1 Atomic Structure

1.1 atomic number and mass

1. The _____ is the number of ____ in the nucleus of an atom.

2. The ______ is the total number of _____ and _____ in the nucleus of an atom.

3.

 $^{1}\mathrm{H}$

What does the 1 mean?

4.

 ${}^4_2{
m He}$

5. What does the 4 mean?

6. What does the 2 mean?

7.

 $_{3}^{7}$ Li

How many protons does Lithium have?

How many neutrons does Lithium have?

8.

 $^{2}\mathrm{H}$

based on this symbol, how many protons does Hydrogen have? ____

How many neutrons? _____

1.2 The Bohr model

- 9. The Bohr Model Bohr proposed that an atom was a nucleus with electrons "orbiting" in different ______.
- 10. The electrons closest to the nucleus have the _____ energy, while those further from away have _____ energy.