

(Ch. 7) Fluid + Thermal Systems

Like electrical systems, many fluid + thermal systems can be modeled with similar differential equations used for mechanical systems. Models of fluid + thermal systems are often more complicated than most mechanical + electrical systems, therefore, we will only consider a few basic examples here.

(7.1) Fluid System Background

Conservation of Mass (Main modeling principle)

$$\dot{m} = \dot{q}_{mi} - \dot{q}_{mo} \quad \dot{m} \left(\frac{dm}{dt} \right) = \text{time rate of change of mass in system (kg/s)}$$

$$\dot{q}_{mi} = \text{mass flow rate in (kg/s)}$$

$$\dot{q}_{mo} = \text{mass flow rate out (kg/s)}$$

Note: for incompressible fluids, conservation of mass is equivalent to conservation of volume.

Mass/Volume Flow Rate

$$\dot{q}_m = \rho \dot{q}_v \quad \dot{q}_m = \text{mass flow rate (kg/s)}$$

$$\rho = \text{density (kg/m}^3\text{)}$$

$$\dot{q}_v = \text{volume flow rate (m}^3\text{/s)}$$

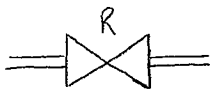
Pressure

$$p = \text{pressure (N/m}^2 \text{ or Pa)}$$

$$p_a = \text{atmospheric pressure (1.0133} \times 10^5 \text{ Pa @ sea level)}$$

$$p_{gh} = \text{hydrostatic pressure, caused by weight of fluid}$$

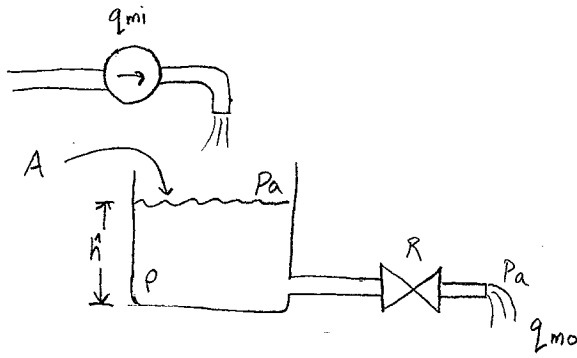
Fluid Resistance



- Fluid meets resistance when flowing through pipes, valves, orifices, etc.
- Mass flow rate, \dot{q}_m , through a resistance is related to pressure diff, \hat{p} , across it.
- While typically nonlinear, the resistance can be approximated around a reference state to yield a linear relation:

$$R \dot{q}_m = \hat{p} \quad \text{where } \hat{p} = \Delta p \text{ across resistance}$$

Liquid Level System

 q_{mi} = mass flow rate in q_{mo} = mass flow rate out R = outlet resistance (linear about reference state) h = fluid height $\Rightarrow \hat{h} = h_r + h$ where h_r = reference height, h is deviation from reference A = area of tank p_a = atmospheric pressure ρ = density of fluidDevelop a model of h , the deviation from reference height

Conservation of mass: $\frac{dm}{dt} = q_{mi} - q_{mo}$ mass in tank: $m = \rho A \hat{h} = \rho A(h + h_r)$

$$\frac{d(\rho A(h + h_r))}{dt} = \frac{d(\rho A h + \rho A h_r)}{dt} = \rho A \frac{dh}{dt} = q_{mi} - q_{mo} \quad (\text{b/c } \rho, A, h_r = \text{cst.})$$

We can find a relation between the outlet resistance & mass flow rate out

$$R q_{mo} = \Delta p, \text{ where } \Delta p = \text{pressure diff. across valve}$$

$$R q_{mo} = (pgh + p_a) - p_a \Rightarrow q_{mo} = \frac{\rho g h}{R}$$

Substituting:

$$\rho A \frac{dh}{dt} + \frac{\rho g}{R} h = q_{mi}$$

which is a classic first order ODE ($ax + bx = f$)

(7.6) Thermal Systems Background

Conservation of Heat Energy (Main modeling principle)

$$\dot{E}_{st} = \dot{E}_{in} - \dot{E}_{out}$$

E_{st} = stored energy in a control volume

E_{in} = energy entering control volume

E_{out} = energy exiting control volume

Thermal Capacitance of a mass

$$E_{st} = C \Delta T$$

where $C = \rho V c_p$

C = thermal capacitance ($J/^\circ C$)

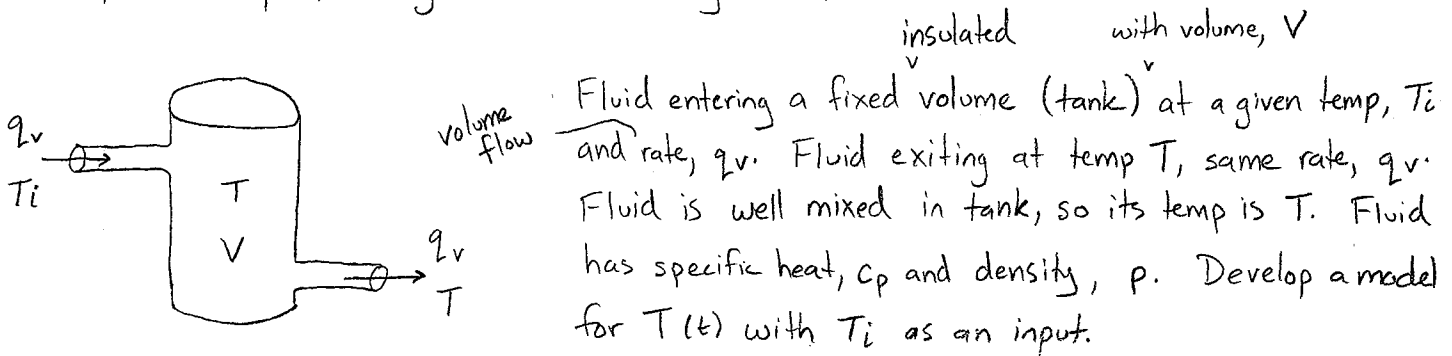
$\Delta T = T - T_r$ = Temperature difference ($^\circ C$)

ρ = density of mass (kg/m^3)

V = volume of mass (m^3)

c_p = specific heat ($J/kg^\circ C$)

Example: Temperature Dynamics of a Mixing Process



Apply conservation of energy to find model.

- Stored energy in tank: $E_{st} = \rho V c_p (T - T_r)$ where T_r is arbitrarily selected reference temperature
- Energy flowing in: $\dot{E}_{in} = \rho q_v c_p (T_i - T_r)$ note, this is a rate of energy flow b/c we know q_v , a volume flow rate
- Energy flowing out: $\dot{E}_{out} = \rho q_v c_p (T - T_r)$

Apply conservation of energy:

$$\dot{E}_{st} = \dot{E}_{in} - \dot{E}_{out} \Rightarrow \frac{d}{dt} [\rho V c_p (T - T_r)] = \rho q_v c_p (T_i - T_r) - \rho q_v c_p (T - T_r)$$

$$\rho V c_p \frac{dT}{dt} = \rho q_v c_p (T_i - T)$$

cancel ρc_p : $V \frac{dT}{dt} = q_v (T_i - T)$

rearrange: $\boxed{\frac{V}{q_v} \frac{dT}{dt} + T = T_i}$

which is a classic 1st order ODE ($ax + bx = f$)