

Name: _____

1. Write the following numbers in scientific notation:

212,000

0.0031300

2. Write the following numbers in standard (non-scientific) notation:

6.440×10^3

9.58×10^{-2}

3. Find the number of significant figures in the answer to this calculation:

$2000.0 - 1999.0$

4. Name what the following units measure:

Kilograms

Liters

Meters

$\frac{g}{mL}$

5. Write whether the following properties are chemical or physical:

The malleability of aluminum foil

The melting point of ice cream

The combustion of paper

6. Write whether the following are homogenous mixtures, heterogeneous mixtures, or pure substances:

Gasoline

Copper

Air

Salt water

A stream with gravel at the bottom

7. Using the formula for specific heat, calculate the amount of energy needed to raise the temperature of 1.6g of gold (specific heat $0.13 \frac{J}{g^{\circ}C}$) from $23^{\circ}C$ to $41^{\circ}C$: