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Propane undergoes a process from state 1, where $p_1 = 1.4 \text{ MPa}$, $T_1 = 60^\circ \text{C}$, to state 2, where $p_2 = 1.0 \text{ MPa}$, during which the change in specific entropy is $s_2 - s_1 = -0.035 \text{ kJ/kg} \cdot \text{K}$. At state 2, determine the temperature, in $^\circ \text{C}$, and the specific enthalpy,...

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