curriculum map

teacher

August 11, 2025

Contents

1	Math for Chemistry						
	1.1 Mathematics Review	2					
	1.1.1 Laws of exponents	2					
	1.2 Measurement	2					
	1.3 Base and Derived Units	2					
	1.4 Tools for measurement	2					
	1.5 Rules of Sig Figs	2					
	1.5.1 practice	2					
2	Matter and Nomenclature 4						
3	Chemical Reactions						
4	Stoichiometry and the Mole						
5	Atomic Structure and Light						
6	The Periodic Table						
7	Bonding and Lewis Structure						
8	Gases						
9	Kenetics						
10	0 Equilibrium						
11	Acids and Bases	4					

12 Thermodynamics	4
13 Electrochemistry	4
14 Organic Chemistry	4
15 Nuclear Chemistry	4

1 Math for Chemistry

1.1 Mathematics Review

1.1.1 Laws of exponents

$$10^a \times 10^b = 10^{a+b}$$

1.2 Measurement

quantity name abbrev length mass time amount(of substance) temperature electric current luminous intensity

factor	prefix	the long way	abbrev
10^{18}	exa		${ m E}$
10^{15}	peta		Р
10^{12}	tera		${ m T}$
10^{9}	giga		G
10^{6}	mega		M

- 1.3 Base and Derived Units
- 1.4 Tools for measurement
- 1.5 Rules of Sig Figs
- 1.5.1 practice

- **1.** 143 cm = ____ m
- **2.** 674 m = _____ mm
- 3. 0.1235 km =____ m
- **4.** 164 m =____ km
- 5. 1.117 mL =_____L
- **6.** $143cm = _{\underline{\hspace{1cm}}} m$
- 7. $143cm = ___m$
- 8. 143cm = ____m

- 2 Matter and Nomenclature
- 3 Chemical Reactions
- 4 Stoichiometry and the Mole
- 5 Atomic Structure and Light
- 6 The Periodic Table
- 7 Bonding and Lewis Structure
- 8 Gases
- 9 Kenetics
- 10 Equilibrium
- 11 Acids and Bases
- 12 Thermodynamics
- 13 Electrochemistry
- 14 Organic Chemistry
- 15 Nuclear Chemistry