# IB Physics Topic E4 Nuclear Fission; HL

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# Table of Contents

1	Ind	uced and Spontaneous Fission	1
2	The	e Fission Mechanism	2
	2.1	Fission Products — Percentage Yield	3
3	Cha	nin Reactions	4
4	Fiss	sion Reactors	5
	4.1	Moderators	6
		Control Rods	
	4.3	Heat Exchangers	7
	4.4	Shielding	8
	4.5	Nuclear Waste Handling	9

### 1 Induced and Spontaneous Fission

Nuclear fission is the splitting of a heavier parent nucleus (with a nucleon number greater than about 230) splits into two or more nuclei. It can occur in two ways:

- 1. **Spontaneous** fission: An extremely rare type of fission that occurs **without any external energy or particles supplied into the system**. This is often naturally occurring in nuclei such as those of thorium-232, uranium-235, and uranium-238. The half life for spontaneous fission is typically around 11 billion years.
- 2. **Neutron-induced** fission: A neutron interacts with a heavy nuclide to produce an unstable isotope of the original, with a neutron number one greater.

### 2 The Fission Mechanism

Two nuclear fuels commonly used nowadays are uranium-235 and plutonium-239.

- Uranium occurs naturally.
- Plutonium are produced in the rocks when natural uranium-238 captures a neutron from the two possible origins below
  - 1. Cosmic rays
  - 2. A natural uranium-235 decay.

A typical example of an induced uranium-235 is as follows

$$\underbrace{{}^{235}_{92}\text{U} + {}^{1}_{0}\text{n}}_{56} \to {}^{144}_{56}\text{Ba} + {}^{90}_{36}\text{Kr} + 2{}^{1}_{0}\text{n} + \Delta E$$

interaction with neutron to form unstable isotope

- The Ba and Kr are the fission products, both have lighter nuclei.
- The example above shows the emission of two electrons, but the number of emitted neutrons can vary. It is typically 2-3 neutrons.
- The energy released  $\Delta E$  is given by the mass defect between the two sides of the equation

$$\Delta E = (m_{\text{U-235}} - (m_{\text{Ba-144}} + m_{\text{Kr-90}} + m_n)) c^2$$

where  $m_{\rm initial}$  is the mass of the initial nucleus, and  $m_{\rm final}$  is the mass of the final nucleus.

### 2.1 Fission Products — Percentage Yield

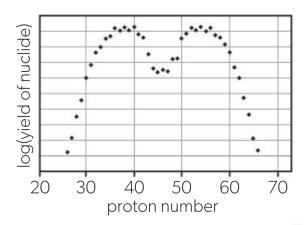


Figure 1: % Yield of fission products of uranium-235 by proton numbers

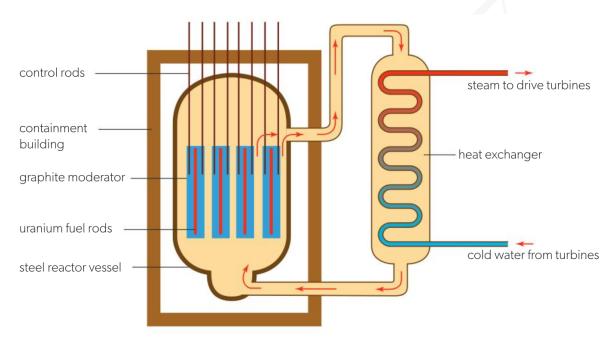
- There are two peaks in the distribution, indicating that fission products tend to cluster around proton numbers 36-38 and 54-58. This means that when U-235 undergoes fission, the resulting nuclei are most likely to have proton numbers in these two ranges.
- The nuclei formed in these regions (36-38, 54-58) are more stable than U-235 itself. This is because the binding energy per nucleon (the energy that holds the nucleus together) is greater for these fission products than for U-235.
- The daughter nuclei, immediately upon fission, are close together and repel each other. The electrostatic repulsion provides the kinetic energy for the nuclei and particles to move in different directions. A portion of the energy is released through photon emission.

## 3 Chain Reactions

When one of the emitted neutrons from a fission event interacts with another nucleus to form a new unstable isotope, triggering a subsequent fission event, and this process carries on with more and more nuclei, we have a chain reaction.

### 4 Fission Reactors

The following is a thermal fission reactor, specifically, a pressurized water reactor (PWR) that uses uranium-235 as fuel. The aim of this reactor is to convert nuclear energy into electrical energy, by heating water to produce steam, which drives a turbine and a generator.



▲ Figure 7 A schematic of a pressurized water reactor (PWR).

Figure 2: A schematic of a pressurized water reactor

- 1. The reactor's fuel is *enriched* if its proportion of U-235 is boosted to a level greater than the natural 0.7%.
- 2. The enriched fuel is placed in fuel rods, which are then placed in the reactor core.
- 3. The neutrons are required to be slowed down to increase the probability of interacting with a uranium nucleus to induce fission. This is done by a moderator, which is typically water or graphite. The typical accepted speed is around  $2.2\,\mathrm{km\,s^{-1}}$  with kinetic energy of  $0.025\,\mathrm{eV}$ . Neutrons at this speed are called *thermal neutrons*, as they would be in equilibrium with matter at about room temperatures.

