Quantum Mechanics - Problem Set #1

Rest mass of the electron

Magnitude of the electron charge

Avogadro's number

Universal gas constant

Boltzmann's constant

Speed of light

Planck's constant

Vacuum permittivity

Vacuum permeability

Universal gravitational constant

Acceleration due to gravity

atmosphere pressure

1 angstrom

 $m_e = 9.11 \times 10^{-31} \text{ kilogram} = 9.11 \times 10^{-25} \text{ gram}$

 $e = 1.60 \times 10^{-19} \text{ coulomb} = 4.80 \times 10^{-10} \text{ stateoulomb (esu)}$

 $N_0 = 6.02 \times 10^{23} \, \text{per mole}$

 $R = 8.32 \text{ joules/(mole \cdot K)}$

 $k = 1.38 \times 10^{-23} \text{ joule/K} = 1.38 \times 10^{-16} \text{ erg/K}$

 $c = 3.00 \times 10^8 \,\mathrm{m/s} = 3.00 \times 10^{10} \,\mathrm{cm/s}$

 $h = 6.63 \times 10^{-34}$ joule · second = 4.14×10^{-15} eV · second

 $\hbar = h/2\pi$

 $\varepsilon_0 = 8.85\,\times\,10^{-12}\,coulomb^2/(newton\cdot meter^2)$

 $\mu_0 = 4\pi \times 10^{-7} \text{ weber/(ampere · meter)}$

 $G = 6.67 \times 10^{-11} \text{ meter}^3/(\text{kilogram} \cdot \text{second}^2)$

 $g = 9.80 \text{ m/s}^2 = 980 \text{ cm/s}^2$

1 atm = 1.0×10^5 newton/meter² = 1.0×10^5 pascals (Pa)

 $1 \text{ Å} = 1 \times 10^{-10} \text{ meter}$

 $1 \text{ weber/m}^2 = 1 \text{ tesla} = 10^4 \text{ gauss}$

- 1. The wave function of a particle is $e^{i(kx-\omega t)}$, where x is distance, t is time, and k and ω are positive real numbers. The x-component of the momentum of the particle is
 - (A) 0
 - (B) ħω
 - (C) ħk
 - (D) $\frac{\hbar\omega}{c}$
 - (E) $\frac{\hbar k}{\omega}$

- 27. If a freely moving electron is localized in space to within $\triangle x_0$ of x_0 , its wave function can be described by a wave packet $\psi(x, t) = \int_{-\infty}^{\infty} e^{i(kx \omega t)} f(k) dk$, where f(k) is peaked around a central value k_0 . Which of the following is most nearly the width of the peak in k?
 - $(A) \ \triangle k = \frac{1}{x_0}$
 - **(B)** $\triangle k = \frac{1}{\triangle x_0}$
 - (C) $\triangle k = \frac{\triangle x_0}{x_0^2}$
 - (D) $\triangle k = \left(\frac{\triangle x_0}{x_0}\right) k_0$
 - (E) $\triangle k = \sqrt{k_0^2 + \left(\frac{1}{x_0}\right)^2}$
- 28. A system is known to be in the normalized state described by the wave function

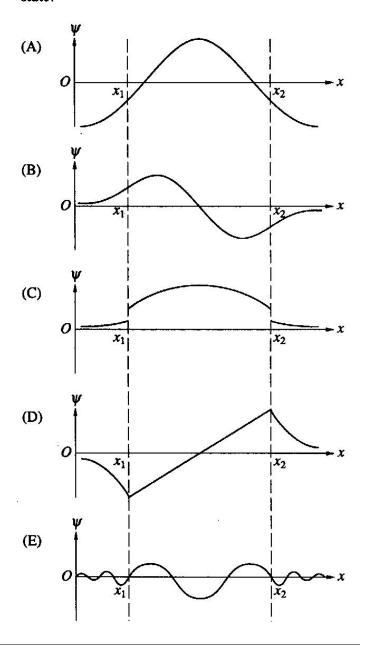
$$\psi(\theta,\,\varphi)\,=\frac{1}{\sqrt{30}}\,(5\,\,Y_4{}^3\,+\,Y_6{}^3\,-\,2\,\,Y_6{}^0)\,,$$

where the $Y \varrho^m(\theta, \varphi)$ are the spherical harmonics. The probability of finding the system in a state with azimuthal orbital quantum number m=3 is

- (A) 0
- **(B)** $\frac{1}{15}$
- (C) $\frac{1}{6}$
- (D) $\frac{1}{3}$
- (E) $\frac{13}{15}$



29. An attractive, one-dimensional square well has depth V_0 as shown above. Which of the following best shows a possible wave function for a bound state?



