

Determination of Zeta Potential by Microelectrophoresis

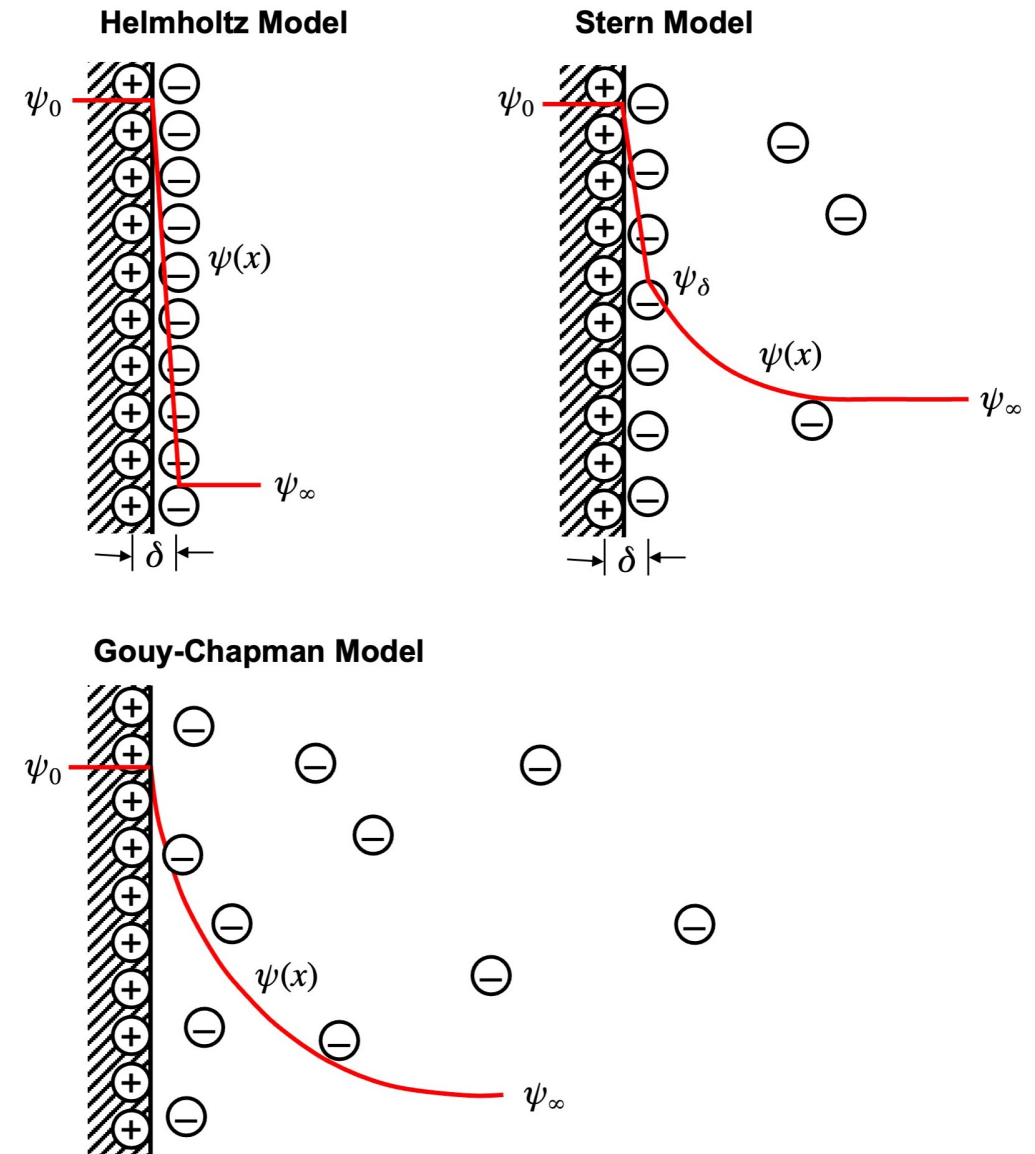
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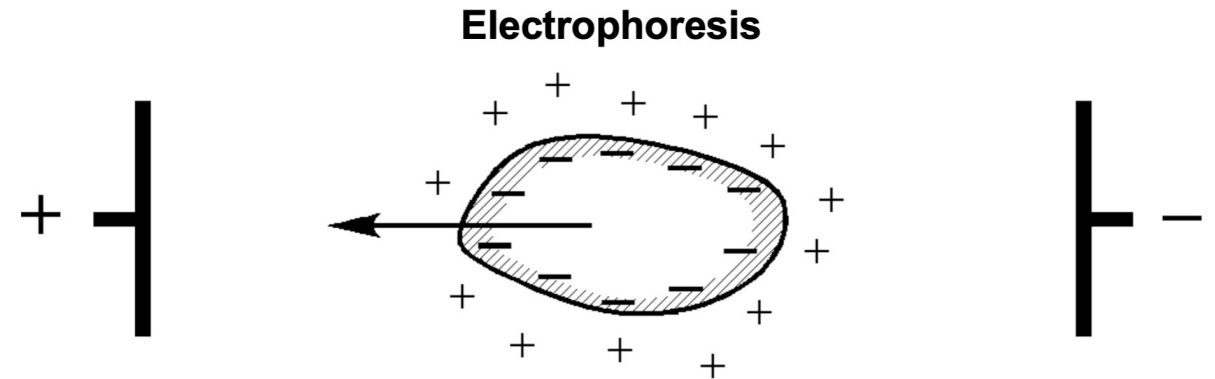
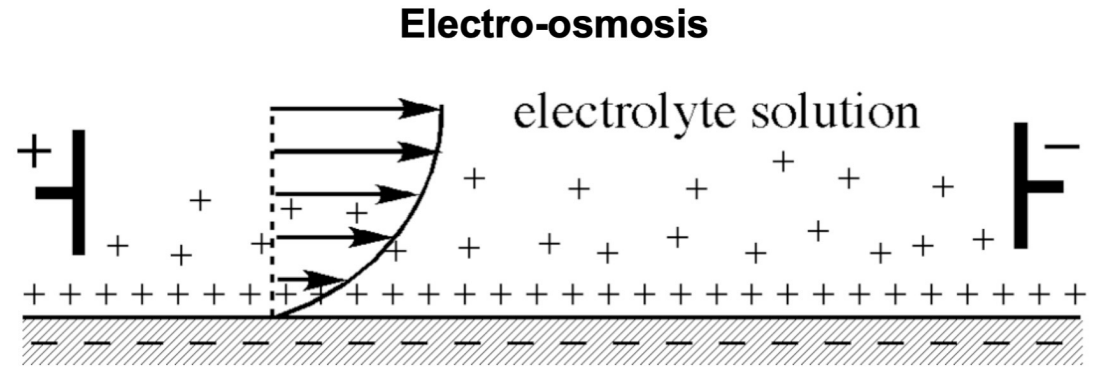
Zeta potential is the electrical potential at the slip plane of the electric double layer

