

# Characterizaiton of Pu Separation by PUREX

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### Outline

#### Motivation

### Background

The PUREX Process

Distribution Coefficients

**Decontamination Factors** 

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Recovery of Pu and U

**Experimental Decontamination Factors** 

#### Future Work



# Big Picture

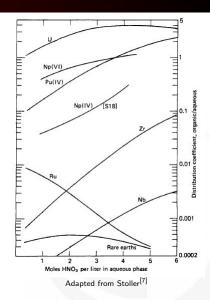
- Weapons-grade Pu can be extracted from reactor discharged fuel with a burnup of about 1 (GWD/tU)
- Pu isotopes produced in irradiated fuel can vary
- Two examples of reactors which can intentionally discharge low burned fuel for extracting weapon-grade Pu are:
  - > Fast Breeder Reactor
  - CANDU Reactor





### Smaller Picture

- Attribution for unpurified Pu has been previously studied [2, 6, 3]
- Interdicted Pu would likely have been processed
- Lack of literature on decontamination factors and distribution coefficients for useful forensic elements (Cs, Sb, Eu, Rb, Sr, Nd, Pm, and Sm)



Motivation Background Previous Work Future Work

# Background



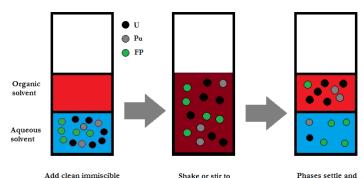
# What is PUREX - A type of laundry detergent?

- Plutonium Uranium Redox EXtraction
  - Liquid-liquid solvent extraction
  - Many stages:
    - 1. Preparation for Dissolution
    - 2. Dissolution
    - 3. Preparation of Dissolved Feed
    - 4. Primary Decontamination Extraction to organic\*
    - 5. Scrubbing
    - 6. Plutonium Partition Back-Extraction to aqueous\*
    - 7. Plutonium Purification



#### Extraction

$$\begin{array}{l} UO_{2(aq)}^{2+} + 2NO_{3(aq)}^{-} + 2TBP_{(o)} \leftrightarrow UO_{2}(NO_{3})_{2} \cdot 2TBP_{(o)}^{[1]} \\ Pu_{(aq)}^{4+} + 4NO_{3(aq)}^{-} + 2TBP_{(o)} \leftrightarrow Pu(NO_{3})_{4} \cdot 2TBP_{(o)} \end{array}$$



allow molecules to

partition

organic solvent

phase

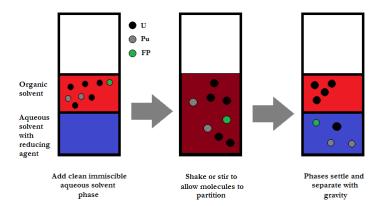
separate with

gravity



#### Back-Extraction

$$Pu(NO_3)_4(\textit{TBP})_{2(o)} + Fe^{2+}_{(aq)} \leftrightarrow Pu^{3+}_{(aq)} + 4NO^-_{3(aq)} + 2\textit{TBP}_{(o)}{}^{[5]}$$





# Distribution Coefficients - The Missing link

 Distribution Coefficient (D): The ratio between the organic and aqueous phases (aka: D-values)

$$D = \frac{c_o}{c_{aq}}$$

- Specific element to element
- Vary widely with:<sup>[7]</sup>
  - Composition of phases
  - Solution saturation
  - ➤ Temperature of the solvent
- \* The fraction of mass,  $f_o$  deposited in the organic phase, assuming a volume ratio between the aqueous and organic phases,  $V_R$ , is:

$$f_o = (1 + D^{-1}V_R^{-1})^{-1}$$



# Decontamination Factors - The Pot of gold

 After several cycles of Pu extraction/scrubbing/back-extraction are completed, the effectiveness of a PUREX cycle is described by the decontamination factor (DF):

$$DF_{j} = \frac{\left|\frac{c_{j}}{c_{Pu}}\right|_{initial}}{\left|\frac{c_{j}}{c_{Pu}}\right|_{final}}$$

- \* DFs are characteristic of different process cycles
- Larger values (10<sup>7</sup>) for industrial scale PUREX (compared to benchtop)<sup>[7, 1]</sup>

Motivation Background Previous Work Future Work

# Previous Work



#### Irradiation

- \*  $12.9 \pm 0.1 \text{ mg of DUO}_2$  was irradiated
  - High Flux Isotope Reactor at Oak Ridge National Laboratory
- \* Burnup was  $4.43 \pm 0.31$  GWd/tHM<sup>[8]</sup>
- \*  $0.196 \pm \text{mg}$  of total Pu was produced as measured by ICP-MS





