

# Predicting Alloyed Mixed-Valence Perovskite Structures

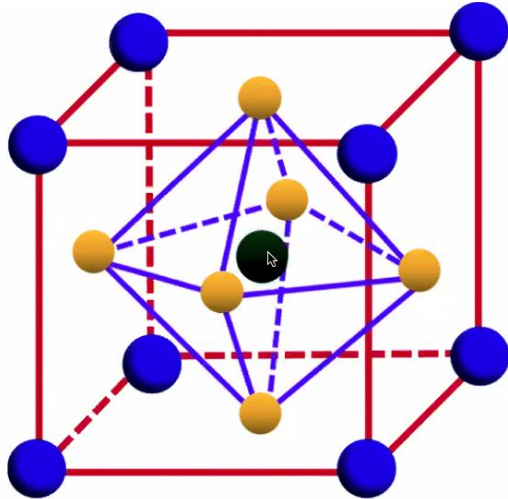
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<sup>1</sup>*Department of Chemistry, Stanford University*

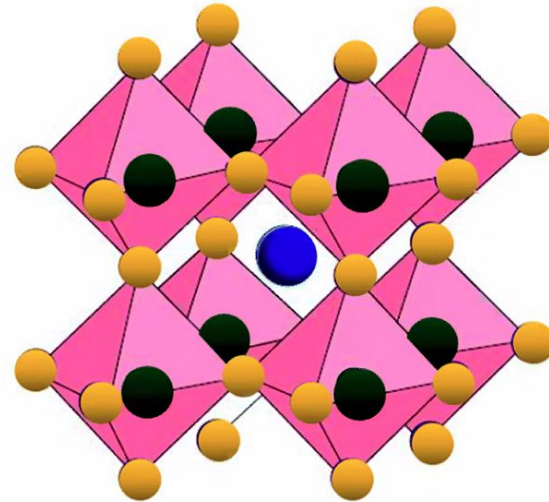
<sup>2</sup>*Department of Chemistry, University of Southern California*

# What Are Perovskites?

**Cubic Perovskites:** Have the general formula  $ABX_3$  where A and B are cations, and X is an anion



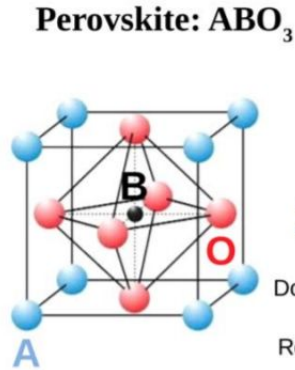
One unit cell (repeating unit)



Multiple unit cells together to make up one crystal structure

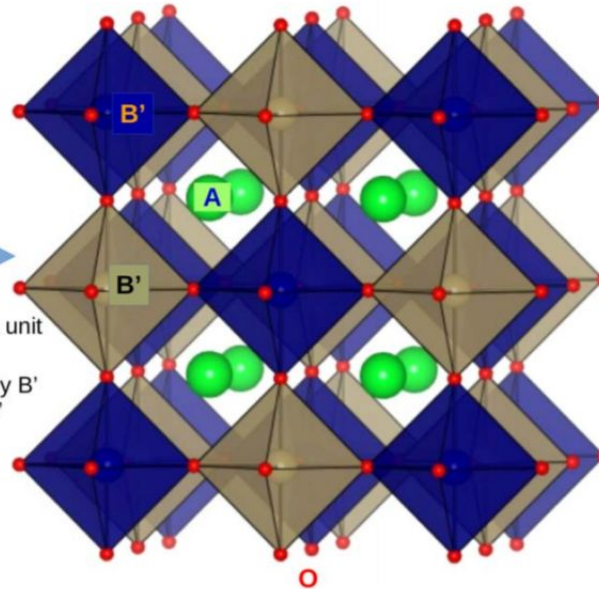
- **Double Perovskites:** Variation of the cubic perovskite - has the general formula of  $A_2BB'O_6$ .
  - Note: the B (central) sites are not the same in each repeating unit - they alternate between B and B'!

