

REFERENCES and SOURCES for HITRAN

(Last undated: 26 August 2004)

The reference 0 (zero) is used for all data surviving from the 1986 HITRAN Database. For further details, refer to: L.S. Rothman, R.R. Gamache, A. Goldman, L.R. Brown, R.A. Toth, H.M. Pickett, R.L. Poynter, J.-M. Flaud, C. Camy-Peyret, A. Barbe, N. Husson, C.P. Rinsland, and M.A.H. Smith, "The HITRAN database: 1986 Edition," *Appl. Opt.* **26**, 4058-4097 (1987).

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H₂O 161, 181, 171, 162, 182, 172

Positions

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Halfwidths (self)

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<u>Temperature-dependence of air-broadened halfwidth</u>

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Pressure-shift (air)

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Halfwidths (air)

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- **1.** K. Chance, K.W. Jucks, D.G. Johnson, and W.A. Traub, "The Smithsonian Astrophysical Observatory Database SAO92," *JQSRT* **52**, 447-457 (1994).
- 2. N. Lacome, A. Levy, and G. Guelachvili, "Fourier transform measurement of self-, N₂-, and O₂-broadening of N₂O lines: temperature dependence of linewidths," *Appl.Opt.* 23, 425-434 (1984).
- 3. Third-oder polynomial fit of experimental results based on the three works: N. Lacome, A.

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Halfwidths (self)

1. Third-oder polynomial fit of experimental results based on: R.A. Toth, "Line strengths (900-3600 cm⁻¹), self-broadened linewidths, and frequency shifts (1800-2660 cm⁻¹) of N₂O," *Appl.Opt.* **32**, 7326-7365 (1993).

<u>Temperature-dependence of air-broadened halfwidth</u>

1. Fixed to a constant value of 0.75 based on the two works: N. Lacome, A. Levy, and G. Guelachvili, "Fourier transform measurement of self-, N₂-, and O₂-broadening of N₂O lines: temperature dependence of linewidths," *Appl. Opt.* **23**, 425-434 (1984); V. Nemtchinov, C. Sun, and P. Varanasi, "Measurements of Line Intensities and Line Widths in the v₃-fundamental Band of Nitrous Oxide at Atmospheric Temperatures," *JQSRT* **83**, 267-284 (2004).

Pressure-shift (air)

1. R.A. Toth, "Linelist of N₂O parameters from 500 to 7500 cm⁻¹," to be published (2004).

CO 26, 36, 28, 27, 38, 37

Positions

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Halfwidths (air)

- 1. T. Nakazawa and M. Tanaka, "Measurements of Intensities and Self- and Foreign gas broadened halfwidths of spectral Lines in the CO fundamental Band," JQSRT **28,** 409-416 (1982); values for transitions having $20 < |m| \le 33$ are extrapolated, while those for |m| > 34 are assumed to be constant (0.0400 cm⁻¹/atm).
- **2.** Polynomial fit of several measurements (M.A.H. Smith, private communication, 2004). For details, see L.S. Rothman, D. Jacquemart, A. Barbe, D.C. Benner, J.W. Brault, J.-P. Bouanich, L.R. Brown, Jr.C. Chackerian, C. Camy-Peyret, M.R. Carleer, V. Dana, V.M. Devi, A. Fayt, J.-M. Flaud, R.R. Gamache, A. Goldman, K.W. Jucks, A.G. Maki, J.-Y. Mandin, M.-F. Mérienne, C.E. Miller, A. Perrin, H.M. Pickett, C.P. Rinsland, M.A.H. Smith, B. Sumpf, Tashkun Sa, R.A. Toth, J. Vander Auwera, and P. Varanasi, "The HITRAN 2004 Molecular Spectroscopic Database," *JQSRT* in preparation.

Halfwidths (self)

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Perrin, H.M. Pickett, C.P. Rinsland, M.A.H. Smith, B. Sumpf, Tashkun Sa, R.A. Toth, J. Vander Auwera, and P. Varanasi, "The HITRAN 2004 Molecular Spectroscopic Database," *JQSRT* in preparation.