- 50. The state of a quantum mechanical system is described by a wave function ψ. Consider two physical observables that have discrete eigenvalues: observable A with eigenvalues {α}, and observable B with eigenvalues {β}. Under what circumstances can all wave functions be expanded in a set of basis states, each of which is a simultaneous eigenfunction of both A and B?
 - (A) Only if the values $\{\alpha\}$ and $\{\beta\}$ are nondegenerate
 - (B) Only if A and B commute
 - (C) Only if A commutes with the Hamiltonian of the system
 - (D) Only if B commutes with the Hamiltonian of the system
 - (E) Under all circumstances

Questions 51-53

A particle of mass m is confined to an infinitely deep square-well potential:

$$\begin{array}{l} V(x) = \infty, x \leq 0, x \geq a \\ V(x) = 0, 0 < x < a. \end{array}$$

The normalized eigenfunctions, labeled by the quantum

number
$$n$$
, are $\psi_n = \sqrt{\frac{2}{a}} \sin \frac{n\pi x}{a}$.

- 51. For any state n, the expectation value of the momentum of the particle is
 - (A) 0
 - (B) $\frac{hn\pi}{a}$
 - (C) $\frac{2\hbar n\pi}{a}$
 - (D) $\frac{\hbar n\pi}{a} (\cos n\pi 1)$
 - $(E) \frac{-i\hbar n\pi}{a} (\cos n\pi 1)$
- 52. The eigenfunctions satisfy the condition

The eigenfunctions assert
$$\int_0^a \psi_n *(x) \psi_{\mathcal{Q}}(x) dx = \delta_n \mathcal{Q}, \, \delta_n \mathcal{Q} = 1$$
 if $n = \mathcal{Q}$, otherwise $\delta_n \mathcal{Q} = 0$. This is a statement that the eigenfunctions are

- (A) solutions to the Schrödinger equation
- (B) orthonormal
- (C) bounded
- (D) linearly dependent
- (E) symmetric
- 53. A measurement of energy E will <u>always</u> satisfy which of the following relationships?

$$(A) E \le \frac{\pi^2 \hbar^2}{8ma^2}$$

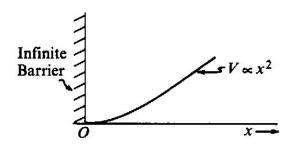
(B)
$$E \ge \frac{\pi^2 \hbar^2}{2ma^2}$$

$$(C) E = \frac{\pi^2 \hbar^2}{8ma^2}$$

$$(D) E = \frac{n^2 \pi^2 \hbar^2}{8ma^2}$$

$$(E) E = \frac{\pi^2 \hbar^2}{2ma^2}$$

- 56. If v is frequency and h is Planck's constant, the ground state energy of a one-dimensional quantum mechanical harmonic oscillator is
 - (A) 0
 - (B) $\frac{1}{3}hv$
 - (C) $\frac{1}{2}hv$
 - (D) hv
 - (E) $\frac{3}{2}hv$



- 89. The energy levels for the one-dimensional harmonic oscillator are $hv\left(n+\frac{1}{2}\right)$, n=0,1,2... How will the energy levels for the potential shown in the graph above differ from those for the harmonic oscillator?
 - (A) The term $\frac{1}{2}$ will be changed to $\frac{3}{2}$.
 - (B) The energy of each level will be doubled.
 - (C) The energy of each level will be halved.
 - (D) Only those for even values of n will be present.
 - (E) Only those for odd values of n will be present.

The matrix
$$A = \begin{bmatrix} 0 & 1 & 0 \\ 0 & 0 & 1 \\ 1 & 0 & 0 \end{bmatrix}$$

has three eigenvalues λ_i defined by $Av_i = \lambda_i v_i$. Which of the following statements is NOT true?

- (A) $\lambda_1 + \lambda_2 + \lambda_3 = 0$
- (B) λ_1 , λ_2 , and λ_3 are all real numbers.
- (C) $\lambda_2 \lambda_3 = +1$ for some pair of roots.
- (D) $\lambda_1\lambda_2 + \lambda_2\lambda_3 + \lambda_3\lambda_1 = 0$
- (E) $\lambda_i^3 = +1, i = 1, 2, 3$
- 99. In perturbation theory, what is the first order correction to the energy of a hydrogen atom (Bohr radius a_0) in its ground state due to the presence of a static electric field E?
 - (A) Zero
 - (B) eEa_0
 - (C) 3eEa₀
 - (D) $\frac{8e^2Ea_0^3}{3}$
 - (E) $\frac{8e^2E^2a_0^3}{3}$