**(A)**



**(B)**

**Double Perovskite:  $A_2BB'O_6$**



*(Single Perovskite)*

B site: 4+

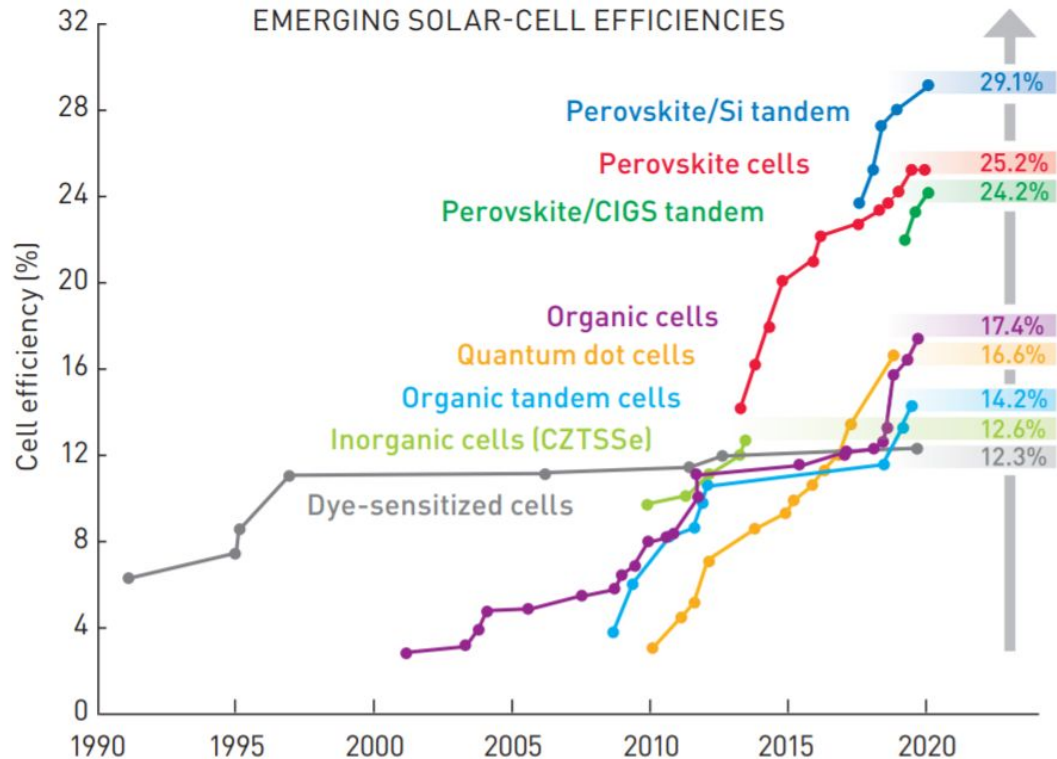
*(Double Perovskite)*

B Site: 3+

B' Site: 1+

Mixed Valence means the ions in the B sites can have different charges!

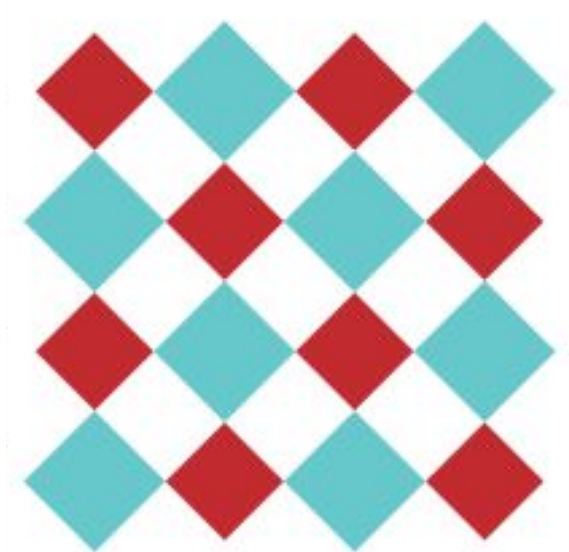
# Why Do We Study Perovskites?



