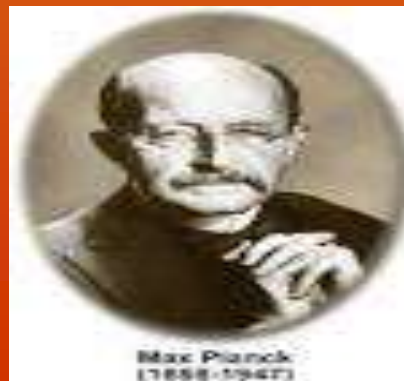


CHAPTER MODERN PHYSICS

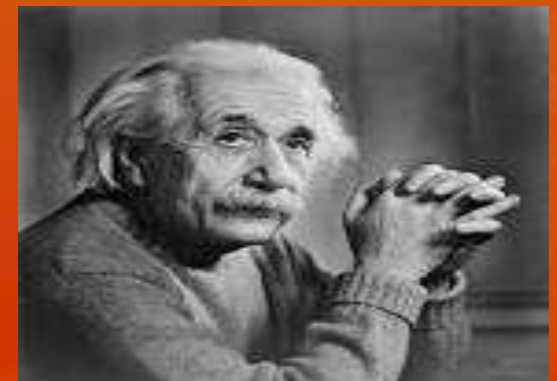
PHOTO ELECTRICITY

Represented by

Prof. Mr. S.N. jadhav

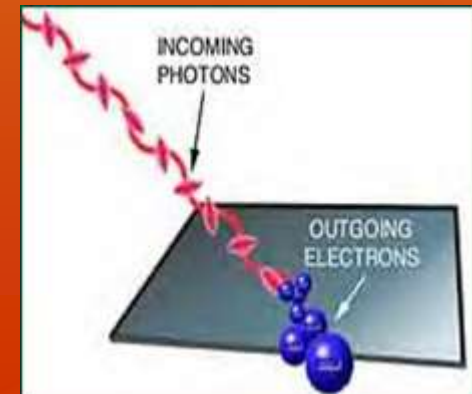
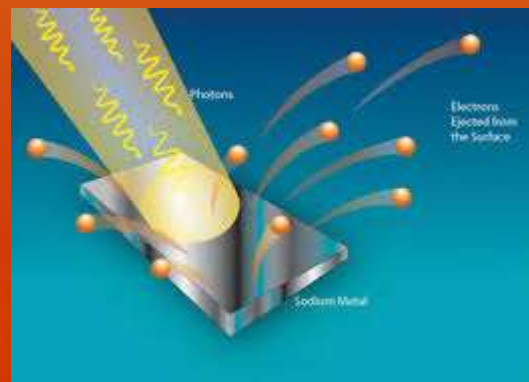
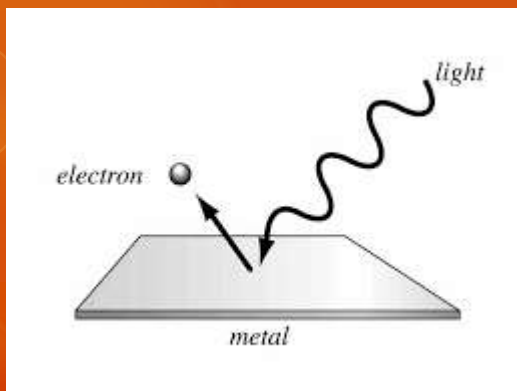


Max Planck
(1858-1947)



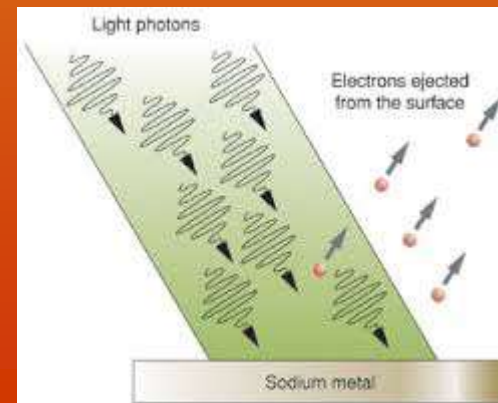
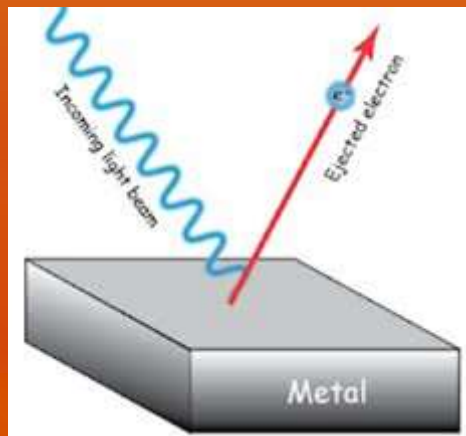
PHOTON (QUANTUM OF LIGHT)

- In 1905, Einstein proposed that electromagnetic radiation or light is made up of photons. Thus the photon is the elementary element of light or light is made up of photons.
- Einstein show that- light energy is not emitted continuously but it is emitted by individual amount of energy called as quantum of energy.



