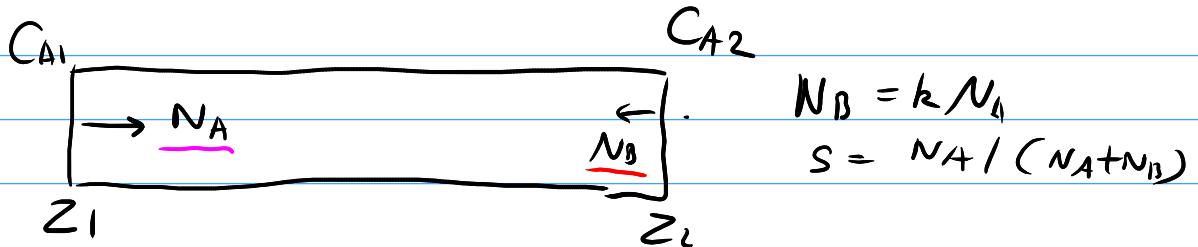


CHE 318 L04 Theories of D_{AB} (gas)

Jan -12 2026

I Recap
General solution

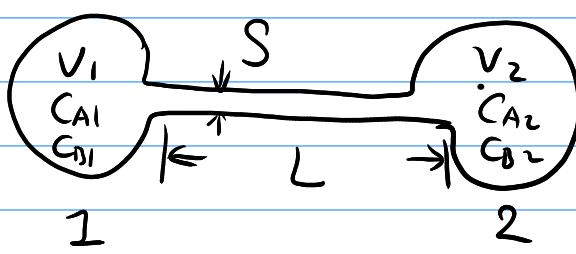


$$N_A = \frac{C_T D_{AB}}{z_2 - z_1} \cdot S \ln \left(\frac{S - x_{A2}}{S - x_{A1}} \right)$$

$N_A = \text{const}$

$x_A = \text{function of } z$

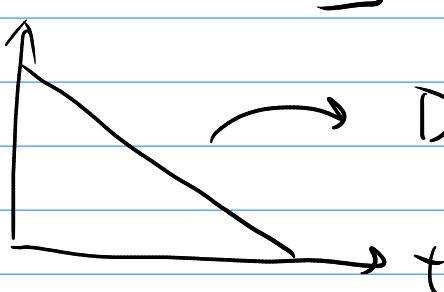
Measuring D_{AB} : two bulb Eudi



$t=0$ pure A in 1 Open
pure B in 2

$t=t$ mixed Close

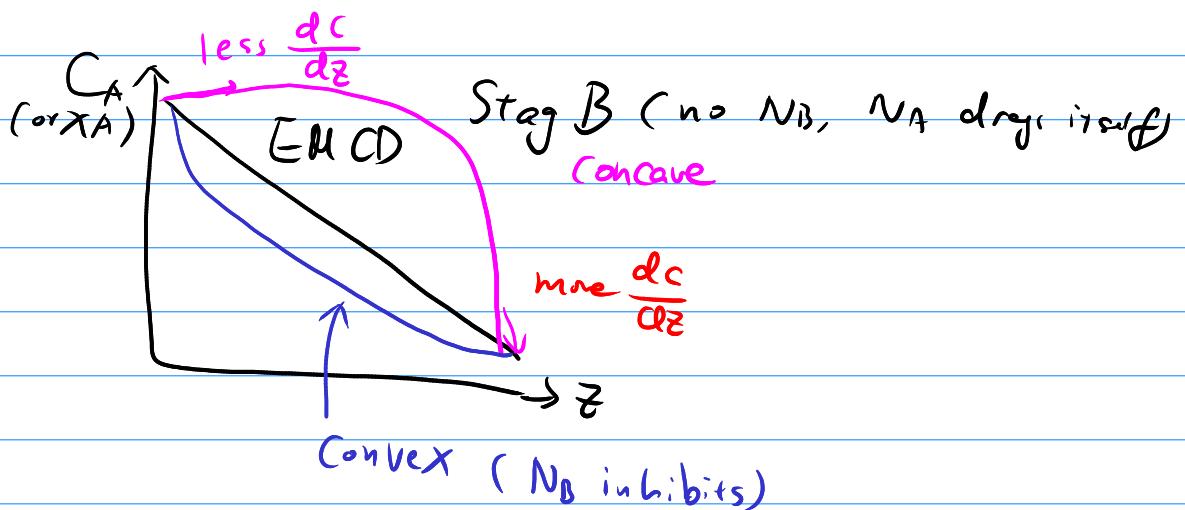
$$\ln \left(\frac{C_{A,av} - C_{A2}^{t=t}}{C_{A,av} - C_{A1}^{t=0}} \right) = - D_{AB} \cdot \frac{V_T S}{V_1 V_2 L} \cdot t$$



