# FINAL JEE-MAIN EXAMINATION - APRIL, 2024

(Held On Tuesday 09th April, 2024)

TIME: 3:00 PM to 6:00 PM

# **MATHEMATICS**

#### **SECTION-A**

- 1.  $\lim_{x\to 0} \frac{e (1 + 2x)^{\frac{1}{2x}}}{x}$  is equal to :
  - (1) e

(2)  $\frac{-2}{e}$ 

(3) 0

 $(4) e-e^2$ 

Ans. (1)

- Sol.  $\lim_{x \to 0} \frac{e e^{\frac{1}{2x}\ln(1+2x)}}{x}$   $= \lim_{x \to 0} (-e) \frac{\left(e^{\frac{\ln(1+2x)}{2x} 1} 1\right)}{x}$   $= \lim_{x \to 0} (-e) \frac{\ln(1+2x) 2x}{2x^2}$ 
  - $=(-e)\times(-1)\frac{4}{2\times2}=e$
- 2. Consider the line L passing through the points (1, 2, 3) and (2, 3, 5). The distance of the point  $\left(\frac{11}{3}, \frac{11}{3}, \frac{19}{3}\right)$  from the line L along the line

$$\frac{3x-11}{2} = \frac{3y-11}{1} = \frac{3z-19}{2}$$
 is equal to :

(1)3

(2)5

(3)4

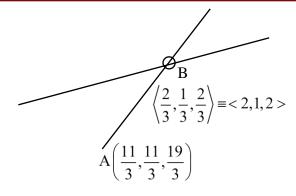
(4)6

Ans. (1)

Sol. 
$$\frac{x-1}{2-1} = \frac{y-2}{3-2} = \frac{z-3}{5-3}$$
  

$$\Rightarrow \frac{x-1}{1} = \frac{y-2}{1} = \frac{z-3}{2} = \lambda$$

# **TEST PAPER WITH SOLUTION**



$$B(1+\lambda,2+\lambda,3+2\lambda)$$

D.R. of AB = 
$$<\frac{3\lambda - 8}{3}, \frac{3\lambda - 5}{3}, \frac{6\lambda - 10}{3}>$$

$$B\left(\frac{5}{3}, \frac{8}{3}, \frac{13}{3}\right) \frac{3\lambda - 8}{3\lambda - 5} = \frac{2}{1} \Rightarrow 3\lambda - 8 = 6\lambda - 10$$

$$3\lambda = 2$$

$$\lambda = \frac{2}{3}$$

$$AB = \frac{\sqrt{36+9+36}}{3} = \frac{9}{3} = 3$$

3. Let 
$$\int_{0}^{x} \sqrt{1-(y'(t))^{2}} dt = \int_{0}^{x} y(t)dt$$
,  $0 \le x \le 3$ ,  $y \ge 0$ ,

y(0) = 0. Then at x = 2, y'' + y + 1 is equal to:

(1) 1

- (2) 2
- (3)  $\sqrt{2}$
- (4) 1/2

Ans. (1)

**Sol.** 
$$\sqrt{1-(y'(x))^2} = y(x)$$

$$1 - \left(\frac{\mathrm{d}y}{\mathrm{d}x}\right)^2 = y^2$$

$$\left(\frac{\mathrm{d}y}{\mathrm{d}x}\right)^2 = 1 - y^2$$

$$\frac{dy}{\sqrt{1-y^2}} = dx \text{ OR } \frac{dy}{\sqrt{1-y^2}} = -dx$$

$$\Rightarrow$$
  $\sin^{-1}y = x + c$ ,  $\sin^{-1}y = -x + c$ 

$$x = 0, y = 0 \implies c = 0$$

$$\sin^{-1} y = x$$
, as  $y \ge 0$ 

$$sinx = y$$

$$\Rightarrow \frac{\mathrm{d}y}{\mathrm{d}x} = \cos x$$

$$\frac{\mathrm{d}^2 y}{\mathrm{d} x^2} = -\sin x$$

$$\Rightarrow$$
 - sinx + sinx + 1 = 1

- Let z be a complex number such that the real part 4. of  $\frac{z-2i}{z+2i}$  is zero. Then, the maximum value of
  - |z-(6+8i)| is equal to:

Ans. (1)

Sol. 
$$\frac{z-2i}{z+2i} + \frac{\overline{z}+2i}{\overline{z}-2i} = 0$$

$$z\overline{z} - 2i\overline{z} - 2iz + 4(-1)$$

$$+z\overline{z}+2zi+2\overline{z}i+4(-1)=0$$

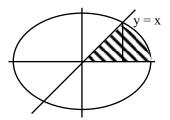
$$\Rightarrow 2|\mathbf{z}|^2 = 8 \Rightarrow |\mathbf{z}| = 2$$

$$|z-(6+8i)|_{\text{maximum}} = 10+2=12$$

- The area (in square units) of the region enclosed by 5. the ellipse  $x^2 + 3y^2 = 18$  in the first quadrant below the line y = x is:
  - (1)  $\sqrt{3}\pi + \frac{3}{4}$
- (3)  $\sqrt{3}\pi \frac{3}{4}$  (4)  $\sqrt{3}\pi + 1$

Ans. (2)

**Sol.** 
$$\frac{x^2}{18} + \frac{y^2}{6} = 1$$



$$\frac{x^2}{18} + \frac{3x^2}{18} = 1 \implies 4x^2 = 18 \implies x^2 = \frac{9}{2}$$

$$\int_{\frac{3}{\sqrt{2}}}^{3\sqrt{2}} \frac{\sqrt{18 - x^2}}{\sqrt{3}} \, dx$$

$$= \frac{1}{\sqrt{3}} \left( \frac{x\sqrt{18 - x^2}}{2} + \frac{18}{2} \sin^{-1} \frac{x}{3\sqrt{2}} \right)_{\frac{3}{\sqrt{2}}}^{3\sqrt{2}}$$

$$= \frac{1}{\sqrt{3}} \left( 9 \times \frac{\pi}{2} - \frac{3}{2\sqrt{2}} \times \frac{3\sqrt{3}}{\sqrt{2}} - 9 \times \frac{\pi}{6} \right)$$

Required Area

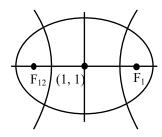
$$= \frac{1}{2} \times \frac{9}{2} + \left(\frac{18\pi}{6} - \frac{9\sqrt{3}}{4}\right) \frac{1}{\sqrt{3}}$$

$$=\sqrt{3}\pi$$

- Let the foci of a hyperbola H coincide with the foci of the ellipse E :  $\frac{(x-1)^2}{100} + \frac{(y-1)^2}{75} = 1$  and the eccentricity of the hyperbola H be the reciprocal of the eccentricity of the ellipse E. If the length of the transverse axis of H is a and the length of its conjugate axis is  $\beta$ , then  $3\alpha^2 + 2\beta^2$  is equal to :
  - (1)242
  - (2)225
  - (3) 237
  - (4)205

Ans. (2)

Sol.



$$e_1 = \sqrt{1 - \frac{75}{100}} = \frac{5}{10} = \frac{1}{2}$$

$$e_2 = 2$$

$$F_1$$
 (6, 1),  $F_2$  (-4, 1)

$$2ae_2 = 10 \Rightarrow a = \frac{5}{2} \Rightarrow 2a = 5$$

$$\Rightarrow \alpha = 5$$

$$4 = 1 + \frac{b^2}{a^2} \Longrightarrow b^2 = 3a^2$$

$$b = \sqrt{3} \times \frac{5}{2}$$

$$\beta = 5\sqrt{3}$$

$$3\alpha^2 + 2\beta^2 = 3 \times 25 + 2 \times 25 \times 3$$

- 7. Two vertices of a triangle ABC are A(3, -1) and B (-2, 3), and its orthocentre is P(1, 1). If the coordinates of the point C are  $(\alpha, \beta)$  and the centre of the circle circumscribing the triangle PAB is (h, k), then the value of  $(\alpha + \beta) + 2(h + k)$  equals :
  - (1)51

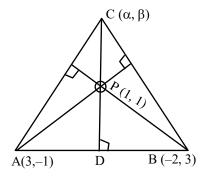
(2)81

(3)5

(4) 15

Ans. (3)

Sol.



$$M_{AB} = \frac{4}{-5} \Rightarrow M_{DP} = \frac{5}{4}$$

Equation of PC is  $y - 1 = \frac{5}{4}(x-1)$  .....(1)

$$M_{AP} = \frac{2}{-2} = -1 \implies M_{BC} = +1$$

Equation of BC is y - 3 = (x + 2) ......(2)

On solving (1) and (2)

$$x + 4 = \frac{5}{4}(x-1) \Rightarrow 4x + 16 = 5x - 5 \Rightarrow \alpha = 21$$

$$\Rightarrow \beta = y = x + 5 = 26$$

$$\alpha + \beta = 47$$

Equation of  $\perp$  bisector of AP

$$y - 0 = (x - 2)$$
 .....(3)

Equation of  $\perp$  bisector of AB

$$y-1=\frac{5}{4}\left(x-\frac{1}{2}\right)$$
....(4)

On solving (3) & (4)

$$(x-3)4 = 5x - \frac{5}{2}$$

$$x = \frac{-19}{2} = h$$

$$y = \frac{-23}{2} = k$$

$$\Rightarrow$$
 2(h + k) = -42

8. If the variance of the frequency distribution is 160, then the value of  $c \in N$  is

X	c	2c	3c	4c	5c	6c
f	2	1	1	1	1	1

(1)5

(2) 8

(3)7

(4) 6

Ans. (3)

Sol.

X	С	2C	3C	4C	5C	6C
f	2	1	1	1	1	1

$$\overline{x} = \frac{(2+2+3+4+5+6)C}{7} = \frac{22C}{7}$$

$$Var (x) = \frac{c^2 (2 + 2^2 + 3^2 + 4^2 + 5^2 + 6^2)}{7}$$

$$-\left(\frac{22c}{7}\right)^2$$

$$= \frac{92c^2}{7} - c^2 \times \frac{484}{49}$$

$$= \frac{(644 - 484)c^2}{49} = \frac{160c^2}{49}$$

$$160 = \frac{160 \times c^2}{49} \Rightarrow c = 7$$

**9.** Let the range of the function

$$f(x) = \frac{1}{2 + \sin 3x + \cos 3x}$$
,  $x \in IR$  be [a, b].

If  $\alpha$  and  $\beta$  are respectively the A.M. and the G.M.

of a and b, then  $\frac{\alpha}{\beta}$  is equal to :

(1) 
$$\sqrt{2}$$

(3) 
$$\sqrt{\pi}$$

$$(4) \pi$$

Ans. (1)

**Sol.** 
$$f(x) = \frac{1}{2 + \sin 3x + \cos 3x}$$

$$\left[\frac{1}{2+\sqrt{2}}, \ \frac{1}{2-\sqrt{2}}\right]$$

$$\frac{\alpha}{\beta} = \frac{a+b}{2\sqrt{ab}} = \frac{1}{2} \left( \sqrt{\frac{a}{b}} + \sqrt{\frac{b}{a}} \right)$$

$$= \frac{1}{2} \left( \sqrt{\frac{2 - \sqrt{2}}{2 + \sqrt{2}}} + \sqrt{\frac{2 + \sqrt{2}}{2 - \sqrt{2}}} \right)$$

$$= \frac{(2 - \sqrt{2}) + (2 + \sqrt{2})}{2 \times \sqrt{2}} = \sqrt{2}$$

**10.** Between the following two statements :

**Statement-I**: Let  $\vec{a} = \hat{i} + 2\hat{j} - 3\hat{k}$  and  $\vec{b} = 2\hat{i} + \hat{j} - \hat{k}$ . Then the vector  $\vec{r}$  satisfying  $\vec{a} \times \vec{r} = \vec{a} \times \vec{b}$  and  $\vec{a} \cdot \vec{r} = 0$  is of magnitude  $\sqrt{10}$ .

