \\ 3(\beta + 1) + 3\alpha & \alpha^{2} \end{bmatrix}$$

$$A^{2} = \begin{bmatrix} 1 & -1 \\ 2 & \alpha \end{bmatrix} \begin{bmatrix} 1 & -1 \\ 2 & \alpha \end{bmatrix}$$

$$= \begin{bmatrix} -1 & -1 - \alpha \\ 2 + 2\alpha & \alpha^{2} - 2 \end{bmatrix}$$

$$\therefore \begin{bmatrix} 1 & -\alpha + 1 \\ 2\alpha + 4 & \alpha^{2} \end{bmatrix} = \begin{bmatrix} (\beta + 1)^{2} & 0 \\ 3(\alpha + \beta + 1) & \alpha^{2} \end{bmatrix}$$

$$\boxed{\alpha = 1} = \alpha_{1}$$

$$B^{2} = \begin{bmatrix} \beta & 1 \\ 1 & 0 \end{bmatrix} \begin{bmatrix} \beta & 1 \\ 1 & 0 \end{bmatrix}$$

$$= \begin{bmatrix} \beta^{2} + 1 & \beta \\ \beta & 1 \end{bmatrix} = \begin{bmatrix} (\beta + 1)^{2} & 0 \\ 3(\beta + 1) + 3\alpha & \alpha^{2} \end{bmatrix}$$

$$\therefore \beta = 0, \alpha = -1 = \alpha_{2}$$

$$|\alpha_{1} - \alpha_{2}| = |1 - (-1)| = 2$$

6. For  $p,q \in R$ , consider the real valued function  $f(x) = (x-p)^2 - q$ ,  $x \in R$  and q > 0. Let  $a_1$ ,  $a_2$ ,  $a_3$  and  $a_4$  be in an arithmetic progression with mean p and positive common difference. If  $|f(a_i)| = 500$  for all i = 1, 2, 3, 4, then the absolute difference between the roots of f(x) = 0 is

Official Ans. by NTA (50)

**Sol.**  $f(x) = 0 \implies (x - p)^2 - q = 0.$ 

Roots are  $p + \sqrt{q}$ ,  $p - \sqrt{q}$  absolute difference between roots  $2\sqrt{q}$ .

Now, 
$$|f(a_i)| = 500$$

Let  $a_1$ ,  $a_2$ ,  $a_3$ ,  $a_4$  are  $a_1 a + d$ , a + 2d, a + 3d

$$|f(a_4)| = 500$$

$$|(a_1 - p)^2 - q| = 500$$

$$\Rightarrow$$
  $(a_1 - p)^2 - q = 500$ 

$$\Rightarrow \frac{9}{4}d^2 - q = 500 \qquad (1)$$

and 
$$|f(a_1)|^2 = |f(a_2)|^2$$

$$((a_1 - p)^2 - q)^2 = ((a_2 - p)^2 - q)^2$$

$$\Rightarrow ((a_1 - p)^2 - (a_2 - p)^2) ((a_1 - p)^2 - q + (a_2 - p)^2 - q) = 0$$

$$\Rightarrow \frac{9}{4} d^2 - q + \frac{d^2}{4} - q = 0$$

$$2q = \frac{10d^2}{4} \Rightarrow q = \frac{5d^2}{4}$$

$$\Rightarrow$$
 d<sup>2</sup> =  $\frac{4q}{5}$ 

From equation (1)  $\frac{9}{4} \cdot \frac{4 \cdot q}{5} - q = 500$ 

$$\frac{4q}{5} = 500$$

$$\frac{4q}{5} = 500$$

and 
$$2\sqrt{q} = 2 \times \frac{50}{2} = 50$$

- 7. For the hyperbola H:  $x^2 y^2 = 1$  and the ellipse E:  $\frac{x^2}{a^2} + \frac{y^2}{b^2} = 1$ , a > b > 0, let the
  - (1) eccentricity of E be reciprocal of the eccentricity of H, and
  - (2) the line  $y = \sqrt{\frac{5}{2}}x + K$  be a common tangent of E and H.

Then  $4(a^2 + b^2)$  is equal to .

Official Ans. by NTA (3)

**Sol.** 
$$e_E = \sqrt{1 - \frac{b^2}{a^2}}$$
,  $e_H = \sqrt{2}$ 

If 
$$\Rightarrow e_E = \frac{1}{e_H}$$

$$\Rightarrow \frac{a^2 - b^2}{a^2} = \frac{1}{2}$$

$$2a^{2-2b}2 = a^2$$

$$a^2 = 2b^2$$

and  $y = \sqrt{\frac{5}{2}}x + k$  is tangent to ellipse then

$$K^2 = a^2 \times \frac{5}{2} + b^2 = \frac{3}{2}$$

$$6b^2 = \frac{3}{2} \Rightarrow b^2 = \frac{1}{4} \text{ and } a^2 = \frac{1}{2}$$

$$\therefore 4.(a^2 + b^2) = 3$$

8. Let  $x_1, x_2, x_3, \ldots, x_{20}$  be in geometric progression with  $x_1 = 3$  and the common ration  $\frac{1}{2}$ . A new data is constructed replacing each  $x_i$  by  $(x_i - i)^2$ . If  $\overline{x}$  is the mean of new data, then the greatest integer less than or equal to  $\overline{x}$  is \_\_\_\_\_\_.