- ψ - electrical potential difference between dispersing medium and ...
 - $\psi_\infty \equiv 0$ - the dispersing medium
 - ψ_0 - the surface of colloidal particle
 - True surface potential
 - ψ_δ - the outer first layer of counterions
 - Effective (Stern) surface potential
 - ζ - the slip plane (medium velocity = 0)
 - Zeta potential, electrokinetic potential
- The slip plane may be slightly further out into the solution than Stern layer
 - $\psi_\delta \approx \zeta$



Electro-osmosis and electrophoresis give the same zeta potential

- **Electro-osmosis** - diffuse layer of ions beside a charged immobile surface move under E field, which sets the liquid into motion by the action of viscosity
 - *Liquid* moves, *solid* at rest
- **Electrophoresis** - diffuse layer of ions beside a charged particle surface move under E field, which sets the particle into motion
 - *Solid* moves, *liquid* at rest



Electrophoretic mobility allows determination of zeta potential

- **Electrophoretic mobility** - $u_E [(\mu\text{m/s})/(\text{V/cm})]$

$$\circ \quad u_E = \frac{V_p}{E_x} = \begin{cases} \frac{\varepsilon\varepsilon_0\zeta}{\mu} & (\kappa a > 200) \\ \frac{2}{3} \frac{\varepsilon\varepsilon_0\zeta}{\mu} & (\kappa a < 0.1) \end{cases}$$

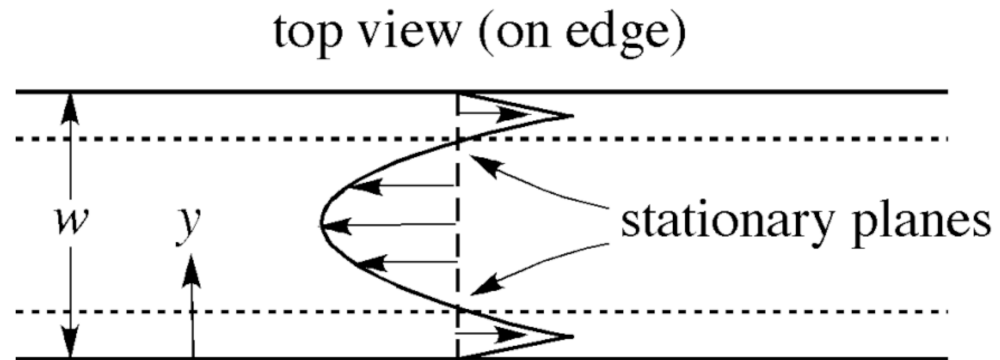
- u_E - Electrophoretic mobility
- V_p - Particle velocity
- E_x - Electric field strength
- ε - Dielectric constant of the medium
- ε_0 - Permittivity of free space
- ζ - Zeta potential
- μ - Viscosity of the medium
- κ - Debye length
- a - Particle radius

