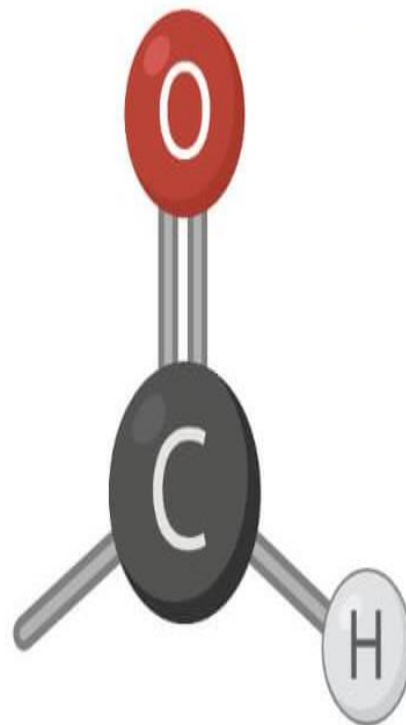


# Chapter 10

Aldehydes & ketones



# Aldehydes

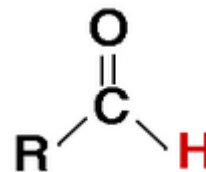
### 1- Molecular formula:



### 2- Molar mass:

$$14n + 16$$

### 3- Strutural formula:



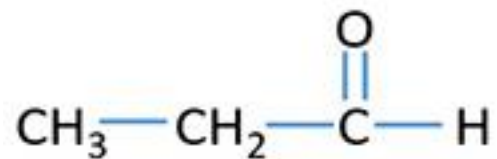
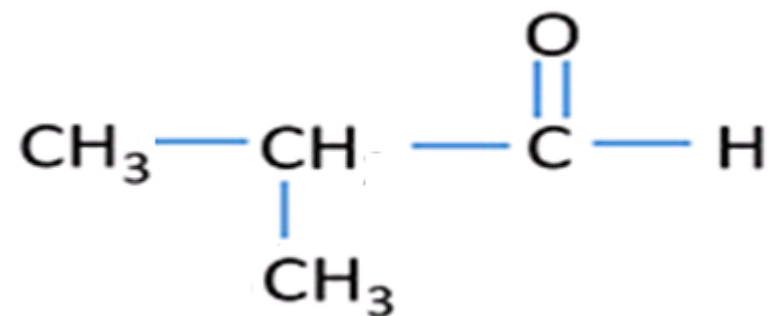
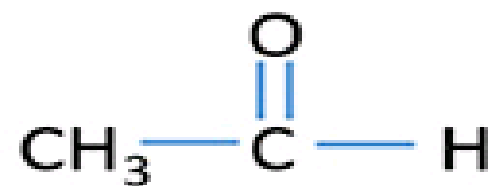
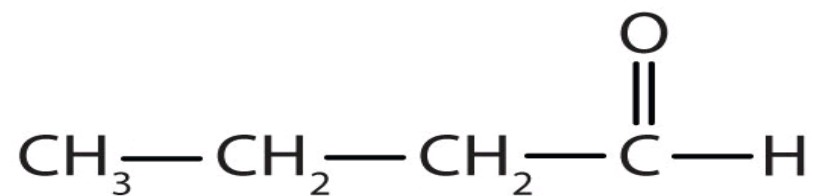
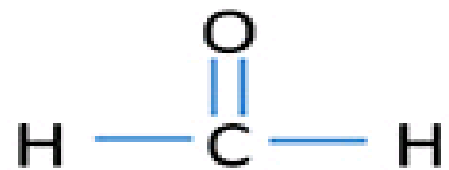
### 4- Functional group:



### 5- Nomenclature :

**Alkane → AlKanal**

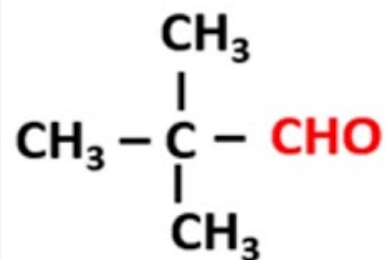
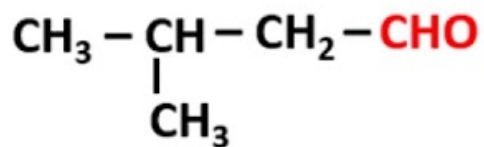
The position of functional group should not be indicated in the nomenclature since it is found on the carbon number one



## 6- isomers of aldehydes:

Aldehydes have skeletal isomer and a functional isomer which is ketone

Write all possible condensed structural formula of an aldehyde of formula  $C_4H_8O$



# ketones

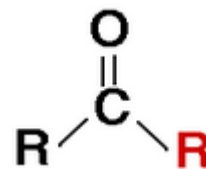
### 1- Molecular formula:



