

CHEM 238 Organic Chemistry

Ch10 Alcohol, Thiol Reactions

Reaction	Reactant	Condition	Product	Overall
Dehydration of alcohol	alcohol	phosphoric acid H ₃ PO ₄	alkene, water	$\text{ROH} \xrightarrow{\text{H}_3\text{PO}_4} \text{alkene} + \text{H}_2\text{O}$
Alcohol and HX	alcohol, hydrogen halide		alkyl halide, water	$\text{ROH} + \text{HX} \longrightarrow \text{RX} + \text{H}_2\text{O}$
Alcohol and sulfonate ester TsCl	alcohol, sulfonate ester TsCl, ROTs	pyridine, NaBr	alkyl bromide	$\text{ROH} + \text{TsCl} \xrightarrow{\text{pyridine}} \text{R-OTs} \xrightarrow{\text{NaBr}} \text{RBr}$
Alkoxide and inorganic acid	alkoxide, inorganic acid (dimethyl sulfate)		ester, conjugate base	$\text{RO}^- + (\text{CH}_3)_2\text{SO}_4 \longrightarrow \text{ROCH}_3 + \text{CH}_3\text{SO}_4^-$
Alcohol and SOCl ₂	alcohol, thionyl chloride SOCl ₂	pyridine	alkyl chloride, sulfate dioxide, hydrogen chloride	$\text{ROH} + \text{SOCl}_2 \xrightarrow{\text{pyridine}} \text{RCl} + \text{SO}_2 + \text{HCl}$
Alcohol and Ph ₃ PBr ₂	alcohol, triphenylphosphine dibromide Ph ₃ PBr ₂	DMF	alkyl bromide, base	$\text{ROH} + \text{Ph}_3\text{PBr}_2 \xrightarrow{\text{DMF}} \text{RBr} + \text{Ph}_3\text{P}^+\text{O}^- + \text{HBr}$
Oxidation of secondary 2nd alcohol	secondary alcohol, Cr(VI) (PCC)		ketone, Cr(III)	$\text{ROH} + \text{Cr(VI)} (\text{CrO}_3, \text{pyridine}) \longrightarrow \text{R=O} + \text{Cr(III)}$
Oxidation of primary 1st alcohol	primary alcohol	Cr ₂ O ₄ ⁻	aldehyde, carboxylic acid	$\text{RCH}_2\text{OH} \xrightarrow[\text{H}_2\text{SO}_4]{\text{Cr}_2\text{O}_4^-} \text{RCH=O} \xrightarrow[\text{H}_2\text{SO}_4]{\text{Cr}_2\text{O}_4^-} \text{RC-OH=O}$
Controlled oxidation of primary 1st alcohol	primary alcohol	PCC	aldehyde	$\text{RCH}_2\text{OH} \xrightarrow[\text{CH}_2\text{Cl}_2]{\text{PCC}} \text{RCH=O}$
Oxidation of primary alcohol	primary alcohol, potassium permanganate KMnO ₄		carboxylic acid's conjugate base	$\text{RCH}_2\text{OH} + \text{KMnO}_4 + \text{OH}^- \longrightarrow \text{MnO}_2 + \text{RCOO}^- \xrightarrow{\text{H}_3\text{O}^+} \text{RCOOH}$
Oxidation of thiols/disulfide	thiols/disulfide, KMnO ₄ /HNO ₃		sulfonic acid	$\text{RSH} \xrightarrow{\text{HNO}_3} \text{RSO}_3\text{H}$
Oxidation of thiols	thiol, oxygen O ₂ /iodine I ₂ /bromine Br ₂		disulfide	$2 \text{RSH} + \text{I}_2 + 2 \text{NaOH} \longrightarrow \text{RSSR} + 2 \text{NaI} + 2 \text{H}_2\text{O}$

| Synthesis of alcohol

Reaction	Reactant	Condition	Product	Overall
Hydroboration-oxidation	alkene	HBR ₂ , H ₂ O ₂ /OH ⁻	alcohol	$\text{alkene} \xrightarrow[\text{(2) H}_2\text{O}_2/\text{OH}^-]{\text{(1) HBR}_2} \text{ROH}$
Oxymercuration-reduction	alkene	Hg(OAc) ₂ /H ₂ O, NaBH ₂ /NaOH	alcohol	$\text{alkene} \xrightarrow[\text{(2) NaBH}_2/\text{NaOH}]{\text{(1) Hg(OAc)}_2/\text{H}_2\text{O}} \text{ROH}$

Reaction	Reactant	Condition	Product	Overall
Acid-catalyzed hydration of alkene	alkene, water H ₂ O	acid	alcohol	alkene + H ₂ O $\xrightarrow{\text{H}_3\text{O}^+}$ ROH