D_{AB} from slope

For gas $D_{AB} \approx 10^{-5} \text{ m}^2/\text{s}$

2. Demo! See how flux change



All depends on S value

$$\text{Stage B } S=1 \quad N_{\text{Stage}} = N_{\text{EMCD}} \frac{1}{x_{Bm}} (x_{A1} - x_{A2})$$

$\text{or log mean } J < 1$

3. Today how do we get D_{AB} ?

without experiment

$$\textcircled{1} \quad D_{AB} = f(T, P)$$

$$\textcircled{2} \quad D_{AB}(z) = \text{const}$$

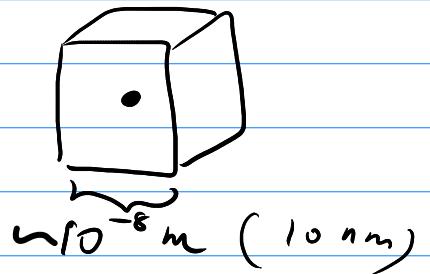
$$\textcircled{3} \quad D_{AB} = D_{BA} \quad (\text{see assignment 1 Q1})$$

Kinetic theory of gas

Diluted gas

$$C = \frac{n}{V} = \frac{P}{RT}$$

at 1 atm $C = 10^{25} \text{ m}^{-3}$



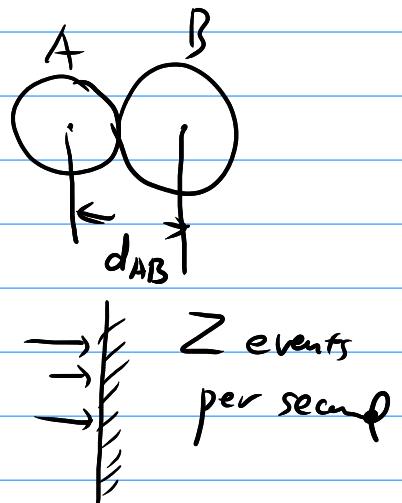
Assumptions

- (1) A, B rigid spheres
- (2) Far away only 2 can collide
- (3) Momentum conservation
- (4) Molecules travel very far (λ_{AB}) before colliding

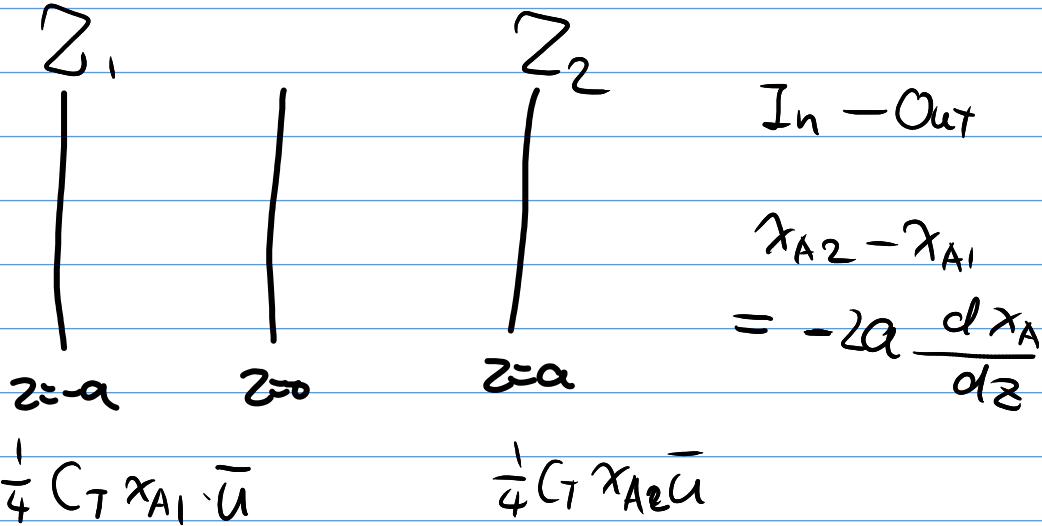
$$u = \sqrt{\frac{8k_B T}{\pi \bar{m}_{AB}}} \quad \bar{m}_{AB} = \left(\frac{1}{m_A} + \frac{1}{m_B} \right)^{-1}$$

$$\lambda_{AB} = \frac{1}{\sqrt{2\pi d_{AB}^2} C_T}$$

$$Z = \frac{1}{4} C_A \bar{u}$$



$$a = \frac{2}{3} \lambda_{AB} \quad \begin{matrix} \text{avg. path} \\ \text{collision} \longleftrightarrow \text{plane} \end{matrix}$$