### 2- Molar mass:

$$14n + 16$$

### 3- Structural formula:



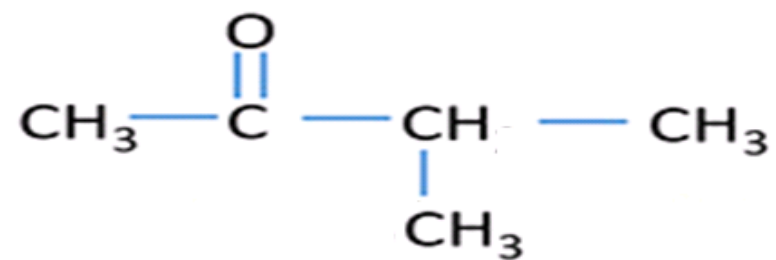
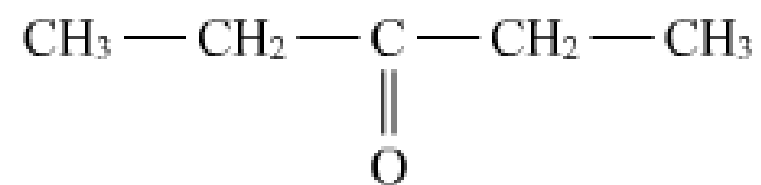
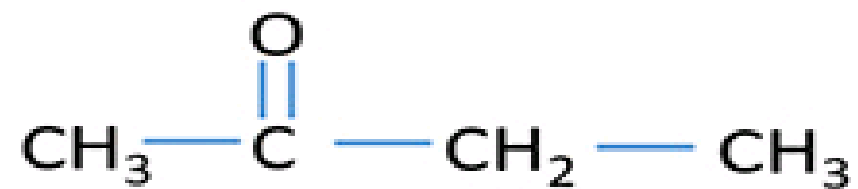
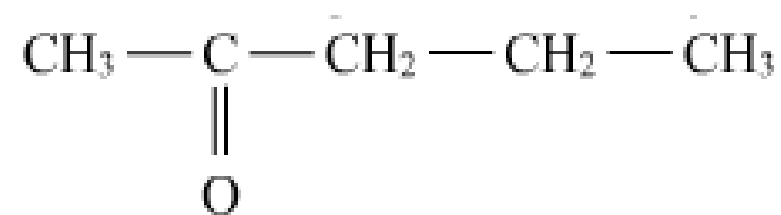
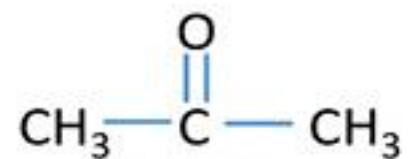
### 4- Functional group:



### 5- Nomenclature :

**Alkane → Alkanone**

The position of the functional group should be indicated if the main chain contains at least 5 carbon atoms :  
x-alkanone

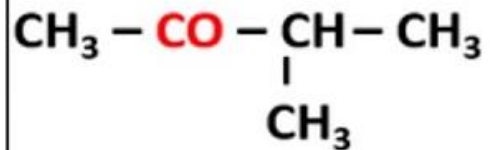
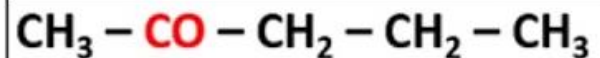




## 6- isomers of ketones:

Ketones have skeletal and positional isomers and a functional isomer which is aldehyde

Write all possible condensed structural formula of a ketone of molecular formula  $C_5H_{10}O$



Identify experimental aldehydes and ketones:

	2,4-DNPH	NaHSO <sub>3</sub>	fehling (blue)	schiff's reagent (colorless)	Tollens reagent (inocolor)	$MnO_4^-$ Purple	$Cr_2O_7^{2-}$ orange
aldehydes	<b>Yellow orange precipitate</b>	<b>White crystals</b>	<b>Brick red precipitate</b>	<b>Pink</b>	<b>Silvery mirror</b>	<b>Colorless</b>	<b>green</b>
ketones	<b>Yellow orange precipitate</b>	<b>White crystals</b>	<b>blue</b>	<b>colorless</b>	<b>Colorless</b>	<b>purple</b>	<b>Orange</b>

2,4-DNPH: 2,4-dinitro-phenyl-hydrazine

NaHSO<sub>3</sub> : Sodium bisulfite

tollens : Ammoniacal silver nitrate

$MnO_4^-$  : Permanganate ion

$Cr_2O_7^{2-}$  : Dichromate ion

## Procedure :

### ***2,4-DNPH & NaHSO<sub>3</sub>***

Add few ml of these substances into a test tube than add few drops of an organic compound

### ***Fehling blue***

Add few ml of Fehling into a test tube than add 1 ml of the organic compound, heat the tube using a flame

### ***tollens reagent***

add few ml of this reagent in a test tube tahn add 1 ml of an organic compound, shake the mixture and put the test tube in a water bath at 50°C

### ***schiff's reagent***

Add some ml of this reagent into a test tube. Cool the contents of the tube in an ice water bath then add few drops of the organic compound