Ch11 Ether, Sulfide, Epoxide, Glycol

| Synthesis, Cleavage of ether

Reaction	Reactant	Condition	Product	Overall
Williamson ether synthesis	alcohol	NaH, THF	ether	$\text{ROH} \xrightarrow[\text{THF}]{\text{NaH}} \text{RO}^- + \text{Na}^+ + \text{H}_2 \xrightarrow{\text{XR}'} \text{ROR}' + \text{NaX} + \text{H}_2$
Williamson sulfide synthesis	thiol	ROTs	sulfide	$\text{RSH} \xrightarrow[\text{CH}_3\text{OH}]{\text{OH}^-} \text{RS}^- + \text{H}_2\text{O} \xrightarrow{\text{R}'\text{OTs}} \text{RSR}' + \text{H}_2\text{O} + \text{OTs}^-$
Alkoxymercuration-reduction	alkene	Hg(OAc) ₂ /HOR, NaBH ₄	ether	alkene $\xrightarrow[(2) \text{NaBH}_4]{(1) \text{Hg}(\text{OAc})_2/\text{HOR}'} \text{ROR}'$
Primary 1st alcohol dehydration	primary alcohol	sulfuric acid H ₂ SO ₄ , heat	ether, water H ₂ O	$2 \text{ROH} \xrightarrow{\text{H}_2\text{SO}_4, \Delta} \text{ROR} + \text{H}_2\text{O}$
Tertiary 3rd alcohol dehydration	tertiary alcohol, primary alcohol	sulfuric acid H ₂ SO ₄	ester, water H ₂ O	$\text{ROH} + \text{R}'\text{OH} \xrightarrow{\text{dilute H}_2\text{SO}_4} \text{ROR}' + \text{H}_2\text{O}$
Alkene addition by tertiary alcohol	alkene, tertiary alcohol	sulfuric acid H ₂ SO ₄	ether	alkene + ROH $\xrightarrow{\text{dilute H}_2\text{SO}_4} (\text{CH}_3)_3\text{COR};$
Ether cleavage	ether, hydrogen halide	heat	alcohol, alkyl halide	$\text{ROR}' + \text{HX} \xrightarrow{\Delta} \text{ROH} + \text{R}'\text{X}$

| Synthesis of epoxide

Reaction	Reactant	Condition	Product	Overall
Oxidation of alkene with RCOOOH	alkene, peroxycarboxylic acid RCOOOH (MCPBA)		epoxide, carboxylic acid	alkene + RCOOOH (MCPBA) \longrightarrow epoxide + RCOOH
Cyclization of halohydrin	halohydrin (alcohol + alkyl halide RX), base		epoxide, water H ₂ O	halohydrin (OH–C–C–X) + NaOH \longrightarrow epoxide + NaX + H ₂ O

| Ring opening of epoxide

Reaction	Reactant	Condition	Product	Overall
Base ring opening	epoxide, alcohol	alkoxide	alcohol-ether	epoxide + R'OH $\xrightarrow{\text{R}'\text{O}^-}$ ester–alcohol
Acid ring opening	epoxide, alcohol	acid	alcohol-ether	epoxide, R'OH $\xrightarrow{\text{H}_3\text{O}^+}$ alcohol–ether
Grignard organometallic ring opening	Grignard reagent MgBr, epoxide	ether, heat, acid H ₃ O ⁺	alcohol	epoxide + R'MgBr $\xrightarrow[\text{H}_3\text{O}^+]{(1) \text{ ether, heat}}$ R'CR ₂ CR ₂ OH

| Preparation, Cleavage of glycol

Reaction	Reactant	Condition	Product	Overall
Acid ring opening (hydrolysis) of epoxide	epoxide, water H2O	acid	glycol	$\text{epoxide} + \text{H}_2\text{O} \xrightarrow{\text{acid}} \text{glycol}$
Oxidation of alkene by OsO4	alkene, osmium tetroxide OsO4	water H2O, NaHSO3	glycol	$\text{R}_2\text{C}=\text{CR}_2 + \text{OsO}_4 \xrightarrow[\text{NaHSO}_3]{\text{H}_2\text{O}} \text{glycol} + \text{OsO}_3(\text{OH})_2$
Alkene + KMnO4	alkene, potassium permanganate KMnO4	water, acetone	glycol, MnO2	$\text{alkene} + \text{KMnO}_4 \xrightarrow[\text{acetone}]{\text{H}_2\text{O}, \text{OH}^-} \text{glycol} + \text{MnO}_2$
Glycol cleavage	glycol, periodic acid H5IO6	HOAc	aldehyde, ketone, water	$\text{glycol} + \text{H}_5\text{IO}_6 \xrightarrow{\text{HOAc}} \text{R}'-\text{CH}=\text{O} + \text{R}-\text{CR}=\text{O} + 2 \text{H}_2\text{O} + \text{HIO}_3\text{H}_2\text{O}$

| Oxonium salt

Reaction	Reactant	Condition	Product	Overall
Oxonium salt	nucleophile, oxonium salt		alcohol	$\text{Nu}^- + \text{R}_3\text{O}^+ \longrightarrow \text{ROH} + \text{H}_2\text{O}$

