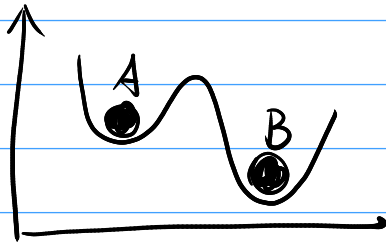


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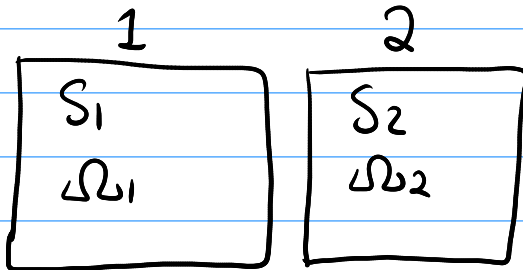
$$\Delta G^\circ = G_B - G_A$$

means reaction free energy

At reaction equilibrium
 $\Delta G = 0$ (change of A, B ratio)
 $\approx \Delta G^\circ + RT \ln \frac{x_B}{x_A}$

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Logarithm of microstate counting func



$$S_T = S_1 + S_2 \Leftarrow S \text{ (entropy) is extensive}$$

$$\Omega_T = \Omega_1 \cdot \Omega_2$$