Temperature-dependence of air-broadened halfwidth

1. Polynomial fit of several measurements (M.A.H. Smith, private communication, 2004). For details, see L.S. Rothman, D. Jacquemart, A. Barbe, D.C. Benner, J.W. Brault, J.-P. Bouanich, L.R. Brown, Jr.C. Chackerian, C. Camy-Peyret, M.R. Carleer, V. Dana, V.M. Devi, A. Fayt, J.-M. Flaud, R.R. Gamache, A. Goldman, K.W. Jucks, A.G. Maki, J.-Y. Mandin, M.-F. Mérienne, C.E. Miller, A. Perrin, H.M. Pickett, C.P. Rinsland, M.A.H. Smith, B. Sumpf, Tashkun Sa, R.A. Toth, J. Vander Auwera, and P. Varanasi, "The HITRAN 2004 Molecular Spectroscopic Database," *JQSRT* in preparation.

Pressure-shift (air)

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CH₄ 211, 311, 212

Positions

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- **10.** J.S. Margolis, "Measured line positions and strengths of methane between 5500 and 6180 cm⁻¹," *Appl.Opt.* **27,** 4038-4051 (1988); J.S. Margolis, "Empirical values of the ground state energies for methane transitions between 5500 and 6150 cm⁻¹," *Appl.Opt.* **29,** 2295-2302 (1990).
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Halfwidths (self)

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Temperature-dependence of air-broadened halfwidth

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Pressure-shift (air)

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- **4.** Molecular Hamiltonian constants for the v''=0 and v'=1 of the $X\Sigma$ state are from G. Rouillé et al (Ref. 1). Vibrational term values are from Krupenie (Ref. 2), upper vibrational state energy shifted to agree with band center of G. Rouillé et al for $(1 \leftarrow 0)$ band.
- 5. Molecular Hamiltonian constants for the v''=1 and v'=1 of the $X\Sigma$ state are from G. Rouillé et al (Ref. 1). Vibrational term values are from P.H. Krupenie (Ref. 2), upper vibrational state energy shifted to agree with band center of G. Rouillé et al for $(1 \leftarrow 0)$ band.
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- 23. Molecular Hamiltonian constants for the v''=0 state of $X\Sigma$ are from G. Rouillé et al (Ref. 1). For the v'=2 of the $b\Sigma$ state, the constants of D.L. Albritton (Ref. 18) are used. Vibrational and electronic term values are from Krupenie (Ref. 2). Dunham zero point energy correction applied to zero point energy of upper vibrational state.
- 24. Molecular Hamiltonian constants for the v''=1 state of $X\Sigma$ are from G. Rouillé et al (Ref. 1). For the v'=1 of the $b\Sigma$ state, the constants of D.L. Albritton (Ref. 18) are used. Vibrational and electronic term values are from Krupenie (Ref. 2). Dunham zero point energy correction applied to zero point energy of upper vibrational state.
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- **26.** Molecular Hamiltonian constants for the v''=0 state of $X\Sigma$ are from M. Mizushima and S. Yamamoto (Ref. 6). For the v'=0 of the $b\Sigma$ state, the constants of H. Babcock and L. Herzberg (Ref. 19) are used. Vibrational and electronic term values are from Krupenie (Ref. 2). Dunham zero point energy correction applied to zero point energy of upper vibrational state. Electronic term value is shifted by (-0.041-0.014) cm⁻¹, unknown reference.
- 27. Molecular Hamiltonian constants for the v''=0 state of $X\Sigma$ are from M. Mizushima and S. Yamamoto (Ref. 6). For the v'=1 of the $b\Sigma$ state, the constants of W.S. Benedict (Ref. 20) are used. Vibrational and electronic term values are from P.H. Krupenie (Ref. 2). Dunham zero point energy correction applied to zero point energy of upper vibrational state. Electronic term value is shifted by (-0.041-0.014) cm⁻¹, unknown reference.
- 28. Molecular Hamiltonian constants for the v''=0 state of $X\Sigma$ are from M. Mizushima and S. Yamamoto (Ref. 6). For the v'=2 of the $b\Sigma$ state, the constants from W.S. Benedict, University of Maryland (private communication) are used. Vibrational and electronic term values are from

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Halfwidths (self)

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Halfwidths (air)

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Halfwidths (self)

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OH 61, 81, 62

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Intensities

- **0.** L.S. Rothman, R.R. Gamache, A. Goldman, L.R. Brown, R.A. Toth, H.M. Pickett, R.L. Poynter, J.-M. Flaud, C. Camy-Peyret, A. Barbe, N. Husson, C.P. Rinsland, and M.A.H. Smith, "The HITRAN database: 1986 Edition," *Appl. Opt.* **26,** 4058-4097 (1987).
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Halfwidths (air)

