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Holt ChemFile: Problem-Solving Workbook 127 Percentage Yield Name Class Date Problem Solving continued Sample Problem 2 Acetylene, C 2H 2, can be used as an industrial starting material for the production of many organic compounds. Sometimes, it is first brominated to form 1,1,2,2-tetrabromoethane, CHBr 2CHBr 2, which can then be

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Holt ChemFile: Problem-Solving Workbook 191 Stoichiometry of Gases Name Class Date Problem Solving continued Rearrange the ideal-gas-law equation to solve for the unknown quantity, V. PV nRT V nR P T COMPUTE EVALUATE Are the units correct? Yes; units canceled to give liters of SO 2. Is the number of significant figures correct?

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Holt ChemFile: Problem-Solving Workbook 318 Electrochemistry Name Class Date Problem Solving continued Sample Problem 2 A cell is constructed in which the following two half-reactions can occur in either direction. Zn2 2e ^ Zn Br 2 2e ^ 2Br Write the full ionic equation for the cell in the spontaneous direction,

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and hydrogen under pressure, at high temperature, and in the presence of a catalyst. The ... Holt ChemFile: Problem-Solving Workbook 102 Stoichiometry Name Class Date Problem Solving continued

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Holt ChemFile: Problem-Solving Workbook 280 Titrations Name Class Date Problem Solving continued Sample Problem 3 A supply of NaOH is known to contain the contaminants NaCl and $MgCl_2$. A 4.955 g sample of this material is dissolved and diluted to 500.00 mL with water. A 20.00 mL sample of this solution is titrated with 22.26 mL

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Holt ChemFile: Problem-Solving Workbook 58 Mole Concept Name Class Date Problem Solving continued CONVERTING NUMBER OF ATOMS OF AN ELEMENT TO MASS Sample Problem 4 uses the progression of steps 1 → 2 → 3 to convert from the mass of an element to the number of atoms.

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