Exponential function: $\frac{e^{kx}}{e^{kx}} = \frac{1}{e^{0}} = \frac{1}{1} = 1$ (When x = 0)
or k = 0 $\frac{-kx}{e} = \frac{1}{e^{kx}} = \frac{1}{a} = 0$ (Where x = x or x = a) $\frac{1}{\int_{z=k_{3}x}^{z=k_{1}x}} \left| k_{3} \right\rangle k_{2} \right\rangle k_{4}$ We use the properties of the exponential function in our derivation: $A \xrightarrow{k_1} \times \xrightarrow{k_2} Z$ $[x] = [A]_0 \frac{k_1}{k_2 - k_1} \left(= k_1 + - e^{k_2 + 1} \right)$ Steady-State Treatment: ke is large ky is very very small ke>>> k, There $k_2 - k_1 \approx k_1$ | Assume $k_2 = 10 \text{ S}$ | $k_2 = 10 \text{ S}$ | $k_1 = 0.001 \text{ S}$ | $k_2 = 10 \text{ S}$ | $k_1 = 0.001 \text{ S}$ | $k_2 = 10 \text{ S}$ | $k_1 = 0.001 \text{ S}$ | $k_2 = 10 \text{ S}$ | $k_2 = 10 \text{ S}$ | $k_3 = 10 \text{ S}$ | $k_4 = 0.001 \text{ S}$ | $k_5 = 0.001 \text{ S}$ | $k_6 = 0.$

Now concentrade on the quantity (= kit_= k2t) At t=0, = k10 = e = 1 and $\bar{e}^{k_2t} = \bar{e}^{k_20} = \bar{e}^0 = 1$ Therefore, (= k1t -= k2t)=(1-1)=0 $[x] = [x]_0 \frac{k_1}{k_2} \left(e^{k_1 t} - e^{k_2 t} \right)$ $= [A]_0 \frac{k_1}{k_2} (1-1) = (A+b)$ After very short time, relative to the duration of the reaction, Follow ten plot on the first page, = k1 t - = k2t) ~1 ~ e 0'001x10 - 10x10 k_2 >> k_1 and $k_1 = 0.00151$ $k_2 = 1051$ ~ 0.99 - 1049 1 to = KIE ~ 0.99 -0 ≈ 0.99 $[x] = [A]_{6} \frac{k_{1}}{k_{2}} \left(e^{k_{1}t} - e^{k_{2}t} \right)$ Again $= [A]_6 \frac{k_1}{k_2}$ Again Change of [x] is constant Since ki <<1 are contant [x] << [A] o