

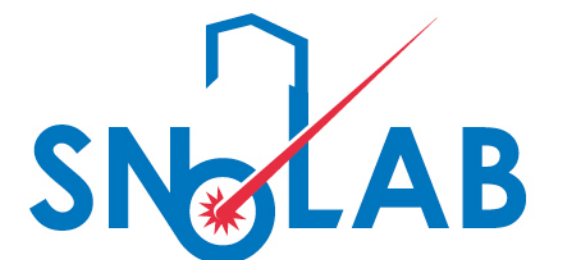
2023/05/11

# Gamma Ray Counting for Low Background Experiments


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SNOLAB Staff Scientist



# InterSpec Gamma Ray Software



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 wcjohns  
 v1.0.11\_rc6  
 77dc7b3

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## v1.0.11 rc6

Pre-release

Likely the final release candidate for v1.0.11.

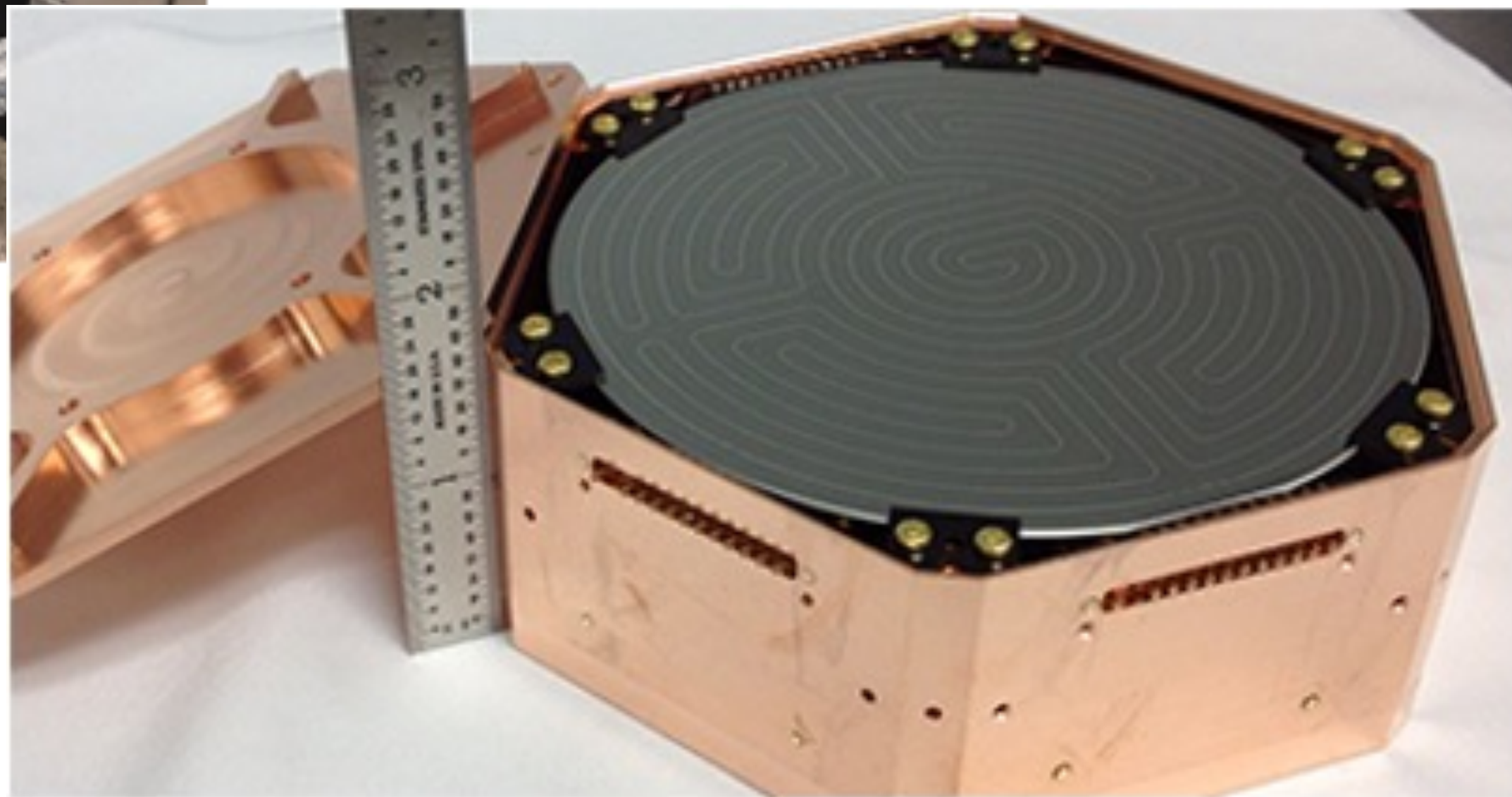
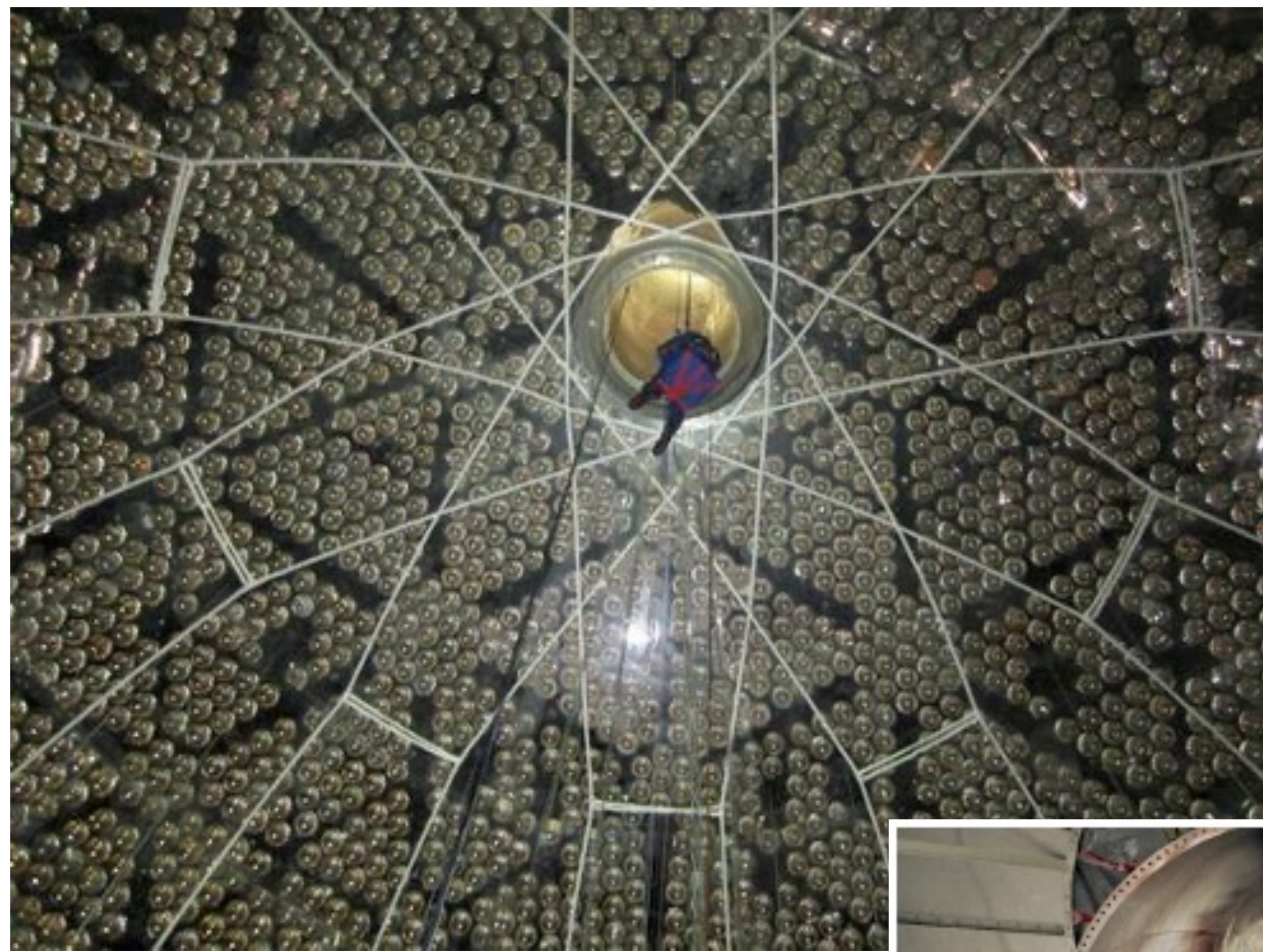
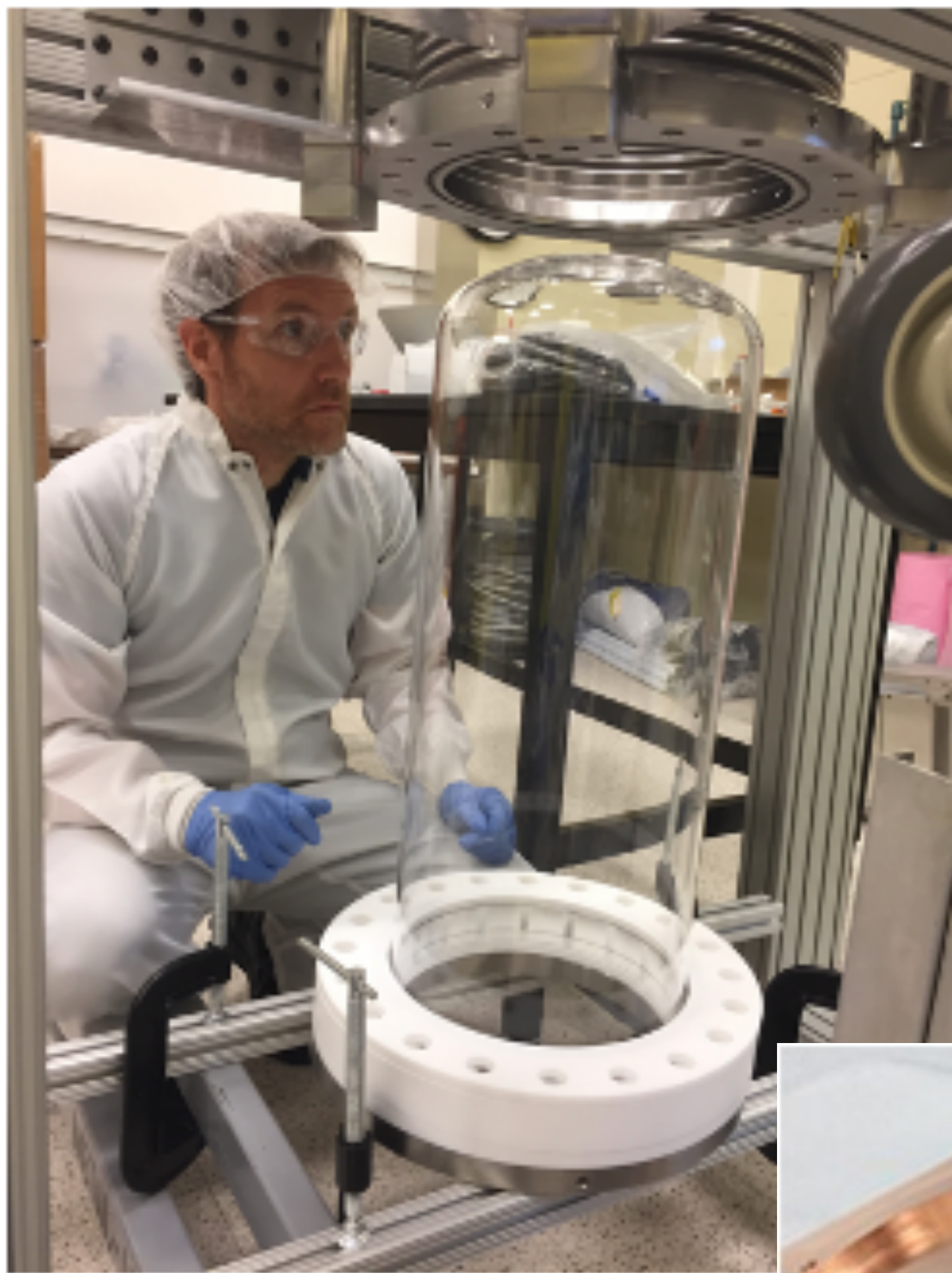
- For Windows 10 and 11 users, the "InterSpec-win32-x64\_WebView2\_v1.0.11\_rc6b" build is recommended; may also work on Windows 7 (64-bit) if the Edge runtime is installed.
- The macOS build requires 10.13 (High Sierra) or newer.
- The Linux build work on Ubuntu 14 and newer.