Energy of photon

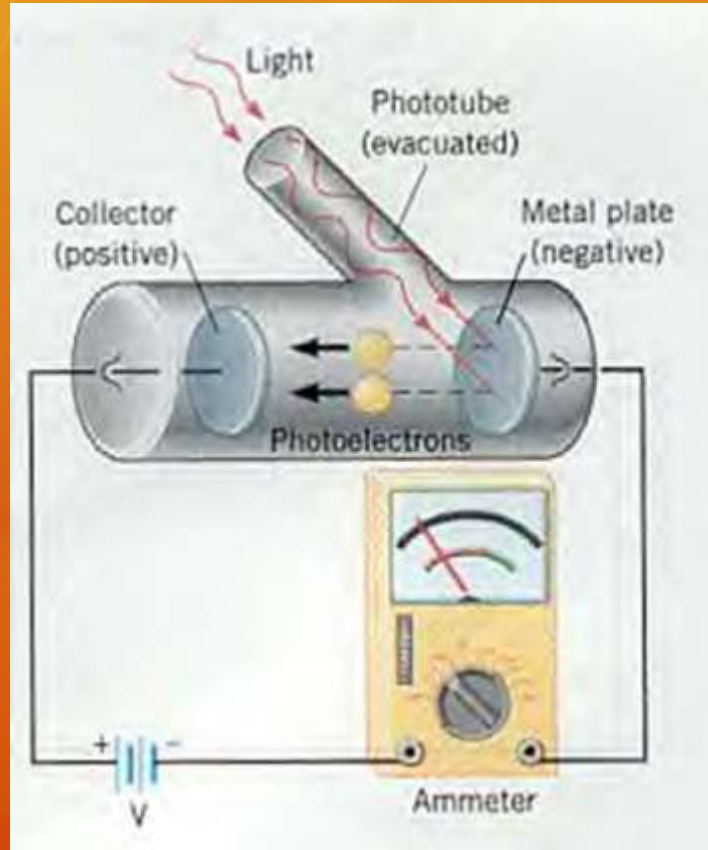
- According to Einstein, each photon of a light wave of frequency has the energy E is given by,
 $E = h\nu$ 1
- where E = energy of photon(joule)
 h = planks constant- 6.626×10^{-34} J.s
 ν = frequency of photon(Hz)



Properties of photon

- A photon does not have any mass.
- A photon does not have any charge and are not deflected in electric field or magnetic field.
- All the quantum numbers are zero for a photon
- In empty space, the photon moves at speed of light.
- In the interaction of radiation with matter, radiation behaves as if it is made up of particles called photons.
- The energy and momentum of a photon are related as follows $E = p \cdot c$ where p - magnitude of momentum and c is the speed of light.
- Photon is called as a virtual particles.
- The energy of a photon is directly proportional to frequency and inversely proportional to its wavelength.

Photo electric effect



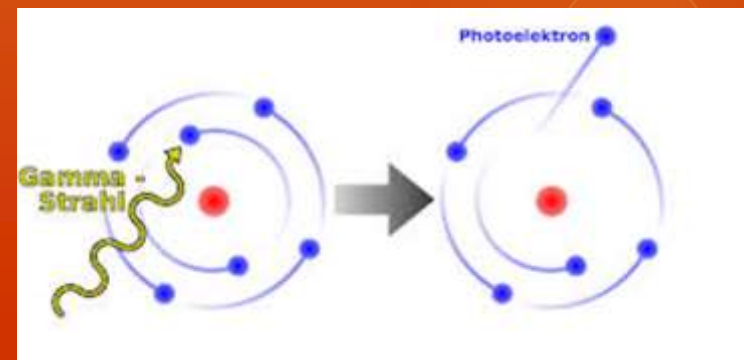
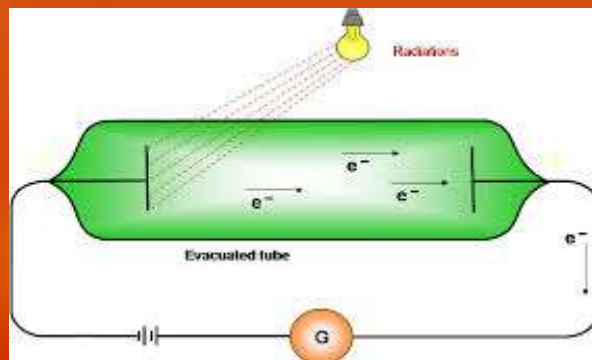
- When a beam of light of sufficiently high frequency onto a clean metal surface then the light will cause electrons to leave the surface.
- Definition : the phenomenon of emission of electrons by the metals when they are exposed to light of suitable frequency is called as the **photo electric effect** and emitted electrons is called as **photoelectrons**.