Official Ans. by NTA (142)

**Sol.** 
$$\sum x_0^1 = \frac{3\left(1 - \left(\frac{1}{2}\right)\right)^{20}}{1 - \frac{1}{2}} = 6\left(1 - \frac{1}{2^{20}}\right)$$

$$= \sum_{i=1}^{20} (x_{i-i})^2$$

$$= \sum_{i=1}^{20} (x_i)^2 + (i)^2 - 2x_i i$$

Now = 
$$\sum_{i=1}^{20} (x_i)^2 = \frac{9(1-(\frac{1}{4}))^{20}}{1-\frac{1}{4}} = 12(1-\frac{1}{2^{40}})$$

$$\sum_{i=1}^{20} i^2 = \frac{1}{6} \times 20 \times 21 \times 41 = 2870$$

$$\sum_{i=1}^{20} x_i \ i = s = 3 + 2.3 \frac{1}{2} + 3.3 \frac{1}{2^2} + 4.3 \frac{1}{2^3} + \dots AGP$$

$$= 6 \left( 2 - \frac{22}{2^{20}} \right)$$

$$\overline{x} = \frac{12 - \frac{12}{2^{40}} + 2870 - 12\left(2 - \frac{22}{2^{20}}\right)}{20}$$

$$\overline{x} = \frac{2858}{20} + \left(\frac{-12}{2^{40}} + \frac{22}{2^{20}}\right) \times \frac{1}{20}$$

$$\lceil \overline{x} \rceil = 142$$

9. 
$$\lim_{x \to 0} \left( \frac{\left(x + 2\cos x\right)^3 + 2\left(x + 2\cos x\right)^2 + 3\sin\left(x + 2\cos x\right)}{\left(x + 2\right)^3 + 2\left(x + 2\right)^2 + 3\sin\left(x + 2\right)} \right)^{\frac{100}{x}}$$

is equal to \_\_\_\_\_

Official Ans. by NTA (1)

Sol.

$$\lim_{x \to 10} \left( \frac{\left(x + 2\cos x\right)^3 + 2\left(x + 2\cos x\right)^2 + 3\sin\left(x + 2\cos x\right)}{\left(x + 2\right)^3 + 2\left(x + 2\right)^2 + 3\sin\left(x + 2\right)} \right)^x$$

Form 1°

$$= e^{\lim_{x\to 0} \left[ \frac{\left( (x+2\cos x)^3 + 2(x+2\cos x)^2 + 3\sin(x+2\cos x)}{(x+2)^3 + 2(x+2)^2 + 3\sin(x+2)} \right) - 1 \right] \times \frac{100}{x}}$$

$$= e^{\lim_{x \to 0} \left[ \frac{100}{x} \left( \frac{(x+2\cos x)^3 + 2(x+2\cos x)^2 + 3\sin(x+2\cos x) - \left( (x+2)^3 + 2(x+2)^2 + 3\sin(x+2) \right)}{(x+2)^3 + 2(x+2)^2 + 3\sin(x+2)} \right]} \right]}$$

$$= e^{\lim_{x\to 0} \frac{100}{x} \left[ \left( \frac{(x+2\cos x)^3 + (x+2)^3 + 2(x+2\cos x)^2 - 2(x+2)^2 + 3\sin(x+2\cos x) - 3\sin(x+2)}{8+8+3\sin^2} \right) \right]}$$

$$=e^{\frac{100}{16+3\sin^2}\lim_{x\to 0}\frac{3(x+2\cos x)^2\times (1+2\sin x)-3(x+2)^2-4(x+2\cos x)}{x(1-2\sin x)-4(x+2)+3\cos(x+2\cos x)\times (1-2\sin x)-3\cos(x+2)}}$$

$$= e^{\frac{100}{16+3\sin 2}} \left( \frac{12-3(4)+8\times 1-8+3\cos 2-3\cos 2}{1} \right)$$

Using L'H rule.

$$= e^{\circ} = 1$$

10. The sum of all real values of x for which  $\frac{3x^2 - 9x + 17}{x^2 + 3x + 10} = \frac{5x^2 - 7x + 19}{3x^2 + 5x + 12}$  is equal to \_\_\_\_.

Official Ans. by NTA (6)

Sol. 
$$\frac{3x^2 - 9x + 17}{x^2 + 3x + 10} = \frac{5x^2 - 7x + 19}{3x^2 + 5x + 12}$$
$$\frac{x^2 + 3x + 10 + 2x^2 - 12x + 7}{x^2 + 3x + 10} = \frac{3x^2 + 5x + 12 + 2x^2 - 12x + 7}{3x^2 + 5x + 12}$$
$$1 + \frac{2x^2 - 12x + 7}{x^2 + 3x + 10} = 1 + \frac{2x^2 - 12x + 7}{3x^2 + 5x + 12}$$

$$(2x^{2} - 12x + 7)\left(\frac{1}{x^{2} + 3x + 10} - \frac{1}{3x^{2} + 5x + 12}\right) = 0$$

$$2x^2 - 12x + 7 = 0$$
 OR  $3x^2 + 5x + 12 = x^2 + 3x + 10$ 

$$x = \frac{12 \pm \sqrt{D}}{4}$$

$$2x^{2} + 2x + 2 = 0$$

$$x^{2} + x + 1 = 0$$

Sum of Roots = 6 No solution.