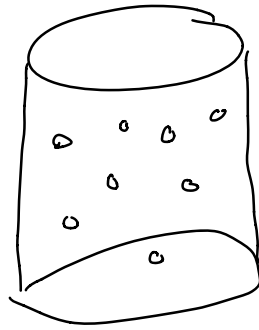


## Macro state VS Microstate

- We have a gas w  $10^{20}$  molecules



How do we describe this system?

- Pressure
- Temperature
- Volume
- Energy ( $U$ )
- etc ...

Another way:


$$m1: \quad \vec{r} = \langle 3, 7, -1 \rangle \\ \quad \quad \vec{p} = \langle 13, 12, 9 \rangle$$

$$m2: \quad \vec{r} = \langle -5, 6, 4 \rangle \quad \vec{p} = \langle -1, 0, 13 \rangle$$

$m^3$ : . . .

Macrostate: Specify  $P, V, T, U$

Microstate: List  $\vec{r} + \vec{p}$  of  
every molecule

  
If we know these,  
we could then calculate  
the macroscopic  
variables

- Can't actually specify a microstate  
entirely (too many molecules!)
- Nonetheless, can use it as a starting  
point for explaining macro-phenomena

If we are in equilibrium, macro variables don't change ( $P, v, T, U$ )

- But micro properties do!

This suggests that multiple microscopic configurations can correspond to one single macroscopic state

- Very basic example

System of 2 particles in 1 dimension



Total Energy  $U = 5$  ( $m=1$ )

$$U = \frac{1}{2}mv_1^2 + \frac{1}{2}mv_2^2$$

$v_1$	$v_2$	$U$
0	3.2	5
0	-3.2	5
3.2	0	5
-3.2	0	5
2	2.45	5
2.45	2	5
-2.45	-2	5

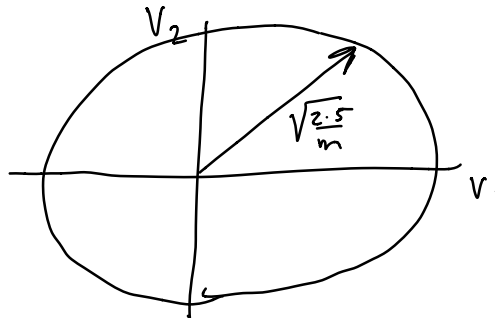
$$\frac{1}{2} m v_2^2 = 5$$

$$v_2 = \sqrt{\frac{10}{m}} = \sqrt{10}$$

...

$$\frac{1}{2} m v_1^2 + \frac{1}{2} m v_2^2 = 5$$

$$v_1^2 + v_2^2 = \frac{2(5)}{m}$$



The question we're getting at:

How many micro states correspond to a given macrostate?

Technically, infinitely many (in the above example). But you can see that different Macrostates have different numbers of microstates (what if I increased the energy of the system above?)

As we will soon see, the relation between micro and macro states is very important in statistical mechanics.

As it turns out, macro states which have more associated micro states are more probable (we will see what this means)

In the previous example, each particle had infinitely many momenta it could assume

Let's consider a system where each particle only has two possible states (even this has some physical applications) and go from there

Let's count micro & macro states of flipped  
coins

Example:

Flipping coins

Flip 3 coins

What are the possible outcomes?

c1	c2	c3
H	H	H
H	H	T
H	T	H
T	H	H
H	T	T
T	T	H
T	H	T
T	T	T

Each row is a microstate

End result (# heads, # tails)  
is a macrostate

Possible macro states:

$3H, 2H1T, 2T1H, 3T$

If you know the microstate,  
you know the macro state

$(HHT = 2H, 1T)$

Reverse is not true

$(1H, 2T)$  has many possible  
microstates

Notation call the # of microstates  
associated with a given macro state

$\Omega$

- What is  $\Omega(2 \text{ heads})$ ?