Statement-II: In a triangle ABC,  $\cos 2A + \cos 2B + \cos 2C \ge -\frac{3}{2}$ .

- (1) Both Statement-I and Statement-II are incorrect
- (2) Statement-I is incorrect but Statement-II is correct
- (3) Both Statement-I and Statement-II are correct
- (4) Statement-I is correct but Statement-II is incorrect

Ans. (2)

**Sol.** 
$$\overline{a} = \hat{i} + 2\hat{j} - 3\hat{k}$$

$$\overline{a} = 2\hat{i} + \hat{j} - \hat{k}$$

$$\overline{a} \times \overline{r} = \overline{a} \times \overline{b}$$
:  $\overline{a} \cdot \overline{r} = 0$ 

$$\Rightarrow \overline{a} \times (\overline{r} - \overline{b}) = \overline{0}$$

$$\Rightarrow \overline{a} = \lambda(\overline{r} - \overline{b})$$

$$\overline{a}.\overline{a} = \lambda(\overline{a}.\overline{r} - \overline{a}.\overline{b})$$

$$14 = -7\lambda \implies \lambda = -2$$

$$\frac{-\overline{a}}{2} = \overline{r} - \overline{b} \implies \overline{r} = \overline{b} - \frac{\overline{a}}{2}$$

$$=\frac{2\overline{b}-\overline{a}}{2}=\frac{3\hat{i}+\hat{k}}{2}$$

Statement (I) is incorrect

$$cos2A + cos 2B + cos 2c \ge -\frac{3}{2}$$

$$2A + 2B + 2C = 2\pi$$

$$\cos 2A + \cos 2B + \cos 2C$$

$$= -1 - 4 \cos A \cdot \cos B \cdot \cos C$$

$$\geq -1 - 4 \times \frac{1}{2} \times \frac{1}{2} \times \frac{1}{2}$$

$$=-\frac{3}{2}$$

Statement (II) is correct.

11. 
$$\lim_{x \to \frac{x}{2}} \left( \frac{\int_{x^3}^{(\pi/2)^3} (\sin(2t^{1/3}) + \cos(t^{1/3})) dt}{\left(x - \frac{\pi}{2}\right)^2} \right) \text{ is equal}$$

to:

$$(1)\;\frac{9\pi^2}{8}$$

(2) 
$$\frac{11\pi^2}{10}$$

$$(3) \ \frac{3\pi^2}{2}$$

(4) 
$$\frac{5\pi^2}{9}$$

Ans. (1)

**Sol.** 
$$\lim_{x \to \frac{\pi}{2}} \frac{0 - \left\{ \sin(2x) + \cos(x) \right\} . 3x^2}{2\left(x - \frac{\pi}{2}\right)}$$

$$= \lim_{x \to \frac{\pi}{2}} \frac{-\{2\sin x \cos x + \cos x\} 3x^2}{2\left(x - \frac{\pi}{2}\right)}$$

$$= \lim_{x \to \frac{\pi}{2}} \left\{ \frac{2\sin x \sin\left(\frac{\pi}{2} - x\right)}{2\left(x - \frac{\pi}{2}\right)} + \frac{\sin\left(\frac{\pi}{2} - x\right)}{2\left(\frac{\pi}{2} - x\right)} \right\} 3x^{2}$$

$$=\left(1(1)+\frac{1}{2}\right)3\left(\frac{\pi}{2}\right)^2$$

$$=\frac{9\pi^2}{8}$$

12. The sum of the coefficient of 
$$x^{2/3}$$
 and  $x^{-2/5}$  in the

binomial expansion of  $\left(x^{2/3} + \frac{1}{2}x^{-2/5}\right)^9$  is:

Ans. (1)

**Sol.** 
$$T_{r+1} = {}^{9}C_{r} \left( {}_{X}^{2/3} \right)^{9-r} \left( \frac{X^{-2/5}}{2} \right)^{r}$$

$$= {}^{9}C_{r} \left(\frac{1}{2}\right)^{r} \left(r\right)^{\left(6 - \frac{2r}{3} - \frac{2r}{5}\right)}$$

for coefficient of 
$$x^{2/3}$$
, put  $6 - \frac{2r}{3} - \frac{2r}{5} = \frac{2}{3}$ 

$$\Rightarrow$$
 r = 5

$$\therefore \text{ Coefficient of } x^{2/3} \text{ is } = {}^{9}C_{5} \left(\frac{1}{5}\right)^{5}$$

For coefficient of 
$$x^{-2/5}$$
, put  $6 - \frac{2r}{3} - \frac{2r}{5} = -\frac{2}{5}$ 

$$\Rightarrow$$
 r = 6

Coefficient of 
$$x^{-2/5}$$
 is  ${}^9C_6\left(\frac{1}{2}\right)^6$ 

Sum = 
$${}^{9}C_{5}\left(\frac{1}{2}\right)^{5} + {}^{9}C_{6}\left(\frac{1}{2}\right)^{6} = \frac{21}{4}$$

13. Let 
$$B = \begin{bmatrix} 1 & 3 \\ 1 & 5 \end{bmatrix}$$
 and A be a 2 × 2 matrix such that

$$AB^{-1} = A^{-1}$$
. If  $BCB^{-1} = A$  and  $C^4 + \alpha C^2 + \beta I = O$ ,

then  $2\beta - \alpha$  is equal to :

**Sol.** 
$$BCB^{-1} = A$$

$$\Rightarrow$$
 (BCB<sup>-1</sup>) (BCB<sup>-1</sup>) = A.A

$$\Rightarrow$$
 BCI CB<sup>-1</sup> = A<sup>2</sup>

$$\Rightarrow$$
 BC<sup>2</sup>B<sup>-1</sup> = A<sup>2</sup>

$$\Rightarrow$$
 B<sup>-1</sup>(BC<sup>2</sup>B<sup>-1</sup>)B = B<sup>-1</sup>(A.A)B

From equation (1)

$$C^2 = A^{-1}.A.B$$

$$C^2 = B$$

Also 
$$AB^{-1} = A^{-1}$$

$$\Rightarrow AB^{-1}.A = A^{-1}A = I$$

$$\Rightarrow A^{-1}(AB^{-1}A) = A^{-1}I$$

$$\mathbf{B}^{-1}\mathbf{A} = \mathbf{A}^{-1}$$

Now characteristics equation of C<sup>2</sup> is

$$|C_2-\lambda I|=0$$

$$|\mathbf{B} - \lambda \mathbf{I}| = 0$$

$$\Rightarrow \begin{vmatrix} 1 - \lambda & 3 \\ 1 & 5 - \lambda \end{vmatrix} = 0$$

$$\Rightarrow (1 - \lambda) (5 - 1) - 3 = 0 \Rightarrow (\lambda^2 - 6\lambda + 5) - 3 = 0$$

$$\Rightarrow \lambda^2 - 6\lambda + 2 = 0$$

$$\Rightarrow \beta^2 - 6B + 2I = 0$$

$$\Rightarrow$$
 C<sup>4</sup> - 6C<sup>2</sup> + 2I = 0

$$\alpha = -6$$

$$\beta = 2$$

$$\therefore 2\beta - \alpha = 4 + 6 = 10$$

14. If  $\log_e y = 3 \sin^{-1} x$ , then  $(1 - x)^2 y'' - xy'$  at  $x = \frac{1}{2}$  is equal to :

(1) 
$$9e^{\pi/6}$$

(2) 
$$3e^{\pi/6}$$

(3) 
$$3e^{\pi/2}$$

(4) 
$$9e^{\pi/2}$$

Ans. (4)

**Sol.** 
$$\ln(y) = 3\sin^{-1} x$$

$$\frac{1}{y} \cdot y' = 3 \left( \frac{1}{\sqrt{1 - x^2}} \right)$$

$$\Rightarrow$$
 y' =  $\frac{3y}{\sqrt{1-x^2}}$  at  $x = \frac{1}{2}$ 

$$\Rightarrow y' = \frac{3e^{3\left(\frac{\pi}{6}\right)}}{\frac{\sqrt{3}}{2}} = 2\sqrt{3}e^{\frac{\pi}{2}}$$

$$\Rightarrow y'' = 3 \left( \frac{\sqrt{1 - x^2}y' - y\frac{1}{2\sqrt{1 - x^2}}(-2x)}{(1 - x^2)} \right)$$

$$\Rightarrow (1-x^2)y'' = 3\left(3y + \frac{xy}{\sqrt{1-x^2}}\right)$$

$$\downarrow$$
 at  $x = \frac{1}{2}$ ,  $y = e^{3\sin^{-1}\left(\frac{1}{2}\right)} = e^{3\left(\frac{\pi}{6}\right)} = e^{\frac{\pi}{2}}$ 

$$(1-x^{2})y"\Big|_{at x=\frac{1}{2}} = 3\left(3e^{\frac{\pi}{2}} + \frac{\frac{1}{2}\left(e^{\frac{\pi}{2}}\right)}{\frac{\sqrt{3}}{2}}\right)$$

$$= 3e^{\frac{\pi}{2}}\left(3 + \frac{1}{\sqrt{3}}\right)$$

$$(1-x^{2})y"-xy'\Big|_{at x=\frac{1}{2}}$$

$$= 3e^{\frac{\pi}{2}}\left(3 + \frac{1}{\sqrt{3}}\right) - \frac{1}{2}\left(2\sqrt{3}e^{\frac{\pi}{2}}\right) = 9e^{\frac{\pi}{2}}$$

15. The integral 
$$\int_{1/4}^{3/4} \cos \left( 2 \cot^{-1} \sqrt{\frac{1-x}{1+x}} \right) dx$$
 is equal

to:

$$(1)-1/2$$

$$(4) - 1/4$$

Ans. (4)

**Sol.** 
$$I = \int_{1/4}^{3/4} \cos \left( 2 \cot^{-1} \left( \sqrt{\frac{1-x}{1+x}} \right) dx \right)$$

$$\int_{1/4}^{3/4} \cos\left(2\left(\tan^{-1}\sqrt{\frac{1+x}{1+x}}\right)\right) dx$$

$$\int_{1/4}^{3/4} \frac{1 - \tan^2\left(\tan^{-1}\sqrt{\frac{1+x}{1-x}}\right)}{1 + \tan^2\left(\tan^{-1}\sqrt{\frac{1+x}{1-x}}\right)} dx$$

$$= \int_{1/4}^{3/4} \frac{1 - \left(\frac{1+x}{1-x}\right)}{1 + \left(\frac{1+x}{1-x}\right)} dx = \int_{1/4}^{3/4} \frac{-2x}{2} dx$$

$$= \int_{1/4}^{3/4} (-x) dx = -\left(\frac{x^2}{2}\right)_{1/4}^{3/4}$$
$$= -\frac{1}{2} \left[\frac{9}{16} - \frac{1}{16}\right]$$

$$=-\frac{1}{4}$$

16. Let a, ar,  $ar^2$ , ......be an infinite G.P. If  $\sum_{n=0}^{\infty} ar^n = 57 \text{ and } \sum_{n=0}^{\infty} a^3 r^{3n} = 9747, \text{ then a} + 18r \text{ is equal to :}$ 