### ▼ Assets 6

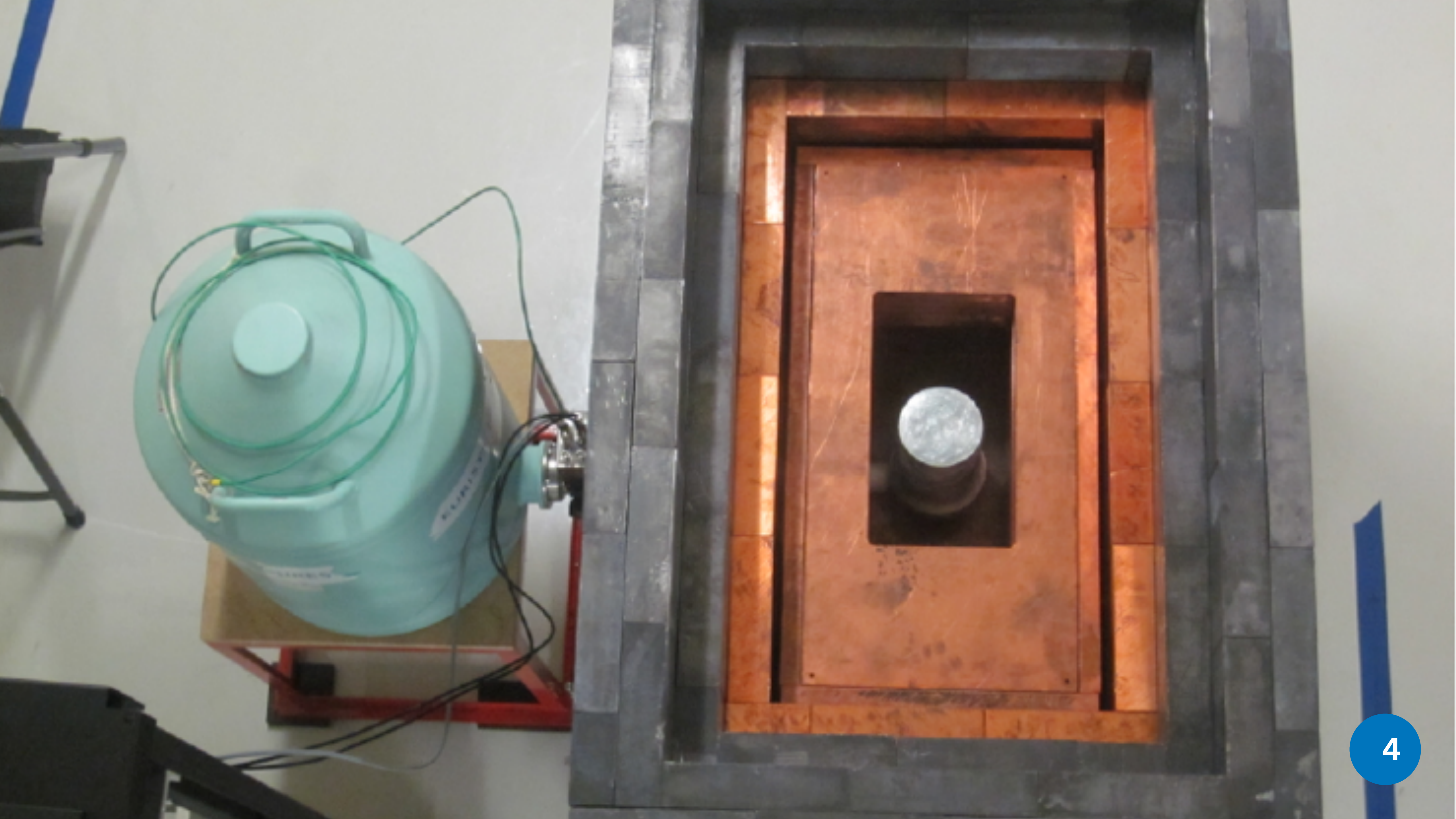
 <a href="#">InterSpec-linux-x64_v1.0.11_rc6.zip</a>	121 MB	2 weeks ago
 <a href="#">InterSpec-macOS_v1.0.11_rc6.dmg</a>	34.8 MB	2 weeks ago
 <a href="#">InterSpec-win32-x64_Electron_v1.0.11_rc6.zip</a>	120 MB	2 weeks ago
 <a href="#">InterSpec-win32-x64_WebView2_v1.0.11_rc6b.zip</a>	29.5 MB	last week

- <https://github.com/sandialabs/interspec/releases>
- Please download and install the appropriate version for your OS.











# Example Radioactivity Levels

- $^{14}\text{C}$ : 1 ppt of natural Carbon
- $^{40}\text{K}$ : 120 ppm of natural Potassium
- $^{238}\text{U}$ : to 10 ppm in rocks and soil
- $^{232}\text{Th}$ : to 10 ppm in rocks and soil
- Thoriated welding rods: 1-4%  $^{232}\text{Th}$
- Screened low-background materials: <1 ppb U, Th
- Rn gas:
  - 1 to 100 Bq/m<sup>3</sup> on surface
  - ~3 Bq/m<sup>3</sup> in Sudbury
  - ~130 Bq/m<sup>3</sup> at SNOLAB

# Sources of Radioactive Backgrounds

- Nearly stable isotopes
  - $^{238}\text{U}$  –  $4.5 \times 10^9$  years
  - $^{232}\text{Th}$  –  $1.4 \times 10^{10}$  years
  - $^{235}\text{U}$  –  $7 \times 10^8$  years
  - $^{40}\text{K}$  –  $1.3 \times 10^9$  years
- Activated isotopes
  - $^{14}\text{C}$  – 5,700 years
  - $^{39}\text{Ar}$  – 270 years
  - $^7\text{Be}$  – 53 days
- Daughter isotopes
  - $^{210}\text{Pb}$  – 22 years
  - $^{208}\text{Tl}$  – 3 minutes
  - $^{228}\text{Ac}$  – 6.2 hours
- Gaseous daughter isotopes
  - $^{220}\text{Rn}$  – 56 seconds
  - $^{222}\text{Rn}$  – 3.8 days
  - $^4\text{He}$  – stable
- Fission products
  - $^{137}\text{Cs}$  – 30 years
  - $^{131}\text{I}$  – 8 days
- Man-made isotopes
  - $^{60}\text{Co}$  – 5.3 years
  - $^3\text{H}$  – 12 years
  - $^{18}\text{F}$  – 110 minutes

