1 Atomic Structure

atomic number and mass 1.1 1. The _____ is the number of ____ in the nucleus of an atom. 2. The _____ is the total number of ____ and ____ in the nucleus of an atom. 3. $^{1}\mathrm{H}$ What does the 1 mean? 4. ${}_{2}^{4}\mathrm{He}$ 5. What does the 4 mean? 6. What does the 2 mean? 7. $_{3}^{7}$ Li How many protons does Lithium have? _____ How many neutrons does Lithium have? 8. $^{2}\mathrm{H}$

based on this symbol, how many protons does Hydrogen have?

How many neutrons? _____

1.2 The Bohr model

- 9. The Bohr Model Bohr proposed that an atom was a nucleus with electrons "orbiting" in different ______.
- 10. The electrons closest to the nucleus have the _____ energy, while those further from away have _____ energy.