Ans. (4)

Sol. 
$$\sum_{n=0}^{\infty} ar^n = 57$$

$$a + ar + ar^2 + \infty = 57$$

$$\frac{a}{1-r} = 57$$
 .....(I)

$$\sum_{n=0}^{\infty}a^{3}r^{3n}=9747$$

$$a^3 + a^3 \cdot r^3 + a^3 \cdot r^6 + \dots = 9746$$

$$\frac{a^3}{1-r^3} = 9746 \dots (II)$$

$$\frac{\text{(I)}^3}{\text{(II)}} \Rightarrow \frac{\frac{\text{a}^3}{\text{(1-r)}^3}}{\frac{\text{a}^3}{1-\text{r}^3}} = \frac{57^3}{9717} = 19$$

On solving,  $r = \frac{2}{3}$  and  $r = \frac{3}{2}$  (rejected)

$$a = 19$$

$$\therefore a + 18r = 19 + 18 \times \frac{2}{3} = 31$$

17. If an unbiased dice is rolled thrice, then the probability of getting a greater number in the  $i^{th}$  roll than the number obtained in the  $(i-1)^{th}$  roll, i=2,3, is equal to:

Ans. (3)

**Sol.** Favourable cases =  ${}^{6}C_{3}$ 

Total out comes =  $6^3$ 

Probability of getting greater number than previous

one = 
$$\frac{{}^{6}\text{C}_{3}}{\mathrm{r}^{3}} = \frac{20}{216} = \frac{5}{54}$$

18. The value of the integral  $\int_{-1}^{2} \log_{e} \left( x + \sqrt{x^{2} + 1} \right) dx$ 

(1) 
$$\sqrt{5} - \sqrt{2} + \log_e \left( \frac{9 + 4\sqrt{5}}{1 + \sqrt{2}} \right)$$

(2) 
$$\sqrt{2} - \sqrt{5} + \log_e \left( \frac{9 + 4\sqrt{5}}{1 + \sqrt{2}} \right)$$

(3) 
$$\sqrt{5} - \sqrt{2} + \log_e \left( \frac{7 + 4\sqrt{5}}{1 + \sqrt{2}} \right)$$

(4) 
$$\sqrt{2} - \sqrt{5} + \log_e \left( \frac{7 + 4\sqrt{5}}{1 + \sqrt{2}} \right)$$

Ans. (2)

**Sol.** 
$$I = \int_{-1}^{2} 1.\log_{e} \left( x + \sqrt{x^2 + 1} \right) dx$$

$$= x \log_e \left( x + \sqrt{x^2 + 1} \right) - \int_{-1}^{2} \left( \frac{1 + \frac{x}{\sqrt{x^2 + 1}}}{x + \sqrt{x^2 + 1}} \right) dx$$

$$= x \log_e \left( x + \sqrt{x^2 + 1} \right) - \int_{-1}^{2} \frac{x}{\sqrt{x^2 + 1}} dx$$

$$= x \log_e \left( x + \sqrt{x^2 + 1} \right) - \sqrt{x^2 + 1} \Big|_{x=0}^{2}$$

$$= \left(2\log_{e}\left(2+\sqrt{5}\right)-\sqrt{5}\right)$$

$$-\left(-\log_{\mathrm{e}}\left(-1+\sqrt{2}\right)-\sqrt{2}\right)$$

$$= \log_{e} \left( 2 + \sqrt{5} \right)^{2} - \sqrt{5} + \log_{e} \left( \sqrt{2} - 1 \right) + \sqrt{2}$$

$$= \log_{e} \left(2 + \sqrt{5}\right)^{2} - \sqrt{5} + \log_{e} \left(\sqrt{2} - 1\right) + \sqrt{2}$$

$$= \sqrt{2} - \sqrt{5} + \log_{e} \left( \frac{\left(2 + \sqrt{5}\right)^{2}}{\sqrt{2 + 1}} \right)$$
$$= \sqrt{2} - \sqrt{5} + \log_{e} \left( \frac{9 + 4\sqrt{5}}{\sqrt{2 + 1}} \right)$$

- 19. Let  $\alpha$ ,  $\beta$ ;  $\alpha > \beta$ , be the roots of the equation  $x^2 \sqrt{2}x \sqrt{3} = 0.$  Let  $P_n = \alpha^n \beta^n$ ,  $n \in \mathbb{N}$ . Then  $\left(11\sqrt{3} 10\sqrt{2}\right) P_{10} + \left(11\sqrt{2} + 10\right) P_{11} 11P_{12} \text{ is equal to :}$ 
  - (1)  $10\sqrt{2}P_9$
  - (2)  $10\sqrt{3}P_9$
  - (3)  $11\sqrt{2}P_{0}$
  - (4)  $11\sqrt{3}P_{0}$

Ans. (2)

$$\begin{aligned} & \textbf{Sol.} \quad x^2 - \sqrt{2x} - \sqrt{3} = 0 \ \Big\langle_{\beta}^{\alpha} \\ & \alpha^{n+2} - \sqrt{2}\alpha^{n+1} - \sqrt{3}\alpha^n = 0 \\ & \text{and } \beta^{n+2} - \sqrt{2}\beta^{n+1} - \sqrt{3}\beta^n = 0 \\ & \text{Subtracting} \\ & \left(\alpha^{n+2} - \beta^{n+2}\right) - \sqrt{2}\left(\alpha^{n+1} - \beta^{n+1}\right) - \sqrt{3}\left(\alpha^n - \beta^n\right) = 0 \\ & \Rightarrow P_{n+2} - \sqrt{2}P_{n+1} - \sqrt{3}P_n = 0 \\ & Put \ n = 10 \\ & P_{12} - \sqrt{2}P_{11} - \sqrt{3}P_{10} = 0 \\ & n = 9 \end{aligned}$$

 $11(\sqrt{3}.P_{10} + \sqrt{2}P_{11} - P_{11}) - 10(\sqrt{2}P_{10} - P_{11})$ 

 $P_{11} - \sqrt{2}P_{10} - \sqrt{3}P_{0} = 0$ 

 $=0-10(-\sqrt{3}P_9)=10\sqrt{3}P_9$ 

20. Let  $\vec{a} = 2\hat{i} + \alpha\hat{j} + \hat{k}$ ,  $\vec{b} = -\hat{i} + \hat{k}$ ,  $\vec{c} = \beta\hat{j} - \hat{k}$ , where  $\alpha$  and  $\beta$  are integers and  $\alpha\beta = -6$ . Let the values of the ordered pair  $(\alpha, \beta)$  for which the area of the parallelogram of diagonals  $\vec{a} + \vec{b}$  and  $\vec{b} + \vec{c}$  is  $\frac{\sqrt{21}}{2}$ , be  $(\alpha_1, \beta_1)$  and  $(\alpha_2, \beta_2)$ .

Then  $\alpha_1^2 + \beta_1^2 - \alpha_2 \beta_2$  is equal to

- (1) 17
- (2) 24

- (3) 21
- (4) 19

Ans. (4)

**Sol.** Area of parallelogram =  $\frac{1}{2} | \vec{d}_1 \times \vec{d}_2$ 

$$A = \frac{1}{2} |(\vec{a} + \vec{b}) \times (\vec{b} + \vec{c})| = \frac{\sqrt{21}}{2}$$

so, 
$$\vec{a} + \vec{b} = \hat{i} + \alpha \hat{j} + 2\hat{k}$$

$$\vec{b} + \vec{c} = -\hat{i} + \beta\hat{j}$$

$$(\vec{a} + \vec{b}) \times (\vec{b} + \vec{c}) = \begin{vmatrix} \hat{i} & \hat{j} & \hat{k} \\ 1 & \alpha & 2 \\ -1 & \beta & 0 \end{vmatrix}$$

$$= \hat{\mathbf{i}}(-2\beta) - \hat{\mathbf{j}}(2) + \hat{\mathbf{k}}(\beta + \alpha)$$

$$|(\vec{a} + \vec{b}) \times (\vec{b} + \vec{c})| = \sqrt{4\beta^2 + 4 + (\alpha + \beta)^2} = \sqrt{21}$$

$$4\beta^2 + 4 + \alpha^2 + \beta^2 + 2\alpha\beta = 21$$

$$\alpha^2 + 5\beta^2 - 12 = 17$$

$$\alpha^2 + 5\beta^2 = 29$$

and 
$$\alpha\beta = -6$$

and given  $\alpha_i\beta$  are integers

so,

$$\alpha = -3, \beta = 2$$

or

$$\alpha = 3$$
,  $\beta = -2$ 

$$(\alpha_1,\beta_1)=(-3,2)$$

$$(\alpha_2,\beta_2)=(3,-2)$$

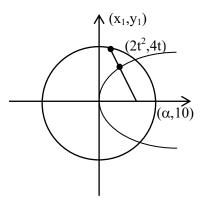
$$\alpha_1^2 + \beta_1^2 - \alpha_2 \beta_2 = 9 + 4 + 6 = 19$$

## **SECTION-B**

21. Consider the circle  $C: x^2 + y^2 = 4$  and the parabola  $P: y^2 = 8x$ . If the set of all values of  $\alpha$ , for which three chords of the circle C on three distinct lines passing through the point  $(\alpha, 0)$  are bisected by the parabola P is the interval (p, q), then  $(2q - p)^2$  is equal to \_\_\_\_\_.

Ans. (80)

Sol.



$$T = S_1$$

$$xx_1 + yy_1 = x_1^2 + y_1^2$$

$$\alpha x_1 = x_1^2 + y_1^2$$

$$\alpha(2t^2) = 4t^4 + 16t^2$$

$$\alpha = 2t^2 + 8$$

$$\frac{\alpha - 8}{2} = t^2$$

Also, 
$$4t^4 + 16t^2 - 4 < 0$$

$$t^2 = -2 + \sqrt{5}$$

$$\alpha = 4 + 2\sqrt{5}$$

$$\alpha \in (8, 4 + 2\sqrt{5})$$

$$(2q - p)^2 = 80$$

22. Let the set of all values of p, for which  $f(x) = (p^2 - 6p + 8) (\sin^2 2x - \cos^2 2x) + 2(2 - p)x + 7$  does not have any critical point, be the interval (a, b). Then 16ab is equal to \_\_\_\_\_.