# 232Th Chain

Thorium Gamma Intensities				$A = 4n$			13.52 1.600 16.2 0.72 12.75 0.304 15.5 0.16	Ra 228 5.75 a	63.823 0.264 204.68 0.021	Th 232 $1.405 \times 10^{10}$ a
								911.204 25.8 968.971 15.8 338.320 11.27 964.766 4.99 463.004 4.40 794.947 4.25 209.253 3.89	Ac 228 6.15 h	
	238.632 43.3 300.087 3.28 115.183 0.592	Pb 212 10.64(1) h	804.9 0.0019	Po 216 145(2) ms	549.76 0.114	Rn 220 55.6(1) s	240.986 4.10	Ra 224 3.66(4) d	84.373 1.220 215.983 0.254 131.613 0.131 166.410 0.104	Th 228 1.9116(16) a
2614.533 99.0 583.191 84.5 510.77 22.6 860.564 12.42 277.351 6.31 763.13 1.81	Tl 208 3.053(4) m	39.858 1.091	Bi 212 60.55(6) m	727.330 6.58 1620.50 1.49 785.37 1.102						
		Pb 208 stable		Po 212 299(2) ns						



# 235U Chain

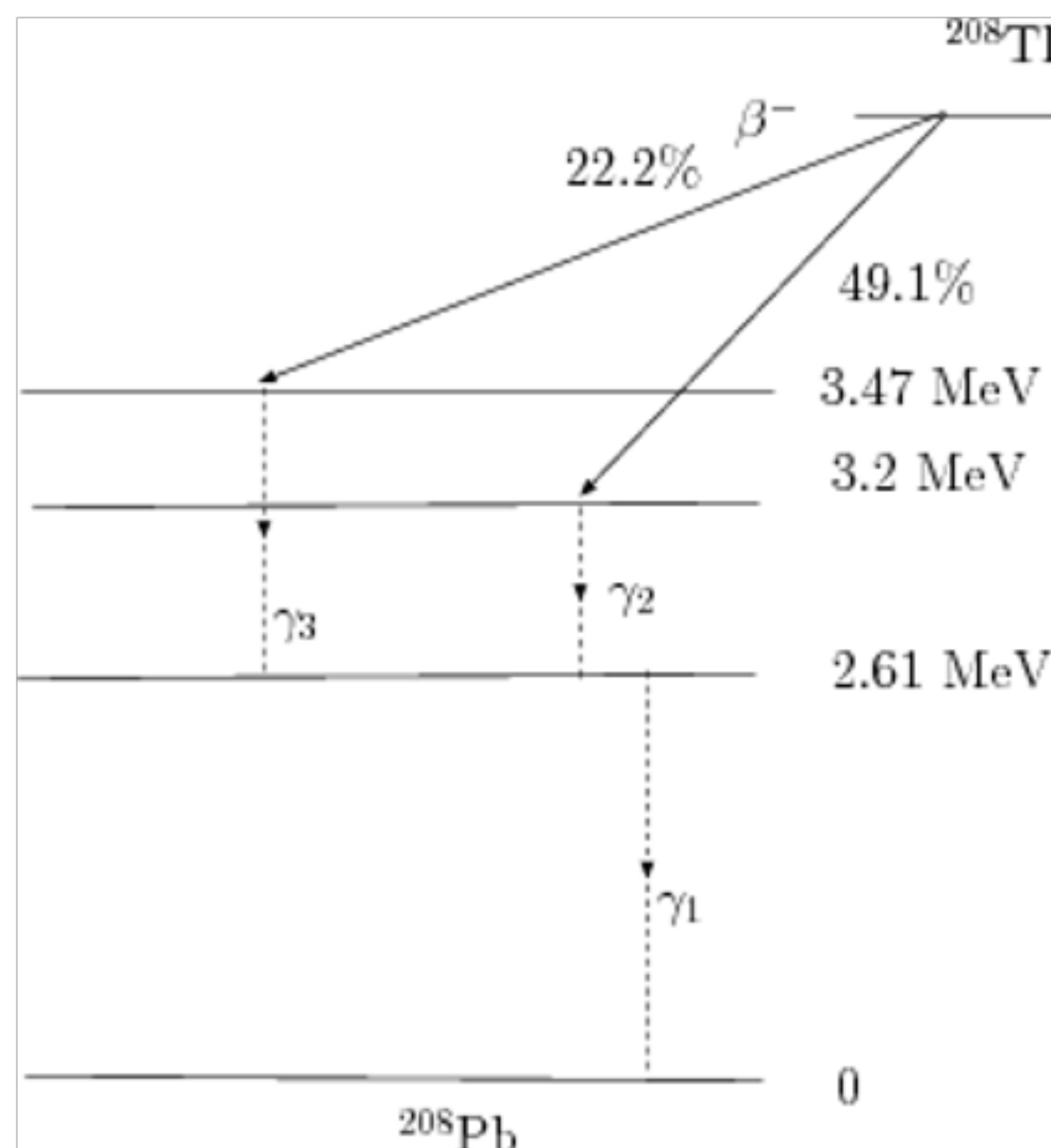
Actinium Gamma Intensities				A = 4n + 3								25.64 14.5 84.214 6.6	Th 231 1.0633 d	185.715 57.2 143.76 10.96 163.33 5.08 205.311 5.01 109.16 1.54 202.11 1.08	U 235 7.028x10 <sup>8</sup> a
		293.56 100 271.23 8.2 517.60 4.3 776.90 3.4 1398.8 3.4 564.09 2.8 608.30 2.8 835.32 2.6	Bi 215 7.6 m	α none β none	At 219 56 s -97% -3%	α none β 50.13 36.0 β 79.72 9.1 β 234.81 3.0 β 49.89 2.7	Fr 223 21.8 m 0.006% 99.994%	α 160.26 0.0059 β none	Ac 227 21.773(3) a 1.380% 98.620%	27.36 10.29 300.07 2.47 302.65 2.19 283.69 1.70 330.06 1.40 19.00 0.374	Pa 231 3.276x10 <sup>4</sup> a				
	404.853 3.78 832.01 3.52 427.088 1.76	Pb 211 36.1(2) m		Po 215 1.781(4) ms		Rn 219 3.96(1) s	269.459 13.70 154.21 5.62 323.871 3.93 144.232 3.22 338.281 2.79 445.031 1.27	Ra 223 11.435(4) d	235.971 12.3 50.13 8.26 256.25 7.01 329.85 2.69 300.00 2.32 286.12 1.53	Th 227 18.72(2) d					
897.80 0.260 569.702 0.00159 328.12 0.00140	Tl 207 4.77 m	α 351.059 12.91 β none	Bi 211 2.14(2) m 99.724% 0.276%												
		Pb 207 stable		Po 211 516 ms											



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# Gamma Ray Generation

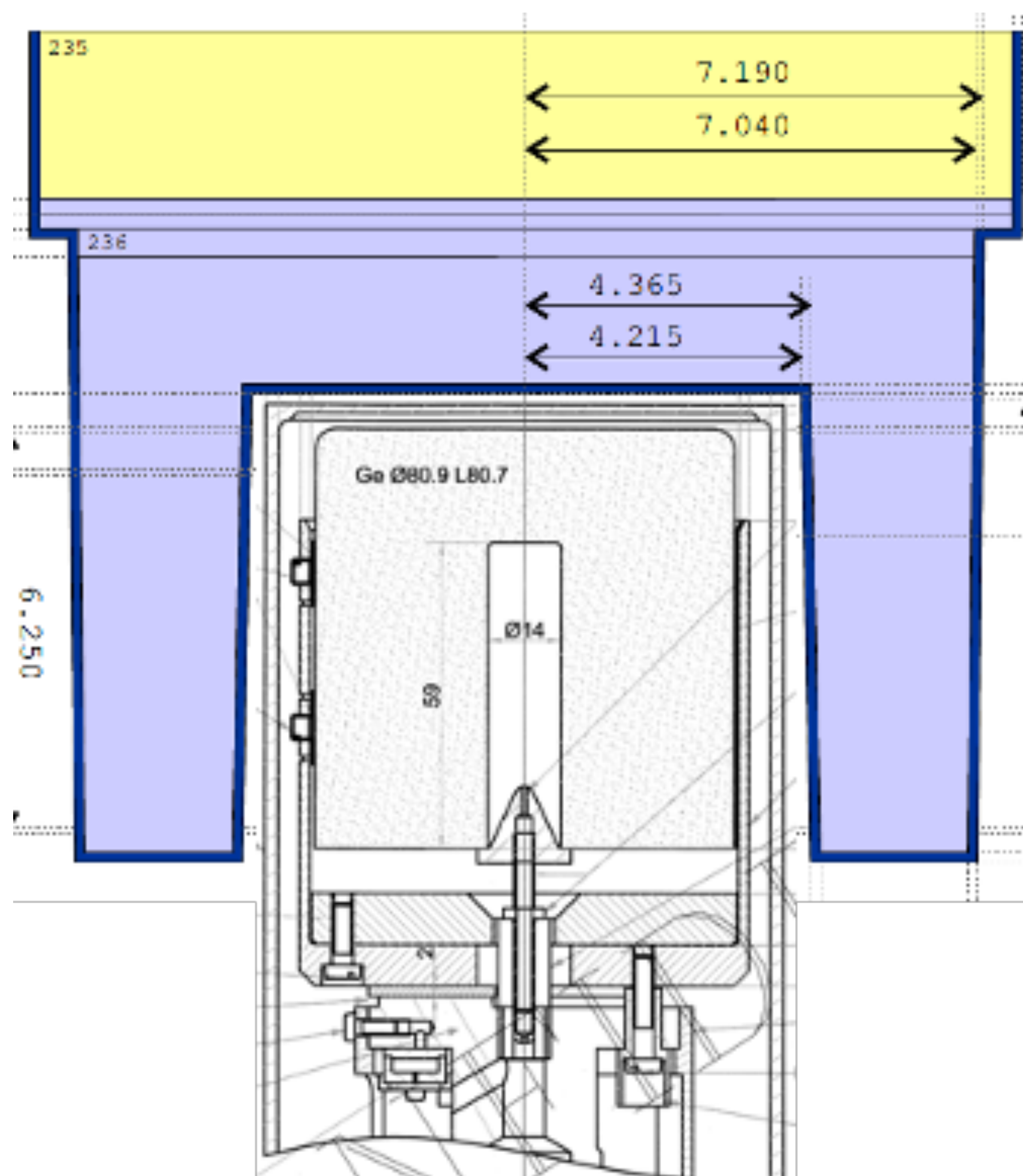


**Gammas from  $^{208}\text{Tl}$  (3.053 m 4)**

Eg (keV)	Ig (%)	Decay mode
211.40 15	0.178 20	$\beta^-$
233.36 15	0.307 20	$\beta^-$
252.61 10	0.69 4	$\beta^-$
277.351 10	6.31 9	$\beta^-$
277.72		$\beta^-$
485.95 15	0.050 5	$\beta^-$
510.77 10	22.6 3	$\beta^-$
583.191 2	84.5 7	$\beta^-$
1381.1 5	0.007 3	$\beta^-$
1647.5 7	0.0020 10	$\beta^-$
1744.0 7	0.0020 10	$\beta^-$
2614.533 13 99		$\beta^-$



