## **Description**

### **Image**





### Caption

1. Siemens toaster in brushed austenitic stainless steel (by Porsche Design) © Granta Design 2. Scissors in ferritic stainless steel; it is magnetic, austenitic stainless is not. © Granta Design

#### The material

Stainless steels are alloys of iron with chromium, nickel, and - often - four of five other elements. The alloying transmutes plain carbon steel that rusts and is prone to brittleness below room temperature into a material that does neither. Indeed, most stainless steels resist corrosion in most normal environments, and they remain ductile to the lowest of temperatures.

### Composition (summary)

Fe/<0.25C/16 - 30Cr/3.5 - 37Ni/<10Mn + Si,P,S (+N for 200 series)

## **General properties**

Density	474	-	506	lb/ft^3
Price	* 2.54	-	2.77	USD/lb
Date first used	1915			

## **Mechanical properties**

Young's modulus	27.4	-	30.5	10^6 psi
Shear modulus	10.7	-	12.2	10^6 psi
Bulk modulus	19.4	-	21.9	10^6 psi
Poisson's ratio	0.265	-	0.275	
Yield strength (elastic limit)	24.7	-	145	ksi
Tensile strength	69.6	-	325	ksi
Compressive strength	24.7	-	145	ksi
Elongation	5	-	70	% strain
Hardness - Vickers	130	-	570	HV
Fatigue strength at 10^7 cycles	* 25.4	-	109	ksi





Hydrochloric acid (36%)

SIEDUPITER				
Fracture toughness	56.4	-	137	ksi.in^0.5
Mechanical loss coefficient (tan delta)	* 2.9e-4	-	0.00148	
Thermal properties				
Melting point	2.51e3	-	2.64e3	F
Maximum service temperature	1.38e3	-	1.51e3	F
Minimum service temperature	-458	-	-456	F
Thermal conductor or insulator?	Poor cor	nduc	tor	
Thermal conductivity	6.93	-	13.9	BTU.ft/h.ft^2.F
Specific heat capacity	0.107	-	0.127	BTU/lb.°F
Thermal expansion coefficient	7.22	-	11.1	µstrain/℉
Electrical properties				
Electrical conductor or insulator?	Good co	ndu	ctor	
Electrical resistivity	64	-	107	μohm.cm
				•
Optical properties	Onegue			
Transparency	Opaque			
Critical Materials Risk				
High critical material risk?	Yes			
Processability				
Castability	3	-	4	
Formability	2	-	3	
Machinability	2	_	3	
Weldability	5			
Solder/brazability	5			
Durability: water and aqueous solutions				
Water (fresh)	Excellen			
Water (salt)	Excellen			
Soils, acidic (peat)	Excellen			
Soils, alkaline (clay)	Excellen			
Wine	Excellen	t		
Durability: acids				
Acetic acid (10%)	Excellen	t		
Acetic acid (glacial)	Excellen	t		
Citric acid (10%)	Excellen	t		
Hydrochloric acid (10%)	Excellen	t		

Limited use



Hydrofluoric acid (40%)	Limited use
Nitric acid (10%)	Excellent
Nitric acid (70%)	Limited use
Phosphoric acid (10%)	Excellent
Phosphoric acid (85%)	Excellent
Sulfuric acid (10%)	Acceptable
Sulfuric acid (70%)	Limited use

# **Durability: alkalis**

Sodium hydroxide (10%)	Excellent
Sodium hydroxide (60%)	Excellent

# **Durability: fuels, oils and solvents**

Amyl acetate	Acceptable
Benzene	Acceptable
Carbon tetrachloride	Excellent
Chloroform	Excellent
Crude oil	Excellent
Diesel oil	Excellent
Lubricating oil	Excellent
Paraffin oil (kerosene)	Excellent
Petrol (gasoline)	Excellent
Silicone fluids	Acceptable
Toluene	Excellent
Turpentine	Acceptable
Vegetable oils (general)	Excellent
White spirit	Excellent