Ch14 Alkyne Reactions

Reaction	Reactant	Condition	Product	Overall
Alkyne addition	alkyne, hydrogen halide		alkene halide	$\text{C}\equiv\text{C} + \text{HX} \longrightarrow \text{XC}=\text{C}$
Hydration of alkyne	alkyne, water	mercury ion, sulfuric acid H2SO4	ketone	$\text{RC}\equiv\text{CH} + \text{H}_2\text{O} \xrightarrow{\text{Hg}^{2+}, \text{H}_2\text{SO}_4} \text{RC}=\text{OCH}_3$
Hydroboration-oxidation of symmetric alkyne	symmetric alkyne	borane BH3, THF, hydrogen peroxide	ketone	$\text{R}-\text{C}\equiv\text{C}-\text{R} \xrightarrow[(2) \text{H}_2\text{O}_2/\text{OH}^-]{(1) \text{BH}_3/\text{THF}} \text{R}-\text{CH}_2-\text{C}=\text{O}-\text{R}$
Hydroboration-oxidation of asymmetric alkyne	asymmetric 1-alkyne	disiamylborane, THF, hydrogen peroxide	aldehyde	$\text{R}-\text{C}\equiv\text{C}-\text{H} \xrightarrow[(2) \text{H}_2\text{O}_2/\text{OH}^-]{(1) (\text{branch})_2\text{BH}/\text{THF}} \text{R}-\text{CH}_2-\text{C}=\text{O}-\text{H}$
Catalytic hydrogenation of alkyne	alkyne	hydrogen gas H2, catalyst	cis alkene	$\text{RC}=\text{CR} \xrightarrow{\text{H}_2, \text{catalyst}} \text{RCH}=\text{HCR}$
Catalytic hydrogenation of alkene	alkene	hydrogen gas H2, catalyst	alkane	$\text{RCH}=\text{HCR} \xrightarrow{\text{H}_2, \text{catalyst}} \text{RCH}_2=\text{H}_2\text{CR}$
Controlled catalytic hydrogenation of alkyne	alkyne	Lindlar catalyst/(Pd/C), pyridine	cis alkene	$\text{RC}=\text{CR} \xrightarrow[\text{or Pd/C, pyridine}]{\text{Lindlar catalyst Pd/CaCO}_3+\text{Pb(OAc)}_2} \text{RCH}=\text{HCR}$
Reduction of alkyne with Na and NH3	alkyne, sodium, liquid ammonia		trans alkene, sodium ion, azanide	$\text{RC}=\text{CR} + 2 \text{Na} + 2 \text{NH}_3 \longrightarrow \text{RCH}=\text{HCR} + 2 \text{Na}^+ + 2 \text{NH}_2^-$

Reaction	Reactant	Condition	Product	Overall
Grignard reaction of alkyne	1-alkyne, Grignard reagent	THF	acetylenic Grignard reagent, hydrocarbon	$\text{RC}\equiv\text{CR} + \text{R}'\text{MgBr} \xrightarrow{\text{THF}} \text{RC}\equiv\text{CMgBr} + \text{R}'$
Alkyne in SN2 rxn	alkane halide, acetylenic anion		alkyne, halide ion	$\text{RX} + \text{C}\equiv\text{CR}^- \longrightarrow \text{RC}\equiv\text{CR} + \text{X}^-$

Ch15 Diene Reactions

Reaction	Reactant	Condition	Product	Overall
Diels-Alder Rxn	diene, alkene (dienophile)		ring	$\text{diene} + \text{dienophile} \longrightarrow \text{ring}$
Addition of HX to conjugated diene	conjugated diene, hydrogen halide		alkene halide	$\text{conj} \cdot \text{diene} + \text{HX} \longrightarrow \text{mixed alkene halide}$