# Gamma Ray Detection

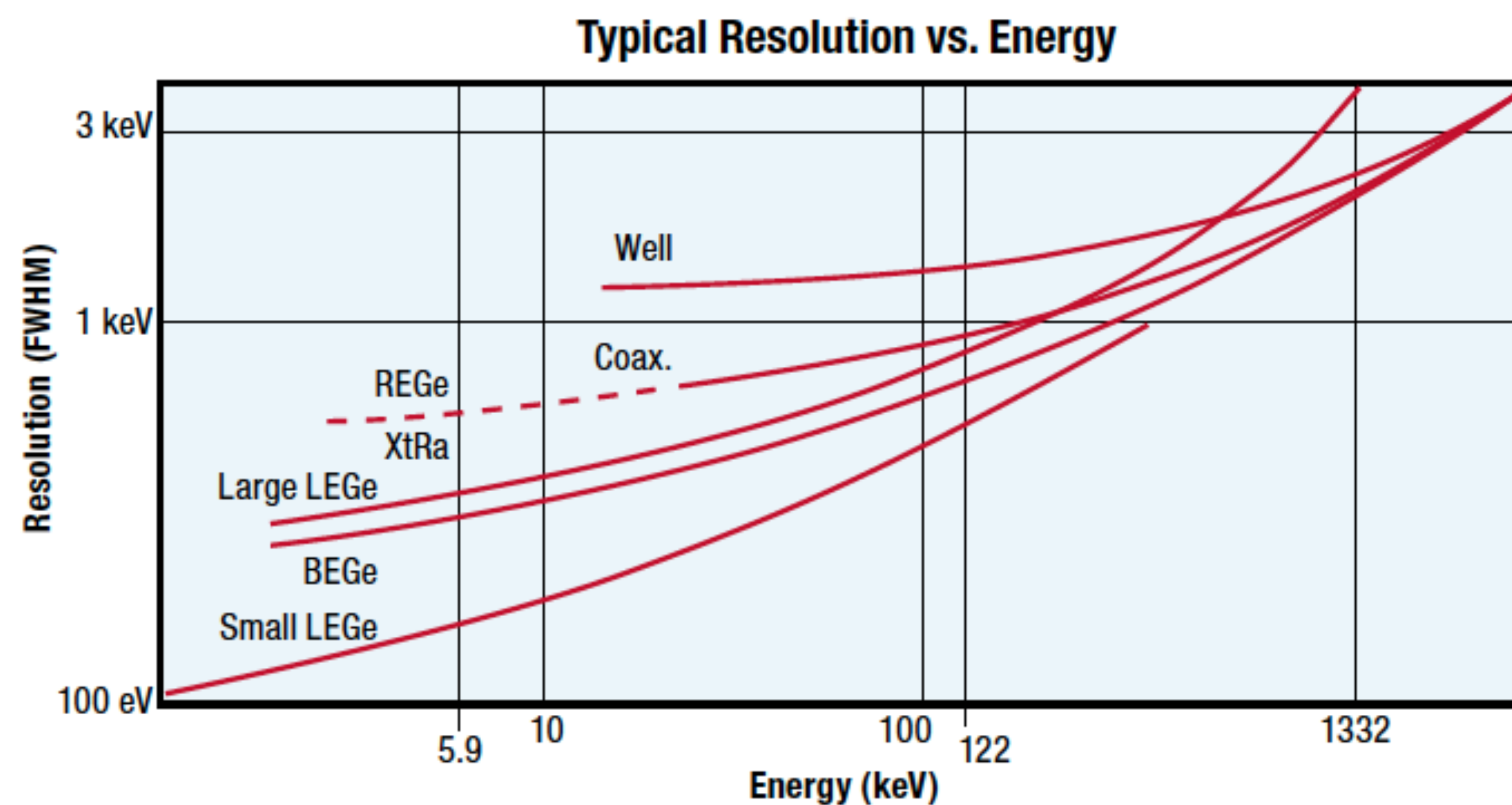
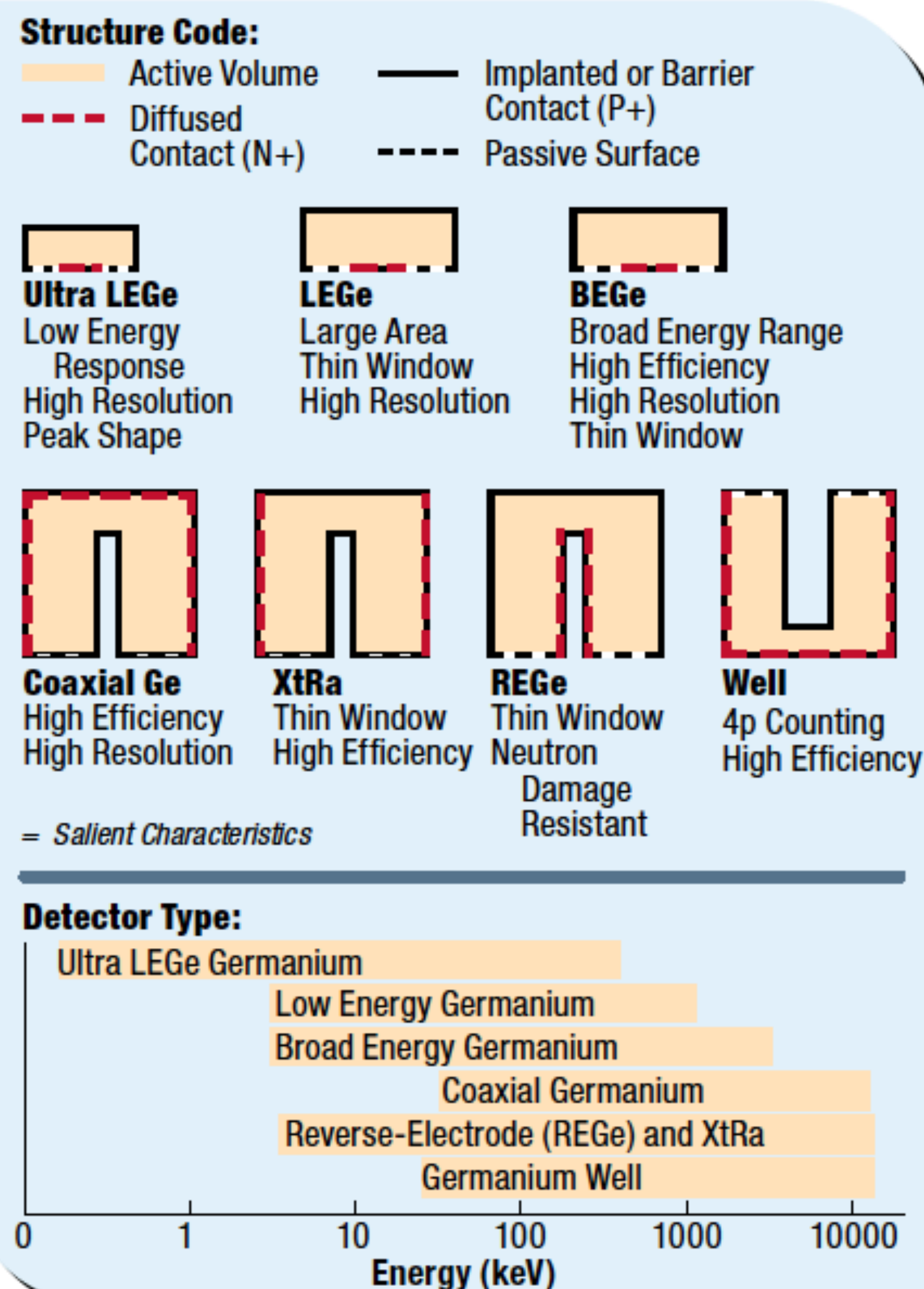


- High-purity germanium crystal detectors: Semiconductor Diodes
- Incoming gammas create holes/e<sup>-</sup> which are drifted to n/p electrodes for read-out
- Shielded with low-activity copper and lead



