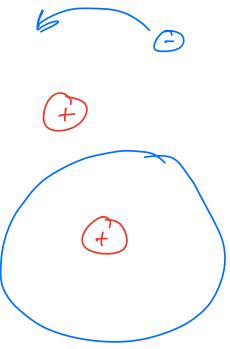
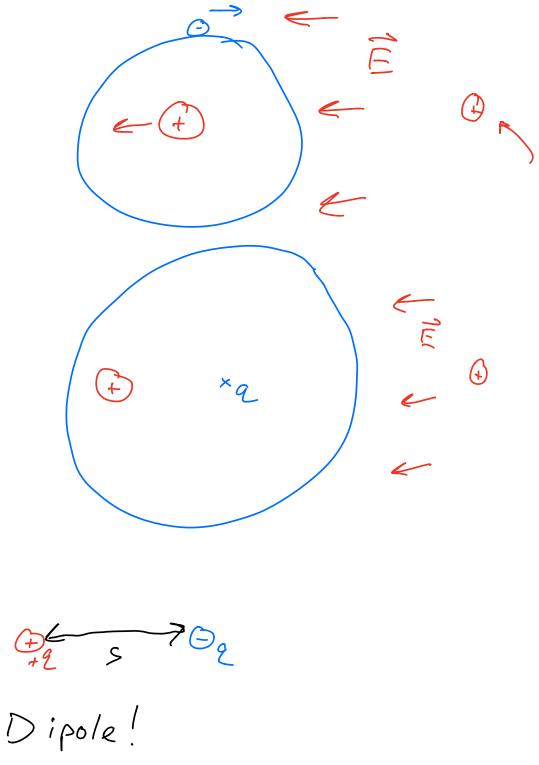
## · Hook:

- Have you ever noticed how, in the winter time, after you walk across the room you end up getting shocked by the doorknob?
- o Ever seen someone rub a balloon on their clothes and then it sticks to them?
- These are both electrostatic phenomena. Our job today is to understand what is happening.
- · Brief review of Big Concepts so far
  - In chapter 13 we introduced the idea of electric charge, electric force and electric fields
  - o Sometimes objects carry an electric charge, similar to a mass. Charges interact with each other.
  - We used the concept of the force between interacting charges to generalize the concept of electric field.
    - Review diagram. Field of q1 causes force F2 on charge q2.
    - Field is always there, whether or not charge feels it.
  - We saw the expression for electric field of a point charge, and discovered that the net electric field of many charges can be found simply by adding the individual field of each charge as a vector sum
  - o At the end of chapter 13, we saw an important application: dipoles

- · Question: what if we had a dipole in the presence of a uniform Electric field?
  - Net charge = 0
  - Net force = 0
  - Net torque > 0
- · Now we want to talk some more about electric charge and how it behaves within matter
  - TAPE DEMO
  - I've got two pieces of tape here. I'm going to stick them facing down on the lectern like this and peel them off quickly.
  - Now when I bring them close to each other, what happens? (they repel)
  - Ask class: what just happened?
  - Clearly, the pieces of tape are charged.
    - · Were they always charged? Repeat if necessary with two ordinary pieces. No repulsion.
  - The tape became charged when we ripped it off of the desk

- This is what we call charging by contact
  - So did the desk just produce electrons?
    - No. Either the tape *stole* some electrons from the desk, or left some its own behind when it was ripped off
    - This is always the case. Net charge does cannot be created or destroyed. Conservation of net charge.
  - I input energy to create relative motion between the two objects, this energy is enough to overcome weak electron-nucleus or ionic bonds
- · How do I know it is electrons that are being transferred and not protons?
  - o Protons are tightly bound deep inside of the atom
  - Much more energy required to kick a proton out of the atom than to kick out an electron (~10 eV /~10^6 eV)
  - o Protons do not move
- I have this rod, which is uncharged, and this piece of cloth, also charge neutral
  - o I've just touched this ball so it, too, is neutral
  - o I'm now going to charge this rod but rubbing it with the cloth
  - Now watch what happens when I bring it near to the ball? (Ball is attracted)
  - We understand how the rod became charged, but why did the ball move?
- · Let's consider the atomic structure of the ball for a moment
  - Ball is made up of many atoms. Each atom is charge neutral (number of protons = number of electrons)





Neutral atom in the presence of an external field resembles an electric dipole
 Still charge neutral, but non-zero electric field

• Typically, the induced dipole moment is proportional to the applied electric field: p = aE

$$\vec{P} = 9\vec{S} = \vec{\lambda}\vec{E}$$

Ex:

Point cha near neutral

Force on Point charge?

De,

$$\frac{\vec{F}}{F} = Q, \vec{E}_{2}$$

$$\frac{\vec{F}}{E_{2}} = \text{dipok on - axis}$$

$$|\vec{F}_{2}| = \frac{1}{4\pi\epsilon_{0}} \frac{2P}{r^{3}}$$

$$P = d\vec{E},$$

$$|\vec{F}_{2}| = \frac{1}{4\pi\epsilon_{0}} \frac{2d\vec{E}_{1}}{r^{3}}$$

$$|\vec{F}_{1}| = \frac{1}{4\pi\epsilon_{0}} \frac{2d\vec{E}_{1}}{r^{3}}$$

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$$|\vec{F}_{2}| = \frac{1}{4\pi\epsilon_{0}} \frac{2d\vec{E}_{1}}{r^{3}}$$

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$$|\vec{F}_{1}| = \frac{1}{4\pi\epsilon_{0}} \frac{2d\vec{E}_{1}}{r^$$

- Repeat pith ball experiment, explaining step by stepCan a charged object ever repel a neutral object?
- Announcements
  - o Read 14.4-14.5 for Friday
  - o HW due Friday
  - Quiz on Friday
  - Lab tomorrow