Ans. (252)

Sol. 
$$f(x) = -(p^2 - 6p + 8) \cos 4n + 2(2-p)n + 7$$
  
 $f^{1}(x) = +4(p^2 - 6p + 8) \sin 4x + (4-2p) \neq 0$   
 $\sin 4x \neq \frac{2p - 4}{4(p - 4)(p - 2)}$ 

$$\sin 4x \neq \frac{2(p-2)}{4(p-4)(p-2)}$$

$$p \neq 2$$

$$\sin 4x \neq \frac{1}{2(p-4)}$$

$$\Rightarrow \left| \frac{1}{2(p-4)} \right| > 1$$

on solving we get

$$\therefore p \in \left(\frac{7}{2}, \frac{9}{2}\right)$$

Hence 
$$a = \frac{7}{2}, b = \frac{9}{2}$$

$$\therefore 16ab = 252$$

**23.** For a differentiable function  $f: IR \rightarrow IR$ , suppose

$$f'(x) = 3f(x) + \alpha$$
, where  $\alpha \in IR$ ,  $f(0) = 1$  and

$$\lim_{x \to -\infty} f(x) = 7. \text{ Then 9f } (-\log_e 3) \text{ is equal to}$$

Ans. (61)

**Sol.** 
$$\frac{dy}{dx} - 3y = \alpha$$

$$If = e^{\int -3 dx} = e^{-3x}$$

$$\therefore y - e^{-3x} = \int e^{-3x} \cdot \alpha \, dx$$

$$y e^{-3x} = \frac{\alpha e^{-3x}}{-3} + c$$

$$(*e^{3x})$$

$$y = \frac{\alpha}{-3} + C \cdot e^{3x}$$

on substituting x = 0, y = 1

$$x \rightarrow -\infty$$
,  $y = 7$ 

we get 
$$y = 7 - 6e^{3x}$$

$$9f(-\log_e 3) = 61$$

The number of integers, between 100 and 1000 24. having the sum of their digits equals to 14, is

Ans. (70)

**Sol.** N = abc

All distinct digits (i) a + b + c = 14 $a \ge 1$  $b, c \in \{0 \text{ to } 9\}$ 

by hit & trial:

8 cases

- (6, 5, 3)
- (8, 6, 0)
- (9, 4, 1)
- (7, 6, 1)
- (8, 5, 1)
- (9, 3, 2)
- (7, 5, 2)(8, 4, 2)
- (7, 4, 3)(9, 5, 0)
- 2 same, 1 diff (ii) a = b : c2a + c = 14by values:
  - (3,8)(4,6)(7,0)
  - = 14 cases
- (iii) all same: 3a = 14 $a = \frac{14}{3} \times rejected$

0 cases

Hence, Total cases:

$$8 \times 3! + 2 \times (4) + 14$$
  
=  $48 + 22$   
=  $70$ 

25. Let  $A = \{(x, y) : 2x + 3y = 23, x, y \in N\}$  and  $B = \{x : (x, y) \in A\}$ . Then the number of one-one functions from A to B is equal to \_\_\_\_\_.

Ans. (24)

**Sol.** 
$$2x + 3y = 23$$

$$x = 1$$
  $y = 7$ 

$$x = 4$$
  $y = 5$ 

$$x = 7 \qquad y = 3$$

$$x = 10 \qquad y = 1$$

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(1, 7)1

(4, 5)4

7 (7, 3)

(10, 1)10

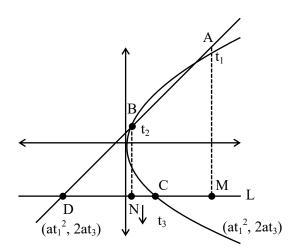
The number of one-one functions from A to B is equal to 4!

Let A, B and C be three points on the parabola **26.**  $y^2 = 6x$  and let the line segment AB meet the line L through C parallel to the x-axis at the point D. Let M and N respectively be the feet of the perpendiculars from A and B on L.

Then 
$$\left(\frac{AM \cdot BN}{CD}\right)^2$$
 is equal to \_\_\_\_\_.

Ans. (36)

Sol.



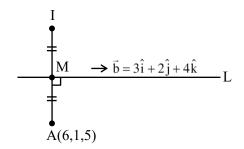
Sol.

$$\begin{split} m_{AB} &= m_{AD} \\ \Rightarrow & \frac{2}{t_1 + t_2} = \frac{2a(t_1 - t_3)}{at_1^2 - \alpha} \\ \Rightarrow & at_1^2 - \alpha = a\{t_1^2 - t_1t_3 + t_1t_2 - t_2t_3\} \\ \Rightarrow & \alpha = a(t_1t_3 + t_2t_3 - t_1t_2) \\ AM &= \left| 2a(t_1 - t_3) \right|, \ BN = \left| 2a(t_2 - t_3) \right|, \\ CD &= \left| at_3^2 - \alpha \right| \end{split}$$

$$\begin{split} CD &= \left| at_3^2 - a(t_1t_3 + t_2t_3 - t_1t_2) \right| \\ &= a \left| t_3^2 - t_1t_3 - t_2t_3 + t_1t_2 \right| \\ &= a \left| t_3(t_3 - t_1) - t_2(t_3 - t_1) \right| \\ CD &= a \left| (t_3 - t_2)(t_3 - t_1) \right| \\ \left( \frac{AM \cdot BN}{CD} \right)^2 &= \left\{ \frac{2a(t_1 - t_3) \cdot 2a(t_2 - t_3)}{a(t_3 - t_2)(t_3 - t_1)} \right\}^2 \\ 16a^2 &= 16 \times \frac{9}{4} = 36 \end{split}$$

27. The square of the distance of the image of the point (6, 1, 5) in the line  $\frac{x-1}{3} = \frac{y}{2} = \frac{z-2}{4}$ , from the origin is \_\_\_\_\_.

Ans. (62)



Sol.

Let 
$$M(3\lambda + 1, 2\lambda, 4\lambda + 2)$$
  
 $\overrightarrow{AM} \cdot \overrightarrow{b} = 0$   
 $\Rightarrow 9\lambda - 15 + 4\lambda - 2 + 16\lambda - 12 = 0$   
 $\Rightarrow 29\lambda = 29$   
 $\Rightarrow \lambda = 1$   
 $M(4, 2, 6), I = (2, 3, 7)$   
Required Distance  $= \sqrt{4 + 9 + 49} = \sqrt{62}$   
Ans. 62

28. If 
$$\left(\frac{1}{\alpha+1} + \frac{1}{\alpha+2} + \dots + \frac{1}{\alpha+1012}\right)$$
  
 $-\left(\frac{1}{2\cdot 1} + \frac{1}{4\cdot 3} + \frac{1}{6\cdot 5} + \dots + \frac{1}{2024\cdot 2023}\right)$   
 $= \frac{1}{2024}$ , then  $\alpha$  is equal to-  
Ans. (1011)

Sol. 
$$\left( \frac{1}{\alpha+1} + \frac{1}{\alpha+2} + \dots + \frac{1}{\alpha+2012} \right)$$

$$-\left\{ \left( \frac{1}{1} - \frac{1}{2} \right) + \left( \frac{1}{3} - \frac{1}{4} \right) + \dots + \left( \frac{1}{2023} - \frac{1}{2024} \right) \right\} = \frac{1}{2024}$$

$$\Rightarrow \left( \frac{1}{\alpha+1} + \frac{1}{\alpha+2} + \dots + \frac{1}{\alpha+2012} \right)$$

$$-\left\{ \left( \frac{1}{1} + \frac{1}{2} + \frac{1}{3} + \frac{1}{4} \right) + \dots + \frac{1}{2023} \right\}$$

$$-\frac{1}{2024} - 2\left( \frac{1}{2} + \frac{1}{4} + \dots + \frac{1}{2022} \right) \right\} = \frac{1}{2024}$$

$$\Rightarrow \left( \frac{1}{\alpha+1} + \frac{1}{\alpha+2} + \dots + \frac{1}{\alpha+2012} \right)$$

$$-\left( \frac{1}{1} + \frac{1}{2} + \dots + \frac{1}{2023} \right)$$

$$+\frac{1}{2024} + \left( \frac{1}{1} + \frac{1}{2} + \dots + \frac{1}{1011} \right) = \frac{1}{2024}$$

$$\Rightarrow \frac{1}{\alpha+1} + \frac{1}{\alpha+2} + \dots + \frac{1}{\alpha+2012}$$

$$= \frac{1}{1012} + \frac{1}{1013} + \dots + \frac{1}{2023}$$

$$\Rightarrow \alpha = 1011$$

29. Let the inverse trigonometric functions take principal values. The number of real solutions of the equation  $2 \sin^{-1} x + 3 \cos^{-1} x = \frac{2\pi}{5}$ , is \_\_\_\_\_.

Ans. (0)

Sol. 
$$2\sin^{-1} x + 3\cos^{-1} x = \frac{2\pi}{5}$$
  

$$\Rightarrow \quad \pi + \cos^{-1} x = \frac{2\pi}{5}$$

$$\Rightarrow \quad \cos^{-1} x = \frac{-3\pi}{5}$$
Not possible
Ans. 0

- **30.** Consider the matrices :  $A = \begin{bmatrix} 2 & -5 \\ 3 & m \end{bmatrix}$ ,  $B = \begin{bmatrix} 20 \\ m \end{bmatrix}$ 
  - and  $X = \begin{bmatrix} x \\ y \end{bmatrix}$ . Let the set of all m, for which the
  - system of equations AX = B has a negative solution (i.e., x < 0 and y < 0), be the interval (a, b).
  - Then  $8\int_{a}^{b} |A| dm$  is equal to \_\_\_\_\_.
  - Ans. (450)
- **Sol.**  $A = \begin{pmatrix} 2 & -5 \\ 3 & m \end{pmatrix}, B = \begin{pmatrix} 20 \\ m \end{pmatrix}$

$$X = \begin{pmatrix} x \\ y \end{pmatrix}$$

$$2x - 5y = 20$$
 ...(1

$$3x + my = m \qquad \dots (2)$$

$$\Rightarrow y = \frac{2m - 60}{2m + 15}$$

$$y < 0 \Rightarrow m \in \left(\frac{-15}{2}, 30\right)$$

$$x = \frac{25m}{2m + 15}$$

$$x < 0 \Rightarrow m \in \left(\frac{-15}{2}, 0\right)$$

$$\Rightarrow$$
 m  $\in \left(\frac{-15}{2}, 0\right)$ 

$$|A| = 2m + 15$$

Now,

$$8\int_{\frac{-15}{2}}^{0} (2m+15) dm = 8\left\{m^2 + 15m\right\}_{\frac{-15}{2}}^{0}$$

$$\Rightarrow 8\left\{-\left(\frac{225}{4} - \frac{225}{2}\right)\right\}$$

$$=8 \times \frac{225}{4} = 450$$

#### **SECTION-A**

- **31.** A nucleus at rest disintegrates into two smaller nuclei with their masses in the ratio of 2:1. After disintegration they will move:-
  - (1) In opposite directions with speed in the ratio of 1:2 respectively
  - (2) In opposite directions with speed in the ratio of 2:1 respectively
  - (3) In the same direction with same speed.
  - (4) In opposite directions with the same speed.

Ans. (1)

**Sol.** By conservation of momentum

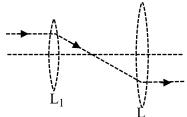
$$p_i = p_f$$

$$O = m_1 u_{1+} m_2 u_2$$

$$\frac{u_1}{u_2} = -\left[\frac{1}{2}\right] \text{ as } \frac{m_1}{m_2} = \frac{2}{1}$$

move in opposite direction with speed ratio 1:2

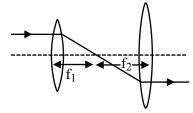
32. The following figure represents two biconvex lenses  $L_1$  and  $L_2$  having focal length 10 cm and 15 cm respectively. The distance between  $L_1$  &  $L_2$  is:



- (1) 10 cm
- (2) 15 cm
- (3) 25 cm
- (4) 35 cm

Ans. (3)

Sol.



$$D = f_1 + f_2 = 25 \text{ cm}$$

Paraxial parallel rays pass through focus and ray from focus of convex lens will become parallel **33.** The temperature of a gas is  $-78^{\circ}$  C and the average translational kinetic energy of its molecules is K. The temperature at which the average translational kinetic energy of the molecules of the same gas becomes 2K is:

- $(1) 39^{\circ}C$
- (2) 117°C
- (3) 127°C
- $(4) 78^{\circ}C$

Ans. (2)

**Sol.** K.E =  $\frac{nf_1RT}{2}$ 

 $T_i = -78^{\circ}C \rightarrow 273 + [-78^{\circ}C] = 195K$ 

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To double the K.E energy temp also

become double

 $T_f = 390 \text{ K}$ 

 $T_{\rm f} = 117^{\circ} C$ 

**34.** A hydrogen atom in ground state is given an energy of 10.2 eV. How many spectral lines will be emitted due to transition of electrons?