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Temperature-dependence of air-broadened halfwidth

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HCl 15, 17

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Halfwidths (air)

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Halfwidths (self)

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Pressure-shift (air)

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HBr 19, 11

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Halfwidths (air)

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HI 17

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Halfwidths (air)

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ClO 56, 76

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OCS 622, 624, 632, 623, 822

Positions

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CH₃Cl 215, 217

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PH₃ 1111

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SF₆ 29

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H₂S 121,141,131

Positions

- 1. A. Goldman and J.R. Gillis, "Line Parameters and Line by Line Calculations for Molecules of Stratospheric Interest," University of Denver Progress Report (1984).
- **2.** J.-M. Flaud, C. Camy-Peyret, and J.W.C. Johns, "The far infrared spectrum of hydrogen sulfide. The (000) rotational constants of H₂³²S, H₂³³S and H₂³⁴S," *Can.J.Phys.* **61,** 1462-1473 (1983).
- 3. L. Sinitsa, Institute of Atmospheric Optics (Tomsk), private communication (1994).
- **4.** L. Lechuga-Fossat, J.-M. Flaud, C. Camy-Peyret, and J.W.C. Johns, "The spectrum of natural hydrogen sulfide between 2150 and 2950 cm⁻¹," *Can.J.Phys.* **62**, 1889-1923 (1984).
- **5.** L.R. Brown, J.A. Crisp, D. Crisp, V. Naumenko, M.A. Smirnov, L.N. Sinitsa, and A. Perrin, "The Absorption Spectrum of H_2S between 2150 and 4260 cm⁻¹: Analysis of the Positions and Intensities in the First [$2v_2$, v_1 and v_3] and Second [$3v_2$, $v_1 + v_2$ and $v_2 + v_3$] Triad Region," *J.Mol. Spectrosc.* **188**, 148-174 (1998).

Intensities

- 1. A. Goldman and J.R. Gillis, "Line Parameters and Line by Line Calculations for Molecules of Stratospheric Interest," University of Denver Progress Report (1984).
- **2.** J.-M. Flaud, C. Camy-Peyret, and J.W.C. Johns, "The far infrared spectrum of hydrogen sulfide. The (000) rotational constants of H₂³²S, H₂³³S and H₂³⁴S," *Can.J.Phys.* **61,** 1462-1473 (1983).
- 3. L. Sinitsa, Institute of Atmospheric Optics (Tomsk), private communication (1994).
- **4.** L. Lechuga-Fossat, J.-M. Flaud, C. Camy-Peyret, and J.W.C. Johns, "The spectrum of natural hydrogen sulfide between 2150 and 2950 cm⁻¹," *Can.J.Phys.* **62**, 1889-1923 (1984).
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Halfwidths (air)

- 1. A. Goldman and J.R. Gillis, "Line Parameters and Line by Line Calculations for Molecules of Stratospheric Interest," University of Denver Progress Report (1984).
- **2.** J. Waschull, F. Kuhnemann, and B. Sumpf, "Self-, air- and Helium Broadening of the v_2 band of H₂S," *J.Mol.Spectrosc.* **165**, 150-158 (1994).
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- **4.** A. Kissel, B. Sumpf, H.-D. Kronfeldt, B.A. Tikhomirov, and Yu.N. Ponomarev, "Molecular-Gas-Pressure-Induced Line-Shift and Line-Broadening in the ν₂-Band of H₂S," *J.Mol.Spectrosc.* **216**, 345-354 (2002).
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Halfwidths (self)

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Pressure-shift (air)

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HCOOH 126

Positions

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- 2. A. Perrin, C.P. Rinsland, and A. Goldman, "Spectral parameters for the v_6 region of HCOOH and its measurement in the infrared tropospheric spectrum," *J. Geophys. Res.* **104**, 18,661-18,666 (1999).
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Halfwidths (air)

1. A. Goldman and J.R. Gillis, "Line Parameters and Line-by-line Calculations for Molecules of Stratospheric Interest," Progress Report, Dept. of Physics, Univ. Denver (1984).

Halfwidths (self)

1. A. Perrin, C.P. Rinsland, and A. Goldman, "Spectral parameters for the v_6 region of HCOOH and its measurement in the infrared tropospheric spectrum," *J. Geophys. Res.* **104**, 18,661-18,666 (1999).

