

Description

Image







Caption

1. Knot tied in a polyamide rope. © Brighterorange at en.wikipedia - (CC BY-SA 3.0) 2. Locking nut with polyamide insert to lock its screw in place. © Cav at en.wikipedia - Public domain 3. Students creating Nylon-6,6 in the laboratory at the University

The material

Back in 1945, the war in Europe just ended, the two most prized luxuries were cigarettes and nylons. Nylon (PA) can be drawn to fibers as fine as silk, and was widely used as a substitute for it. Today, newer fibers have eroded its dominance in garment design, but nylon-fiber ropes, and nylon as reinforcement for rubber (in car tires) and other polymers (PTFE, for roofs) remains important. It is used in product design for tough casings, frames and handles, and reinforced with glass - as bearings gears and other load-bearing parts. There are many grades (Nylon 6, Nylon 11....) each with slightly different properties.

Composition (summary)

(NH(CH2)5C0)n

General properties

| General properties | | | | |
|---|----------|---|--------|------------|
| Density | 69.9 | - | 71.2 | lb/ft^3 |
| Price | * 1.85 | - | 2.04 | USD/lb |
| Date first used | 1938 | | | |
| Mechanical properties | | | | |
| Young's modulus | 0.38 | - | 0.464 | 10^6 psi |
| Shear modulus | * 0.141 | - | 0.172 | 10^6 psi |
| Bulk modulus | 0.537 | _ | 0.566 | 10^6 psi |
| Poisson's ratio | 0.34 | - | 0.36 | · |
| Yield strength (elastic limit) | 7.25 | - | 13.7 | ksi |
| Tensile strength | 13.1 | _ | 23.9 | ksi |
| Compressive strength | 7.98 | _ | 15.1 | ksi |
| Elongation | 30 | _ | 100 | % strain |
| Hardness - Vickers | 25.8 | - | 28.4 | HV |
| Fatigue strength at 10^7 cycles | * 5.22 | _ | 9.57 | ksi |
| Fracture toughness | * 2.02 | - | 5.11 | ksi.in^0.5 |
| Mechanical loss coefficient (tan delta) | * 0.0125 | - | 0.0153 | |
| Thermal properties | | | | |
| Melting point | 410 | - | 428 | °F |
| Glass temperature | 111 | - | 133 | °F |
| Maximum service temperature | 230 | - | 284 | °F |



Polyamides (Nylons, PA)

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| Minimum service temperature | * -19099.7 | °F |
|---------------------------------|-----------------|-----------------|
| Thermal conductor or insulator? | Good insulator | |
| Thermal conductivity | 0.135 - 0.146 | BTU.ft/h.ft^2.F |
| Specific heat capacity | * 0.382 - 0.398 | BTU/lb.°F |
| Thermal expansion coefficient | 80 - 83 | µstrain/°F |

Electrical properties

| Electrical conductor or insulator? | Good insulator | |
|--|------------------------|-----|
| Electrical resistivity | * 1.5e19 - 1.4e20 µohm | .cm |
| Dielectric constant (relative permittivity) | 3.7 - 3.9 | |
| Dissipation factor (dielectric loss tangent) | 0.014 - 0.03 | |
| Dielectric strength (dielectric breakdown) | 384 - 417 V/mil | |

Optical properties

Machinability

Weldability

| Transparency | Translucent | | |
|------------------|-------------|--|--|
| Refractive index | 1.52 - 1.53 | | |
| Processability | | | |
| Castability | 1 - 2 | | |
| Moldability | 4 - 5 | | |

Durability: water and aqueous solutions

| Water (fresh) | Acceptable |
|------------------------|-------------|
| Water (salt) | Acceptable |
| Soils, acidic (peat) | Acceptable |
| Soils, alkaline (clay) | Limited use |
| Wine | Excellent |

Durability: acids

| Acceptable |
|--------------|
| Acceptable |
| Acceptable |
| Unacceptable |
| Limited use |
| Unacceptable |
| Unacceptable |
| Unacceptable |
| |

Durability: alkalis

| Sodium hydroxide (10%) | Limited use |
|------------------------|-------------|
| Sodium hydroxide (60%) | Limited use |

Durability: fuels, oils and solvents

| Amyl acetate | Unacceptable |
|----------------------|--------------|
| Benzene | Excellent |
| Carbon tetrachloride | Excellent |
| Chloroform | Acceptable |
| Crude oil | Acceptable |
| Diesel oil | Excellent |



Polyamides (Nylons, PA)

| Lubricating oil | Acceptable |
|--------------------------|------------|
| Paraffin oil (kerosene) | Excellent |
| Petrol (gasoline) | Excellent |
| Silicone fluids | Acceptable |
| Toluene | Excellent |
| Turpentine | Excellent |
| Vegetable oils (general) | Acceptable |
| White spirit | Acceptable |

Durability: alcohols, aldehydes, ketones

Acceptable
Acetone Excellent
Ethyl alcohol (ethanol) Excellent
Ethylene glycol Excellent
Formaldehyde (40%) Excellent
Glycerol Excellent
Methyl alcohol (methanol) Excellent

Durability: halogens and gases

Chlorine gas (dry)

Fluorine (gas)

O2 (oxygen gas)

Sulfur dioxide (gas)

Unacceptable

Excellent

Durability: built environments

Industrial atmosphereAcceptableRural atmosphereExcellentMarine atmosphereExcellentUV radiation (sunlight)Fair