(1)6

- (2) 3
- (3) 10
- (4) 1

Ans. (4)

**Sol.** Hydrogen will be in first excited state therefore it will emit one spectral line corresponding to transition b/w energy level 2 to 1

35. The magnetic field in a plane electromagnetic wave is  $B_y = (3.5 \times 10^{-7}) \sin (1.5 \times 10^3 x + 0.5 \times 10^{11} t)T$ . The corresponding electric field will be

- (1)  $E_v = 1.17 \sin (1.5 \times 10^3 x + 0.5 \times 10^{11} t) \text{Vm}^{-1}$
- (2)  $E_z = 105 \sin (1.5 \times 10^3 x + 0.5 \times 10^{11} t) \text{Vm}^{-1}$
- (3)  $E_z = 1.17 \sin (1.5 \times 10^3 x + 0.5 \times 10^{11} t) \text{Vm}^{-1}$
- (4)  $E_v = 10.5 \sin(1.5 \times 10^3 x + 0.5 \times 10^{11} t) \text{Vm}^{-1}$

Ans. (2)

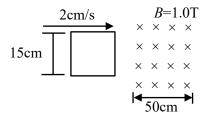
**Sol.**  $E_0 = B_0C$ 

$$E_0 = 3 \times 10^8 \times (3.5 \times 10^{-7}) \sin(1.5 \times 10^3 x + 0.5 \times 10^{11} t)$$

$$E_0 = 105 \sin (1.5 \times 10^3 x + 0.5 \times 10^{11} t) Vm^{-1}$$

Data inconsistent while calculating speed of wave. You can challenge for data.

36. A square loop of side 15 cm being moved towards right at a constant speed of 2 cm/s as shown in figure. The front edge enters the 50 cm wide magnetic field at t = 0. The value of induced emf in the loop at t = 10 s will be:



- (1) 0.3 mV
- (2) 4.5 mV
- (3) zero
- (4) 3 mV

- Ans. (3)
- **Sol.** At t = 10 sec complete loop is in magnetic field therefore no change in flux

$$B\ell \overline{\overline{V}}$$
  $B\ell V$ 

$$e = \frac{d\phi}{dt} = 0$$

e = 0 for complete loop

- 37. Two cars are travelling towards each other at speed of 20 m s<sup> $^{-1}$ </sup> each. When the cars are 300 m apart, both the drivers apply brakes and the cars retard at the rate of 2 m s<sup> $^{-2}$ </sup>. The distance between them when they come to rest is:
  - (1) 200 m
- (2) 50 m
- (3) 100 m
- (4) 25 m

#### Ans. (3)

$$\left| \vec{\mathbf{u}}_{\mathrm{BA}} \right| = 40 \; \mathrm{m/s}$$

$$\left| \vec{a}_{BA} \right| = 4 \text{ m/s}$$

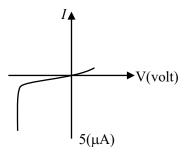
Apply 
$$(v^2 = u^2 + 2as)_{relative}$$

$$O = (40)^2 + 2(-4)(S)$$

$$S = 200 \text{ m}$$

Remaining distance = 300 - 200 = 100 m

**38.** The *I-V* characteristics of an electronic device shown in the figure. The device is :



- (1) a solar cell
- (2) a transistor which can be used as an amplifier
- (3) a zener diode which can be used as voltage regulator
- (4) a diode which can be used as a rectifier

#### Ans. (3)

**Sol.** Theory

Zener diode used as voltage regulator

- 39. The excess pressure inside a soap bubble is thrice the excess pressure inside a second soap bubble. The ratio between the volume of the first and the second bubble is:
  - (1)1:9
- (2) 1:3
- (3) 1:81
- (4) 1: 27

#### Ans. (4)

Sol.





$$P_1 - P_0 = \frac{4T}{r_1}$$

$$P_2 - P_0 = \frac{4T}{r_2}$$

$$P_1 - P_0 = 3(P_2 - P_0)$$

$$\frac{4T}{r_1} = 3\frac{4T}{r_2}$$

$$r_2 = 3r_1$$

$$\frac{V_1}{V_2} = \frac{\frac{4}{3}\pi r_1^3}{\frac{4}{3}\pi r_2^3} = \frac{1}{27}$$

- **40.** The de-Broglie wavelength associated with a particle of mass m and energy E is  $h/\sqrt{2mE}$ . The dimensional formula for Planck's constant is:
  - $(1) [ML^{-1}T^{-2}]$
- $(2) [ML^2T^{-1}]$
- $(3) [MLT^{-2}]$
- (4)  $[M^2L^2T^{-2}]$

Ans. (2)

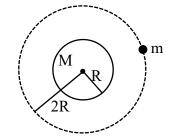
- Sol.  $\lambda = \frac{h}{\sqrt{2mE}}$  or E = hv $[ML^2T^{-2}] = h[T^{-1}]$   $h = [ML^2T^{-1}]$
- 41. A satellite of  $10^3$  kg mass is revolving in circular orbit of radius 2R. If  $\frac{10^4 \text{R}}{6}J$  energy is supplied to the satellite, it would revolve in a new circular orbit of radius:

(use 
$$g = 10 \text{m/s}^2$$
,  $R = \text{radius of earth}$ )

- (1) 2.5 R
- (2) 3 R
- (3) 4 R
- (4) 6 R

Ans. (4)

Sol.



Total energy = 
$$\frac{-GMm}{2(2R)}$$

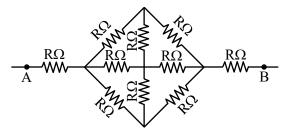
if energy = 
$$\frac{10^4 \text{R}}{6}$$
 is added then

$$\frac{-GMm}{4R} + \frac{10^4 R}{6} = \frac{-GMm}{2r}$$

where r is new radius of revolving and  $g = \frac{GM}{R^2}$ 

$$-\frac{mgR}{4} + \frac{10^4 R}{6} = -\frac{mgR^2}{2r} \quad (m = 10^3 \text{ kg})$$
$$-\frac{10^3 \times 10 \times R}{4} + \frac{10^4 R}{6} = -\frac{10^3 \times 10 \times R^2}{2r}$$
$$-\frac{1}{4} + \frac{1}{6} = -\frac{R}{2r}$$
$$r = 6R$$

**42.** The effective resistance between A and B, if resistance of each resistor is R, will be

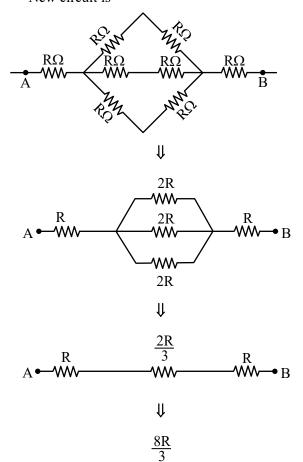


- (1)  $\frac{2}{3}$  R
- (2)  $\frac{8R}{3}$
- (3)  $\frac{5R}{3}$
- (4)  $\frac{4R}{3}$

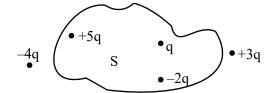
Ans. (2)

**Sol.** From symmetry we can remove two middle resistance.

New circuit is



**43.** Five charges +q, +5q, -2q, +3q and -4q are situated as shown in the figure. The electric flux due to this configuration through the surface S is:



- $(1) \frac{5q}{\epsilon_0}$
- $(2) \frac{4q}{\epsilon_0}$
- $(3) \ \frac{3q}{\epsilon_0}$
- $(4) \frac{q}{\epsilon_0}$

Ans. (2)

Sol. As per gauss theorem,

$$\phi = \frac{q_{in}}{\epsilon_0} = \frac{q + (-2q) + 5q}{\epsilon_0}$$

 $\frac{4q}{\epsilon_0}$ 

- 44. A proton and a deutron (q= +e, m = 2.0u) having same kinetic energies enter a region of uniform magnetic field  $\vec{B}$ , moving perpendicular to  $\vec{B}$ . The ratio of the radius  $r_d$  of deutron path to the radius  $r_p$  of the proton path is:
  - (1) 1 : 1
- (2)  $1:\sqrt{2}$
- $(3)\sqrt{2}:1$
- (4) 1:2

Ans. (3)

Sol. In uniform magnetic field,

$$R = \frac{mv}{qB} = \frac{\sqrt{2m(K.E)}}{qB}$$

Since same K.E

$$R \propto \frac{\sqrt{m}}{q}$$

$$\therefore \frac{R_{deutron}}{R_{proton}} = \sqrt{\frac{m_d}{m_p}} \times \frac{q_p}{q_d}$$

$$=\sqrt{2}\times 1$$

$$\therefore \, \gamma_{\rm d} : \gamma_{\rm p} = \sqrt{2} : 1$$

- **45.** UV light of 4.13 eV is incident on a photosensitive metal surface having work function 3.13 eV. The maximum kinetic energy of ejected photoelectrons will be:
  - (1) 4.13 eV
- (2) 1 eV
- (3) 3.13 eV
- (4) 7.26 eV

Ans. (2)

**Sol.**  $E_{photon} = (work function) + K.E_{max}$ 

$$\therefore 4.13 = 3.13 + K.E_{max}$$

$$\therefore K.E_{max} = 1 \text{ eV}$$

**46.** The energy released in the fusion of 2 kg of hydrogen deep in the sun is  $E_H$  and the energy released in the fission of 2 kg of  $^{235}U$  is  $E_{U}$ . The

ratio 
$$\frac{E_{\rm H}}{E_{\rm U}}$$
 is approximately :

(Consider the fusion reaction as  $4_1^1 \text{H} + 2\text{e}^- \rightarrow_2^4 \text{He} + 2\text{v} + 6\gamma + 26.7 \text{ MeV}$ , energy

released in the fission reaction of  $^{235}$ U is 200 MeV per fission nucleus and  $N_A = 6.023 \times 10^{23}$ )

- (1) 9.13
- (2) 15.04
- (3) 7.62
- (4) 25.6

Ans. (3)

**Sol.** In each fusion reaction,  $4 {}_{1}^{1}$ H nucleus are used.

Energy released per Nuclei of  ${}_{1}^{1}H = \frac{26.7}{4} \text{MeV}$ 

∴ Energy released by 2 kg hydrogen (E<sub>H</sub>)

$$= \frac{2000}{1} \times N_A \times \frac{26.7}{4} \text{ MeV}$$

∴ Energy released by 2 kg Vranium (E<sub>v</sub>)

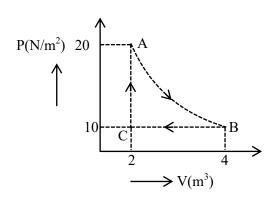
$$= \frac{2000}{235} \times N_A \times 200 MeV$$

So,

$$\frac{E_H}{E_V} = 235 \times \frac{26.7}{4 \times 200} = 7.84$$

:. Approximately close to 7.62

47. A real gas within a closed chamber at 27°C undergoes the cyclic process as shown in figure.
The gas obeys PV<sup>3</sup> = RT equation for the path A to B. The net work done in the complete cycle is (assuming R = 8J/molK):



- (1) 225 J
- (2) 205 J
- $(3) \ 20 J$
- (4) -20 J

Ans. (2)

Sol.  $W_{AB} = \int PdV$  (Assuming T to be constant)  $= \int \frac{RTdV}{V^3}$   $= RT \int_2^4 V^{-3} dV$ 

$$= 8 \times 300 \times \left( -\frac{1}{2} \left[ \frac{1}{4^2} - \frac{1}{2^2} \right] \right)$$

= 225 J

$$W_{BC} = P \int_{4}^{2} dV = 10(2-4) = -20J$$

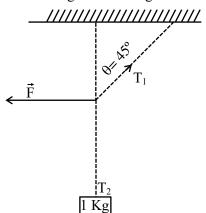
 $W_{_{CA}} = 0$ 

$$\therefore W_{cycle} = 205 J$$

Note: Data is inconsistent in process AB.