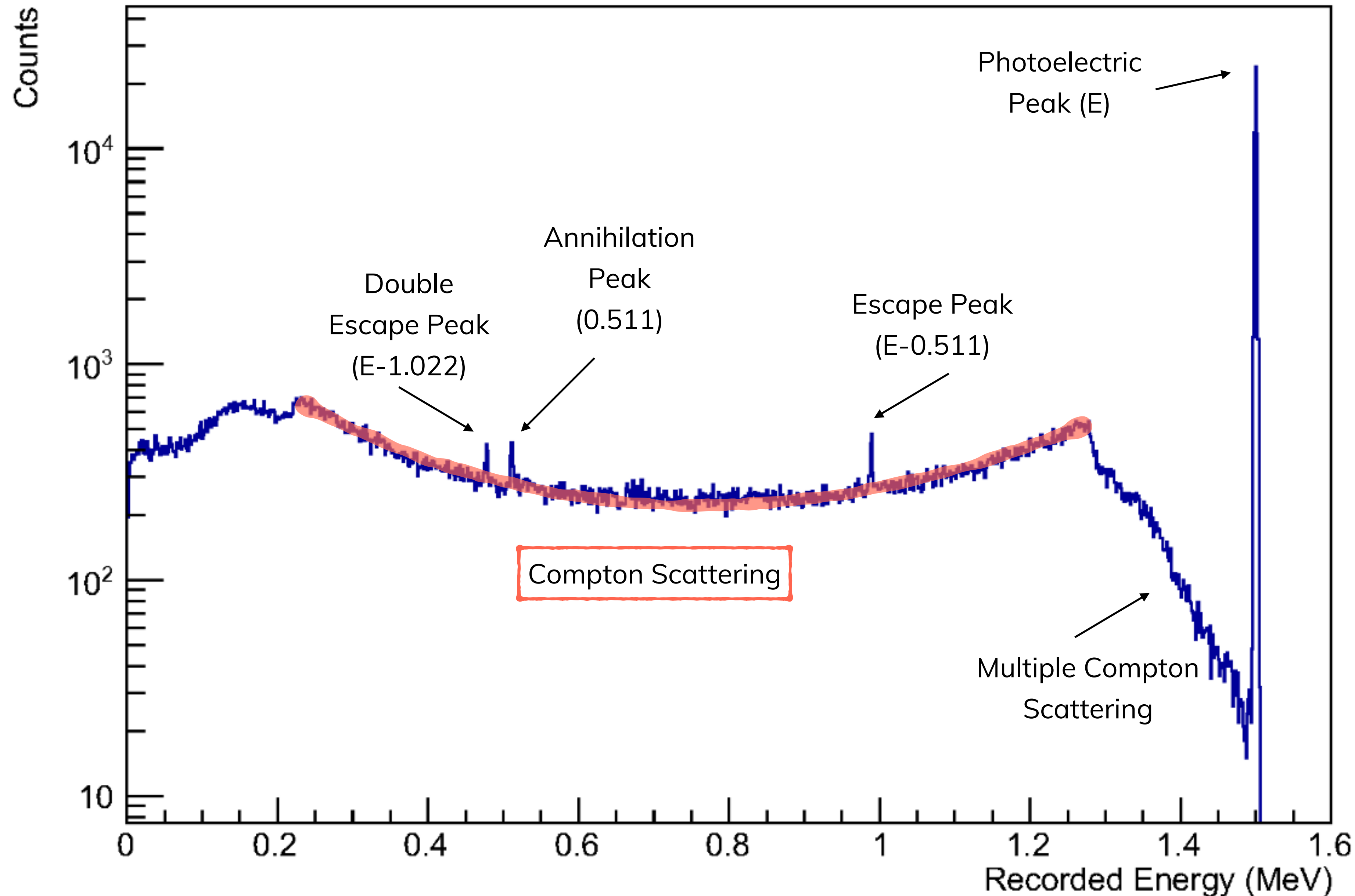
# Germanium Detector Function

- Detectors must be cooled to reduce the thermal generation of charge carriers
- Liquid nitrogen, which has a temperature of 77 K is the common cooling medium
- Detector is mounted in a vacuum chamber, which is attached to or inserted into an LN<sub>2</sub> Dewar.





# Simulation of 1.5 MeV Gamma Rays

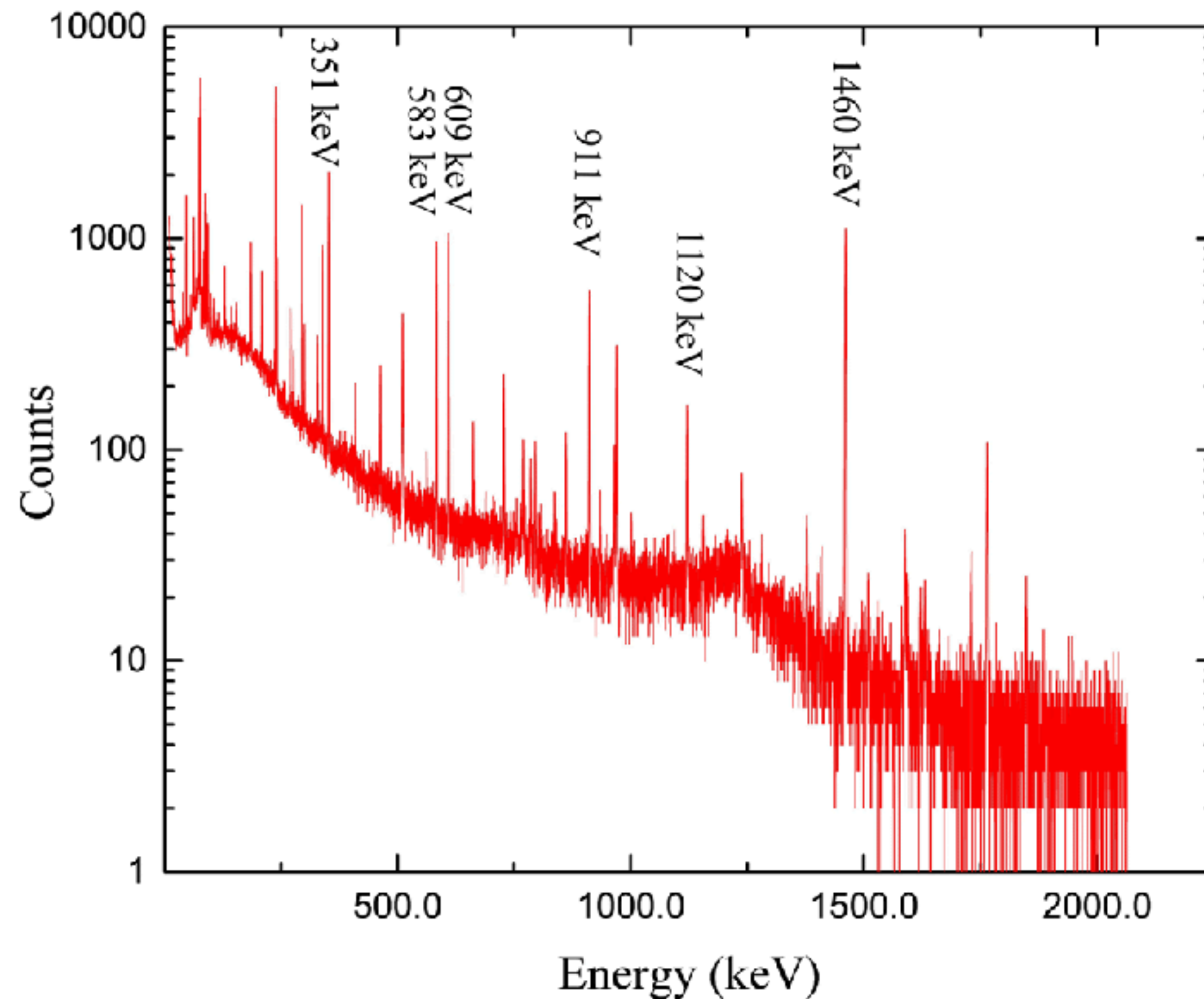


- Line widths are usually set by the resolution of the detector.
- Germanium detectors have resolutions of 1 to 3 keV.



# Analysis Method

- Focus only on photoelectric peak
- Calculate number of observed events in peak
- Convert to activity for that isotope