Point of zero charge and isoelectric point define pH at which potentials are zero

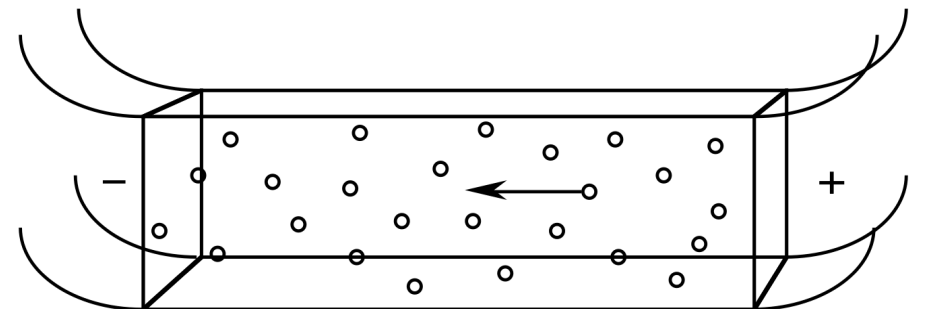
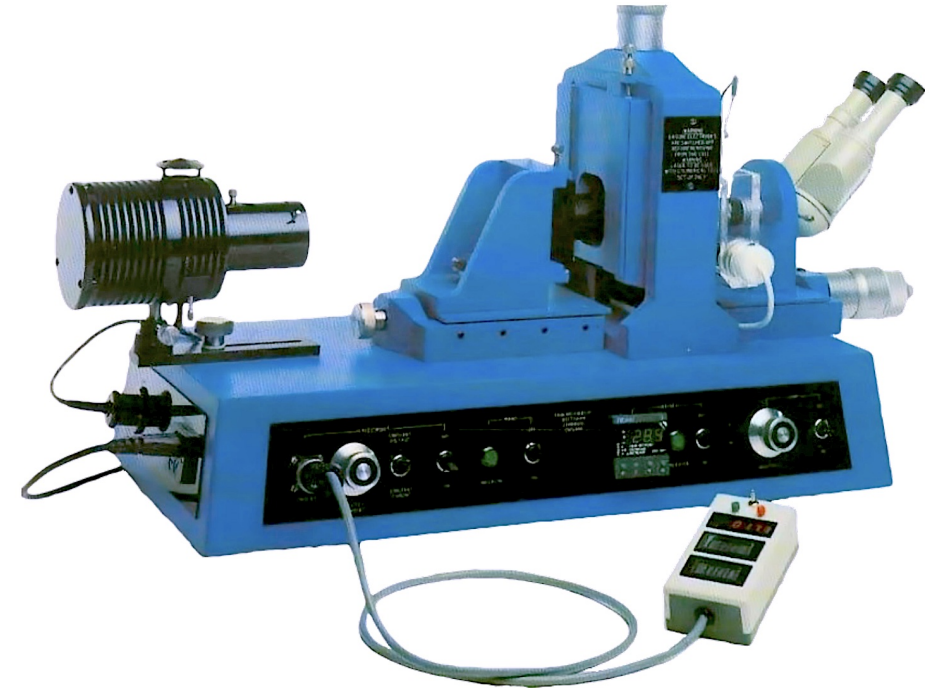
- **Potential determining ions** - ions whose concentration determines surface potential
 - Crystalline solid - lattice ions
 - Oxides - H_3O^+ , OH^- (pH)
- **Point of zero charge (PZC)** - pH at which $\psi_0 = 0$
 - $\text{pH} < \text{PZC}$: $\psi_0 > 0$
 - $\text{pH} > \text{PZC}$: $\psi_0 < 0$
- **Isoelectric point (IEP, pl)** - pH at which $\zeta \approx \psi_\delta = 0$
 - $u_E = 0 \Rightarrow V_p = 0$

Darkfield illumination microscopy visualizes colloidal particles under E field

- $u_E = \frac{V_p}{E_x} = \frac{\epsilon\epsilon_0\zeta}{\mu} \quad (\kappa a > 200)$
 - $\zeta = \frac{\mu V_p}{\epsilon\epsilon_0 E_x}$
- Electrophoresis has solid particles moving when liquid is at rest ($v_x = 0$)



- $v_x = 0$ when $y = \begin{cases} 0.205w \\ 0.795w \end{cases}$



Laser Doppler electrophoresis determines zeta potential with more sensitivity

