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Holt ChemFile: Problem-Solving Workbook 48 Mole Concept Mole Concept Suppose you want to carry out a reaction that requires combining one atom of iron with one atom of sulfur. How much iron should you use? How much sulfur? When you look around the lab, there is no device that can count numbers of atoms.

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Holt ChemFile: Problem-Solving Workbook 313 Electrochemistry Name Class Date Problem Solving continued You can use reduction potentials to predict the direction in which any redox reaction will be spontaneous. A spontaneous reaction occurs by itself, without outside influence. The redox reaction will proceed in the direction for which the

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Holt ChemFile: Problem-Solving Workbook 246 Equilibrium of Salts, K<sub>sp</sub> Equilibrium of Salts, K<sub>sp</sub> When you try to dissolve a solid substance in water, you expect the solid form to disappear, forming ions or molecules in solution. Most substances, however, are only slightly soluble in water. For example, when you stir silver chloride in water,

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Holt ChemFile B v Introduction to the Lab Program mc06SE\_cFMme\_j-xviii.qxd 1/31/05 11:59 AM Page v. CHEMICALS ARE NOT TOYS. Any chemical can be dangerous if it is misused. Always follow the instructions ... Holt ChemFile B viii Safety in the Chemistry Laboratory mc06SE\_cFMme\_j-xviii.qxd 1/31/05 11:59 AM Page viii. 19.

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