

参考答案

1-5 BBAA 6-7 CB

8. (1) ②⑥; ⑦

(2) ③④; ①; ⑤⑥⑧

(3) $\text{Cu(OH)}_2 \rightleftharpoons \text{Cu}^{2+} + 2\text{OH}^-$

$\text{H}_2\text{S} \rightleftharpoons \text{H}^+ + \text{HS}^-$

$\text{KHSO}_4 = \text{K}^+ + \text{H}^+ + \text{SO}_4^{2-}$

$\text{KClO}_3 = \text{K}^+ + \text{ClO}_3^-$

$\text{NH}_4\text{NO}_3 = \text{NH}_4^+ + \text{NO}_3^-$

9.(1) ⑤⑥⑦⑧ (2)D (3)D