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Determination of 20 synthetic dyes in chili powders and syrup-preserved fruits by liquid chromatography/tandem mass spectrometry



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ABSTRACT

A liquid chromatography/tandem mass spectrometry (LC-MS/MS) method is developed to simultaneously determine 20 synthetic dyes (New Coccine, Indigo Carmine, Erythrosine, Tartrazine, Sunset Yellow FCF, Fast Green FCF, Brilliant Blue FCF, Allura Red AC, Amaranth, Dimethyl Yellow, Fast Garnet GBC, Para Red, Sudan I, Sudan II, Sudan III, Sudan IV, Sudan Orange G, Sudan Red 7B, Sudan Red B, and Sudan Red G) in food samples. This method offers high sensitivity and selectivity through the selection of two fragment ion transitions under multiple reaction monitoring mode to satisfy the requirements of both quantitation and qualitation. Using LC-MS/MS, the newly developed extraction protocol used in this study is rapid and simple and does not require the use of solid-phase extraction cartridges. The linearities and recoveries of the method are observed at the concentration range of $0.10-200 \mu g/kg$ and more than 90% for all dyes, respectively. The method has been successfully applied to screen 18 commercial chili powders and six commercial syruppreserved fruits purchased from retail establishments in Taipei City. The results show that three legal food dyes, Tartrazine, and/or Sunset Yellow FCF, and/or New Coccine, are present in some syrup-preserved fruits. Amaranth, an illegal food dye in certain countries but declared illegal in Taiwan, is found in an imported syrup-preserved fruit.

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1. Introduction

Food dye additives are defined as dyes, pigments, or substances that impart colors into foods, drugs, or cosmetics. Originally, natural dyes present in foods are unstable and altered rapidly during food processing and storage. Hence, synthetic dyes that outperform natural ones at various aspects such as low price, high effectiveness, and excellent stability are widely used by food companies all over the world [1,2].

The toxicities of synthetic azo type food dyes, such as Sudan I–IV, have been confirmed [3-8] and classified as category 3 carcinogen to humans by the International Agency for Research on Cancer [1]. Although Sudan dyes have been

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Fig. 1 - Structures of dyes.

Compound	Precursor ion (m/z)	Product ion (m/z)	Declustering potential (V)	Entrance potential (V)	CEP (V)	Collision energy (V)	CXP (V)
Sudan Orange G	215.1	93.1	31	8	30	31	4
	215.1	122.1	31	8	30	21	4
Dimethyl Yellow	226.1	120.1	41	8.5	30	43	4
	226.1	105.1	41	8.5	30	25	4
Sudan I	249.1	93	26	4.5	30	33	4
	249.1	156.1	26	4.5	30	21	4
Sudan II	277.1	121.1	26	4.5	30	25	4
	277.1	106.1	26	4.5	30	55	4
Sudan Red G	279.1	123.1	26	4.5	30	23	4
	279.1	108.1	26	4.5	30	47	4
Para Red	294.1	156.1	31	5	30	21	4
	294.1	128.1	31	5	30	35	4
Sudan III	353.1	197.1	51	10.5	30	27	4
	353.1	128.1	51	10.5	30	51	4
Sudan Red 7B	380.2	183.1	31	9	30	21	4
buduii ilea / b	380.2	115.1	31	9	30	65	4
Sudan IV	381.1	224.1	51	9	30	31	4
Juduii I V	381.1	225.1	51	9	30	25	4
	381.1	143.1	51	9	30	37	4
	381.1	104.1	51	9	30	83	4
Sudan Red B	381.2	224.1	56	10	30	29	4
Juduii iicu D	381.2	225.1	56	10	30	29	4
	381.2	156.1	56	10	30	27	4
	381.2	134.1	56	10	30	27	4
Fast Garnet GBC	226.1	91	46	9.5	30	29	4
rast darriet dbd	226.1	107	46	9.5	30	31	4
Fast Green FCF	381	80	-40	-10	-33	_98	-8
rast Green r Gr	381	170	-40 -40	-10 -10	-33 -33	-38 -38	_8 _8
Tartrazine	244	80	-21	-10 -10	-29.5	− 62	_8
I di li dZiiie	244	198	-21 -21	-10 -10	-29.5 -29.5	_02 _20	-8 -8
New Coccine	206	80	-21 -35	-10 -10	-29.5 -28.4	-20 -40	-o -8
New Coccine	268	206	–35 –25	-10 -10	-20. 4 -30	- 1 0 -18	-8 -8
Indian Cormina	208	198	-25 -42	-10 -10	-30 -29	-18 -27	−8 −8
Indigo Carmine	226	105	-42 -42	-10 -10	-29 -29	-27 -53	-o -8
Brilliant Blue		80	-42 -45		-29 -33	-53 -92	-8 -8
Billiant Blue	373			-10			
Cumant Vallary	373	170	-45 27	-10 10	-33	-42 20	-8
Sunset Yellow	203	171	-27	-10	-28	-20	-8
	203	207	-27	-10	-28	-20	-8
Erythrosine	834	227	-22	-10	-45 45	-91	-8
.11 5 1	834	127	-80	-10	-45	-84	-8
Allura Red	225	80	-32	-10	-29	-59	-8
	225	136	-32	-10	-29	-34	-8
Amaranth	279	206	-32	-10	-30	-30	-8
	279	221	-32	-10	-30	-22	-8