Photoelectric effect- construction and working

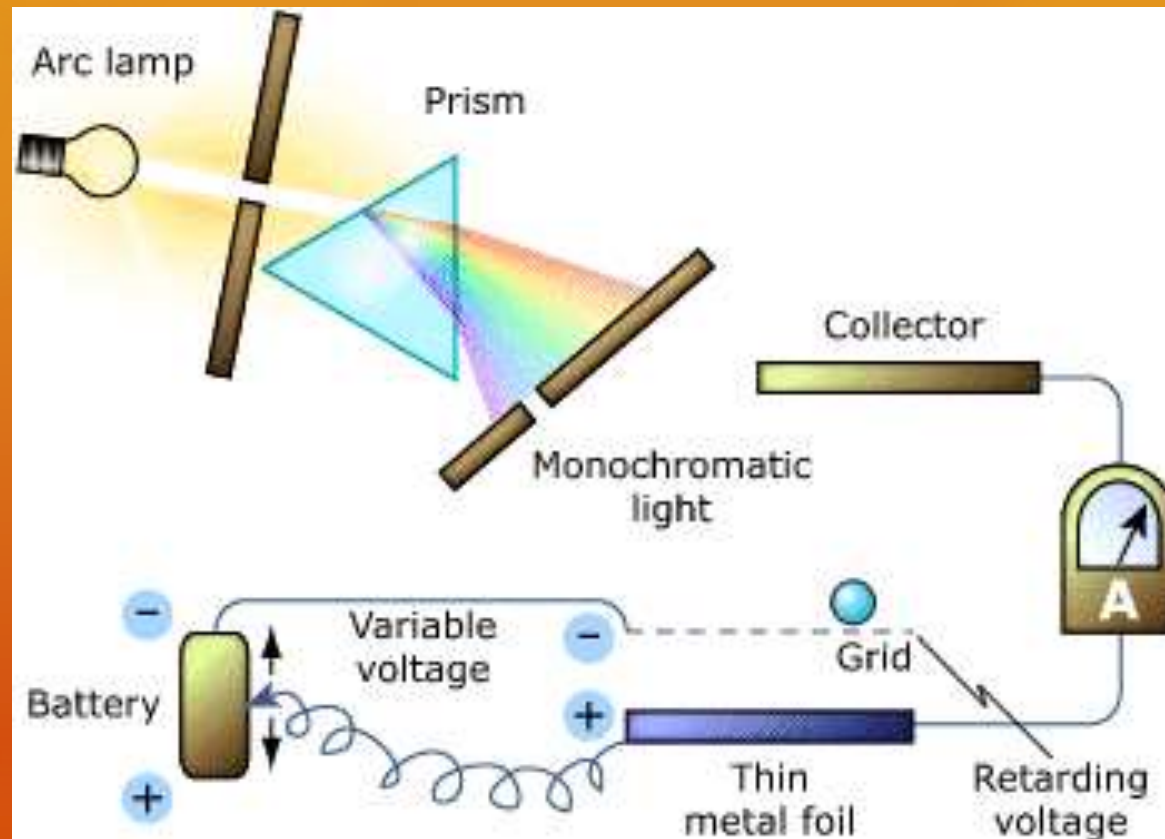
- Construction: 1) the Hertz experimental set up used for studying the photoelectric effect is shown in above fig.
- 2) the set up consist of an evacuated glass tube that has a photosensitive metal plate C and another metal plate A as shown.
- A monochromatic light source emerging from the source S of sufficiently short wavelength enters the glass window W and fall on photosensitive plate C, is called emmitter.