- Used in solar-cells, fuel cells, and other sustainable energy applications
- More efficient
- Relatively cheap

# Mixing Double and Single Perovskites

Double Perovskite **before** mixing

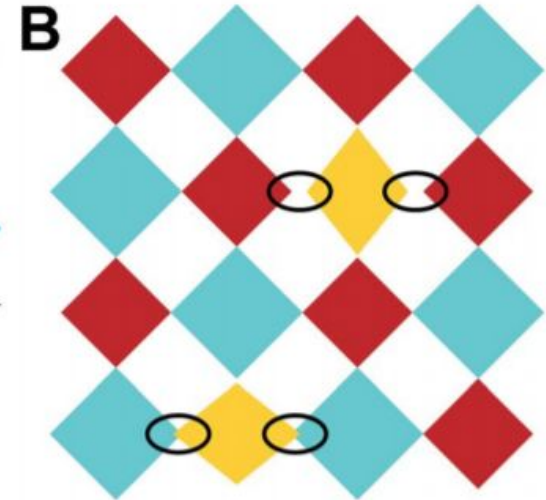


All ions aligned perfectly

- Key:
- Red =  $\text{In}^{3+}$
  - Blue =  $\text{Cu}^{+}$
  - Yellow =  $\text{Cu}^{2+}$



Double Perovskite **after** mixing



Less stable because of the mismatches  
(gaps and overlaps between ions)!

The perovskite materials with the top-down view of the B sites only

# Mixing Double and Single Perovskites

Single Perovskite **before** mixing

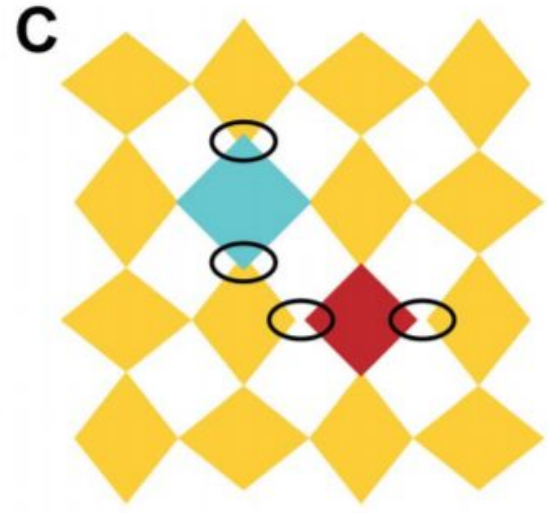


All ions aligned perfectly

- Key:
- Red =  $\text{In}^{3+}$
  - Blue =  $\text{Cu}^{+}$
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Single Perovskite **after** mixing

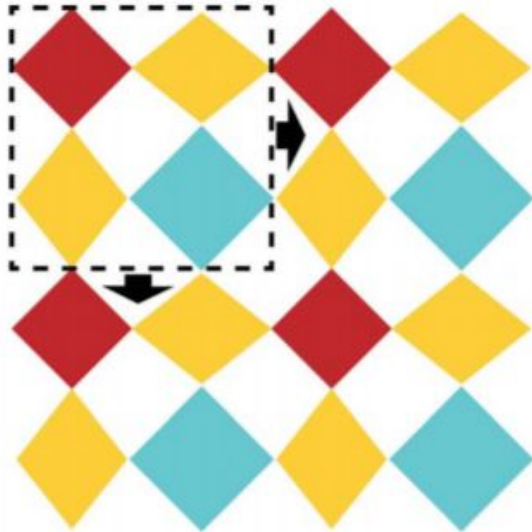


Less stable because of the mismatches (gaps and overlaps between ions)!

# Mixing Double and Single Perovskites

Now.. we need to find a stable arrangement with less mismatches for our mixed perovskite....

# Mixing Double and Single Perovskites



No mismatches - finally,  
something stable!

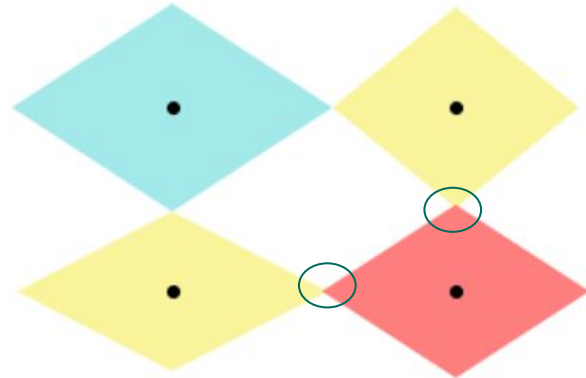
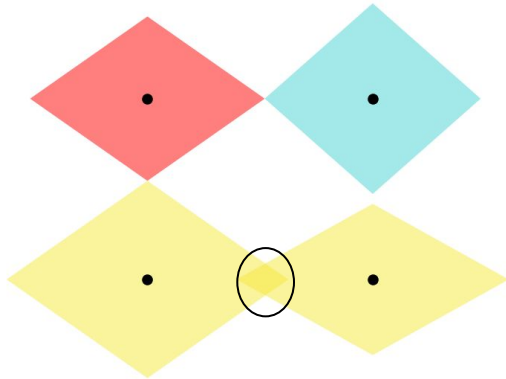


