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Modern Chemistry 7 Solutions CHAPTER 12 REVIEW Solutions SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Describe the errors made by the following students in making molar solutions. a. James needs a 0.600 M solution of KCl. He measures out 0.600 g of KCl and adds 1 L of water to the solid.

CHAPTER 12 REVIEW Solutions - Weebly

of solution. Molality is the number of moles of solute per kilogram of solvent. 27. Solutions are homogeneous mixtures. Suspensions and colloids are heteroge-neous mixtures. Solutions have the

smallest particle size, followed by the size of colloid particles and suspension particles. Solutions and colloids do not settle out on standing, but ...

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