Photoelectric effect- working

- Working: when a beam of light fall on photosensitive metal plate c which is called emitter.
- The plate c emits photoelectrons due to photoelectric effect. The photo electrons emitted by plate c will be attracted towards the positive plate A. these electron flows in the external ckt to cause an electric current in the ckt.
- Such a current is known as the photoelectric current and measured by the micrometer connected in the ckts.



Lenard's Photoelectric Apparatus:

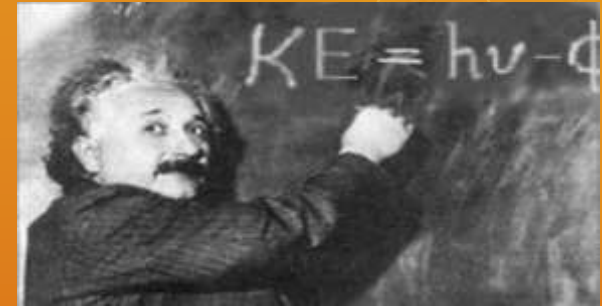


Characteristic of photoelectric effect

- **Photoelectric effect-** When light of suitable frequency is incident on metal surface then electrons are emitted from surface called as Photoelectric effect.
- **Characteristics of photoelectric effect-**
 - 1. Threshold frequency is different for different material.
 - 2. Photoelectric current is directly proportional to intensity of light.
 - 3. The K.E. of photoelectrons is directly proportional to frequency of light.
 - 4. Stopping potential is directly proportional to frequency.
 - 5. The process is instantaneous.

Einstein photoelectric equation

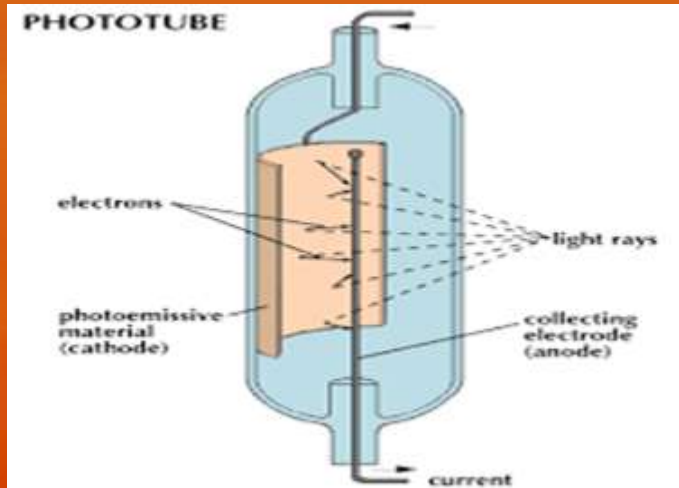
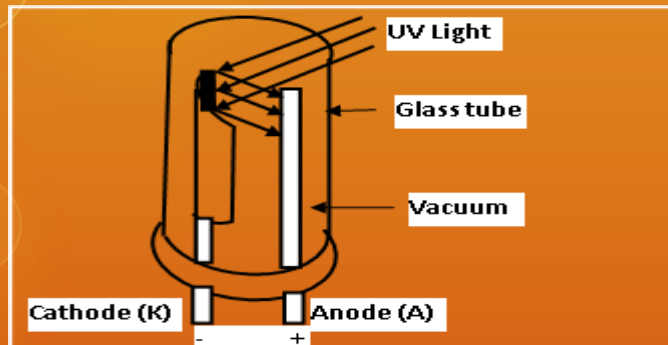
- **Einstein's photoelectric function-** According to quantum theory, radiation is considered as shower of particles called photons.
- Energy of photon absorbed by the atom ($h\nu$) is
 - 1. Used to detach the electron (W_0) and
 - 2. K.E. is given to electron.
- $h\nu = W_0 + \text{K.E.}$



- $h\nu = W_0 + \frac{1}{2}mv^2$
- $K.E. = h\nu - W_0$ Where, W_0 = Photoelectric work function = $h\nu_0$
- $K.E. = h\nu - h\nu_0$
- $K.E. = h(\nu - \nu_0)$
- Where, m = Mass of electron v = Velocity of electron h = Planck's constant
- ν = Frequency of radiation ν_0 = Threshold frequency
- **Significance-**
 - 1. If $\nu < \nu_0$ - Kinetic energy is negative. i.e. No emission.
 - 2. If $\nu = \nu_0$ - Kinetic energy is zero. i.e. Emission just begins.
 - 3. If $\nu > \nu_0$ - Kinetic energy is positive. i.e. Emission takes place.