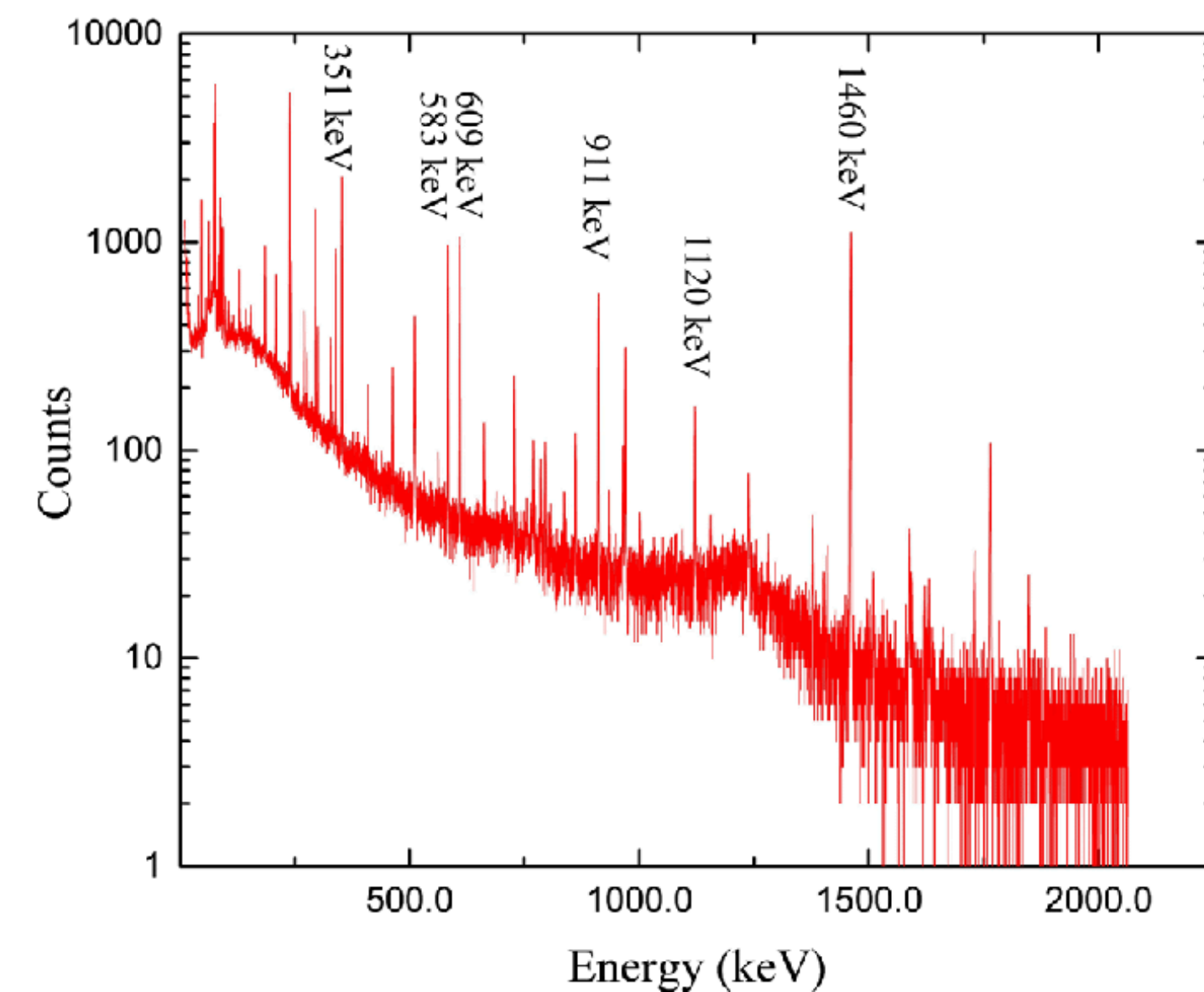




# Analysis Method

$$N = \Omega \cdot \epsilon \cdot \frac{I_g}{100} \cdot M \cdot t \cdot A$$

- N = Number of Counts
- $\Omega$  = Solid Angle factor (from Simulations)
- $\epsilon$  = Photoelectric peak efficiency (from Simulations)
- $I_g$  = Branch Probability (from Lund database)
- M = Sample mass
- t = Data acquisition time
- A = Activity (units such as Bq/kg)



# Analysis Goals

$$N = \Omega \cdot \epsilon \cdot \frac{I_g}{100} \cdot M \cdot t \cdot A$$

- What radioactive elements are present in your samples?
- What is the activity (with uncertainty) of each of these elements?
- Are these activities consistent with other measurements?
- Are there any decay chains present?
- Are the elements in the chains in equilibrium?
- Or have certain elements been concentrated or diluted?
- What are the dominant uncertainties in your analysis?



<https://www.snolab.ca/~tjsonley/CAPSS2023/>

## CAPSS 2023 Gamma Ray Counting Exercise

### Interspec Software

Each group need a copy of Interspec on one of their computers.

Windows [link](#)

Mac [link](#)

Linux [link](#)

The download website is [here](#)

### Data Files

Download all of the data files in a [.zip file](#)

The unzipped data is [here](#) if you have trouble unzipping the main file.

### Python

We will be using python to perform part of the analysis.

There is a simple python interface using "Jupyter Notebooks" in your web browser available at <https://jupyter.org/try-jupyter/lab/>

Follow that link, then use the file browser to upload all of the files in the "PythonFitting" directory.

Then open CAPSS2023Notebook.ipynb.

Now, highlight each of the cells in order and click the "play" button above to execute the code.

The first box should take 30 seconds or so to run, but the others should be fast.

You will see the marker in front of the box go from [] before hitting play,

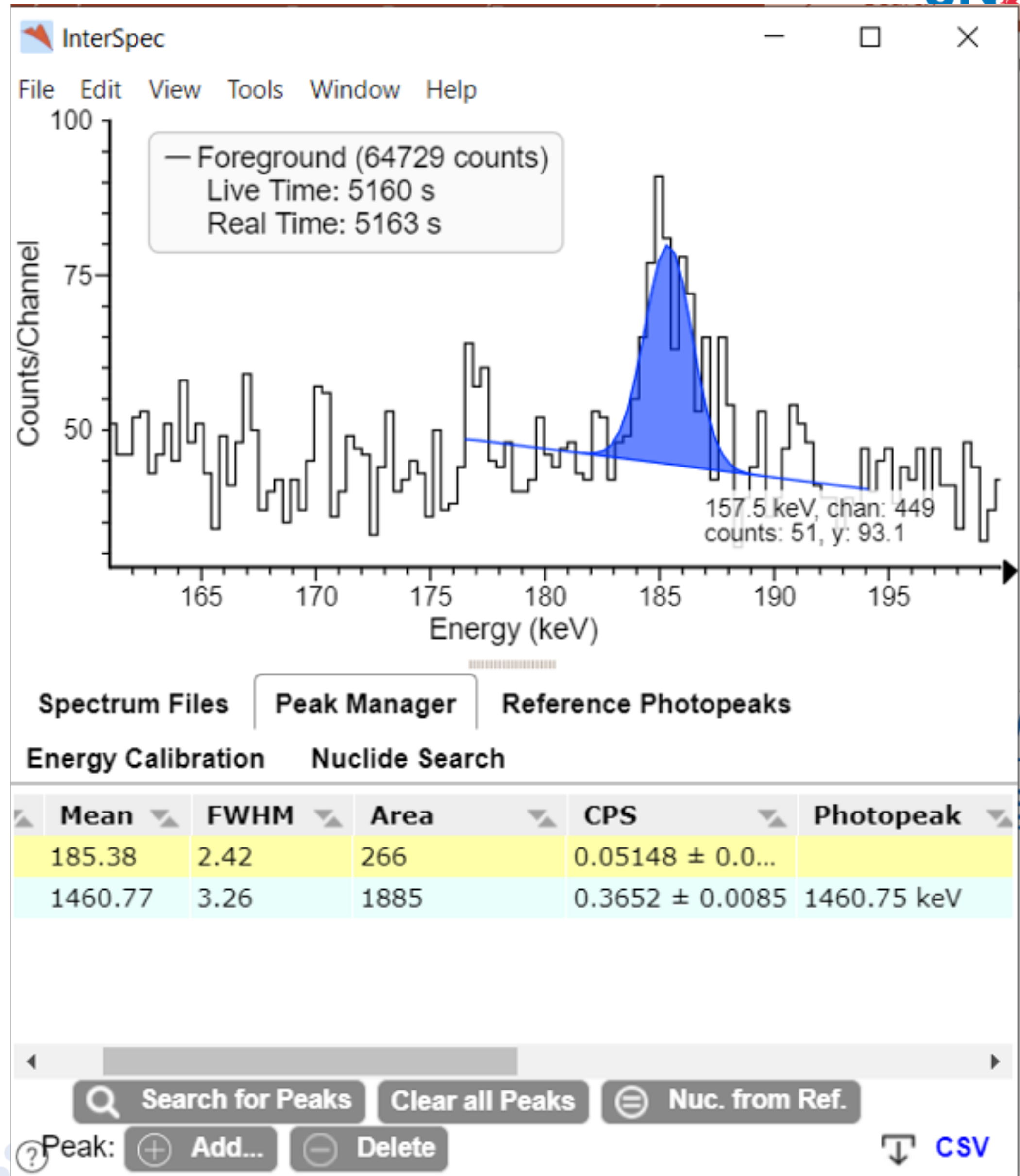
to [\*] while the code is running,

to [1] or [2] when the code has finished.

You can change any of the lines and rerun that cell to see how it changes the output.

# InterSpec

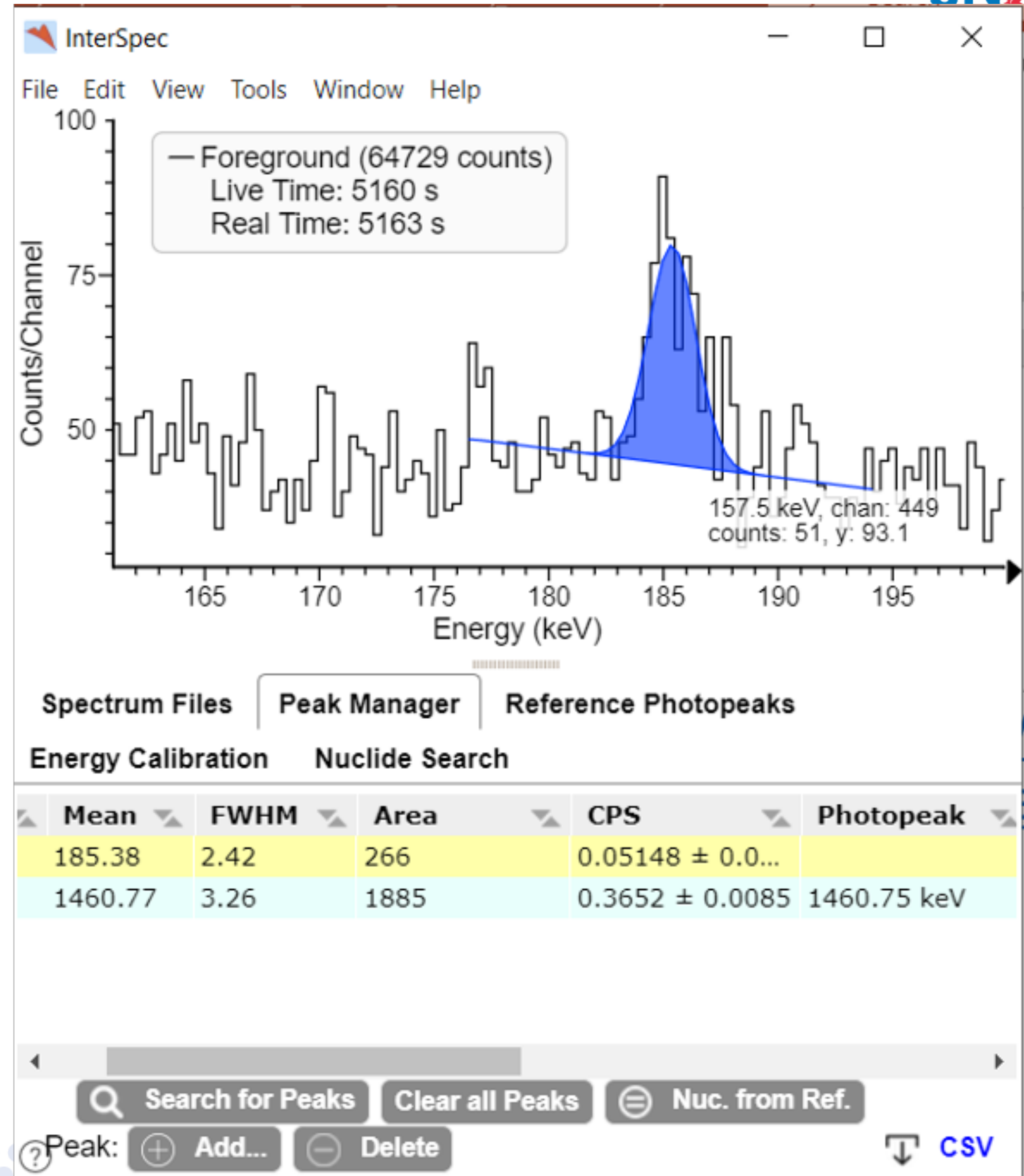
- <https://github.com/sandialabs/interspec/releases>
- Can open many data formats
- Automatic Peak Fitting
- Automatic Background Subtraction
- Library of Gamma Ray Nuclides