Temperature-dependence of air-broadened halfwidth

1. A. Goldman, private communication (1996).

HO₂ 166

Positions

- **1.** C. Yamada, Y. Endo, and E. Hirota, "Difference frequency laser spectroscopy of the v_1 band of the HO₂ radical," *J. Chem. Phys.* **78**, 4379-4384 (1983).
- **2.** K. Nagai, Y. Endo, and E. Hirota, "Diode Laser Spectroscopy of the HO₂ v₂ Band," *J.Mol.Spectrosc.* **89**, 520-527 (1981).
- **3.** D.D. Nelson, Jr., and M.S. Zahniser, "Diode Laser Spectroscopy of the ν₃ Vibration of the HO₂ Radical," *J.Mol.Spectrosc.* **150**, 527-534 (1991).
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Halfwidths (air)

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<u>Positio</u>ns

1. L.R. Zink, K.M. Evenson, F. Matsuchima, T. Nelis, and R. L. Robinson, "Atomic oxygen fine-structure splittings with tunable far-infrared spectroscopy," *Astrophys. J.* **371,** L85 (1991).

Intensities

1. H.M. Pickett, R.L. Poynter, E.A. Cohen, M.L. Delitsky, J.C. Pearson, and H.S.P. Müller, "Submillimeter, Millimeter, and Microwave Spectral Line Catalog," JPL Publication 800-23, rev. 4 (1996).

Halfwidths (air)

1. Does not have the standard HITRAN definition of Lorentz air broadening, but a default value of 0.05 cm⁻¹/atm was appended.

CIONO₂ 5646, 7646

Positions

1. W. Bell, G. Duxbury, and D.D. Stuart, "High-Resolution Spectra of the ν₄ Band of Chlorine Nitrate," *J.Mol.Spectrosc.* **152**, 283-297 (1992); A. Goldman, C.P. Rinsland, F.J. Murcray, R.D. Blatherwick, and D.G. Murcray, "High Resolution Studies of Heavy NOy Molecules in Atmospheric Spectra," *JQSRT* **52**, 367-377 (1994).

Intensities

1. A. Goldman, C.P. Rinsland, F.J. Murcray, R.D. Blatherwick, and D.G. Murcray, "High Resolution Studies of Heavy NOy Molecules in Atmospheric Spectra," *JQSRT* **52**, 367-377 (1994).

Halfwidths (air)

1. A. Goldman, C.P. Rinsland, F.J. Murcray, R.D. Blatherwick, and D.G. Murcray, "High Resolution Studies of Heavy NOy Molecules in Atmospheric Spectra," *JQSRT* **52**, 367-377 (1994).

NO⁺ 46

Positions

1. Positions based on a fit by D.R. Smith, AF Phillips Lab, using data of F.P. Billingsley, *Chem.Phys.Lett.* 23, 160-166 (1973), K.P. Huber and G. Herzberg, "Molecular Spectra and Molecular Structure IV. Constants of Diatomic Molecules," Van Nostrand Reinhold Co., NY (1979), and D.R. Smith, E.R. Huppi, and R.M. Nadile, "Improved Rotational Constants for the Ground Electronic State of NO⁺ from Atmospheric Emission Spectra," in preparation; D.R. Smith, E.R. Huppi, and J.O. Wise, "Observation of highly rotationally excited NO⁺ emissions in the themosphere," *J.Atmos.Solar-Terrestrial Phys.* 62, 1189-1198 (2000).

Intensities

1. H.-J. Werner and P. Rosmus, "Ab Initio Calculations of Radiative Transition Probabilities in the $X^1\Sigma^+$ Ground State of the NO⁺ Ion," *J.Mol.Spectrosc.* **96**, 362-367 (1982).

Halfwidths (air)

1. Default value of 0.06 cm⁻¹/atm chosen, but applications are most likely not required to work in Lorentzian regime.

HOBr 169, 161

Positions

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Intensities

1. Y. Koga, H. Takeo, S. Kondo, M. Sugie, C. Matsumura, G.A. Rae, and E.A. Cohen, "The Rotational Spectra, Molecular Structure, Dipole Moment, and Hyperfine Constants of HOBr and DOBr," *J.Mol.Spectrosc.* **138**, 467-481 (1989).

Halfwidths (air)

1. A constant value of $0.06 \text{ cm}^{-1}/\text{atm}$ has been assumed for the air-broadened halfwidth with a temperature-dependence coefficient n = 0.67.

C_2H_4 221, 231

J.Mol.Spectrosc. 189, 64-73 (1998).