Photo electric cell and LDR

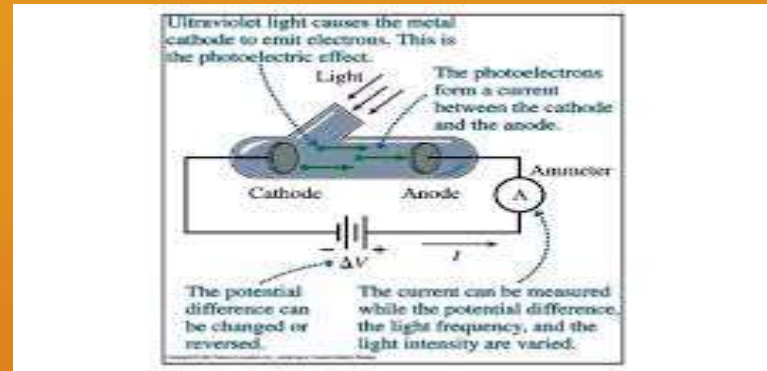
Photo electric cell



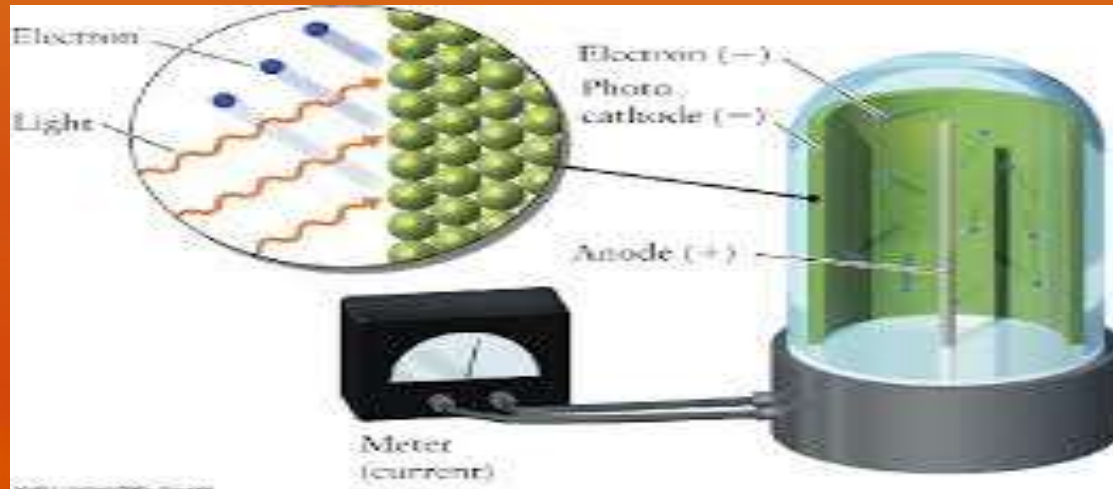
PRINCIPLE AND CONSTRUCTION

- Principle : a photoelectric cell is a device which converts the light energy into an electrical energy.
- Construction : it consists of an evacuated glass tube. Inside this tube photosensitive metal plate which is called emitter and a wire loop A which acts as collector.