CEP = collision cell entrance potential; CXP = collision cell exit potential; LC-MS/MS = liquid chromatography/tandem mass spectrometry.

forbidden in recent decades, it can be still found in various imported food products in many countries [4]. In Taiwan, food dyes are strictly regulated by the Standards for Specification, Scope, Application, and Limitation of Food Additives [9], and food additives require governmental approvals for use prior to their inclusion in food. To ensure consumers' health, a reliable screening method for the determination of synthetic dyes in foods is required. In the literature, thin layer chromatography (TLC) [10–12], spectrophotometry [13,14], high-performance liquid chromatography (HPLC) [2,15], capillary electrophoresis [16], liquid chromatography/tandem mass spectrometry (LC-MS/MS) [17–21], and gas chromatography/mass spectrometry (GC-MS) [22] have been used in the determination of dyes in foods. Among these mentioned methods, TLC and spectrophotometry are widely used for the determination of water-

soluble dyes because of their low cost although they often suffer from poor sensitivity and interference from the food matrix. GC or LC methods offer good separation but suffer from time-consuming procedures and complicated instrumentation. In this work, we developed an LC-MS/MS method applied as a single-step extraction protocol offering simple and rapid sample preparation for the determination of dyes in foods.

2. Materials and methods

2.1. Reagents

New Coccine, Indigo Carmine, Erythrosine, Tartrazine, Sunset Yellow FCF, Fast Green FCF, Brilliant Blue FCF, Allura Red AC,

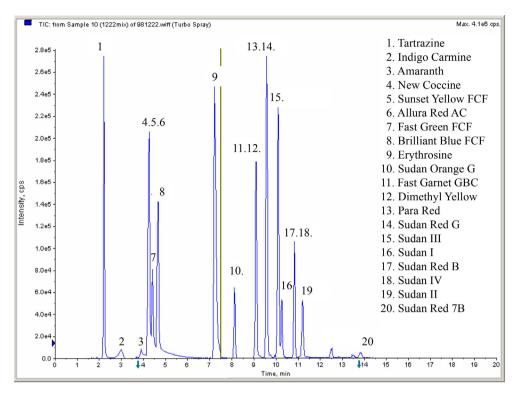


Fig. 2 - Liquid chromatography/tandem mass spectrometry (LC-MS/MS) chromatogram of 20 dyes.

Amaranth, Sudan II, Sudan IV, Dimethyl Yellow, Fast Garnet GBC, Para Red, Sudan Orange G, Sudan Red 7B, Sudan Red B and Sudan Red G were purchased from Sigma-Aldrich (St. Louis, MO, USA). Sudan I was purchased from Tokyo Chemical Industry (Tokyo, Japan). Sudan II was purchased from Kanto Chemical (Tokyo, Japan). Methanol of HPLC grade was obtained from Mallinckrodt (Paris, KY, USA). Dimethyl sulfoxide (DMSO), acetonitrile, and acetic acid of HPLC grade were obtained from Merck (Darmstadt, Germany). Ammonia acetate was from Nacalai Tesque (Kyoto, Japan).