So needs to be challenged.

48. A 1 kg mass is suspended from the ceiling by a rope of length 4m. A horizontal force 'F' is applied at the mid point of the rope so that the rope makes an angle of 45° with respect to the vertical axis as shown in figure. The magnitude of F is:



- $(1) \; \frac{10}{\sqrt{2}} \, N$
- (2) 1 N
- $(3) \ \frac{1}{10 \times \sqrt{2}} N$
- (4) 10 N

Ans. (4)

**Sol.**  $T_1 \sin 45^\circ = F$ 

$$T_1 \cos 45^\circ = T_2 = 1 \times g$$

$$\therefore \tan 45^\circ = \frac{F}{g}$$

$$\therefore F = 10N$$

**49.** A spherical ball of radius  $1 \times 10^{-4}$  m and density  $10^{5}$  kg/m<sup>3</sup> falls freely under gravity through a distance h before entering a tank of water, If after entering in water the velocity of the ball does not change, then the value of h is approximately:

(The coefficient of viscosity of water is  $9.8 \times 10^{-6}$ 

 $N s/m^2$ )

- (1) 2296 m
- (2) 2249 m
- (3) 2518 m
- (4) 2396 m

Ans. (3)

Sol. 
$$V_T = \frac{2g}{9} \frac{R^2 \left[ \rho_B - \rho_L \right]}{\eta}$$

$$\Rightarrow V_T = \frac{2}{9} \times \frac{10 \times \left(10^{-4}\right)^2}{9.8 \times 10^{-6}} \left[10^5 - 10^3\right]$$

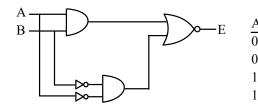
$$\Rightarrow$$
 V<sub>T</sub> = 224.5

when ball fall from height (h)

$$V = \sqrt{2gh}$$

$$h = \left(\frac{V^2}{2g}\right) = 2518m$$



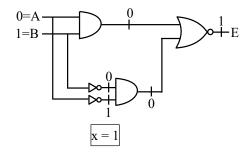


In the truth table of the above circuit the value of X and Y are:

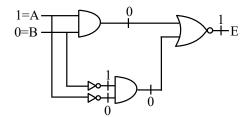
- (1) 1, 1
- (2) 1, 0
- (3) 0, 1
- (4) 0, 0

# Ans. (1)

#### Sol. For x



For y

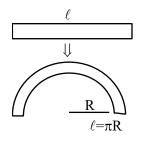


#### **SECTION-B**

**51.** A straight magnetic strip has a magnetic moment of 44 Am². If the strip is bent in a semicircular shape, its magnetic moment will be ........ Am²

(Given 
$$\pi = \frac{22}{7}$$
)

- Ans. (28)
- **Sol.** Magnetic moment of straight wire =  $mx \ell = 44$



Magnetic moment of arc

$$= m \times 2 r$$

$$= m \times \frac{2\ell}{\pi}$$

$$=\frac{44\times2}{\pi}=\frac{88}{\pi}=28$$

52. A particle of mass 0.50 kg executes simple harmonic motion under force  $F = -50(Nm^{-1})x$ . The time period of oscillation is  $\frac{x}{35}$ s. The value of x is

(Given 
$$\pi = \frac{22}{7}$$
)

Ans. (22)

**Sol.** 
$$m = 0.5 \text{ kg}$$

$$F = -50 (x)$$

$$ma = (-50x)$$

$$0.5 a = -50x$$

$$a = (-100x)$$

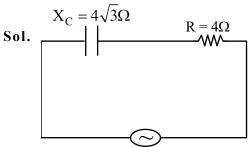
$$W^2 = 100 \Rightarrow (w = 10)$$

$$T = \frac{2\pi}{10} = \left(\frac{\pi}{5}\right) = \frac{22}{7 \times 15} = \left(\frac{22}{35}\right)$$

$$\frac{\pi}{35} = \frac{22}{35} \Rightarrow \boxed{x = 22}$$

53. A capacitor of reactance  $4\sqrt{3}\Omega$  and a resistor of resistance  $4\Omega$  are connected in series with an ac source of peak value  $8\sqrt{2}V$ . The power dissipation in the circuit is ......W.

Ans. (4)



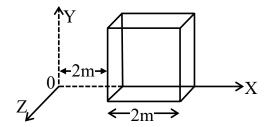
$$Z = \sqrt{R^2 + X^2 L}$$

$$Z = \sqrt{4^2 + (4\sqrt{3})^2} = 8\Omega$$

$$V_{rms} = \frac{V}{\sqrt{2}} = \frac{8\sqrt{2}}{\sqrt{2}} = (8V)$$

$$I_{rms} = \frac{V_{rms}}{Z} = \frac{8}{8} = 1A$$

Power dissipated =  $I_{rms}^2 \times R = 1 \times 4 = (4W)$ 



Ans. (16)

Sol. 
$$\vec{E} = 4\hat{i}$$
  $\vec{E} = 8$   $(0,0)$   $x=2$   $x=4$ 

$$\vec{E} = 2x\hat{i}$$
  
 $\phi = \vec{E} \cdot \vec{A}$ 

$$\phi_{in} = -4 \times 4 = -16 \text{ Nm}^2 / c$$

$$\phi_{out} = 8 \times 4 = 32 \text{Nm}^2 / c$$

$$d_{net} = \phi_{in} + \phi_{out} = -16 + 32 = 16 \text{ Nm}^2 / c$$

55. A circular disc reaches from top to bottom of an inclined plane of length l. When it slips down the plane, if takes t s. When it rolls down the plane then it takes  $\left(\frac{\alpha}{2}\right)^{1/2}$  t s, where  $\alpha$  is ............

Ans. (3)

**Sol.** For slipping  $a = g sin \theta$ 

$$\ell = \frac{1}{2} \operatorname{at}^2 \implies t = \sqrt{\frac{2\ell}{g \sin \theta}}$$

For rolling

$$a' = \frac{g\sin\theta}{1 + \frac{k^2}{R^2}} \left[ k = \frac{R}{\sqrt{2}} \right]$$

$$\Rightarrow a' = \frac{2g\sin\theta}{3}$$

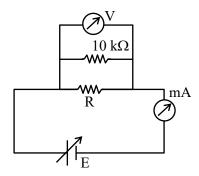
$$\ell = \frac{1}{2}a'(t')^2$$

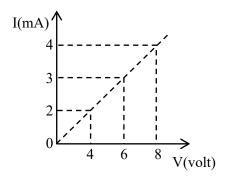
$$\ell = \frac{1}{2} a'(t')^{2}$$

$$\Rightarrow t' = \sqrt{\frac{6\ell}{2g\sin\theta}} = \sqrt{\frac{\alpha}{2}} \sqrt{\frac{2\ell}{g\sin\theta}}$$

$$\Rightarrow \alpha = 3$$

56. To determine the resistance (R) of a wire, a circuit is designed below, The V-I characteristic curve for this circuit is plotted for the voltmeter and the ammeter readings as shown in figure. The value of R is ......Ω.





Ans. (2500)

**Sol.** Req = 
$$\frac{10^4 \text{R}}{10^4 + \text{R}}$$

$$E = 4V, I = 2mA$$

$$I = \frac{E}{Req} \Rightarrow 2 \times 10^{-3} = \frac{4(10^4 + R)}{10^4 R}$$

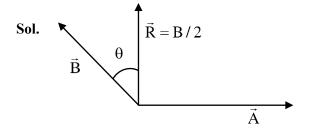
$$\Rightarrow$$
 20R = 40000 + 4R

$$16R = 40000$$

$$R = 2500\Omega$$

57. The resultant of two vectors  $\vec{A}$  and  $\vec{B}$  is perpendicular to  $\vec{A}$  and its magnitude is half that of  $\vec{B}$ . The angle between vectors  $\vec{A}$  and  $\vec{B}$  is

Ans. (150)



$$B\cos\theta = \frac{B}{2}$$

$$\Rightarrow \theta = 60^{\circ}$$

So, angle between  $\vec{A}$  &  $\vec{B}$  is  $90^{\circ} + 60^{\circ} = 150^{\circ}$ 

58. Monochromatic light of wavelength 500 nm is used in Young's double slit experiment. An interference pattern is obtained on a screen When one of the slits is covered with a very thin glass plate (refractive index = 1.5), the central maximum is shifted to a position previously occupied by the 4<sup>th</sup> bright fringe. The thickness of the glass-plate is ......μm.

Ans. (4)

Sol. 
$$(\mu - 1) t = n\lambda$$
  
 $(1.5 - 1) t = 4 \times 500 \times 10^{-9} m$   
 $t = 4000 \times 10^{-9} m$   
 $t = 4 \mu m$ 

**59.** A force  $(3x^2 + 2x - 5)$  N displaces a body from x = 2 m to x = 4m. Work done by this force is ......J.

Ans. (58)

Sol. 
$$W = \int_{x_1}^{x_2} F dx$$
  
 $W = \int_{2}^{4} (3x^2 + 2x - 5) dx$   
 $W = \left[x^3 + x^2 - 5x\right]_{2}^{4}$   
 $W = \left[60 - 2\right] J = 58J$ 

60. At room temperature (27°C), the resistance of a heating element is  $50\Omega$ . The temperature coefficient of the material is  $2.4 \times 10^{-4}$  °C<sup>-1</sup>. The temperature of the element, when its resistance is  $62 \Omega$ , is ......°C.

