# Interfacial Tension Measurement and Adsorption Isotherm Determination Using the Inverted Drop Weight Method

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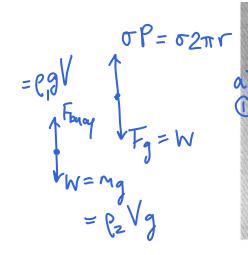
Surface and Colloid Science

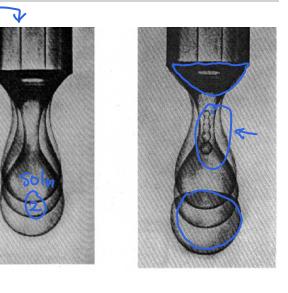
## Surface tension calculation from the drop weight method

• Tate (1864)

$$W=2\pi r\sigma$$

 $W=2\pi r\sigma \leftarrow$ • Harkins and Brown (1919)  $f=\frac{W \text{ of liq breaks away}}{W \text{ of pendant drop}}$   $W=2\pi r\sigma f=\frac{r\sigma}{F}$   $F\equiv\frac{1}{2}\pi f$   $\sigma=\frac{WF}{F}=\frac{|\mathfrak{l}_1-\mathfrak{l}_2|}{\mathbb{Q}^{V-F}}$ 





• Heertjes et al. (1971)

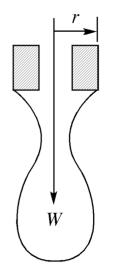
$$\begin{bmatrix} F = 0.14782 + 0.27896 \left(\frac{r}{V^{1/3}}\right) - 0.1662 \left(\frac{r}{V^{1/3}}\right)^2 = \text{f(r,V)} \\ \text{Constraint:} \left(\frac{r}{V^{1/3}}\right) \in (0.3,1.2) \end{bmatrix}$$

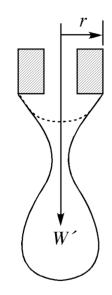
Surface tension

$$\sigma = rac{V | 
ho_2 - 
ho_1 | gF}{r} = rac{V 
ho_2 = 
ho}{r}$$

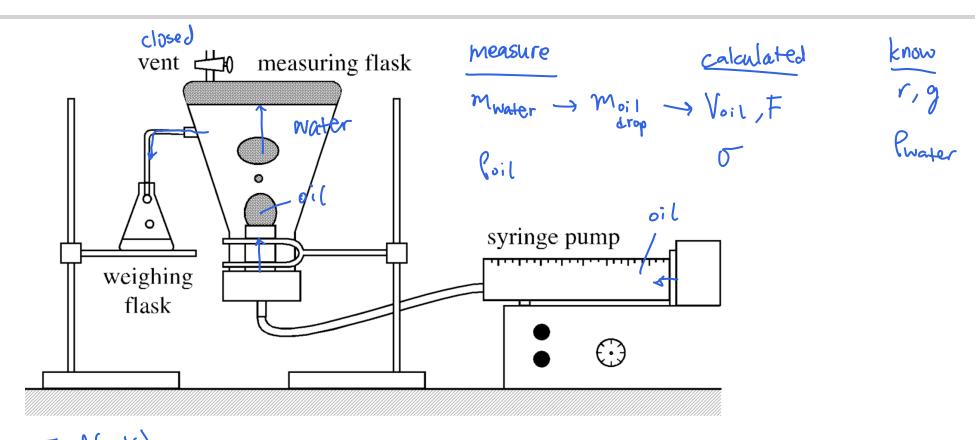
if 
$$l_1 \rightarrow 0$$
,  $l_2 = l$ 

$$= \frac{V l g F}{\Gamma} = \frac{mg F}{\Gamma}$$





### Interfacial tension between liquids is measured with inverted drop weight method



$$\circ \stackrel{\longleftarrow}{\sigma} = \frac{V|\rho_2 - \rho_1|gF^{\checkmark}}{r \checkmark}$$

# Szyszkowski equation describes surface tension of binary aqueous solutions

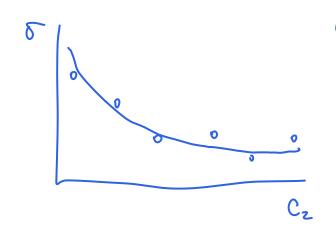
• Szyszkowski equation

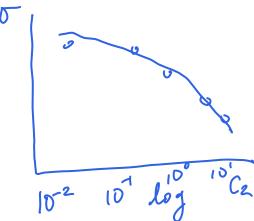
$$\sigma = \sigma_0 - RTB \ln \left(1 + \frac{C_2}{a}\right)^{\text{Cohcentration}}$$
surface ideal also of solute tension of const pure solvent  $a = 8.314 \text{ J/mol/K}$ 

Lauricacid

Mydrophobic hydrophilic

a, B empirical const





Hover wecoult
lauricated
oil +

### Adsorption isotherm is modeled by Gibbs adsorption equation

Gibbs adsorption equation: ideal dilute solution

relative adsorption of solute 
$$\Gamma_{2,1}=-\frac{C_2}{RT}\left|\frac{d\sigma}{dC_2}\right|$$
 with respect to solvent  $T_2$ 

Finite difference method

$$rac{d\sigma}{dC_2} = rac{\Delta\sigma}{\Delta C_2} = rac{\sigma_2 - \sigma_1}{C_{2,2} - C_{2,1}}$$

Szyszkowski equation

$$\sigma = \sigma_0 - RTB \ln \left(1 + rac{C_2}{a}
ight)$$

$$\frac{d\sigma}{dCz} = \frac{d}{dCz} \left( \sigma_0 - RTB \ln \left( 1 + \frac{Cz}{a} \right) \right)$$

$$= -RTB \frac{1/a}{1 + Cz}$$

$$\frac{C_{2}}{C_{2}} = \frac{1}{3} = 1$$

$$\frac{C_{2}}{C_{2}} = \frac{3}{3-1} = 1$$

$$\frac{3}{4} = \frac{3}{4} = 1$$