Durability: flammability

Flammability Slow-burning

Durability: thermal environments

Tolerance to cryogenic temperatures

Tolerance up to 150 C (302 F)

Tolerance up to 250 C (482 F)

Tolerance up to 450 C (842 F)

Tolerance up to 850 C (1562 F)

Tolerance above 850 C (1562 F)

Unacceptable
Unacceptable
Unacceptable
Unacceptable
Unacceptable

Geo-economic data for principal component

Annual world production 3.64e6 - 3.69e6 ton/yr Reserves * 9.05e8 - 9.15e8 I. ton

Primary material production: energy, CO2 and water

Embodied energy, primary production * 1.25e4 1.38e4 kcal/lb CO2 footprint, primary production * 7.58 8.38 lb/lb * 21.1 Water usage 23.2 gal(US)/lb Eco-indicator 95 630 millipoints/kg Eco-indicator 99 495 millipoints/kg

Material processing: energy

Polymer extrusion energy * 638 - 706 kcal/lb Polymer molding energy * 2.24e3 - 2.48e3 kcal/lb



| Coarse machining energy (per unit wt removed) | * 142 | _ | 157 | kcal/lb |
|--|-----------|---|--------|---------|
| Fine machining energy (per unit wt removed) | * 956 | _ | 1.06e3 | kcal/lb |
| Grinding energy (per unit wt removed) | * 1.86e3 | | 2.06e3 | kcal/lb |
| Gillialing energy (per ariit wt removed) | 1.0063 | - | 2.0063 | KCal/ID |
| Material processing: CO2 footprint | | | | |
| Polymer extrusion CO2 | * 0.442 | - | 0.489 | lb/lb |
| Polymer molding CO2 | * 1.55 | - | 1.72 | lb/lb |
| Coarse machining CO2 (per unit wt removed) | * 0.0982 | - | 0.109 | lb/lb |
| Fine machining CO2 (per unit wt removed) | * 0.662 | - | 0.731 | lb/lb |
| Grinding CO2 (per unit wt removed) | * 1.29 | - | 1.42 | lb/lb |
| Material recycling: energy, CO2 and recycle from | action | | | |
| Recycle | ✓ | | | |
| Embodied energy, recycling | * 4.37e3 | - | 4.82e3 | kcal/lb |
| CO2 footprint, recycling | * 3.17 | - | 3.5 | lb/lb |
| Recycle fraction in current supply | * 0.5 | - | 1 | % |
| Downcycle | 1 | | | |
| Combust for energy recovery | ✓ | | | |
| Heat of combustion (net) | * 3.26e3 | - | 3.42e3 | kcal/lb |
| Combustion CO2 | * 2.28 | - | 2.39 | lb/lb |
| Landfill | 1 | | | |
| Biodegrade | × | | | |
| Toxicity rating | Non-toxic | | | |
| A renewable resource? | × | | | |

Environmental notes

Nylons have no known toxic effects, although they are not entirely inert biologically. Nylons are oil-derivatives, but this will not disadvantage them in the near future. With refinements in polyolefin catalysis, nylons face stiff competition from less expensive polymers.

Recycle mark



Supporting information

Design guidelines

Nylons are tough, strong and have a low coefficient of friction, with useful properties over a wide range of temperature (-80 to +120 C). They are easy to injection mold, machine and finish, can be thermally or ultrasonically bonded, or joined with epoxy, phenol-formaldehyde or polyester adhesives. Certain grades of nylon can be electroplated allowing metallization, and most accept print well. A blend of PPO/Nylon is used in fenders, exterior body parts. Nylon fibers are strong, tough, elastic and glossy, easily spun into yarns or blended with other materials. Nylons absorb up to 4% water; to prevent dimensional changes, they must be conditioned before molding, allowing them to establishing equilibrium with normal atmospheric humidity. Nylons have poor resistance to strong acids, oxidizing agents and solvents, particularly in transparent grades.

Technical notes

The density, stiffness, strength, ductility and toughness of Nylons all lie near the average for unreinforced polymers. Their thermal conductivities and thermal expansion are a little lower than average. Reinforcement with mineral, glass powder or glass fiber increases the modulus, strength and density. Semi-crystalline nylon is distinguished by a numeric code for the material class indicating the number of carbon atoms between two nitrogen atoms in the molecular chain. The amorphous material is transparent; the semi-crystalline material is opal white.

Typical uses

Polyamides (Nylons, PA)



Light duty gears, bushings, sprockets and bearings; electrical equipment housings, lenses, containers, tanks, tubing, furniture casters, plumbing connections, bicycle wheel covers, ketchup bottles, chairs, toothbrush bristles, handles, bearings, food packaging. Nylons are used as hot-melt adhesives for book bindings; as fibers - ropes, fishing line, carpeting, car upholstery and stockings; as aramid fibers - cables, ropes, protective clothing, air filtration bags and electrical insulation.

Tradenames

Adell, Akulon, Albis, Amilan, Ashlene, Capron, Celanese, Chemlon, Durethan, Gapex, Grilon, Grivory, Hylon, Kopa, Latamid, Lubrilon, Magnacomp, Maranyl, Minlon, NSC, Nivionplast, Novamid, Nydur, Nylamid, Nylene, Nypel, Orgamide, Radilon, Schulamid, Selar, Sniamid, Star-C, Star-L, Staramide, Texalon, Ultramid, Vestamid, Wellamid, Zytel

Links

Reference

ProcessUniverse

Producers