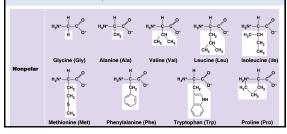


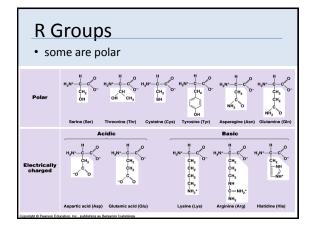
Amino Acid Structure

- 4 Parts:
 - 1. Carboxyl end acts like an acid, can release the hydrogen
 - 2. Amine end acts like a base, accepting a 3rd hydrogen
 - 3. Alpha Carbon central Carbon
 - 4. R group variant, determines the properties of the amino acid

R Groups

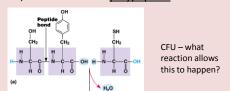
- side chains that determine the properties of the amino acid, each one is different
- · some are nonpolar

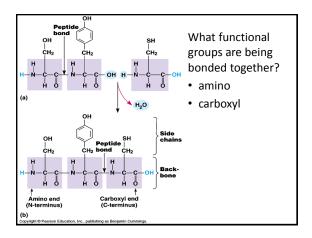


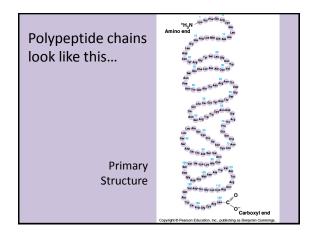


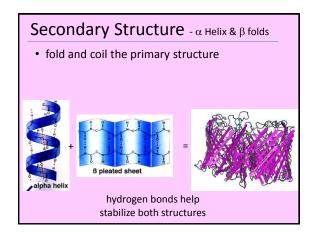
Making of a Protein...

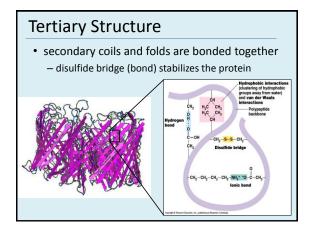
- individual amino acids (monomers) can covalently bond
 - bond is a called a peptide bond
- many amino acids bonded together make a protein, also known as a <u>polypeptide</u>

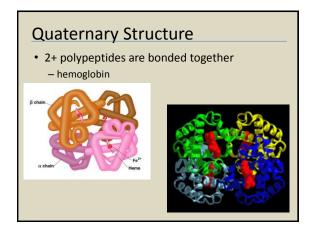




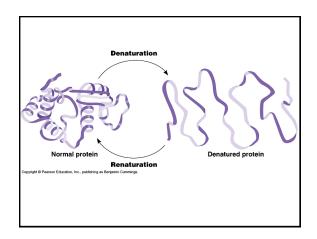


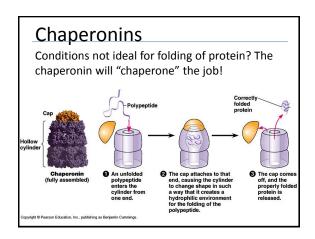






unraveling of the protein cannot function caused by changes in pH – lower pH causes amino acid to lose etemp – disrupts hydrogen bonds salt concentration – can cause attraction between R groups





Proteomics

- large-scale study of protein structure & function
- genome & environment will cause changes of the proteome