Positions

- 1. I. Cauuet, J. Walrand, G. Blanquet, A. Valentin, L. Henry, Ch. Lambeau, M. DeVleeschouwer, and A. Fayt, "Extension to Third-Order Coriolis Terms of the Analysis of ν₁₀, ν₇, and ν₄ Levels of Ethylene on the Basis of Fourier Transform and Diode Laser Spectra," *J.Mol.Spectrosc.* 139, 191-214 (1990); J. Legrand, M. Azizi, F. Herlemont, and A. Fayt, "Saturation Spectroscopy of C₂H₄ Using a CO₂ Laser Sideband Spectrometer," *J.Mol.Spectrosc.* 171, 13-21 (1995); E. Rusinek, H. Fichoux, M. Khelkhal, F. Herlemont, J. Legrand, and A. Fayt, "Subdoppler study of the ν₇ band of C₂H₄ with a CO₂ Laser Sideband Spectrometer,"
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Intensities

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Halfwidths (air)

1. J.F. Brannon, Jr. and P. Varanasi, A Tunable Diode Laser Measurements on the 951.7393 cm⁻¹ Line of ¹²C₂H₄ at Planetary Atmospheric Temperatures," *JQSRT* **47**, 237-242 (1992).

Halfwidths (self)

Temperature-dependence of air-broadened halfwidth

1. J.F. Brannon, Jr. and P. Varanasi, A Tunable Diode Laser Measurements on the 951.7393 cm⁻¹ Line of ¹²C₂H₄ at Planetary Atmospheric Temperatures," *JQSRT* **47**, 237-242 (1992).

CH₃OH 2161

Positions

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- **2.** H.S.P. Müller, S. Thorwirth, D.A. Roth, G Winnewisser, "The Cologne Database for Molecular Spectroscopy, CDMS," *A&A* **370**, L49-L52 (2001).

Intensities

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Halfwidths (air)

1. L.H. Xu, R.M. Lees, P. Wang, L.R. Brown, I. Kleiner, J.W.C. Johns, "New assignments, line intensities and HITRAN database for CH₃OH at 10 μm," *J.Mol.Spectrosc.* (in press).

Halfwidths (self)

1. L.H. Xu, R.M. Lees, P. Wang, L.R. Brown, I. Kleiner, J.W.C. Johns, "New assignments, line intensities and HITRAN database for CH₃OH at 10 μm," *J.Mol.Spectrosc.* (in press).

Temperature-dependence of air-broadened halfwidth

1. L.H. Xu, R.M. Lees, P. Wang, L.R. Brown, I. Kleiner, J.W.C. Johns, "New assignments, line intensities and HITRAN database for CH₃OH at 10 μm," *J.Mol.Spectrosc.* (in press).

** Cross-section files **

- **1.** S.T. Massie, A. Goldman, D.G. Murcray, and J.C. Gille, "Approximate absorption cross sections of F12, F11, ClONO₂, N₂O₅, HNO₃, CCl₄, CF₄, F21, F113, F114, and HNO₄," *Appl.Opt.* **24,** 3426-3427 (1985).
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- **4.** J.J. Orlando, G.S. Tyndall, A. Huang, and J.G. Calvert, "Temperature Dependence of the Infrared Absorption Cross Sections of Carbon Tetrachloride," *Geophys.Res.Lett.* **19,** 1005-1008 (1992).
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- **11.** P. Varanasi and V. Nemtchinov, "Thermal Infrared Absorption Coefficients of CFC-12 at Atmospheric Conditions," *JQSRT* **51**, 679-687 (1994).
- **12.** K. Smith, D. Newnham, M. Page, J. Ballard, and G. Duxbury, "Infrared Absorption Cross-sections and Integrated Absorption Intensities of HCF-134 and HCF-143a Vapour," *JQSRT* **59**, 437-451 (1998).
- **13.** K. Smith, D. Newnham, M. Page, J. Ballard, and G. Duxbury, "Infrared Band Strengths and Absorption Cross-Sections of HFC-32 Vapour," *JQSRT* **56**, 73-82 (1996).
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- A. Jenouvrier, and B. Coquart, "Measurements of the NO₂ absorption cross-section from 42000 cm⁻¹ to 10000 cm⁻¹ (238-1000 nm) at 220 K and 294 K," *JOSRT* **59**, 171-184 (1997).
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- **21.** V. Nemtchinov and P. Varanasi, "Absorption cross-sections of HFC-134a in the spectral region between 7 and 12 μm," *JQSRT* **83**, 243-265 (2004).
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- 24. C.P. Rinsland, S.W. Sharpe, and R.L. Sams, "Temperature-dependent cross-sections in the thermal infrared bands of SF₅CF₃," *JQSRT* 82,483-490 (2003).
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Isotopic Abundances Used for HITRAN