# Durability: alcohols, aldehydes, ketones

Acetaldehyde	Excellent
Acetone	Excellent
Ethyl alcohol (ethanol)	Excellent
Ethylene glycol	Acceptable
Formaldehyde (40%)	Acceptable
Glycerol	Excellent
Methyl alcohol (methanol)	Excellent

# **Durability: halogens and gases**

Chlorine gas (dry)	Excellent
Fluorine (gas)	Excellent
O2 (oxygen gas)	Excellent



BEDUPITCK						
Sulfur dioxide (gas)	Ex	xcellent	t			
Durability: built environments						
Industrial atmosphere	Ex	Excellent				
Rural atmosphere	Ex	xcellent	t			
Marine atmosphere	Ex	xcellent	t			
UV radiation (sunlight)	Ex	xcellent	t			
Durability: flammability						
Flammability	No	on-flam	mab	le		
Durability: thermal environments						
Tolerance to cryogenic temperatures	Ex	xcellent	t			
Tolerance up to 150 C (302 F)	Ex	xcellent	t			
Tolerance up to 250 C (482 F)	Ex	xcellent	t			
Tolerance up to 450 C (842 F)	Ex	xcellent	t			
Tolerance up to 850 C (1562 F)	Ex	xcellent	t			
Tolerance above 850 C (1562 F)	Ur	naccept	table	)		
Geo-economic data for principal component						
Annual world production, principal component	2.2	26e9			ton/yr	
Reserves, principal component	1.9	57e11			I. ton	
Primary material production: energy, CO2 and w	water					
Embodied energy, primary production	* 8.7	7e3	-	9.62e3	kcal/lb	
CO2 footprint, primary production	* 4.7	73	-	5.23	lb/lb	
Water usage	* 15	5.5	-	17	gal(US)/lb	
Eco-indicator 95	91	0			millipoints/kg	
Eco-indicator 99	30	18			millipoints/kg	
Material processing: energy						
Casting energy	* 1.	17e3	-	1.29e3	kcal/lb	
Extrusion, foil rolling energy	* 1.5	59e3	-	1.77e3	kcal/lb	
Rough rolling, forging energy	* 81	3	-	898	kcal/lb	
Wire drawing energy	* 5.8	89e3	-	6.51e3	kcal/lb	
Metal powder forming energy	* 3.9	91e3	-	4.48e3	kcal/lb	
Vaporization energy	* 1.	18e6	-	1.3e6	kcal/lb	
Coarse machining energy (per unit wt removed)	* 16	39	-	186	kcal/lb	
Fine machining energy (per unit wt removed)	* 1.2	22e3	-	1.35e3	kcal/lb	
Grinding energy (per unit wt removed)	* 2.3	39e3	-	2.64e3	kcal/lb	
Non-conventional machining energy (per unit wt removed	1.	18e4	-	1.3e4	kcal/lb	



Casting CO2	* 0.809	-	0.894	lb/lb
Extrusion, foil rolling CO2	* 1.1	-	1.22	lb/lb
Rough rolling, forging CO2	* 0.562	-	0.621	lb/lb
Wire drawing CO2	* 4.08	-	4.51	lb/lb
Metal powder forming CO2	* 2.89	-	3.31	lb/lb
Vaporization CO2	* 815	-	900	lb/lb
Coarse machining CO2 (per unit wt removed)	* 0.117	-	0.129	lb/lb
Fine machining CO2 (per unit wt removed)	* 0.847	-	0.936	lb/lb
Grinding CO2 (per unit wt removed)	* 1.66	-	1.83	lb/lb
Non-conventional machining CO2 (per unit wt removed	8.15	-	9	lb/lb

### Material recycling: energy, CO2 and recycle fraction

3,7,7,7,7,7,7,7,7,7,7,7,7,7,7,7,7,7,7,7		
Recycle	✓	
Embodied energy, recycling	* 1.82e3 - 2.02e	3 kcal/lb
CO2 footprint, recycling	* 1.32 - 1.46	lb/lb
Recycle fraction in current supply	35 - 40	%
Downcycle	✓	
Combust for energy recovery	×	
Landfill	✓	
Biodegrade	×	
Toxicity rating	Non-toxic	
A renewable resource?	×	

### **Environmental notes**

Stainless steels are FDA approved -- indeed, they are so inert that they can be implanted in the body, and are widely used in food processing equipment. All can be recycled.