# Objective

- Develop a program using Python to predict possible stable structures with minimal mismatches using different combinations of ions at the B/B' sites

# Results:

- Program predicts all possible structural arrangements with various sets of ions (can specify with ions you want to use to use to form a perovskite crystal)
- Calculates the total mismatch for each arrangement using the length of each bond in the material
- Ex: these are considered different “combinations”/arrangements and have different mismatches

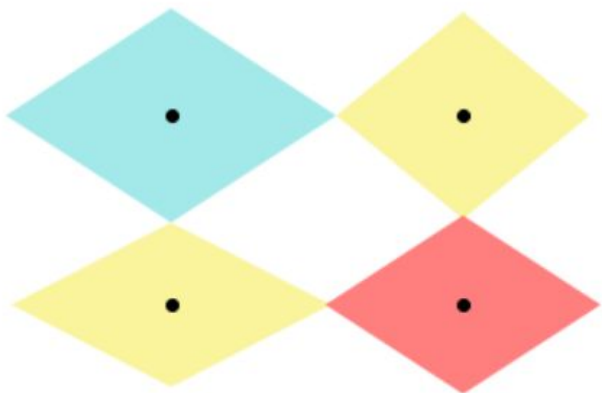


# Results

Key:

- Red =  $\text{In}^{3+}$
- Blue =  $\text{Cu}^{+}$
- Yellow =  $\text{Cu}^{2+}$

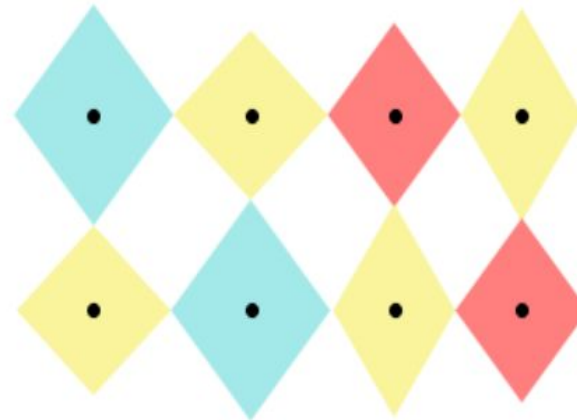
## 2x2 Arrangements:



*Optimal arrangement for a 2x2 structure*

- 48 possible arrangements

## 2x4 Arrangements:

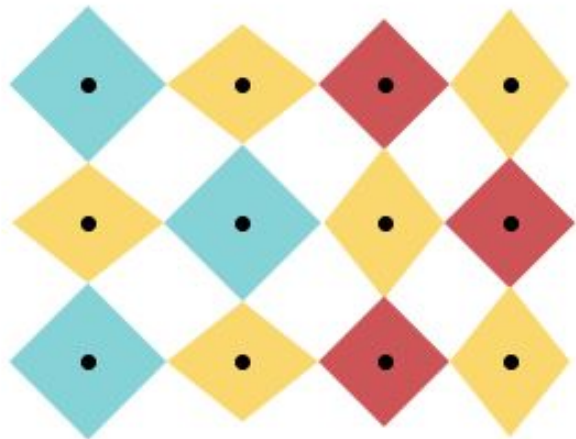


*Optimal arrangement for a 2x4 structure*

- 6720 possible arrangements

# Results

## 3x4 Combinations:



*Optimal combination for a 3x4*

- 1,182,720 possible arrangements

## Takeaways:

- MANY possible combinations with just 1 set of 3 ions - imagine the possibilities when we start to also generate combinations for other sets of ions!
- Computers can save us A LOT of time when planning synthesis reactions

# Future Directions:

## *Long-Term Goal:*

- Add more parameters to make sure materials are stable in experimental conditions
- Account for edge conditions (far right column and bottom row) in geometric calculations
  - Sometimes, the bond lengths are distorted from bonds across multiple unit cells

# Acknowledgments

- Professor Hemamala Karunadasa, Stanford University
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