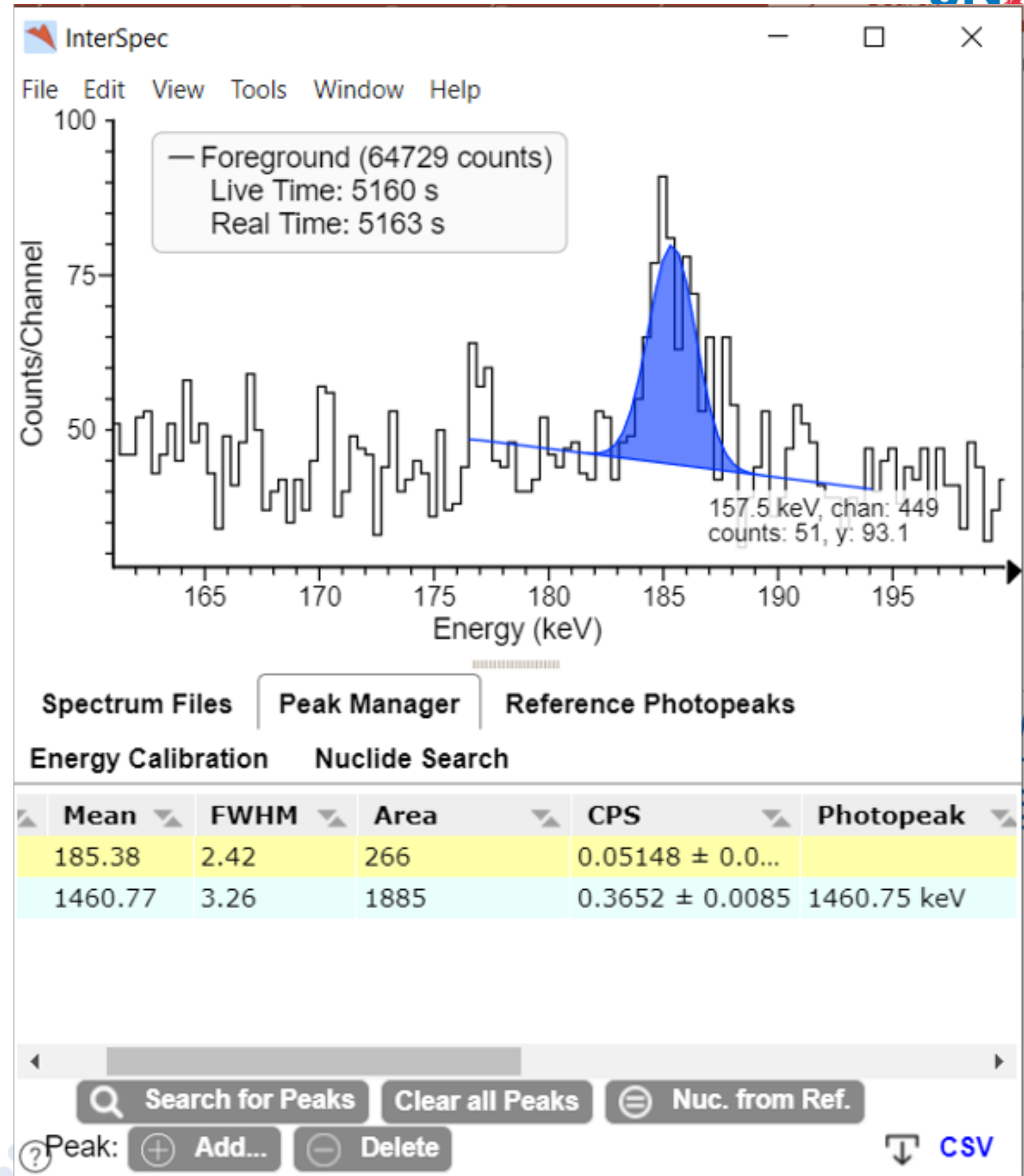
# InterSpec

- Open the Rain\_Water.chn file
- View -> Chart Options -> Linear Y-Scale
- Zoom in by dragging right
- Zoom out by dragging left
- Double-click on a peak to fit it
- Details show up in the bottom peak manager
- Right-click on a peak to refit
- Library of Gamma Ray Nuclides



# InterSpec

- Use Nuclide Search tab to identify peaks
- Click on peak to automatically fill energy
- Search for decays with high “Rel B.R.” = Relative Branching Ratio
- InterSpec will sometimes list multiple “Parents” for the same decay. Just choose one.

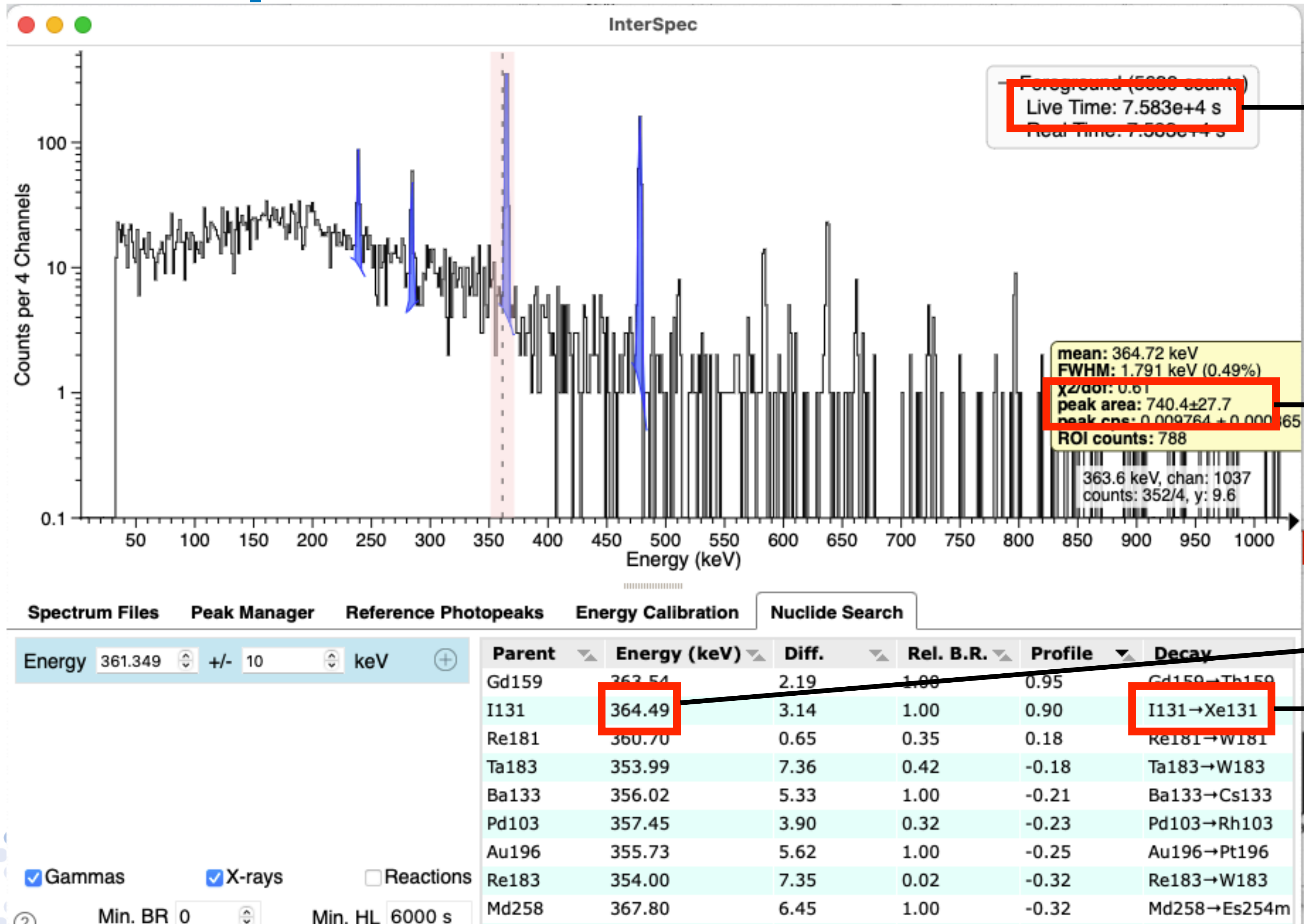




# InterSpec Parameters (1)

$$N = \Omega \cdot \epsilon \cdot \frac{I_g}{100} \cdot M \cdot t \cdot A$$

SNOLAB



t=Time of sample taking

N = Peak Area

Energy &  
Decay

# InterSpec Parameters (2)



Spectrum FilesPeak ManagerReference PhotopeaksEnergy CalibrationNuclide Search

Energy 361.349 +/- 10 keV

☒ Gammas☒ X-rays☐ Reactions

Min. BR 0Min. HL 6000 s

Parent	Energy (keV)	Diff.	Rel. B.R.	Profile	Decay
Gd159	363.54	2.19	1.00	0.95	Gd159→Th159
I131	364.49	3.14	1.00	0.90	I131→Xe131
Re181	360.70	0.65	0.35	0.18	Re181→W181
Ta183	353.99	7.36	0.42	-0.18	Ta183→W183
Ba133	356.02	5.33	1.00	-0.21	Ba133→Cs133
Pd103	357.45	3.90	0.32	-0.23	Pd103→Rh103
Au196	355.73	5.62	1.00	-0.25	Au196→Pt196
Re183	354.00	7.35	0.02	-0.32	Re183→W183
Md258	367.80	6.45	1.00	-0.32	Md258→Es254m