Ans. (1027)

**Sol.** 
$$R = R_0(1 + \alpha \Delta T)$$
  
 $62 = 50 [1 + 2.4 \times 10^{-4} \Delta T]$   
 $\Delta T = 1000^{\circ} C$   
 $\Rightarrow T - 27^{\circ} = 1000^{\circ} C$   
 $\boxed{T = 1027^{\circ} C}$ 

# **CHEMISTRY**

#### **SECTION-A**

- 61. The candela is the luminous intensity, in a given direction, of a source that emits monochromatic radiation of frequency 'A'  $\times 10^{12}$  hertz and that has a radiant intensity in that direction of  $\frac{1}{B'}$  watt per steradian. 'A' and 'B' are respectively
  - (1) 540 and  $\frac{1}{683}$
  - (2) 540 and 683
  - (3) 450 and  $\frac{1}{68^2}$
  - (4) 450 and 683

Ans. (2)

- **Sol.** The candela is the luminous intensity of a source that emits monochromatic radiation of frequency radiation of frequency  $540 \times 10^{12}$  Hz and has a radiant intensity in that direction of  $\frac{1}{683}$  w/sr. It is unit of Candela.
- **62.** The correct stability order of the following resonance structures of  $CH_3$ –CH = CH–CHO is

$$CH_{3}-CH-CH=C-H \leftrightarrow CH_{3}-CH-CH=C-H$$

$$I \qquad II$$

$$CH_{3}-CH-CH=C-H \leftrightarrow CH_{3}-CH-CH-C-H$$

$$I \qquad III$$

- $(1) \parallel > \parallel \parallel > \parallel$
- (2) III > II > I
- (3) I > II > III
- (4) II > I > III

Ans. (2)

#### **TEST PAPER WITH SOLUTION**

Sol. CH,-CH=CH-CH (III)

> Non Polar R.S. More No of covalent bond

$$O^{\Theta}$$
 $CH_3$ -CH-CH=CH (II)

Having -ve charge on more electronegative atom

Having –ve charge on less electronegative atom

Stability order III > II > I

**63.** Total number of stereo isomers possible for the given structure:

(1) 8

(3)4

(4)3

Ans. (1)

There are three stereo center So No of stereoisomer =  $2^3 = 8$ 

- 64. The correct increasing order for bond angles among BF<sub>3</sub>, PF<sub>3</sub> and  $C\ell F_3$  is :

  - (1)  $PF_3 < BF_3 < C\ell F_3$  (2)  $BF_3 < PF_3 < C\ell F_3$
  - (3)  $C\ell F_3 < PF_3 < BF_3$  (4)  $BF_3 = PF_3 < C\ell F_3$

Ans. (3)

Sol.

Order of bond angle is  $ClF_3 < PF_3 < BF_3$ 

#### 65. Match List I with List II

	LIST-I	LIST-II		
	(Test)	(Observation)		
A.	Br <sub>2</sub> water test	I.	Yellow orange or	
			orange red	
			precipitate	
			formed	
B.	Ceric	II.	Reddish orange	
	ammonium		colour	
	nitrate test		disappears	
C.	Ferric chloride	III.	Red colour	
	test		appears	
D.	2, 4-DNP test	IV.	Blue, Green,	
			Violet or Red	
			colour appear	

Choose the correct answer from the options given below:

- (1) A-I, B-II, C-III, D-IV
- (2) A-II, B-III, C-IV, D-I
- (3) A-III, B-IV, C-I, D-II
- (4) A-IV, B-I, C-II, D-III

Ans. (2)

**Sol.** (A) Br<sub>2</sub> water test is test of unsaturation in which reddish orange colour of bromine water disappears.

- (B) Alcohols given Red colour with ceric ammonium nitrate.
- (C) Phenol gives Violet colour with natural ferric chloride.
- (D) Aldehyde & Ketone give Yellow/Orange/Red Colour compounds with 2, 4-DNP i.e., 2, 4-Dinitrophenyl hydrazine.

#### **66.** Match List I with List II

	LIST-I	LIST-II		
	(Cell)	(Use/Property/Reaction)		
A.	Leclanche	I.	Converts energy	
	cell		of combustion into	
			electrical energy	
B.	Ni-Cd cell	II.	Does not involve	
			any ion in solution	
			and is used in	
			hearing aids	
C.	Fuel cell	III.	Rechargeable	
D.	Mercury	IV.	Reaction at anode	
	cell		$Zn \rightarrow Zn^{2+} + 2e^-$	

Choose the correct answer from the options given below:

- (1) A-I, B-II, C-III, D-IV
- (2) A-III, B-I, C-IV, D-II
- (3) A-IV, B-III, C-I, D-II
- (4) A-II, B-III, C-IV, D-I

Ans. (3)

Sol. A-IV, B-III, C-I, D-II

**67.** Match List I with List II

	LIST-I	LIST-II	
A.	$K_2[Ni(CN)_4]$	I.	sp <sup>3</sup>
B.	[Ni(CO) <sub>4</sub> ]	II.	$sp^3d^2$
C.	$[Co(NH_3)_6]Cl_3$	III.	dsp <sup>2</sup>
D.	Na <sub>3</sub> [CoF <sub>6</sub> ]	IV.	$d^2sp^3$

Choose the correct answer from the options given below:

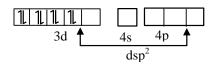
- (1) A-III, B-I, C-II, D-IV
- (2) A-III, B-II, C-IV, D-I
- (3) A-I, B-III, C-II, D-IV
- (4) A-III, B-I, C-IV, D-II

Ans. (4)

Sol. (A)  $K_2 [Ni(CN)_4]$ 

 $Ni^{2+}$ : [Ar]3d $^8$ 4s $^o$ ,(CN $^-$ is S.F.L)

Pre hybridization state of Ni<sup>+2</sup>

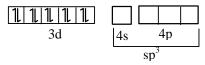


**(B)** [Ni(CO)<sub>4</sub>]

 $Ni : [Ar] 3d^8 4s^2$ 

CO is S.F.L, so pairing occur

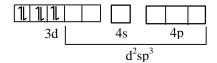
Pre hybridization state of Ni



(C)  $\left[\operatorname{Co(NH_3)_6}\right]\operatorname{Cl_3}$ 

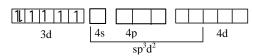
 $Co^{+3}: [Ar] 3d^6 4s^0$ 

With Co<sup>3+</sup>, NH<sub>3</sub> act as S.F.L



(d)  $Na_3 [CoF_6]$ 

 $Co^{3+}$ : [Ar]  $3d^{6}(F^{\Theta}: W.F.L)$ 



- **68.** The coordination environment of  $Ca^{2+}$  ion in its complex with  $EDTA^{4-}$  is:
  - (1) tetrahedral
  - (2) octahedral
  - (3) square planar
  - (4) trigonal prismatic

Ans. (2)

Sol. EDTA<sup>4-</sup>  $\rightarrow$  Hexadentate ligand [Ca(EDTA)]<sup>2-</sup>

So Coordination environment is octahedral

- **69.** The **incorrect** statement about Glucose is :
  - (1) Glucose is soluble in water because of having aldehyde functional group
  - (2) Glucose remains in multiple isomeric form in its aqueous solution
  - (3) Glucose is an aldohexose
  - (4) Glucose is one of the monomer unit in sucrose

Ans. (1)

**Sol.** Glucose is soluble in water due to presence of alcohol functional group and extensive hydrogen bonding.

Glucose exist is open chain as well as cyclic forms in its aqueous solution.

Glucose having 6C atoms so it is hexose and having aldehyde functional group so it is aldose.

Thus, aldohexose.

Glucose is monomer unit in sucrose with fructose.

70. 
$$OCH_3 \xrightarrow{KCN(alc)} Major Product 'P'$$

In the above reaction product 'P' is

$$(1) \bigcirc CN$$

$$(2) \bigcirc CN$$

$$(3) \bigcirc CN$$

$$(3) \bigcirc CH_3$$

$$(4) \bigcirc CN$$

$$(4) \bigcirc CH_3$$

Ans. (1)

Sol.

Due to NGP effect of phenyl ring Nucleophilic substitution of Br will occurs.

**71.** Which of the following compound can give positive iodoform test when treated with aqueous KOH solution followed by potassium hypoiodite.

$$\begin{array}{c} \text{O} \\ \text{II} \\ \text{(1) CH}_{3}\text{CH}_{2}\text{-C-CH}_{2}\text{CH}_{3} \end{array}$$

(3) CH<sub>3</sub>CH<sub>2</sub>CH<sub>2</sub>CHO

Ans. (2)

Sol.

$$CH_{3}-CH_{2}-C-CH_{3} \xrightarrow{aq. KOH} CH_{3}-CH_{2}-C-CH_{3}$$

$$-H_{2}O \downarrow O \downarrow O \downarrow O$$

$$CH_{3}-CH_{2}-C-CH_{3}$$

$$KOI \downarrow CH_{3}-CH_{2}-COOK+CHI_{3}\downarrow$$

$$Yellow ppt$$

- 72. For a sparingly soluble salt  $AB_2$ , the equilibrium concentrations of  $A^{2+}$  ions and  $B^-$  ions are  $1.2\times 10^{-4}$  M and  $0.24\times 10^{-3}$  M, respectively. The solubility product of  $AB_2$  is :
  - $(1)\ 0.069\times 10^{-12}$
  - (2)  $6.91 \times 10^{-12}$
  - (3)  $0.276 \times 10^{-12}$
  - (4)  $27.65 \times 10^{-12}$

Ans. (2)

**Sol.**  $AB_{2(s)} \rightleftharpoons A^{+2}_{(aq)} + 2B^{-}_{(aq)}$ 

$$K_{sp} = [A^{+2}][B^{-}]^{2}$$

$$= 1.2 \times 10^{-4} \times (2.4 \times 10^{-4})^{2}$$

$$= 6.91 \times 10^{-12} \text{ M}^{3}$$

**73.** Major product of the following reaction is

Ans. (2)

Sol.

$$C \equiv N$$

$$CH_{3} \longrightarrow NMgBn$$

$$CH_{3} \longrightarrow NMgB$$

# **74.** Given below are two statements:

**Statement I:** The higher oxidation states are more stable down the group among transition elements unlike p-block elements.

**Statement II:** Copper can not liberate hydrogen from weak acids.

In the light of the above statements, choose the correct answer from the options given below:

- (1) Both Statement I and Statement II are false
- (2) Statement I is false but Statement II is true
- (3) Both Statement I and Statement II are true
- (4) Statement I is true but Statement II is false

# Ans. (3)

**Sol.** On moving down the group in transition elements, stability of higher oxidation state increases, due to increase in effective nuclear charge.

$$\Rightarrow E^{o}_{Cu^{+2}/Cu} = 0.34 \text{ V}$$

$$\Rightarrow E^{\circ}_{H^+/H_2} = 0$$

SRP : 
$$Cu^{2+} > H^{+}$$

Cu can't liberate hydrogen gas from weak acid.

#### **75.** The **incorrect** statement regarding ethyne is

- (1) The C–C bonds in ethyne is shorter than that in ethene
- (2) Both carbons are sp hybridised
- (3) Ethyne is linear
- (4) The carbon-carbon bonds in ethyne is weaker than that in ethene

#### Ans. (4)

**Sol.** The carbon-carbon bonds in ethyne is stronger than that in ethene.

(H–C≡C–H) Ethyne is linear and carbon atoms are SP hybridised.