$$\Omega = 3$$
$$(HTT, THT, TTH)$$

$$\Omega(1 \text{ heads}) = 3$$

$$\Omega(2 \text{ heads}) = 3$$

$$\Omega(3 \text{ heads}) = 1$$

$$\Omega(0 \text{ heads}) = 1$$

$$3 + 3 + 1 + 1 = 8$$

We relate this to probability

$$\text{prob}(2 \text{ heads}) = \frac{\Omega(2)}{\Omega(0) + \Omega(1) + \Omega(2) + \Omega(3)}$$

$$\text{prob}(2) = \frac{3}{8}$$

$$\text{prob}(3) = \frac{1}{8} \quad \text{etc ...}$$



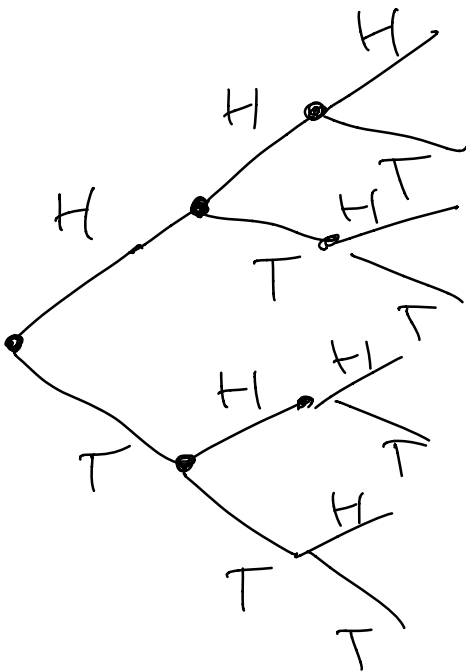
$$\text{prob}(\text{macrostate}) = \frac{\# \text{ microstates}}{\text{all microstates}}$$

Now let's flip 100 coins

There are 101 macrostates

$$n_{\text{heads}} = 0, 1, 2, \dots, 100$$

Can't <sub>1</sub> list <sub>2</sub> all <sub>3</sub> microstates



After  $n$  flips,  
there are  $2^n$   
possible outcomes

For 100 coins,  
 $2^{100} \approx 1.3 \times 10^{30}$   
outcomes

$$\Omega(\text{all}) = 2^{100}$$

# microstates per macrostate?

Consider

$$\Omega(n=0)$$

Every coin must be tails

only 1 state

$$\Omega(0) = 1 \quad (\Omega(100) = 1)$$

$$\Omega(1) ?$$

1. H T T T T ...

T H T T T ...

T T H T T ...  $\Omega = 100$

Each coin could  
be the single  
heads.

$$\Omega(1) = 100 \quad (\Omega(99) = 100)$$

$$\Omega(2) ?$$

Take each  $\Omega(1)$  state &  
Choose one more head

H T T T ... T T T

H H T T ... T T T

H T H T ... T T T

H T T T ... T T H

} 99

T H T T ... T T T

H H T T ... T T T

T H H T ... T T T

T H T H ... T T T

T H T T ... T T H

} 99

100 microstates for one head,  
each with 99 outcomes for  
the second head

$$\Omega(n=2) = 100 \cdot 99$$

But! Some of the states are  
duplicates!

H T T T ... T T T  
 Counted twice!  $\left. \begin{array}{l} H H T T \dots T T T \\ H T H T \dots T T T \\ H T T T \dots T T H \end{array} \right\} 99$   
T H T T ... T T T  
 $\left. \begin{array}{l} H H T T \dots T T T \\ T H H T \dots T T T \\ T H T H \dots T T T \\ T H T T \dots T T H \end{array} \right\} 99$

How many duplicates?

Well, how did this happen...

We considered choosing a first head, & then choosing a second

1)  $T T T H_1 T T T H_2 T \dots$

$T T T H_2 T T T H_1 T \dots$

these are the same!

Each microstate will have one  
duplicate

↓

$$\Omega(2) = \frac{100 \cdot 99}{2} = 4950$$

$$\Omega(3) = ?$$

100 ways to choose first coin

99 for second

98 for third

$$\Omega(3) = 100 \cdot 99 \cdot 98$$

But what about duplicates?