2.2. Instrumentation

The HPLC system consists of an UltiMate 3000 Standard LC System (Thermo Fisher Scientific Inc., Waltham, MA, USA) coupled with a triple quadrupole mass spectrometer API 3200 (AB SCIEX, Framingham, MA, USA) with a Turbo V ion source and Analyst software. Gas nitrogen was supplied by a nitrogen generator (Peak Scientific Instruments Ltd., Chicago, IL, USA). Nitrogen was used as curtain gas, nebulizer gas, and collision gas on the MS. Separation was carried out on an Acclaim Polar Advantage C16 (3 μ m, 120 Å, 4.6 \times 150 mm) column (Thermo Fisher Scientific Inc.). Allegra 25R centrifuge was from Beckman Coulter (Brea, CA, USA).

2.3. Standard solutions

A stock solution of individual dyes was prepared by dissolving 50 mg of the compound in 50 mL DMSO (1 mg/mL). This stock solution was further diluted with methanol to obtain a working solution of 10 μ g/mL and stored in the dark at -18 °C.

2.4. Sample preparation and extraction

Each homogenized chili powder or syrup-preserved fruit sample (10 g) obtained from retail stores was extracted with acetonitrile twice (50 mL \times 2, 5 minutes). Two extracts were combined and then centrifuged at 4000 rpm (5 minutes, 15°C). Supernatant was filtered through a 0.45- μm nylon membrane filter and then transferred into an amber autosampler vial. Sample vials were stored at 4°C prior to analysis by LC-MS/MS.

2.5. LC-MS/MS analysis

The mobile phase of the HPLC system consisted of acetonitrile (A) and 20 mM ammonia acetate buffer with 1% acetic acid (B). Linear gradient elution was programmed as follows: 0-3 minutes, 98-50% B; 3-8 minutes, 50-0% B; 8-15 minutes, 0% B; 15-15.5 minutes, 0-98% B; 15.5-20 minutes, 98% B. The flow rate was set at 1 mL/min. Electrospray ionization mass spectra (ESI-MS) were acquired in the positive (ESI +) and then negative (ESI -) ion mode. Other MS parameters were as follows: turbo gas, 50 psi; curtain gas, 10 psi; collision gas, 5 psi; source temperature, 300°C; nebulizer gas, 60 psi; interface heater, on; needle current, 4000 V; entrance potential, 10 V. A dwell time of 200 milliseconds was set between transitions of the ions. A total of 11 dyes—Dimethyl Yellow, Fast Garnet GBC, Para Red, Sudan I, Sudan II, Sudan III, Sudan IV, Sudan Orange G, Sudan Red 7B, Sudan Red B, and Sudan Red G-were detected by positive ESI, and nine other dyes—New Coccine, Indigo Carmine, Erythrosine, Tartrazine, Sunset Yellow FCF, Fast Green FCF, Brilliant Blue FCF, Allura Red AC, and

0.9987

0.9966

Compound	Linear range	Chili powder		Raisins	
		Linear equation	R ²	Linear equation	R ²
Sudan Orange G	0.1–200 ng/mL	y = 314.460x + 449.8	0.9985	y = 313.574x + 1190.2	0.998
Dimethyl Yellow	0.1-200 ng/mL	y = 582.447x + 794.7	0.9965	y = 813.147x + 855.1	0.996
Sudan I	0.1-200 ng/mL	y = 290.223x + 129.3	0.9961	y = 211.347x + 127.5	0.996
Sudan II	0.1-200 ng/mL	y = 466.798x + 548.3	0.9956	y = 430.291x + 711.5	0.995
Sudan Red G	0.1–200 ng/mL	y = 633.668x + 1262.5	0.9986	y = 614.833x + 802.5	0.998
Para Red	0.1-200 ng/mL	y = 112.067x + 427.6	0.9985	y = 1030.80x + 136.9	0.998
Sudan III	0.1-200 ng/mL	y = 394.643x + 297.7	0.9968	y = 335.834x + 376.1	0.996
Sudan Red 7B	0.1-200 ng/mL	y = 205.369x + 540.8	0.9978	y = 192.426x + 289.2	0.997
Sudan IV	0.1-200 ng/mL	y = 637.006x + 318.4	0.9975	y = 603.386x + 105.7	0.997
Sudan Red B	0.1-200 ng/mL	y = 871.714x + 442.1	0.9943	y = 861.649x + 531.0	0.994
Fast Garnet GBC	0.1-200 ng/mL	y = 771.906x + 356.4	0.9979	y = 680.185x + 289.2	0.997
Fast Green FCF	0.1–50 μg/mL	y = 33,738x + 15,134	0.9972	y = 35,358x + 19,933	0.997
Tartrazine	0.1–50 μg/mL	y = 61,987x + 24,647	0.9991	y = 63,975x + 25,357	0.999
New Coccine	0.1–50 μg/mL	y = 12,869x + 20,973	0.9983	y = 12,943x + 28,456	0.998
Indigo Carmine	0.1–50 μg/mL	y = 23,832x + 45,322	0.9968	y = 21,445x + 48,732	0.996
Brilliant Blue FCF	0.1–50 μg/mL	y = 30,091x + 33,557	0.9954	y = 29,630x + 31,787	0.995
Sunset Yellow FCF	0.1–50 μg/mL	y = 12,162x + 44,064	0.9947	y = 12,936x + 44,523	0.994
Erythrosine	0.1–50 μg/mL	y = 79,847x + 16,843	0.9949	y = 73,580x + 14,818	0.994

