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Reg. No. :

Name :

Sixth Semester B.Sc. Degree Examination, April 2018
Career Related First Degree Programme Under CBCSS
Physics with Computer Applications
Core Course
PC 1642: QUANTUM MECHANICS
(2014 Admission)

Time: 3 Hours

Max. Marks: 80

SECTION - A

Answer all questions in one or two sentences. Each question carries one mark.

- 1. Give Planck's distribution law for blackbody radiation.
- 2. What is meant by threshold frequency of photoelectric emission?
- 3. Write down the time-dependent Schrodinger equation.
- Give the expressions for position and momentum operators in one-dimensional space.
- 5. Is the wave function of a free particle with definite energy normalizable? Why?
- 6. Distinguish between phase velocity and group velocity.
- 7. Give the energy levels of a particle confined in an infinite square well.
- 8. Define Dirac delta function.
- 9. Explain transmission coefficient in connection with a finite potential well.
- 10. What is scattering matrix?

(10×1=10 Marks)

SECTION - B

Answer eight questions not exceeding a paragraph. Each question carries two marks.

- 11. Show that, in the low frequency limit, Planck's law gives Rayleigh-Jean's law.
- Describe how Einstein's photoelectric equation explains the laws of photoelectric effect.

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- 13. Briefly explain the Bohr atom model.
- 14. What is meant by expectation value of an observable? Give expectation values of position and momentum.
- 15. Discuss the correspondence principle.
- 16. Define probability current density. What is its value when the wave function is real?
- 17. State and explain Heisenberg's uncertainty principle.
- 18. Distinguish between a bound state and a scattering state.
- 19. Explain quantum mechanical tunneling.
- 20. Explain orthogonality of wave functions with examples.
- 21. What is zero point energy of a harmonic oscillator? Discuss how it originates from the uncertainty principle.
- 22. Explain symmetric (even) and anti-symmetric (odd) wave functions using the bound state solutions of a particle moving in a square potential well.

 (8x2=16 Marks)

SECTION - C

Answer any six questions. Each question carries four marks.

Useful data: Planck's constant h = 6.626 x 10⁻³⁴ Js;

mass of the electron m = 9.1×10^{-31} kg ; charge of the electron e = 1.6×10^{-19} C.

- Find the frequency and wavelength corresponding to a photon of energy 1keV.
- 24. Threshold frequency for photo electric emission in copper is 1.1×10^{15} Hz. Find the maximum kinetic energy of photoelectrons emitted when a radiation of frequency 1.5×10^{15} Hz strikes the copper surface.
- Find the de Broglie wavelengths of (a) 46g golf ball with speed 3 m/s. (b) electron
 moving with a speed of 10⁶ m/s.



- 26. Normalize the wave function $\psi(x) = e^{-\frac{\alpha^2 x^2}{2}}$ where α is a constant. Use $\int_{-\infty}^{\infty} e^{-ax^2} = \sqrt{\frac{\pi}{a}}.$
- 27. The ground state energy of a particle confined in an infinite square well potential is 34 eV. Find the energy of the particle in the first two excited levels.
- 28. Calculate the ground state energy of a harmonic oscillator vibrating with a frequency of 5 x 10¹⁴ Hz.
- 29. Calculate the expectation value of position in the ground state of a particle confined in an infinite potential well located in the region $0 \le x \le L$.
- 30. An electron is held in a square potential well of width 1 \times 10⁻¹⁰ m. Find the possible values of the well's depth V_0 such that there is exactly one bound energy state of the electron.
- 31. Determine the group velocity V_g and phase velocity V_{ph} associated with a free particle of mass m moving with a velocity v. (use non-relativistic relation)
 (6×4=24 Marks)

SECTION - D

Answer any two questions. Each question carries fifteen marks.

