

Description

Image



Caption

1. 1950's telephone made using phenolics. © F_l_a_n_k_e_r at en.wikipedia - (CC BY 3.0) 2. Lamp made using phenolics. © Chris Lefteri

The material

Bakelite, commercialized in 1909, triggered a revolution in product design. It was stiff, fairly strong, could (to a muted degree) be colored, and - above all - was easy to mold. Products that, earlier, were handcrafted from woods, metals or exotics such as ivory, could now be molded quickly and cheaply. At one time the production of phenolics exceeded that of PE, PS and PVC combined. Now, although the ration has changed, phenolics still have a unique value. They are stiff, chemically stable, have good electrical properties, are fire-resistant and easy to mold - and they are cheap.

General properties

| | | | | |
|-----------------|--------|---|--------|-------------------|
| Density | 1.24e3 | - | 1.32e3 | kg/m ³ |
| Price | 1.76 | | | USD/kg |
| Date first used | 1909 | | | |

Mechanical properties

| | | | | |
|--|-----------|---|--------|----------------------|
| Young's modulus | 2.76 | - | 4.83 | GPa |
| Shear modulus | * 0.996 | - | 1.74 | GPa |
| Bulk modulus | 5.2 | - | 5.4 | GPa |
| Poisson's ratio | * 0.378 | - | 0.394 | |
| Yield strength (elastic limit) | * 27.6 | - | 49.7 | MPa |
| Tensile strength | 34.5 | - | 62.1 | MPa |
| Compressive strength | * 30.4 | - | 54.6 | MPa |
| Elongation | 1.5 | - | 2 | % strain |
| Hardness - Vickers | 8.3 | - | 14.9 | HV |
| Fatigue strength at 10 ⁷ cycles | * 13.8 | - | 24.8 | MPa |
| Fracture toughness | * 0.787 | - | 1.21 | MPa.m ^{0.5} |
| Mechanical loss coefficient (tan delta) | * 0.00828 | - | 0.0145 | |

Thermal properties

| | | | | |
|---------------------------------|----------------|---|--------|------------|
| Glass temperature | 167 | - | 267 | °C |
| Maximum service temperature | * 200 | - | 230 | °C |
| Minimum service temperature | * -123 | - | -73.2 | °C |
| Thermal conductor or insulator? | Good insulator | | | |
| Thermal conductivity | 0.141 | - | 0.152 | W/m.°C |
| Specific heat capacity | * 1.47e3 | - | 1.53e3 | J/kg.°C |
| Thermal expansion coefficient | 120 | - | 125 | μstrain/°C |

Electrical properties

| | | | | |
|--|----------------|---|------|-------------|
| Electrical conductor or insulator? | Good insulator | | | |
| Electrical resistivity | 3.3e18 | - | 3e19 | μohm.cm |
| Dielectric constant (relative permittivity) | * 4 | - | 6 | |
| Dissipation factor (dielectric loss tangent) | * 0.005 | - | 0.01 | |
| Dielectric strength (dielectric breakdown) | 9.84 | - | 15.7 | 1000000 V/m |

Optical properties

| | | | | |
|------------------|--------|---|-----|--|
| Transparency | Opaque | | | |
| Refractive index | 1.59 | - | 1.6 | |

Processability

| | | | | |
|---------------|---|---|---|--|
| Castability | 3 | - | 4 | |
| Moldability | 3 | - | 5 | |
| Machinability | 4 | | | |
| Weldability | 1 | | | |

Durability: water and aqueous solutions

| | |
|------------------------|--------------|
| Water (fresh) | Excellent |
| Water (salt) | Excellent |
| Soils, acidic (peat) | Excellent |
| Soils, alkaline (clay) | Unacceptable |
| Wine | Excellent |

Durability: acids

| | |
|-------------------------|--------------|
| Acetic acid (10%) | Excellent |
| Acetic acid (glacial) | Excellent |
| Citric acid (10%) | Excellent |
| Hydrochloric acid (10%) | Excellent |
| Hydrochloric acid (36%) | Excellent |
| Hydrofluoric acid (40%) | Unacceptable |
| Nitric acid (10%) | Excellent |

| | |
|-----------------------|--------------|
| Nitric acid (70%) | Unacceptable |
| Phosphoric acid (10%) | Excellent |
| Phosphoric acid (85%) | Excellent |
| Sulfuric acid (10%) | Excellent |
| Sulfuric acid (70%) | Limited use |

Durability: alkalis

| | |
|------------------------|--------------|
| Sodium hydroxide (10%) | Unacceptable |
| Sodium hydroxide (60%) | Unacceptable |

Durability: fuels, oils and solvents

| | |
|--------------------------|------------|
| Amyl acetate | Excellent |
| Benzene | Excellent |
| Carbon tetrachloride | Excellent |
| Chloroform | Excellent |
| Crude oil | Excellent |
| Diesel oil | Excellent |
| Lubricating oil | Excellent |
| Paraffin oil (kerosene) | Excellent |
| Petrol (gasoline) | Excellent |
| Silicone fluids | Acceptable |
| Toluene | Excellent |
| Turpentine | Excellent |
| Vegetable oils (general) | Excellent |
| White spirit | Acceptable |

Durability: alcohols, aldehydes, ketones

| | |
|---------------------------|-----------|
| Acetaldehyde | Excellent |
| Acetone | Excellent |
| Ethyl alcohol (ethanol) | Excellent |
| Ethylene glycol | Excellent |
| Formaldehyde (40%) | Excellent |
| Glycerol | Excellent |
| Methyl alcohol (methanol) | Excellent |