From File Info:  
**M** = 856.9 g

Energy & Decay give  
Branch Probability

$$N = \Omega \cdot \epsilon \cdot \frac{I_g}{100} \cdot M \cdot t \cdot A$$

Gammas from <sup>131</sup> I (8.02070 d 11)		
Eg (keV)	Ig (%)	Decay mode
80.185 2	2.62 3	b <sup>-</sup>
85.9 2	0.00009 5	b <sup>-</sup>
163.930 8		b <sup>-</sup>
177.214 2	0.270 3	b <sup>-</sup>
232.18 15	0.0032 4	b <sup>-</sup>
272.498 17	0.0578 11	b <sup>-</sup>
284.305 5	6.14 5	b <sup>-</sup>
295.8 2	0.0018 8	b <sup>-</sup>
302.4 2	0.0047 6	b <sup>-</sup>
318.088 16	0.0776 16	b <sup>-</sup>
324.651 25	0.0212 25	b <sup>-</sup>
325.789 4	0.274 21	b <sup>-</sup>
358.4 2	0.016 6	b <sup>-</sup>
364.489 5	81.7 6	b <sup>-</sup>
404.814 4	0.0547 10	b <sup>-</sup>
503.004 4	0.360 3	b <sup>-</sup>
636.989 4	7.17 9	b <sup>-</sup>
642.719 5	0.217 4	b <sup>-</sup>
722.911 5	1.773 25	b <sup>-</sup>



# Simulation Parameters

$$N = \Omega \cdot \epsilon \cdot \frac{I_g}{100} \cdot M \cdot t \cdot A$$

- The product ( $\Omega \cdot \epsilon$ ) is calculated by simulation
- <https://www.snolab.ca/~tjsonley/CAPSS2023/>

## Python

We will be using python to perform part of the analysis.

There is a simple python interface using "Jupyter Notebooks" in your web browser available at <https://jupyter.org/try-jupyter/lab/>

Follow that link, then use the file browser to upload all of the files in the "PythonFitting" directory.

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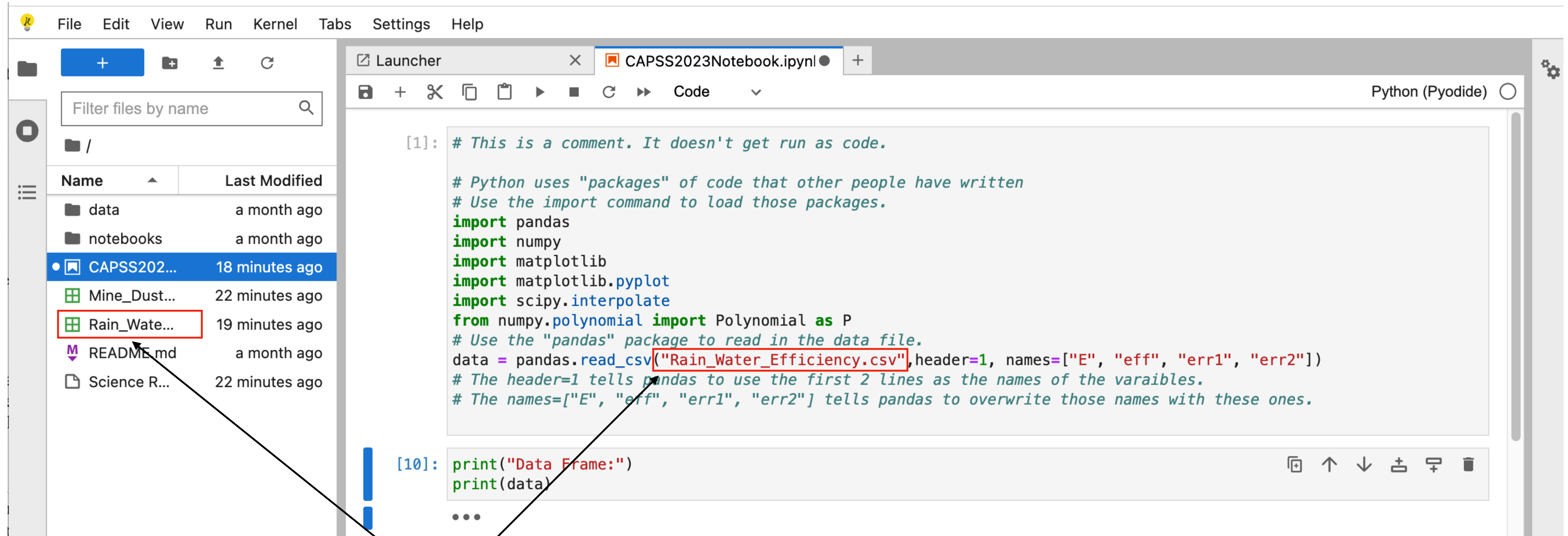
to [\*] while the code is running,

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You can change any of the lines and rerun that cell to see how it changes the output.

$$N = \Omega \cdot \epsilon \cdot \frac{I_g}{100} \cdot M \cdot t \cdot A$$

# Simulation Parameters



```
[1]: # This is a comment. It doesn't get run as code.

# Python uses "packages" of code that other people have written
# Use the import command to load those packages.
import pandas
import numpy
import matplotlib
import matplotlib.pyplot
import scipy.interpolate
from numpy.polynomial import Polynomial as P
# Use the "pandas" package to read in the data file.
data = pandas.read_csv("Rain_Water_Efficiency.csv", header=1, names=["E", "eff", "err1", "err2"])
# The header=1 tells pandas to use the first 2 lines as the names of the variables.
# The names=["E", "eff", "err1", "err2"] tells pandas to overwrite those names with these ones.

[10]: print("Data Frame:")
print(data)
```

- Change the input Efficiency file to the one for the data you are analyzing
- Go through the notebook and find the efficiency at the energy of the given peak
- Repeat for every peak!



# Analysis Goals

$$N = \Omega \cdot \epsilon \cdot \frac{I_g}{100} \cdot M \cdot t \cdot A$$

- What radioactive elements are present in your samples?
- What is the activity (with uncertainty) of each of these elements?
- Are these activities consistent with other measurements?
- Are there any decay chains present?
- Are the elements in the chains in equilibrium?
- Or have certain elements been concentrated or diluted?
- What are the dominant uncertainties in your analysis?

# Error Propagation

$$F = f(a, b)$$

$$\sigma_F^2 = \left( \frac{\partial F}{\partial a} \cdot \sigma_a \right)^2 + \left( \frac{\partial F}{\partial b} \cdot \sigma_b \right)^2$$

$$F = \frac{ab}{c}$$

$$\sigma_f^2 = F^2 \left( \frac{\sigma_a^2}{a^2} + \frac{\sigma_b^2}{b^2} + \frac{\sigma_c^2}{c^2} \right)$$



Measurement precision is given by  $\frac{\sigma_f}{F}$ :



# Error Comparisons

$$\frac{\sigma_f^2}{F^2} = \left( \frac{\partial F}{\partial a} \cdot \frac{\sigma_a}{F} \right)^2 + \left( \frac{\partial F}{\partial b} \cdot \frac{\sigma_b}{F} \right)^2$$

Error Source	Uncertainty (%)
Uncertainty on Counts	10.2
Detection Efficiency	5
Branching Ratio	1
Sample Mass	0.1
Livetime	0.001
Total	11.4

- Usually summarized in an Error Table
- Since they are added in quadrature, only the largest ones matter.
- e.g. “Uncertainty on Counts” contributes 80% of the total

Measurement precision is given by  $\frac{\sigma_f}{F}$ :

# Error Rules of Thumb

$$\frac{\sigma_f^2}{F^2} = \left( \frac{\partial F}{\partial a} \cdot \frac{\sigma_a}{F} \right)^2 + \left( \frac{\partial F}{\partial b} \cdot \frac{\sigma_b}{F} \right)^2$$

- Dominant Uncertainties
  - > 1/3 of max uncertainty
  - These must be addressed to improve the measurement
  - Papers and reports must describe how these were estimated
- Secondary Uncertainties
  - 1/3 to 1/10 of max uncertainty
  - These could become dominant in an improved measurement
  - Reports should describe the general method used to estimate them (repeated measurements, simulations, etc.)
- Negligible Uncertainties
  - < 1/10 of max uncertainty
  - They must go in the error table
  - Reports should spend less than 1 sentence describing them.