$TH_1 \quad TT H_2 \quad TH_3 \quad TT$

$H_2$

$H_1$

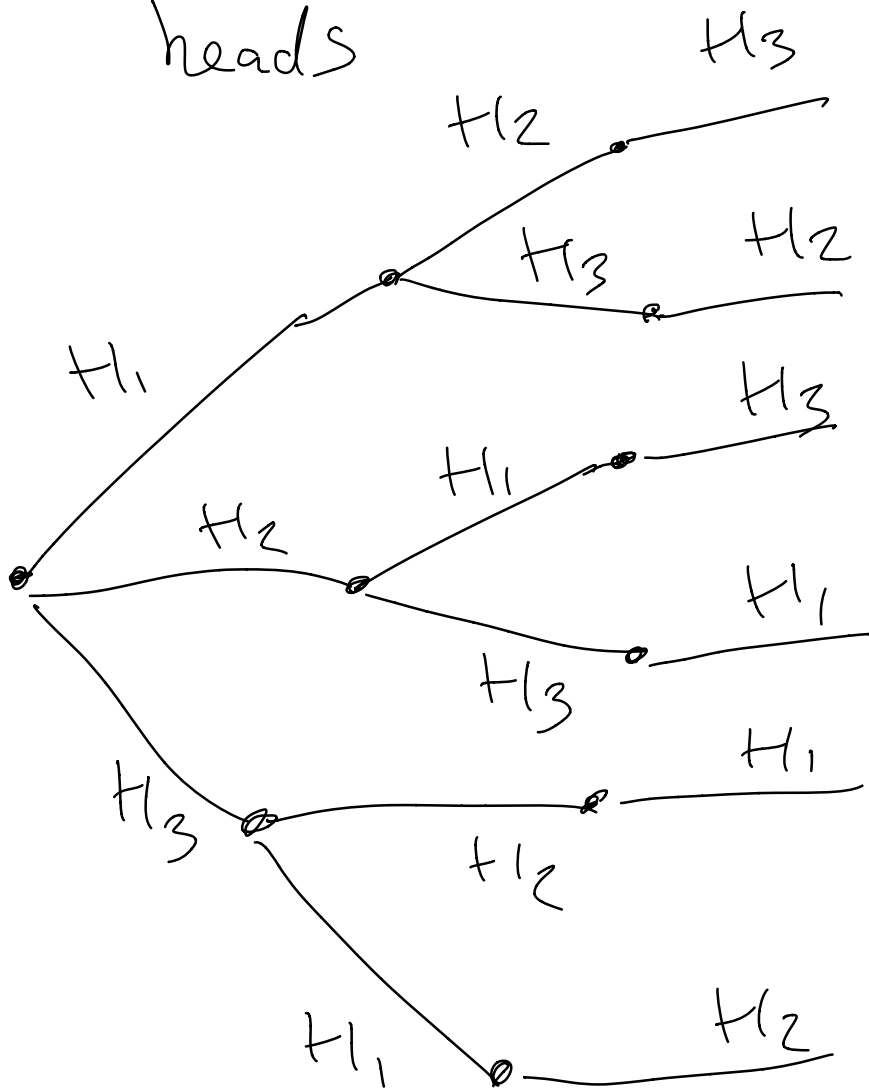
$H_3$

$H_3$

$H_2$

$H_1, \text{ etc...}$

# of duplicates = # of  
ways to arrange my 3  
heads



$$\# = 6 \quad (3 \cdot 2 \cdot 1)$$

$$\Omega(3) = \frac{100 \cdot 99 \cdot 98}{3 \cdot 2 \cdot 1} =$$

161700

Pattern?

$$\Omega(2) = \frac{100 \cdot 99}{2 \cdot 1}$$

$$\Omega(3) = \frac{100 \cdot 99 \cdot 98}{3 \cdot 2 \cdot 1}$$



$$\Omega(2) = \frac{100!}{(100-2)!} \cdot \frac{1}{2!}$$

$$= \frac{100 \cdot 99 \cdot \cancel{98} \cdot \cancel{97} \cdot \cancel{96} \dots}{\cancel{98} \cdot \cancel{97} \cdot \cancel{96} \dots} \cdot \frac{1}{2 \cdot 1}$$

$$\Omega(3) = \frac{100!}{(100-3)!} \cdot \frac{1}{3!}$$

$$= \frac{100 \cdot 99 \cdot 98}{3 \cdot 2 \cdot 1}$$

$$\Omega(n) = \frac{100!}{n! (100-n)!}$$

$N$  coins instead of 100

$$\Omega(N, n) = \frac{N!}{n! (N-n)!}$$

This is just the # of combinations

$N$  things to choose from, want to pick  $r$ .

$$\Omega(N, n) = \binom{N}{n}$$

Applicable to paramagnetic  
Systems.

Will do this later!

Einstein Solids