y = 34,718x + 20,513

y = 14,927x + 37,365

Table 2 – Linear ranges, equations, and determination of coefficients (R^2) of 20 dyes in chili powder and syrup-preserved

Amaranth (Fig. 1)—were detected by negative ion ESI. Target dyes were identified by their retention times (Rt) and selected ions in multiple reaction monitoring (MRM) mode summarized in Table 1. Quantification was conducted with the MRM transition by selecting two characteristic ions that gave the most intense signal/noise ratio.

 $0.1-50~\mu g/mL$

 $0.1-50 \mu g/mL$

2.6. Analytical performance parameters

Allura Red AC

Amaranth

Three replicates of chili powder and syrup-preserved fruit samples spiked with dyes at different concentrations-5, 10, and 50 mg/kg for Indigo Carmine, Sunset Yellow FCF, Tartrazine, Brilliant Blue FCF, Fast Green FCF, New Coccine, Erythrosine, Allura Red AC, and Amaranth (group A), and 10, 100, and 200 μg/kg for Sudan Red G, Sudan Red 7B, Dimethyl Yellow, Sudan II, Sudan Orange G, Sudan I, Para Red, Fast Garnet GBC, Sudan Red B, Sudan III, and Sudan IV (group B) were extracted and detected as described above.

3. Results and discussion

The aim of this work was to develop and validate a simple and rapid method for the detection of dyes in foods.

3.1. LC-MS/MS determination and quantitation

Each target analyte (10 μg/mL, in DMSO/acetate buffer solution/ acetonitrile = 1:50:49, v/v/v) was tuned individually in order to obtain stable precursor and product ion abundance. These solutions were introduced into the mass spectrometer via a syringe pump at a flow rate of 5-20 μL/min. The analyte-dependent parameters, such as declustering potential, entrance potential, and collision energy, were optimized to increase sensitivity (summarized in Table 1). According to the European guidelines EC/657/2002 [8], each analyte will earn 4 identification points (IPs) in this study based on the determination of one precursor (1 IP) and two product ions (1.5 \times 2 = 3 IPs) using the LC-MS/MS technique. The developed mass spectrometric condition in this study met the EU confirmation requirement [23]. Satisfactory separation and response for all analytes were obtained under the gradient elution described above. The LC-MS/MS chromatograms of all dyes are shown in Fig. 2.

y = 36,704x + 20,688

y = 17,963x + 32,978

3.2. Method validation

0.9987

0.9966

The selectivity of the LC method was investigated by observing the potential interferences between the analytes and impurities/food matrix in the sample extracts. Chili powder and syrup-preserved fruit samples spiked with various concentrations of dyes were analyzed, and the results showed that there was no significant interference observed at the corresponding retention time of each target analyte.

3.2.1. Linearity

Standard curves were made in triplicate for each concentration of all dyes. Good linearities were achieved at the concentrations of 0.1-200 ng/mL for Sudan Orange G, Dimethyl Yellow, Sudan I, Sudan II, Sudan Red G, Para Red, Sudan III, Sudan Red 7B, Sudan IV, Sudan Red B, Fast Garnet GBC (group A dyes), and 0.1-50 μg/mL for Fast Green FCF, Tartrazine, New Coccine, Indigo Carmine, Brilliant Blue FCF, Sunset Yellow FCF, Erythrosine, Allura Red AC, and Amaranth (group B dyes) (Table 2). The regression equation and determination of the coefficient (R²) of each dye are shown in Table 2. All R² values exceeded 0.994, indicating good linearities.