#### **76.** Match List I with List II

	List-I (Element)		List-II (Electronic Configuration)		
A.	N	I.	[Ar] $3d^{10}4s^2 4p^5$		
B.	S	II.	[Ne] $3s^2 3p^4$		
C.	Br	III.	[He] $2s^2 2p^3$		
D	Kr	IV.	[Ar] $3d^{10} 4s^2 4p^6$		

Choose the correct answer from the options given below:

- (1) A-IV, B-III, C-II, D-I
- (2) A-III, B-II, C-I, D-IV
- (3) A-I, B-IV, C-III, D-II
- (4) A-II, B-I, C-IV, D-III

### Ans. (2)

**Sol.** (A)  $_{7}$ N:[He]2s<sup>2</sup>2p<sup>3</sup>

(B)  $_{16}$ S:[Ne]2s<sup>2</sup>3p<sup>4</sup>

(C)  $_{35}$ Br:[Ar]3d $^{10}$ 4s $^{2}$ 4p $^{5}$ 

(D)  $_{36}$ Kr:[Ar]3d $^{10}$ 4s $^{2}$ 4p $^{6}$ 

#### 77. Match List I with List II

List-I		List-II		
A.	Melting point [K]	I.	Tl > In > Ga > Al > B	
B.	Ionic Radius [M <sup>+3</sup> /pm]	II.	$B > Tl > Al \approx Ga > In$	
C.	$\Delta_{i}H_{1}$ [kJ mol <sup>-1</sup> ]	III.	Tl > In > Al > Ga > B	
D	Atomic Radius [pm]	IV.	B > Al > Tl > In > Ga	

Choose the correct answer from the options given below:

- (1) A-III, B-IV, C-I, D-II
- (2) A-II, B-III, C-IV, D-I
- (3) A-IV, B-I, C-II, D-III
- (4) A-I, B-II, C-III, D-IV

Ans. (3)

**Sol.** Melting point :  $B > A\ell > T\ell > In > Ga$ 

Ionic radius (M<sup>+3</sup>/pm) :  $T\ell > In > Ga > A\ell > B$ 

$$(\Delta_{IE}H)_1\left[\frac{kJ}{mol}\right]: B \ge T\ell \ge A\ell \approx Ga \ge In$$

Atomic radius (in pm) :  $T\ell > In > A\ell > Ga > B$ 

- **78.** Which of the following compounds will give silver mirror with ammoniacal silver nitrate?
  - (A) Formic acid
  - (B) Formaldehyde
  - (C) Benzaldehyde
  - (D) Acetone

Choose the correct answer from the options given below:

- (1) C and D only
- (2) A, B and C only
- (3) A only
- (4) B and C only

Ans. (2)

Sol. Apart from aldehyde, Formic acid



also gives silver mirror test with ammonical silver nitrate.

79. Which out of the following is a correct equation to show change in molar conductivity with respect to concentration for a weak electrolyte, if the symbols carry their usual meaning:

$$(1) \Lambda_{m}^{2} C - K_{a} \Lambda_{m}^{2} + K_{a} \Lambda_{m} \Lambda_{m}^{2} = 0$$

(2) 
$$\Lambda_{\rm m} - \Lambda_{\rm m}^{\circ} + AC^{\frac{1}{2}} = 0$$

(3) 
$$\Lambda_{\rm m} - \Lambda_{\rm m}^{\circ} - AC^{\frac{1}{2}} = 0$$

$$(4) \Lambda_{\mathrm{m}}^{2}C + K_{\mathrm{a}}\Lambda_{\mathrm{m}}^{^{2}} - K_{\mathrm{a}}\Lambda_{\mathrm{m}}\Lambda_{\mathrm{m}}^{^{2}} = 0$$

Ans. (1)

**Sol.**  $HA(aq) \rightleftharpoons H^{+}(aq) + A^{-}(aq)$ 

$$K_a = \frac{\alpha^2 C}{1 - \alpha}$$

$$\alpha^2 C + K_a \alpha - K_a = 0$$

$$\left(\frac{\lambda_{m}}{\lambda_{m}^{\infty}}\right)^{2}C+K_{a}\frac{\lambda_{m}}{\lambda_{m}^{\infty}}-K_{a}=0$$

$$\lambda_{m}^{2}C + K_{a}\lambda_{m}\lambda_{m}^{\infty} - K_{a}\left(\lambda_{m}^{\infty}\right)^{2} = 0$$

**80.** The electronic configuration of Einsteinium is : (Given atomic number of Einsteinium = 99)

- $(1) [Rn] 5f^{12} 6d^0 7s^2$
- (2) [Rn]  $5f^{11} 6d^0 7s^2$
- (3) [Rn]  $5f^{13} 6d^0 7s^2$
- (4) [Rn]  $5f^{10} 6d^0 7s^2$

Ans. (2)

**Sol.** Einsteinium (atomic No = 99) :  $[Rn] 5f^{11} 6d^0 7s^2$ 

#### **SECTION-B**

**81.** Number of oxygen atoms present in chemical formula of fuming sulphuric acid is

Ans. (7)

**Sol.** Fuming sulphuric acid is a mixture of conc.  $H_2SO_4 + SO_3$  Or  $H_2S_2O_7$ So, Number of Oxygen atoms = 7

82. A transition metal 'M' among Sc, Ti, V , Cr, Mn and Fe has the highest second ionisation enthalpy.
The spin only magnetic moment value of M<sup>+</sup> ion is \_\_\_\_\_\_BM (Near integer)

(Given atomic number Sc : 21, Ti : 22, V : 23, Cr : 24, Mn : 25, Fe : 26)

Ans. (6)

Sol. Among given metals, Cr has maximum  $IE_2$  because Second electron is removed from stable configuration  $3d^5$ 

 $Cr^{+}$ : [Ar]  $3d^{5} 4s^{0}$ 

 $\therefore$  No of unpaired e<sup>-</sup> in Cr<sup>+</sup> is 5, n = 5

So, Magnetic moment =  $\sqrt{n(n+2)}$  B.M

$$=\sqrt{5(5+2)}=5.92 \text{ BM} \approx 6$$

83. The vapour pressure of pure benzene and methyl benzene at 27°C is given as 80 Torr and 24 Torr, respectively. The mole fraction of methyl benzene in vapour phase, in equilibrium with an equimolar mixture of those two liquids (ideal solution) at the same temperature is  $\times 10^{-2}$  (nearest integer)

Ans. (23)

**Sol.** 
$$X_{\text{methylbenzene}} = 0.5$$

$$Y_{\text{methylbenzene}} = \frac{P_{\text{methylbenzene}}}{P_{\text{total}}}$$

$$Y_{\text{methylbenzene}} = \frac{0.5 \times 24}{0.5 \times 80 + 0.5 \times 24}$$
$$= \frac{12}{40 + 12} = 0.23 = 23 \times 10^{-2}$$

**84.** Consider the following test for a group-IV cation.

$$M^{2+} + H_2S \rightarrow A$$
 (Black precipitate) + byproduct

$$A + aqua regia \rightarrow B + NOCl + S + H_2O$$

$$B + KNO_2 + CH_3COOH \rightarrow C + byproduct$$

The spin only magnetic moment value of the metal complex C is \_\_\_\_\_BM.

(Nearest integer)

Ans. (0)

**Sol.** 
$$Co^{2+} + H_2S \rightarrow CoS \downarrow (Black)$$

(A)

$$CoS + Aqua-regia \rightarrow Co^{2+} (aq) + NOCl + S + H_2O$$

$$Co^{2+}$$
 (aq) + KNO<sub>2</sub> + CH<sub>3</sub>COOH

 $\downarrow$ 

$$K_3[Co(NO_2)_6] + NO + S + H_2O$$

In 
$$K_3[Co(NO_2)_6]$$
,  $Co^{+3}: 3d^6 4s^0$ 

Co<sup>3+</sup>: d<sup>2</sup>sp<sup>3</sup> Hybridisation

Number of unpaired e<sup>-</sup>= 0

Magnetic moment =  $\sqrt{n(n+2)}$  = 0 B.M

**85.** Consider the following first order gas phase reaction at constant temperature

$$A(g) \rightarrow 2B(g) + C(g)$$

If the total pressure of the gases is found to be 200 torr after 23 sec. and 300 torr upon the complete decomposition of A after a very long time, then the rate constant of the given reaction is  $\times 10^{-2} \, \text{s}^{-1}$  (nearest integer)

[Given:  $log_{10}(2) = 0.301$ ]

Ans. (3)

**Sol.** 
$$A(g) \rightarrow 2B(g) + C(g)$$

$$P_{23} = P_0 + 2x = 200$$

$$P_{\infty} = 3P_0 = 300$$

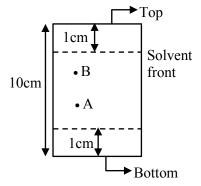
$$P_0 = 100$$

$$K = \frac{1}{t} \ln \frac{P_{\infty} - P_0}{P_{\infty} - P_t}$$

$$K = \frac{2.3}{23} \log \frac{300 - 100}{300 - 200}$$

$$=\frac{2.3\times0.301}{23}=0.0301=3.01\times10^{-2}\,\mathrm{sec}^{-1}$$

86.



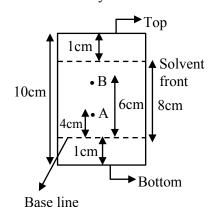
In the given TLC, the distance of spot A & B are 5 cm & 7 cm, from the bottom of TLC plate, respectively.

 $R_f$  value of B is  $x \times 10^{-1}$  times more than A. The value of x is\_\_\_\_.

Ans. (15)

Sol.

 $R_f = \frac{\text{Distance moved by substance from base line}}{\text{Distance moved by solvent from base line}}$ 



$$(R_f)_A = \frac{4}{8} \qquad (R_f)_B = \frac{6}{8}$$
$$\frac{(R_f)_B}{(R_f)_A} = \frac{6}{8} \times \frac{8}{4}$$
$$(R_f)_B = 1.5 (R_f)_A$$
$$x = 15$$

87. Based on Heisenberg's uncertainty principle, the uncertainty in the velocity of the electron to be found within an atomic nucleus of diameter  $10^{-15}$  m is \_\_\_\_ ×  $10^9$  ms<sup>-1</sup> (nearest integer) [Given: mass of electron =  $9.1 \times 10^{-31}$  kg, Plank's constant (h) =  $6.626 \times 10^{-34}$  Js] (Value of  $\pi = 3.14$ )

Ans. (58)

Sol. 
$$m\Delta V.\Delta x = \frac{h}{4\pi}$$
  

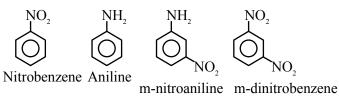
$$\Delta V = \frac{6.626 \times 10^{-34}}{9.1 \times 10^{-31} \times 10^{-15} \times 4 \times 3.14}$$

$$= 57.97 \times 10^{+9} \text{ m/sec}$$

**88.** Number of compounds from the following which **cannot** undergo Friedel-Crafts reactions is :\_\_\_\_ toluene, nitrobenzene, xylene, cumene, aniline, chlorobenzene, m-nitroaniline, m-dinitrobenzene

Ans. (4)

**Sol.** Compounds which can not undergo Friedel Crafts reaction are



**89.** Total number of electron present in  $(\pi^*)$  molecular orbitals of  $O_2$ ,  $O_2^+$  and  $O_2^-$  is\_\_\_\_\_.

Ans. (6)

**Sol.** 
$$O_2(16e): (\sigma_{1s})^2 (\sigma_{1s}^*)^2 (\sigma_{2s})^2 (\sigma_{2s}^*)^2$$

$$(\sigma_{2p})^2 \Big[ (\pi_{2p})^2 = (\pi_{2p})^2 \Big], \Big[ (\pi_{2p}^*)^1 = (\pi_{2p}^*)^1 \Big]$$
Number of  $e^-$  present in  $(\pi^*)$  of  $O_2 = 2$ 
Number of  $e^-$  present in  $(\pi^*)$  of  $O_2^+ = 1$ 
Number of  $e^-$  present in  $(\pi^*)$  of  $O_2^- = 3$ 
So total  $e^-$  in  $(\pi^*) = 2 + 1 + 3 = 6$ 

90. When  $\Delta H_{\text{vap}} = 30 \text{ kJ/mol}$  and  $\Delta S_{\text{vap}} = 75 \text{ J mol}^{-1} \text{ K}^{-1}$ , then the temperature of vapour, at one atmosphere is \_\_\_\_\_K.

Ans. (400)

Sol. At equilibrium 
$$\Delta G_{PT} = 0$$
  
 $\Delta H_{vap} = T\Delta S_{vap}$   
 $30 \times 1000 = T \times 75$   
 $T = 400 K$