Ch16 Benzene Reactions

Reaction	Reactant	Condition	Product	Overall
Halogenation of benzene	benzene, halide gas X ₂	iron/iron(iii) bromide FeBr ₃	benzene halide, hydrogen halide	$\text{Ph} + \text{X}_2 \xrightarrow{\text{FeBr}_3/\text{Fe}} \text{PhX} + \text{HX}$
Nitration of benzene	benzene, nitric acid HNO ₃	sulfuric acid H ₂ SO ₄	nitrobenzene, water	$\text{Ph} + \text{HNO}_3 \xrightarrow{\text{H}_2\text{SO}_4} \text{PhNO}_2 + \text{H}_2\text{O}$
Sulfonation of benzene	benzene, sulfur trioxide	fuming sulfuric acid H ₂ SO ₄	benzenesulfuric acid	$\text{Ph} + \text{SO}_3 \xrightarrow{\text{fuming H}_2\text{SO}_4} \text{PhSO}_3\text{H}$
Friedel-Crafts alkylation of benzene	benzene, alkyl chloride	aluminum chloride AlCl ₃	alkyl benzene, hydrogen chloride	$\text{Ph} + \text{RCl} \xrightarrow{\text{AlCl}_3} \text{PhR} + \text{HCl}$
Friedel-Crafts acylation of benzene	benzene, acyl chloride	aluminum chloride AlCl ₃ , water	ketone, hydrogen chloride	$\text{Ph} + \text{Cl}-\text{CR}=\text{O} \xrightarrow[(2)\text{H}_2]{(1)\text{AlCl}_3} \text{Ph}-\text{C}=\text{OCH}_3 + \text{HCl}$
Hydrogenation of benzene	benzene, hydrogen gas H ₂	Ni	cyclohexane	$\text{Ph} + 3\text{H}_2 \xrightarrow[175^\circ\text{C}, 180\text{ atm}]{\text{Ni}} \text{cyclohexane}$
Hydrogenation of benzene derivatives	benzene derivative, hydrogen gas H ₂	Ni	substituted cyclohexane	$\text{PhR} + 3\text{H}_2 \xrightarrow[175^\circ\text{C}, 180\text{ atm}]{\text{Ni}} \text{cyclohexane}-\text{R}$

Ch17 Allylic/Benzylic Reactions

Reaction	Reactant	Condition	Product	Overall
Radical bromonation of allylic/benzylic hydrogen	allylic/benzylic species, bromine gas Br ₂	light hv	allylic/benzylic bromide, hydrogen bromide	$\text{PhCH}_3 + \text{Br}_2 \xrightarrow{\text{light hv}} \text{PhCH}_2\text{Br} + \text{HBr}$ $\text{CH}_3-\text{CH}_2=\text{CH}_2 \xrightarrow{\text{light hv}} \text{BrCH}_2-\text{CH}_2=\text{CH}_2$

Reaction	Reactant	Condition	Product	Overall
Controlled radical bromination of allylic/benzylic hydrogen	allylic/benzylic species, NBS (N-bromosuccinimide)	heat/light, peroxide	allylic/benzylic bromide, succinimide	$\text{PhCH}_3 + \text{NBS} \xrightarrow[\text{CCl}_4]{\text{heat/light, peroxide}} \text{PhCH}_2\text{Br}$ $\text{CH}_3-\text{CH}_2=\text{CH}_2 + \text{NBS} \xrightarrow{\text{light hv}} \text{BrCH}_2-\text{CH}_2=\text{CH}_2 + \text{succinimide}$
Allylic/benzylic Grignard Reagent	alkene bromide Br, magnesium Mg		alkene, MgBrOH	→ see Ch 17 allylic Grignard reagent
Allylic/benzylic E2 elimination	allylic/benzylic bromide	EtOH, Na ⁺ , EtO ⁻	alkene	$\text{Ph}-\text{CH}_2\text{CH}_2-\text{Br} \xrightarrow[\text{EtOH}]{\text{Na}^+, \text{EtO}^-} \text{Ph}-\text{CH}_2\text{CH}_2-\text{OEt}$
Oxidation of allylic/benzylic alcohol with MnO2	allylic/benzylic alcohol, manganese dioxide MnO2	CH2Cl2	aldehyde/ketone, Mn(OH)2	$\text{PhCH}_2\text{OH} + \text{MnO}_2 \xrightarrow{\text{CH}_2\text{Cl}_2} \text{PhCH}=\text{O} + \text{Mn(OH)}_2$ $\text{H}_2\text{C}=\text{CH}-\text{CH}_2\text{OH} + \text{MnO}_2 \xrightarrow{\text{CH}_2\text{Cl}_2} \text{H}_2\text{C}=\text{CH}-\text{CH}_2=\text{O} + \text{Mn(OH)}_2$
Benzylic oxidation of alkylbenzene	alkylbenzene	Cr(VI): Na2CrO7/CrO3; Mn(VII): KMnO4; O2 + catalyst	benzylic carboxylic acid	$\text{PhR} \xrightarrow[\text{or KMnO}_4 \text{ or O}_2 + \text{catalyst}]{\text{Na}_2\text{CrO}_7/\text{CrO}_3} \text{PhCOOH}$
Biosynthesis of terpene	isopentenyl pyrophosphate IPP, gamma,gamma-dimethylallyl pyrophosphate DMAP	prenyl transferase	terpene, HOPP	$\text{IPP} + \text{DMAP} \xrightarrow{\text{prenyl transferase SN}_1} \text{terpene, HOPP}$