[based on P. De Bievre, M. Gallet, N.E. Holden, and I.L. Barnes, "Isotopic Abundances and Atomic Weights of the Elements," *J. Phys. Chem. Ref. Data* 13, 809-891 (1984)]

Molecule	Isotopologue	Abundance	Molecule	Isotopologue	Abundance
H₂O (1)	161	0.997317	CO (5)	26	0.98654
	181	0.00199983		36	0.01108
	171	0.000372		28	0.0019782
	162	0.00031069		27	0.000368
	182	0.000000623		38	0.00002222
	172	0.000000116		37	0.00000413
CO ₂ (2)	626	0.98420	CH ₄ (6)	211	0.98827
	636	0.01106		311	0.01110
	628	0.0039471		212	0.00061575
	627	0.000734		66	0.995262
	638	0.00004434	$\mathbf{O}_{2}\left(7\right)$	68	0.00399141
	637	0.00000825		67	0.000742
	828	0.0000039573		46	0.993974
	728	0.00000147	NO (8)	56	0.0036543
O ₃ (3)	666	0.992901		48	0.00199312
	668	0.00398194	SO (0)	626	0.94568
	686	0.00199097	$SO_2(9)$	646	0.04195
	667	0.000740	$NO_2(10)$	646	0.991616
	676	0.000370		4111	0.9958715
N ₂ O (4)	446	0.990333	NH ₃ (11)	5111	0.0036613
	456	0.0036409	HNO ₃ (12)	146	0.989110
	546	0.0036409		61	0.997473
	448	0.00198582	ОН (13)	81	0.00200014
	447	0.000369		62	0.00015537

Molecule	Isotopologue	Abundance	Molecule	Isotopologue	Abundance
HF (14)	19	0.99984425	CH Cl (24)	215	0.74894
HCl (15)	15	0.757587	CH ₃ Cl (24)	217	0.23949
	17	0.242257	$H_2O_2(25)$	1661	0.994952
HBr (16)	19	0.50678	CH (20)	1221	0.97760
	11	0.49306	$\mathbf{C_2H_2}$ (26)	1231	0.02197
HI (17)	17	0.99984425	$C_2H_6(27)$	1221	0.97699
CIO (18)	56	0.75591	PH ₃ (28)	1111	0.99953283
	76	0.24172	COF ₂ (29)	269	0.98654
OCS (19)	622	0.93739	SF₆ (30)	29	0.95018
	624	0.04158		121	0.94988
	632	0.01053	$H_2S(31)$	141	0.04214
	623	0.007399		131	0.007498
	822	0.001880	НСООН (32)	126	0.983898
H ₂ CO (20)	126	0.98624	HO ₂ (33)	166	0.995107
	136	0.01108	O (34)	6	0.997628
	128	0.0019776	CIONO (25)	5646	0.74957
HOCl (21)	165	0.75579	ClONO ₂ (35)	7646	0.23970
	167	0.24168	NO ⁺ (36)	46	0.993974
N_2 (22)	44	0.9926874	HOD (27)	169	0.5056
HCN (23)	124	0.98511	HOBr (37)	161	0.4919
	134	0.01107	CH (29)	221	0.9773
	125	0.0036217	$C_2H_4(38)$	231	0.02196
			СН ₃ ОН (39)	2161	0.98593

Uncertainty Codes used in HITRAN Database

Line position and Pressure shift (cm ⁻¹)		Intensity, Halfwidths, and Temperature-dependence		
Code	Uncertainty Range	Code	Uncertainty Range	
0	≥1. or Unreported	0	Unreported or Unavailable	
1	≥0.1 and <1.	1	Default or Constant	
2	≥0.01 and <0.1	2	Average or Estimate	
3	≥0.001 and <0.01	3	≥ 20%	
4	≥0.0001 and <0.001	4	≥ 10% and < 20%	
5	≥0.00001 and <0.0001	5	≥ 5% and < 10%	
6	<0.00001	6	≥ 2% and < 5%	
		7	≥ 1% and < 2%	
		8	< 1%	