Molecular: $a = \frac{2}{3} \lambda_{AB}$

$$J = -\frac{1}{4} C_T \cdot 2a \cdot \bar{u} \frac{dx_A}{dz}$$

Macro

$$J = -D_{AB} C_T \frac{dx_A}{dz}$$

$$\Rightarrow D_{AB} = \frac{1}{3} \bar{u} \lambda_{AB}$$

$$\propto \sqrt{\frac{T}{m_{AB}}} \frac{1}{d_{AB}^2 C_T} \quad C_T \propto \frac{P_T}{T}$$

$$\propto \sqrt{T^3} \left(\frac{1}{m_{AB}}\right)^{\frac{1}{2}} \cdot \frac{1}{d_{AB}^2} \cdot \frac{1}{P_T}$$

$$\propto T^{1.5} \left(\frac{1}{m_{AB}}\right)^{\frac{1}{2}} \frac{1}{d_{AB}^2} \frac{1}{P_T}$$

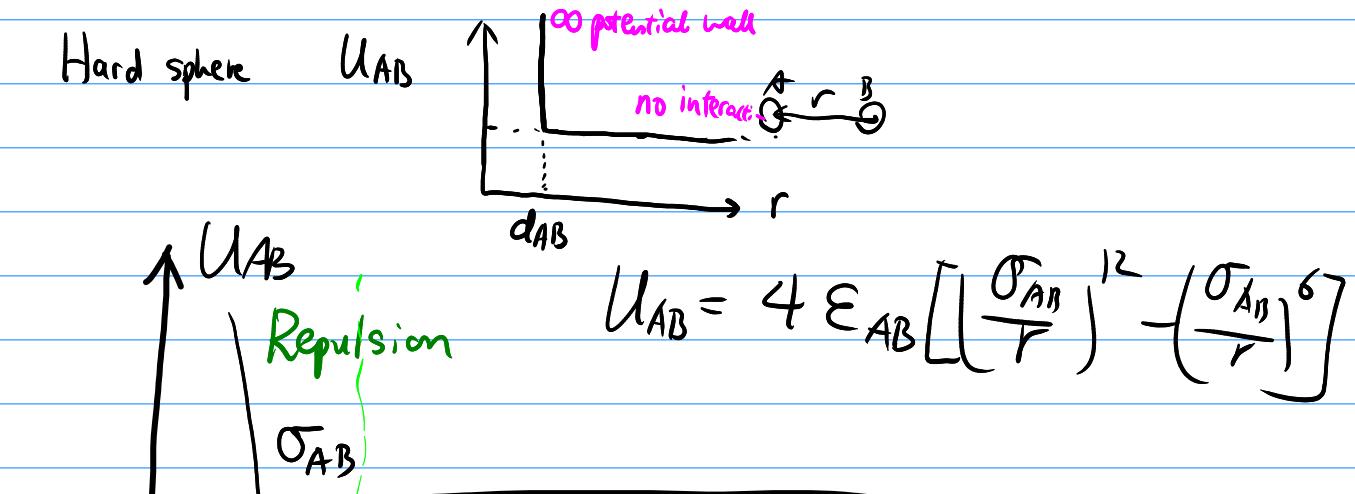
Note the powers!

Chapman - Enskog Theory

$$D_{AB} = \frac{1.8583 \times 10^{-7} T^{1.5}}{P \sigma_{AB}^2 \Omega_{DAD}}$$

↓ ↓ ↓ ↓
 In m²/s In atm In Å kg/(kg mol)

(Chapman - Enskog theory is just hard-sphere kinetic theory on steroid!)



$$U_{AB} = 4 \epsilon_{AB} \left[\left(\frac{\sigma_{AB}}{r} \right)^12 - \left(\frac{\sigma_{AB}}{r} \right)^6 \right]$$

$$\sigma_{AB} = \frac{\sigma_A + \sigma_B}{2}$$

$$\epsilon_{AB} = \sqrt{\epsilon_A \epsilon_B}$$

$$T^* = \frac{k_B T}{\epsilon_{AB}}$$

$$\Omega_{DAD}(T^*)$$

Fuller Method

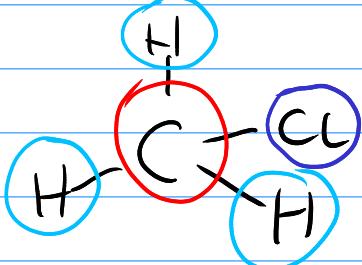
$$D_{AB} = \frac{1.0 \times 10^{-7} T^{1.75}}{PT \left(\left(\sum v_{Ai} \right)^{\frac{1}{3}} + \left(\sum v_{Bi} \right)^{\frac{1}{3}} \right)^2}$$

$\sum v_{Ai}$ has meaning of volume $\Rightarrow [Length]^3$

$\left[\sum v_{Ai}^{\frac{1}{3}} + \sum v_{Bi}^{\frac{1}{3}} \right]^2 \Rightarrow$ has same meaning/unit
as d_{AB}

$$\left(\frac{1}{m_A} + \frac{1}{m_B} \right)^{\frac{1}{2}}$$

$\sum v_{Ai} =$ sum of volum contrib



$$\begin{aligned} \sum v_{CH_3Cl} &= v_C + 3 \cdot v_H \\ &\quad + 2 \cdot v_{Cl} \end{aligned}$$