Gauge-invariant expectation values of the energy of a molecule in an electromagnetic field

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For a molecule interacting with an electromagnetic field, we show that the Hamiltonian can be separated into a molecular Hamiltonian and a field Hamiltonian, both with gauge-invariant expectation values. The expectation value of the molecular Hamiltonian gives physically meaningful result for the energy of a molecule in a time-dependent field. The usual partitioning of the full Hamiltonian results in gauge dependence of both the molecular Hamiltonian and the field Hamiltonian.^{2,3} With the usual partitioning of the Hamiltonian, gauge dependence arises even in the absence of an applied field, as we have shown explicitly by considering a gauge transformation from zero applied field and zero external potentials to zero applied field, but nonzero vector and scalar potentials.^{1,4} By repartitioning the full Hamiltonian, we resolve this problem and also remove the gauge-dependence from the Hamiltonian for a molecule in a nonzero applied field. We treat the electromagnetic field classically and treat the molecule quantum mechanically, but nonrelativistically. We construct the full Hamiltonian using a Lagrange multiplier method originally suggested by Fermi⁵ and later applied by Goldman, ⁶ partition this Hamiltonian into a molecular term H_m and a field term H_f, and then show that both H_m and H_f have gauge-invariant expectation values. Any gauge may be chosen for the calculation; but following our partitioning, the expectation values of the molecular Hamiltonian are identical to those obtained in the Coulomb gauge. As a corollary of this result, the power absorbed by a molecule from a time-dependent, applied electromagnetic field is equal to the time-derivative of the nonadiabatic term in the molecular energy, in any gauge. ^{7,8}

- 1. A. Mandal and K. L. C. Hunt, *J. Chem. Phys.* **144**, 044109 (2016).
- 2. H. A. Kramers, *Quantum Mechanics* (North Holland, Amsterdam, 1957), pp. 268 269.
- 3. C. Cohen-Tannoudji, B. Diu, and F. Laloë, *Quantum Mechanics* (Wiley, New York, 1977), Vol. 1, pp. 315 326.
- 4. D. H. Kobe and S. M. Golshan, J. Phys. A: Math. Gen. 20, 2813 (1987).
- 5. E. Fermi, Rev. Mod. Phys. 4, 87 (1932).
- 6. T. Goldman, *Phys. Rev. D* **15**, 1063 (1977).
- 7. A. Mandal and K. L. C. Hunt, J. Chem. Phys. 137, 164109 (2012).
- 8. A. Mandal and K. L. C. Hunt, *J. Chem. Phys.* **143**, 034102 (2015).