### **Supporting information**

### Design guidelines

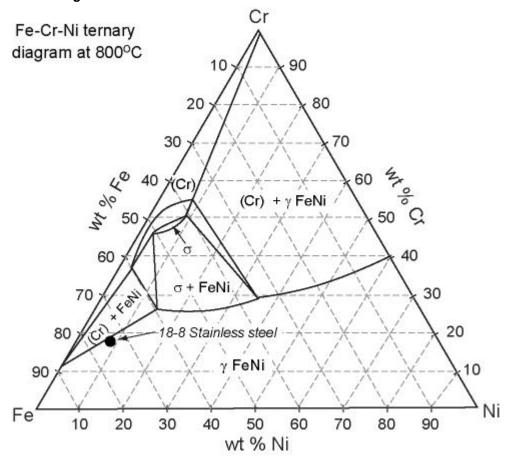
Stainless steel must be used efficiently to justify its higher costs, exploiting its high strength and corrosion resistance. Economic design uses thin, rolled gauge, simple sections, concealed welds to eliminate refinishing, and grades that are suitable to manufacturing (such as free machining grades when machining is necessary). Surface finish can be controlled by rolling, polishing or blasting. Stainless steels are selected, first, for their corrosion resistance, second, for their strength and third, for their ease of fabrication. Most stainless steels are difficult to bend, draw and cut, requiring slow cutting speeds and special tool geometry. They are available in sheet, strip, plate, bar, wire, tubing and pipe, and can be readily soldered and braised. Welding stainless steel is possible but the filler metal must be selected to ensure an equivalent composition to maintain corrosion resistance. The 300 series are the most weldable; the 400 series are less weldable.

#### **Technical notes**



Stainless steels are classified into four categories: the 200and 300 series austenitic (Fe-Cr-Ni-Mn) alloys, the 400 series ferritic (Fe-Cr) alloys, the martensitic (Fe-Cr-C) alloys that also form part of the 400 series, and precipitation hardening or PH (Fe-Cr-Ni-Cu-Nb) alloys with designations starting with S. Typical of the austenitic grades of stainless steel is the grade 304: 74% iron, 18% chromium and 8 % nickel. Here the chromium protects by creating a protective Cr2O3 film on all exposed surfaces, and the nickel stabilizes face-centered cubic austenite, giving ductility and strength both at high and low temperatures; they are non-magnetic (a way of identifying them). The combination of austenitic and ferritic structures (the duplex stainless steels) provide considerably slower growth of stress-induced cracks, they can be hot-rolled or cast and are often heat treated as well. Austenitic stainless steel with high molybdenum content and copper has excellent resistance to pitting and corrosion. High nitrogen content austenitic stainless steel gives higher strength. Superferrites (over 30% chromium) are very resistant to corrosion, even in water containing chlorine. More information on designations and equivalent grades can be found on the Granta Design website at www.grantadesign.com/designations

#### Phase diagram



#### Phase diagram description

Most stainless steels are alloys of iron (Fe) with chromium (Cr) and nickel (Ni). This is the ternary phase diagram, at a temperature of 800 C, for those three elements. The position of AISI 302 stainless steel (Fe-18%Cr-8%Ni) is

### Typical uses

Railway cars, trucks, trailers, food-processing equipment, sinks, stoves, cooking utensils, cutlery, flatware, scissors and knives, architectural metalwork, laundry equipment, chemical-processing equipment, jet-engine parts, surgical tools, furnace and boiler components, oil-burner parts, petroleum-processing equipment, dairy equipment, heat-treating equipment, automotive trim. Structural uses in corrosive environments, e.g. nuclear plants, ships, offshore oil installations, underwater cables and pipes.



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