#### 4.1 Moderators

Its aim is to remove the excess of KE from neutrons to facilitate the fission process. Typical moderators may include water, heavy water, and graphite.

When fast-moving neutrons strike the atom within the moderator elastically, they transfer some of their KE to the atom, slowing down themselves. After a succession of such collisions, the neutrons become thermal neutrons.

When a neutron of mass m collides head-on with moderator nucleus of mass M, the loss in KE is

$$\Delta E_k = \frac{4mM}{(m+M)^2} E_k$$

Let  $r = \frac{\Delta E_k}{E_k}$  denote the fraction of KE lost. For one collision with each of the common moderator nuclei

• Hydrogen:  $r \approx 1$  — roughly all KE of the neutron is lost.

• Deuterium:  $r \approx 0.9$ ; Carbon:  $r \approx 0.3$ 

In reality, the number of collisions required to slow down a neutral by a certain factor is well below the theoretical prediction because not all predictions are head-on.

A potential problem with moderators is that U-238 can absorb even fast-moving neutrons. To tackle this problem, the moderators are placed closer to the fuel rods. The neutrons move from the fuel rods to the moderator at random, and the probability of a neutron being absorbed by U-238 is reduced.

Properties of a good moderator include

- Poor absorber of neutrons
- Being inert (chemically inactive) in the extreme conditions of a reactor

Hydrogen absorbs a massive portion of KE, but it is a good absorber of neutrons; thus, it cannot be used directly as a moderator.

#### 4.2 Control Rods

Handles the turning-on and shutting-down of the reactor. They are made of boron and good absorbers of thermal neutrons. When the rods are inserted into the reactor core, they absorb the neutrons, lowering the rate of fission. When the rods are removed, the rate of fission increases back up.

#### 4.3 Heat Exchangers

The heat energy cannot be directly converted into electrical energy; e.g. the steam cannot be piped directly into the turbine through the reactor vessel because the steam would be radioactive. There must be a closed-system mechanism in place to handle the transformation of energy.

- 1. A primary loop of water passes through the reactor core, absorbing the heat energy.

  This water is kept at high pressure to prevent boiling, although its temperature already exceeds the boiling point. It is then passed through a heat exchanger the steam generator.
- 2. When the primary loop passes through the tubes in the steam generator, the heat is transferred to a secondary loop of water initially kept at a lower temperature. Throughout this process, the primary loop remains **separate from the secondary loop**, preventing the radioactive water from contaminating the secondary loop.
- 3. This secondary loop of water is then heated to produce steam, which drives the turbine and generator.

#### 4.4 Shielding

Safety measures need to be implemented to prevent the penetrative radiation from escaping the reactor. This is typically done in the following ways

- The reactor vessel is made of **thick steel** to absorb alpha and beta radiation, together with some of the gamma photons as well as stray neutrons.
- The vessel is also surrounded by a thick **concrete shield** to further absorb escaping gamma photons and neutrons.
- The building is also constructed such that it can withstand natural disasters such as earthquakes and tsunamis. In the even of a disaster and destruction, the reactor would be trapped in this construction.
- Emergency safety mechanisms immediate operations that are triggered in the event of an issue to shut the reactor down.
- The insertion and withdrawal of control rods are operated by machines instead of humans to reduce radioactive exposure.

## 4.5 Nuclear Waste Handling

This is an important aspect that must be considered when operating a nuclear reactor.

Problem	Descripion	Mitigation Strategy
Useless daughter	As more and more fission	Store the waste in a secure location,
nuclei from fission	events take place, there	such as a deep underground facility.
	will be more and more	
	daughter nuclei (e.g. kryp-	
	ton, barium, molybdenum,	
	etc.); these are neutron-	
	rich, will likely undergo $\beta^-$	
	decay, and have long half-	
	lives.	
Distortion of fuel	The creation of two new	Regularly removing the fission
rod's physical	nuclei changes the struc-	products from the rod and recy-
structure	ture of the fuel rod — the	cling the remaining U-235 into a
	smaller nuclei take up a	new rod.
	different amount of space.	
	In a worst-case scenario,	
	this distortion could cause	
	the fuel rod to jam in its	
	channel inside the reactor.	
Spent fuel rods		Reprocessing — storing the rods
		in cooling underwater storage for
		five to ten years before undergoing
		treatment to recycle the uranium
		into new rods.

Long half-life	Some of the waste prod-	Store the waste in a secure location,
waste	ucts have half-lives of	such as a deep underground vault or
	thousands of years, which	well-sealed drums.
	means that they will re-	
	main radioactive for a long	
	time. But their activity is	
	relatively low, compared	
	to short-half-life waste.	
Low-level waste	This includes the gloves	Buried under secure conditions.
	and over-shoes used by the	
	workers as well as medi-	
	cal equipment, which are	
	contaminated by the radi-	
	ation.	
End of life of the	The reactor has a finite	The reactor is dismantled, and the
reactor	lifespan, and at the end of	radioactive parts are stored in a
	its life, it must be decom-	larger shell of concrete, which will
	missioned.	be buried underground.