Ch18 Vinylic/Aryl Halide Reactions

Reaction	Reactant	Condition	Product	Overall
Vinylic/aryl halide under SN1, SN2 conditions	Vinylic/aryl halide	SN1, SN2	no reaction	$\text{CH}_2=\text{CHX} + \text{Nu} \xrightarrow{\text{SN}_1, \text{SN}_2} \text{no reaction}$ $\text{PhX} + \text{Nu} \xrightarrow{\text{SN}_1, \text{SN}_2} \text{no reaction}$
Elimination of vinylic halide	vinylic halide, nucleophile (hydroxide)	harsh conditions, high temp	vinylic alkyne, bromide, water	$\text{Ph}-\text{CH}=\text{CH}-\text{Br} + \text{KOH} \xrightarrow{200^\circ\text{C}} \text{Ph}-\text{C}\equiv\text{C}-\text{H} + \text{KBr} + \text{H}_2\text{O}$
Nucleophilic aromatic substitution of aryl halide	nitro aryl halide, nucleophile		nitro aryl nucleophile: halide ion	$\text{O}_2\text{N}-\text{Ph}-\text{X} + \text{Nu}^- \longrightarrow \text{O}_2\text{N}-\text{Ph}-\text{Nu} + \text{X}^-$
Heck reaction (aryl)	aryl bromide/iodide, alkene	Pd(0) catalyst (Pd(OAc)2, Pd(PPh3)4)	aryl alkene, hydrogen bromide	$\text{PhBr} + \text{alkene} \xrightarrow{\text{Pd(0)}} \text{Ph}-\text{alkene} + \text{HBr}$
Heck reaction (general)	alkyl halide + alkene	Pd(0) catalyst (Pd(OAc)2, Pd(PPh3)4)	alkene, hydrogen bromide	$\text{R}_1\text{X} + \text{alkene}-\text{R}_2 \xrightarrow{\text{PdL}_4} \text{R}_1-\text{alkene}-\text{R}_2 + \text{HX}$
Suzuki coupling	aryl boronic acid, aryl halide, sodium hydroxide	Pd(OAc)2, PPh3, Na2CO3	biaryl compound, NaBr, B(OH)2	$\text{RB(OH)}_2 + \text{R}'\text{X} + \text{NaOH} \xrightarrow{\text{Pd(OAc)}_2, \text{PPh}_3, \text{Na}_2\text{CO}_3} \text{RR}' + \text{NaBr} + \text{B(OH)}_2$

Reaction	Reactant	Condition	Product	Overall
Oxidation of phenol	phenol -> semiquinone	Na2Cr2O7, H2SO4	quinone	$\text{Ph}(\text{OH})_2 \xrightarrow{\text{Na}_2\text{Cr}_2\text{O}_7, \text{H}_2\text{SO}_4} \text{quinone}$
Bromination of phenol	phenol, bromine gas	CCl4, Hbr, water H2O	phenol bromide	$\text{PhOH} + n \text{Br}_2 \xrightarrow{\text{various conditions}} \text{PhOH}-\text{Br}_n + n \text{HBr}$
Nitration of phenol	phenol	nitric acid HNO3	nitric phenol	$\text{PhOH} \xrightarrow{\text{HNO}_3} \text{PhOH}-\text{NO}_2 + \text{H}_2\text{O}$
Friedel-Crafts alkylation	phenol, alcohol	sulfuric acid H2SO4	alkyl phenol, water H2O	$\text{PhOH} + \text{ROH} \xrightarrow{\text{H}_2\text{SO}_4} \text{RPhOH} + \text{H}_2\text{O}$
Friedel-Crafts acylation	phenol, AlCl3, acyl chloride	PhNO2	acyl phenol	$\text{PhOH} + \text{AlCl}_3 + \text{RC}=\text{O}-\text{Cl} \xrightarrow{\text{PhNO}_2} \text{PhOH}-\text{CR}=\text{O}$