

## Relationship Between Energy, Momentum, Wavelength

Momentum (p) to classical kinetic energy (E):

$$E = \frac{p^2}{2m}$$

Energy to wavelength (De Broglie) [1]:

$$\lambda = \frac{h}{p}$$

## Bibliography

- [1] “The Schrödinger Equation.” [Online]. Available: [https://math.libretexts.org/Bookshelves/Differential\\_Equations/Differential\\_Equations\\_\(Chasnov\)/09%3A\\_Partial\\_Differential\\_Equations/9.08%3A\\_The\\_Schrodinger\\_Equation](https://math.libretexts.org/Bookshelves/Differential_Equations/Differential_Equations_(Chasnov)/09%3A_Partial_Differential_Equations/9.08%3A_The_Schrodinger_Equation)