3.2.2. Sample extraction procedure and recovery

Based on the chemical structures and polarities of all dyes, acetonitrile was chosen as an extraction solvent for chili

Table 3 $-$ Recoveries and coefficient of variatio	n (CV, n $=$ 3) of 20 dyes in spike	d chili powder and syrup-p	reserved fruit
samples.			

Compound	Spiked levels (μg/g)	Chili powder		Raisins	
		Recoveries (%)	CV (%)	Recoveries (%)	CV (%)
Sudan Orange G	0.01	99.9	2.2	100.9	2.9
, and the second	0.1	99.7	3.2	99.8	2.1
	0.2	100.2	3.6	100.6	6.5
Dimethyl Yellow	0.01	98.0	1.4	99.3	13.4
	0.1	97.0	2.3	98.3	4.3
	0.2	98.4	11.8	98.3	7.1
Sudan I	0.01	94.4	8.3	94.7	5.9
	0.1	94.2	1.7	93.1	5.2
Cudon II	0.2 0.01	94.7 91.4	2.3 13.6	93.3 97.4	3.0 14.2
Sudan II	0.01	91.4	8.5	96.7	9.2
	0.2	91.9	9.5	97.0	8.8
Sudan Red G	0.01	92.3	13.0	94.0	9.6
budan ilea d	0.1	91.8	13.3	93.2	4.9
	0.2	92.7	6.8	93.8	4.6
Para Red	0.01	91.2	15.2	100.5	7.3
	0.1	91.1	13.5	99.4	2.4
	0.2	91.2	10.6	99.7	9.3
Sudan III	0.01	92.2	11.6	94.5	4.6
	0.1	92.0	12.0	93.1	12.1
	0.2	92.7	2.5	93.8	8.3
Sudan Red 7B	0.01	100.0	3.4	95.2	8.2
	0.1	99.9	12.8	94.2	13.4
	0.2	100.4	9.0	94.4	5.4
Sudan IV	0.01	99.2	1.7	97.5	8.4
	0.1	98.7	3.3	96.2	4.7
	0.2	99.3	8.8	96.8	4.8
Sudan Red B	0.01	99.2	1.5	96.0	15.7
	0.1	98.8	1.8	94.9	4.9
	0.2	99.6	7.1	95.6	1.5
Fast Garnet GBC	0.01	95.3	8.3	92.4	6.1
	0.1	94.4	1.1	91.3	10.9
	0.2	95.8	3.2	92.2	2.8
Fast Green FCF	5	92.7	13.7	100.7	2.8
	10	91.8	7.0	99.2	6.3
	50	93.2	4.9	99.4	7.8
Tartrazine	5	91.9	13.9	96.1	3.4
	10	91.2	13.1	94.9	5.4
New Coccine	50 5	92.5 99.2	12.8 1.2	95.3 98.4	9.3 9.4
New Goccine	10	98.8	14.2	97.7	2.3
Indigo Carmine	50 5	99.7 93.2	8.3 10.4	98.2 91.1	2.8 4.8
naigo Garinine	10	93.0	2.7	90.1	5.6
	50	93.4	8.6	90.7	7.9
Brilliant Blue FCF	5	99.0	5.2	92.5	5.5
oriniant blue I Gi	10	98.5	9.8	91.4	5.2
	50	99.2	5.5	91.6	4.4
Sunset Yellow FCF	5	92.1	14.0	100.2	3.9
	10	91.7	4.6	98.9	6.3
	50	92.5	5.8	99.4	4.5
Erythrosine	5	95.1	6.6	91.7	6.8
	10	95.0	13.2	90.6	3.7
	50	95.7	9.0	91.4	7.4
Allura red AC	5	93.4	9.2	96.1	5.5
	10	93.2	4.7	95.6	3.9
	50	93.7	8.8	95.9	6.4
Amaranth	5	94.6	9.1	91.9	4.0
	10	93.6	9.1	91.4	4.4
	50	95.2	3.8	92.1	8.3

powder and syrup-preserved fruit samples. Furthermore, acetonitrile offered other advantages such as good extraction yield, less fat solubility, precipitation of carbohydrates, and precipitation of proteins. Satisfactory recovery rates were obtained using this rapid and simple sample preparation procedure (Table 3).