# Dissolution of the spent fuel pellet





## Glovebox











## **Experiments**

- Single stage extraction and back-extraction
  - Purpose: quantify product recovery, D-values and DF values for single stage extraction and back extraction
  - Conditions:

Starting Solution	Extraction Solution	Back extraction solution		
4 M nitric acid	30% vol.% TBP, 70 vol.% kerosene	0.024 M ferrous sulfamate in 0.75 M nitric acid		

- Multi-contact extraction and back-extraction
  - Purpose: Maximize recovery of Pu with 4 extractions, 3 back extractions
  - Conditions:

Starting Solution Extraction Solution		Back extraction solution		
4 M nitric acid 30% vol.% TBP, 70 vol.% kerosene		0.024 M ferrous sulfamate in 4 M nitric acid		



# Previous Experiment Results

## Recoveries of U and Pu

	Pu Recovery	U Recovery
Single stage	(83.4±9.5)%	$(11.2\pm1.3)\%$
Multi-contact Cycle 1	$(99.7 \pm 4.2)\%$	$(6.8\pm0.3)\%$
Multi-contact Cycle 2	$(93.0\pm4.6)\%$	$(6.6\pm0.3)\%$
Overall Experiment 2	$(92.7\pm6.0)\%$	$(0.45\pm0.03)\%$



# Previous Experiment Results

#### **Decontamination Factors**

Element (Z)	SS	Error	MC Cycle 1	Error	Isotopes Used
Rb(37)	39.0	5.9	11.8	0.8	<sup>85</sup> Rb
Sr(38)	283	43	84.6	5.9	<sup>90</sup> Sr
Mo(42)	5.7	0.8	1.9	0.2	<sup>97,98,100</sup> Mo
Ru(44)	59.2	6.4	16.6	2.5	<sup>101,102,104</sup> Ru
Pd(46)	65	14	8.9	1.2	<sup>110</sup> Pd
Cd(48)	74	17	22.1	2.5	<sup>112</sup> Cd
Cs(55)	177	28	52.9	3.9	<sup>133</sup> Cs
Ce(58)	43	16	11.5	4.9	<sup>140,142</sup> Ce
Nd(60)	19.2	2.1	5.9	0.4	<sup>143</sup> Nd
Pm(61)	12.8	1.9	3.9	0.3	<sup>147</sup> Pm
Sm(62)	11.5	1.5	3.6	0.3	<sup>151</sup> Sm
Eu(63)	10.0	1.4	3.6	0.3	<sup>154</sup> Eu
U(92)	7.4	1.2	14.7	0.9	<sup>238</sup> U



#### Conclusions

- Two PUREX experiments were conducted
  - Single stage: Determined DC values for Pu, U and several FP
  - Multi-contact: Utilized Experiment 1 to recover over 92% of Pu while leaving less than 1% of the U
- DF values were measured for 12 FP elements
- DF values were lower than those typically found in industrial scale PUREX plants due to multiple extraction and back-extraction steps without an intermittent scrubbing step.
- This work provide DF data that will be built upon for nuclear forensic investigations of interdicted Pu.

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# Future Work



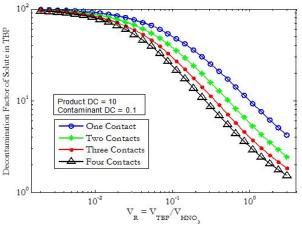
#### Future Work

- Modify Multi-contact extraction, to recover a larger fraction of Pu
- Investigation of how D-values for (Cs, Sb, Eu, Rb, Sr, Nd, Pm, and Sm) change as a function of nitric acid concentration
- Determine statistical uncertainty of D and DF values.
  - ➤ Repeat above experiments 3-5 times
- Connect D-values with process information to DF values

# Questions?



# Previous Experiment Results



Decontamination Factors for multi-contact extraction.



#### References I

- [1] M Benedict, H Levi, and T Pigford. Nuclear chemical engineering. *Nucl. Sci. Eng.*; (United States), 82(4), 1982.
- [2] Sunil S Chirayath, Jeremy M Osborn, and Taylor M Coles. Trace fission product ratios for nuclear forensics attribution of weapons-grade plutonium from fast and thermal reactors. Science & Global Security, 23(1):48–67, 2015.
- [3] Alexander Glaser. Isotopic signatures of weapon-grade plutonium from dedicated natural uranium-fueled production reactors and their relevance for nuclear forensic analysis. *Nuclear Science and Engineering*, 163(1):26–33, 2009.
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#### References II

[8] Mathew Wayne Swinney. Experimental and Computational Assessment of Trace Nuclide Ratios in Weapons Grade Plutonium for Nuclear Forensics Analysis. PhD thesis, 2015.



# Mass Spec

