

## **Description**

## **Image**



### Caption

Open-cell alumina foam sample. © Granta

### The material

If you wanted to filter dross out of a molten metal at -- say -- 1800 C, how would you do it? The answer is to pour it through a ceramic foam. A range of such foams is now available with relative densities between 5% and 40% (the volume fraction of solid in the foam), with pore sizes between 0.05 and 2 mm. They are available in alumina, zirconia, and a number of other ceramics including the bio-ceramic hydroxyapatite. Their great merit is their stability at high temperatures: zirconia foams, for instance, can be use at up to 2000 Cathie data given here is for a medium-density alumina foam

## **Compositional summary**

Al2O3, ZrO2 and many other compositions.

## **General properties**

Density	24.5	-	41.8	lb/ft^3
Price	* 15.9	-	24	USD/lb
Date first used	1970			

## **Mechanical properties**

Young's modulus	0.276	-	0.551	10^6 psi
Shear modulus	* 0.087	-	0.203	10^6 psi
Bulk modulus	* 0.276	-	0.551	10^6 psi
Poisson's ratio	0.26	-	0.27	
Yield strength (elastic limit)	* 0.087	-	0.305	ksi
Tensile strength	* 0.087	-	0.305	ksi
Compressive strength	0.116	-	0.406	ksi
Elongation	0			% strain
Hardness - Vickers	0.08	-	0.3	HV
Fatigue strength at 10^7 cycles	* 0.0725	-	0.261	ksi



Fracture toughness	0.0273	-	0.091	ksi.in^0.5
Mechanical loss coefficient (tan delta)	* 0.003	-	0.01	

# **Thermal properties**

Melting point	3.59e3 - 3.8e3	°F
Maximum service temperature	2.73e3 - 3.27e3	°F
Minimum service temperature	-459	°F
Thermal conductor or insulator?	Good insulator	
Thermal conductivity	0.289 - 0.404	BTU.ft/h.ft^2.F
Specific heat capacity	0.186 - 0.229	BTU/lb.°F
Thermal expansion coefficient	3.86 - 4.94	μstrain/°F

# **Electrical properties**

Electrical conductor or insulator?	Good in	sula	tor	
Electrical resistivity	1e20	-	1e23	µohm.cm
Dielectric constant (relative permittivity)	1.7	-	2.4	
Dissipation factor (dielectric loss tangent)	0.001	-	0.02	
Dielectric strength (dielectric breakdown)	254	-	432	V/mil

# **Optical properties**

Transparency	Translucent

# **Processability**

*		
Machinability	3 - 4	

# **Durability: water and aqueous solutions**

Water (fresh)	Excellent
Water (salt)	Excellent
Soils, acidic (peat)	Excellent
Soils, alkaline (clay)	Excellent
Wine	Excellent

# **Durability: acids**

Acetic acid (10%)	Excellent
Acetic acid (glacial)	Excellent
Citric acid (10%)	Excellent
Hydrochloric acid (10%)	Excellent
Hydrochloric acid (36%)	Excellent
Hydrofluoric acid (40%)	Unacceptable
Nitric acid (10%)	Excellent
Nitric acid (70%)	Excellent
Phosphoric acid (10%)	



	Excellent
Phosphoric acid (85%)	Excellent
Sulfuric acid (10%)	Excellent
Sulfuric acid (70%)	Acceptable

# **Durability: alkalis**

Sodium hydroxide (10%)	Excellent
Sodium hydroxide (60%)	Excellent

# **Durability: fuels, oils and solvents**

Amyl acetate	Excellent
Benzene	Excellent
Carbon tetrachloride	Excellent
Chloroform	Excellent
Crude oil	Excellent
Diesel oil	Excellent
Lubricating oil	Excellent
Paraffin oil (kerosene)	Excellent
Petrol (gasoline)	Excellent
Silicone fluids	Excellent
Toluene	Excellent
Turpentine	Excellent
Vegetable oils (general)	Excellent
White spirit	Excellent

# Durability: alcohols, aldehydes, ketones

Acetaldehyde	Excellent
Acetone	Excellent
Ethyl alcohol (ethanol)	Excellent
Ethylene glycol	Excellent
Formaldehyde (40%)	Excellent
Glycerol	Excellent
Methyl alcohol (methanol)	Excellent

# **Durability: halogens and gases**

Chlorine gas (dry)	Excellent
Fluorine (gas)	Excellent
O2 (oxygen gas)	Excellent
Sulfur dioxide (gas)	Excellent

# **Durability: built environments**



Industrial atmosphere	Excellent
Rural atmosphere	Excellent
Marine atmosphere	Excellent
UV radiation (sunlight)	Excellent

# **Durability: flammability**

Flammability Non-	flammable
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# **Durability: thermal environments**

Tolerance to cryogenic temperatures	Excellent
Tolerance up to 150 C (302 F)	Excellent
Tolerance up to 250 C (482 F)	Excellent
Tolerance up to 450 C (842 F)	Excellent
Tolerance up to 850 C (1562 F)	Excellent
Tolerance above 850 C (1562 F)	Excellent

# Primary material production: energy, CO2 and water

Embodied energy, primary production	* 1.31e4	-	1.45e4	kcal/lb
CO2 footprint, primary production	* 6.53	-	7.22	lb/lb
Water usage	* 20	-	22.2	gal(US)/lb

# **Material processing: energy**

Grinding energy (per unit wt removed)	* 185	- 205	kcal/lb		
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## **Material processing: CO2 footprint**

Grinding CO2 (per unit wt removed)	* 0.128	- 0.142	lb/lb
Gilliand CO2 (bei anit wi teniovea)	0.120	- 0.142	ID/ID

# Material recycling: energy, CO2 and recycle fraction

Recycle	×
Recycle fraction in current supply	0.1 %
Downcycle	✓
Combust for energy recovery	×
Landfill	✓
Biodegrade	×
Toxicity rating	Non-toxic
A renewable resource?	×

## **Environmental notes**

Ceramics, once fired, are inert. Some can even be used for human

# **Supporting information**

Design guidelines



Most ceramic foams are used as high-temperature filters for liquids and gasses (removing carbon particulates from diesel exhaust, for instance), and for high temperature thermal insulation, replacing materials like asbestos. They can be cut with saws, but this leaves an uneven surface. Better is the use of abrasive water-jet cutting. Interpenetrating composites ("3-3 composites") are made by infiltrating the ceramic with a polymer or a metal, giving a fully dense body. They have potential for wear-resistant parts and as armor to resist projectiles. Hydroxyapatite foams are biocompatible and have a pore size that encourages attachment and in-growth of cell.

### **Technical notes**

Ceramic foams are made in a number of ways. Typical of them is co-foaming of a ceramic-polymer slurry. Fine ceramic powder is mixed with an acrylate monomer, a catalyst and a foaming agent. The concoction foams then polymerizes in much the same way that polymer foams are made, but the cell edges are full of ceramic powder. When this is sintered (heated at 1500 to 2000 C) the polymer burns off and the powder particles bond to give a consolidated ceramic foam.

## Typical uses

High temperature filters for liquids and gases; catalyst carriers for chemical processing and catalytic converters; high temperature thermal insulation; as a first step in making 3-dimensionally interpenetrating composites; as a bioactive medium for cell attachment in implants.

# Links

ProcessUniverse

Reference