- Discuss the theory of the Compton effect and describe the experimental observations.
- 33. Discuss the statistical interpretation of the wave function. Describe the normalization of a wave function and show that it is independent of time.
- 34. From the time-dependent Schrodinger equation, deduce the time-independent Schrodinger equation. Discuss the general features of stationary states.
- 35. Obtain the energy eigenvalues of a simple harmonic oscillator. (2x15=30 Marks)

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Reg. No.:

Name :

Sixth Semester B.Sc. Degree Examination, April 2019
Career Related First Degree Programme under CBCSS
PHYSICS WITH COMPUTER APPLICATIONS
Core Course

PC 1642 : Statistical Mechanics and Quantum Mechanics (2015 Admission Onwards)

Time: 3 Hours

Max. Marks: 80

SECTION - A

Answer all the questions:

(10×1=10 Marks)

- 1. What is the nature of the particles which can be treated with B.E. Statistics?
- 2. Why the photo electrons do not have as much energy as the quantum of light which cases its ejection?
- 3. Give the characteristics of Black body radiation.
- 4. What do you mean by tree particle and bound particle?
- 5. What is degeneracy?
- 6. Is M.B. Statistics applicable to electron gas?
- 7. Which Statistics is followed by the following particles ? Electron, ideal gas molecules, proton, photon, helium atom, neutron, π meson, μ meson ?
- Both photoelectric and Compton effects arise due to the action of photons on electrons, but the two effects are not the same. Explain this.
- 9. What do you understand by the terms Eigen value and Eigen function ?
- 10. State the correspondence principle.

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SECTION - B



(8×2=16 Marks)

Answer any 8 questions:

- 11. Explain Micro-Canonical ensemble.
- 12. Can a particle with zero energy exist in the box ? What is zero point energy ?
- 13. What are Micro and Macro States?
- 14. What do you mean by Stationary States?
- 15. Obtain the relation between entropy and thermodynamic probability.
- 16. Distinguish Canonical ensemble and grand canonical ensemble.
- 17. Explain normalization of wave function.
- 18. What are the basic differences between Wien's law and Reyleigh-Jeans law of blackbody radiation?
- 19. Explain why an electron in Compton effect cannot be scattered at an angle greater than 90°.
- 20. Compare MB, BE and FD Statistics mentioning at least three characteristics.
- 21. State Bohr postulates.
- 22. Write the properties of a Wave function.

SECTION - C

Answer any six questions :

(6x4=24 Marks)

- 23. The work function of a metal is 3.45 eV. Find the maximum wavelength of photon that can eject an electron from the metal.
- 24. The life time of a typical excited state of an atom is 1×10⁻⁸ sec. The atom emits a photon of wavelength 6000 A°. What is the energy uncertainty of this photon? What is the wavelength uncertainty of this photon?
- 25. Obtain Schrodinger time dependent equation.

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- 26. A particle trapped is one dimensional infinite potential well of width L is given by $\psi = A \sin\left(\frac{n\pi x}{2}\right)$ in the region : $\begin{cases} x > 0 \\ x < L \end{cases}$. Find the normalization constant A.
- 27. Find out the possible arrangements of two particles A and B in three cells according to MB-Statistics.
- 28. Estimate the maximum error in the determination of velocity of an electron if it is constrained to more in the X-direction and if its X-coordinate is known with an uncertainty of 10⁻⁶ m.
- 29. Obtain the energy of electron in a Bohr orbit.
- 30. The photoelectric threshold for a metal is 3000 \mathring{A} . Find the K $\stackrel{\cdot}{E}$ of an electron ejected from it by radiation of wavelength 1200 \mathring{A} .
- 31. X-rays of wavelength 1.0 \mathring{A} are scattered from a carbon block. Find the wavelength of the scattered beam in a direction making 90° with the incident beam. What is the K-E of the recoiling electron? Given $h = 6.63 \times 10^{-34} \, \text{J-S}$, $e = 3 \times 10^8 \, \text{m/sec.}$ and $1 \text{ev} = 1.6 \times 10^{-19} \, \text{J.}$

SECTION - D

Answer any two questions:

(2×15=30 Marks)

- 32. Explain the phenomenon of BE condensation and show graphically how the condensate fraction varies with temperature ?
- 33. What is photoelectric effect? Give an account of Einstein's explanation of photoelectric effect on the basis of quantum theory.
- 34. Give an account of Heisenberg uncertainty principle. Outline an idealised experiment to bring out its significance.
- 35. Setup Schrodinger wave equation for a linear harmonic oscillator. Solve the equation and obtain its energy eigen values. Also deduce its zero point energy.