Durability: halogens and gases

| | |
|----------------------|--------------|
| Chlorine gas (dry) | Unacceptable |
| Fluorine (gas) | Unacceptable |
| O2 (oxygen gas) | Unacceptable |
| Sulfur dioxide (gas) | Excellent |

Durability: built environments

| | |
|-------------------------|------------|
| Industrial atmosphere | Acceptable |
| Rural atmosphere | Excellent |
| Marine atmosphere | Excellent |
| UV radiation (sunlight) | Good |

Durability: flammability

| | |
|--------------|--------------------|
| Flammability | Self-extinguishing |
|--------------|--------------------|

Durability: thermal environments

| | |
|-------------------------------------|--------------|
| Tolerance to cryogenic temperatures | Unacceptable |
| Tolerance up to 150 C (302 F) | Excellent |
| Tolerance up to 250 C (482 F) | Acceptable |
| Tolerance up to 450 C (842 F) | Unacceptable |
| Tolerance up to 850 C (1562 F) | Unacceptable |
| Tolerance above 850 C (1562 F) | Unacceptable |

Geo-economic data for principal component

| | | | | |
|--|----------|---|--------|----------|
| Annual world production, principal component | 1e7 | - | 1.05e7 | tonne/yr |
| Reserves, principal component | * 2.51e8 | - | 2.55e8 | tonne |

Primary material production: energy, CO2 and water

| | | | | |
|-------------------------------------|--------|---|------|-------|
| Embodied energy, primary production | * 75.3 | - | 83.3 | MJ/kg |
| CO2 footprint, primary production | * 3.44 | - | 3.81 | kg/kg |
| Water usage | * 49.2 | - | 54.4 | l/kg |

Material processing: energy

| | | | | |
|---|--------|---|------|-------|
| Polymer molding energy | * 26.6 | - | 29.4 | MJ/kg |
| Coarse machining energy (per unit wt removed) | * 1.16 | - | 1.28 | MJ/kg |
| Fine machining energy (per unit wt removed) | * 7.33 | - | 8.1 | MJ/kg |
| Grinding energy (per unit wt removed) | * 14.2 | - | 15.7 | MJ/kg |

Material processing: CO2 footprint

| | | | | |
|--|---------|---|--------|-------|
| Polymer molding CO2 | * 2.13 | - | 2.35 | kg/kg |
| Coarse machining CO2 (per unit wt removed) | * 0.087 | - | 0.0962 | kg/kg |
| Fine machining CO2 (per unit wt removed) | * 0.55 | - | 0.607 | kg/kg |
| Grinding CO2 (per unit wt removed) | * 1.06 | - | 1.18 | kg/kg |

Material recycling: energy, CO2 and recycle fraction

| | | | | |
|------------------------------------|-----|---|---|---|
| Recycle | ✗ | | | |
| Recycle fraction in current supply | 0.5 | - | 1 | % |
| Downcycle | ✓ | | | |

| | | | | |
|-----------------------------|-----------|---|------|-------|
| Combust for energy recovery | ✓ | | | |
| Heat of combustion (net) | * 31.5 | - | 33.1 | MJ/kg |
| Combustion CO2 | * 2.86 | - | 3.01 | kg/kg |
| Landfill | ✓ | | | |
| Biodegrade | ✗ | | | |
| Toxicity rating | Non-toxic | | | |
| A renewable resource? | ✗ | | | |

Environmental notes

Thermosetting phenolics are recyclable, but by a different means than that for thermoplastics. Molded phenolic, ground into a fine powder, can be added to the raw material stream. 4% to 12% ground phenolic does not degrade properties.

<http://www.phenolics.org/Publications/recycling2.htm>

Supporting information

Design guidelines

Phenolic resins hard, tolerate heat and resist most chemicals except the strong alkalis. Phenolic laminates with paper have excellent electrical and mechanical properties and are cheap; filled with cotton the mechanical strength increases and a machined surface is finer; filled with glass the mechanical strength increases again and there is improved chemical resistance. Fillers play three roles: extenders (such as wood flour and mica) are inexpensive and reduce cost; functional fillers add stiffness, impact resistance and limit shrinkage; reinforcements (such as glass, graphite and polymer fibers) increase strength, but cost increases too. Phenolic resins have creep resistance, and they self-extinguish in a fire. They can be cast (household light and switch fittings) and are available as rod and sheet. Impregnated into paper (Nomex) and cloth (Tufnol), they have exceptional durability, chemical resistance and bearing properties. Phenolics accept paint, electroplating, and melamine overlays.

Technical notes

Phenolic resins are formed by a condensation, generating water in the process, involving a reaction between phenol and formaldehyde to form the A-stage resin. Fillers, colorants, lubricants and chemicals to cause cross-linking are added to form the B-stage resin. This resin is then fused under heat and pressure converting to the final product - a C-stage resin - or completely cross-linked polymer.

Typical uses

Electrical parts - sockets, switches, connectors, general industrial, water-lubricated bearings, relays, pump impellers, brake pistons, brake pads, microwave cookware, handles, bottles tops, coatings, adhesives, bearings, foams and sandwich structures.

Tradenames

Bakelite, Durez, Ferroreg, Fiberite, Norsophen, Plaslok, Plenco, Polychem, Reliapreg, Resinoid, Texolite, Trolitan, Vyncolite, Tufnol

Links

Reference

ProcessUniverse

Producers