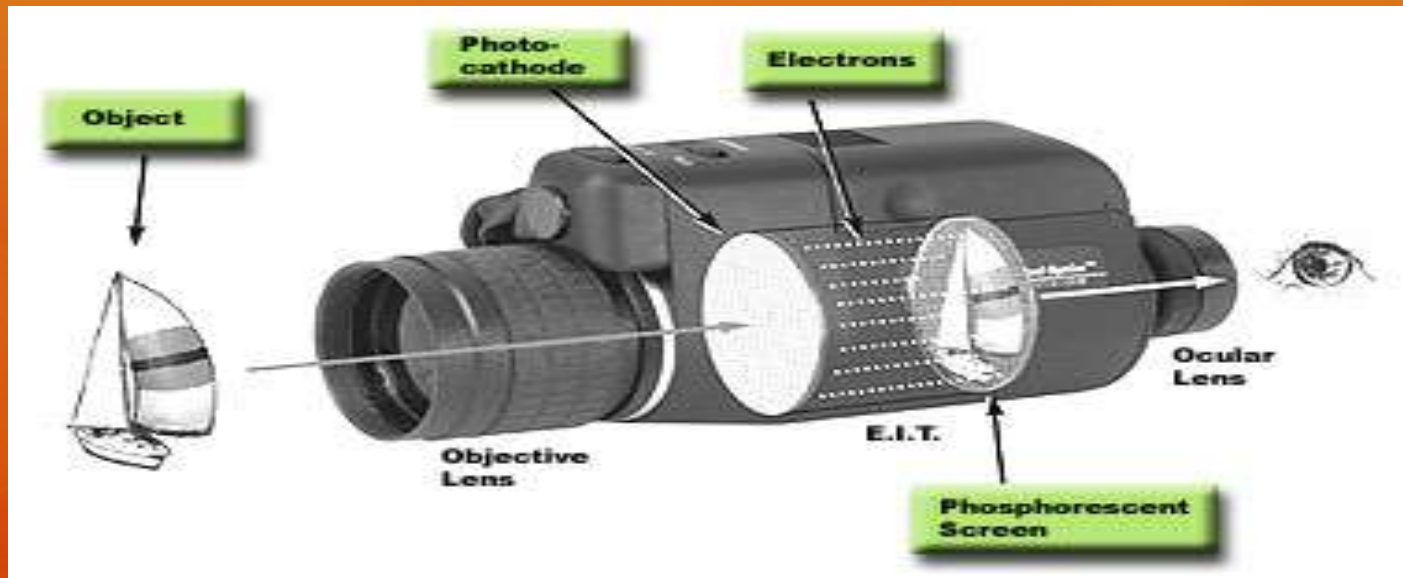
Working of photoelectric cell



- A high tension battery and micro ammeter (UA) are connected to plate C and loop A as shown in fig.
- Working : when the light of suitable frequency is incident on the emitter C, it start emitted the photoelectrons are attracted towards the collector A.



APPLICATION OF PHOTOELECTRIC EFFECT:- 1) photoelectric cell used in camera for exposure meter.



Photocell

- ☼ The photoelectric effect is commonly used to measure light.
 - Camera light meter
- ☼ It can also generate electricity.
 - Photovoltaic cell

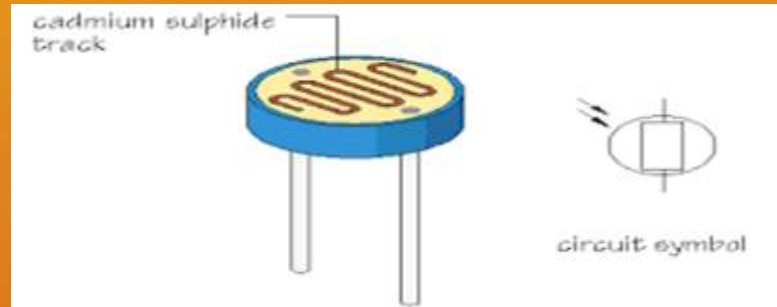


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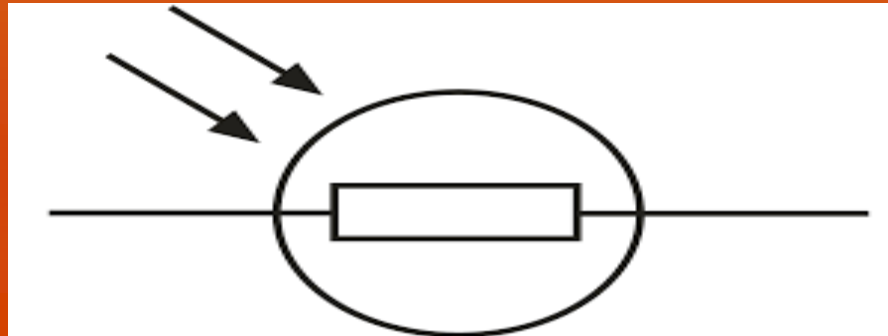
3) Photoelectric cell are also used for burglar alarm.



Photo resistor (light dependent resistor)



- Principle : A photoresistor or LDR is a resistor whose resistance decreases with increasing incident light intensity.
- Symbol for LDR



CONSTRUCTION AND OPERATION OF LDR

- Construction: a light sensitive material such as cadmium sulphide(cds) is deposited on a ceramic substrate.
- Then the substrate along with the photosensitive layer of cds is enclosed in a metal container.
- Light is incident on the light sensitive material through the glass cap or lens which is added at the top of this assembly.
- OPERATION :when light is incident on the photosensitive semiconductor material, the incident photons collide with the atoms of light sensitive material and impart energy to them.
- Due to this energy, the valence electron will cross the forbidden energy gap and enters into conduction band.
- Due to more number of electron entering the conduction band, conductivity increases and resistivity decreases. Thus resistance of the photosensitive material decreases with increases in intensity of light.

