# Unit 3

#### Modeling Atomic Structure

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PACS

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#### **Atomic Number**

The is the number of

in the nucleus of an atom.

#### **Atomic Number**

The atomic number is the number of

in the nucleus of an atom.

#### **Atomic Number**

The atomic number is the number of protons in the nucleus of an atom.

The the total number of and in the nucleus of an atom.

The mass number the total number of of an atom.

and

in the nucleus

The mass number the total number of protons and of an atom.

in the nucleus

The mass number the total number of protons and neutrons in the nucleus of an atom.

# Hydrogen

 $^{1}H$ 

# Hydrogen

 $^{1}{\rm H}$ 

What does the 1 mean?

# Hydrogen

 $^{1}H$ 

What does the 1 mean?

1 is the total number of neutrons and protons.

 ${}^4_2{
m He}$ 

 $_2^4\mathrm{He}$ 

What does the 4 mean?

 $_2^4\mathrm{He}$ 

What does the 4 mean?

4 is the total number of neutrons and protons.

 $_2^4\mathrm{He}$ 

What does the 4 mean?

4 is the total number of neutrons and protons.

What does the 2 mean?

 $_2^4\mathrm{He}$ 

What does the 4 mean?

4 is the total number of neutrons and protons.

What does the 2 mean?

2 is the number of protons.

 $^{7}_{3}\mathrm{Li}$ 

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How many protons does Lithium have?

 $^{7}_{3}\mathrm{Li}$ 

How many protons does Lithium have?

 $^{7}_{3}\mathrm{Li}$ 

How many protons does Lithium have?

How many neutrons?

$$^{7}_{3}\mathrm{Li}$$

How many protons does Lithium have?

How many neutrons?

$$7 - 3 =$$

$$^{7}_{3}\mathrm{Li}$$

How many protons does Lithium have?

How many neutrons?

$$7 - 3 = 4$$

#### The Great Dane



The Bohr Model - Bohr proposed that an atom was a nucleus with electrons "orbiting" in different

#### The Great Dane



The Bohr Model - Bohr proposed that an atom was a nucleus with electrons "orbiting" in different energy levels.

### **Energy Levels**

The electrons closest to the nucleus have the further from away have energy.

energy, while those

### **Energy Levels**

The electrons closest to the nucleus have the lowest energy, while those further from away have energy.

### **Energy Levels**

The electrons closest to the nucleus have the lowest energy, while those further from away have higher energy.

# Nitrogen