This LC-MS/MS method was validated in different matrices, including chili powder and syrup-preserved fruit. To determine the recovery rates of the developed method for 20 dyes, standard dye solutions in three different concentrations (0.01, 0.1, and 0.2 μ g/g in group A dyes, and 5, 10, and 50 μ g/g in group B dyes) were spiked individually into blank chili powder and syrup-preserved fruit samples in triplicate. The results showed that the recoveries ranged from 90.1% to 100.4% for group A dyes, and from 91.1% to 100.9% for group B dyes (Table 3). The coefficient of variation of most data was below 10% and below 15.7% for some owing to the effects of interference in low-concentration samples. Overall, the recovery results were satisfactory. This developed LC-MS/MS method has many advantages over traditional TLC or HPLC methods

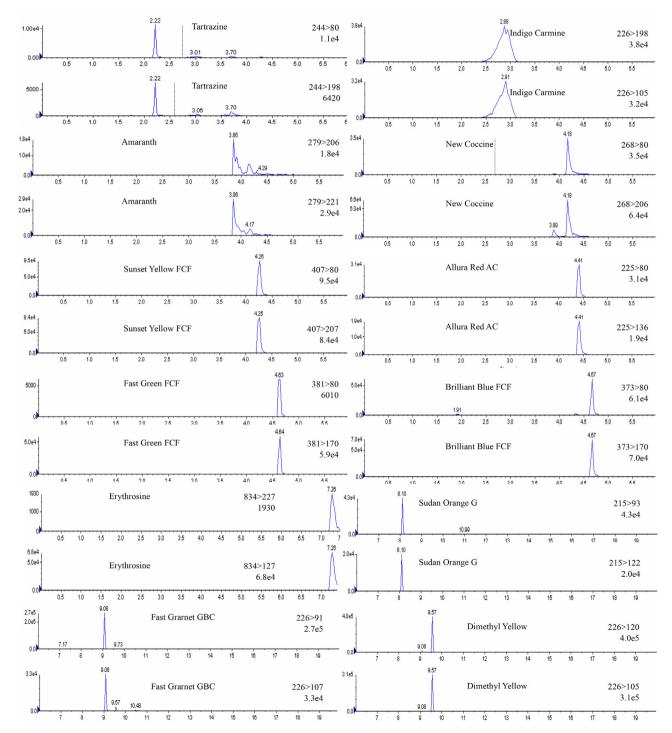


Fig. 3 – Multiple reaction monitoring (MRM) chromatograms of 20 dyes (10 μ g/mL).

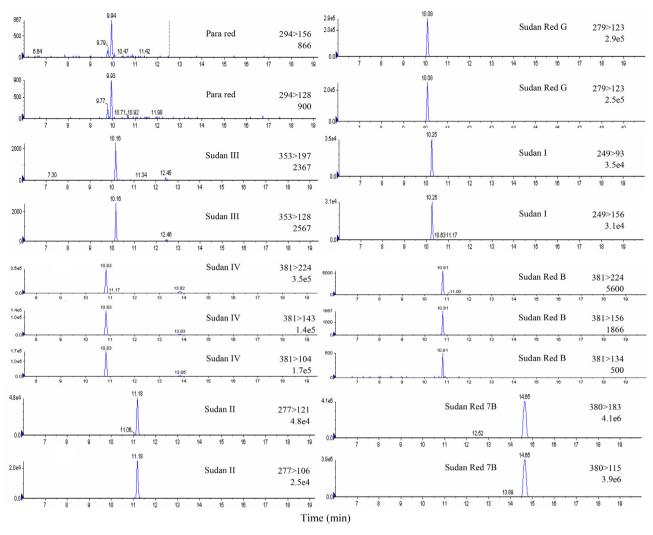


Fig. 3 – (continued).

[24] such as high sensitivity, no extra cleanup procedure, multidye analysis, and simultaneous quantitation and confirmation. In the traditional TLC method, 20 dyes required four developing conditions, and each took 60 minutes developing time. In the traditional HPLC method, 20 dyes required different LC conditions in order to obtain good separation, and each run took 1–2 hours to complete. In this work, the LC-MS/MS analysis time was significantly reduced from hours to 20 minutes. Moreover, the sample extraction time was shortened by a simple procedure and simultaneous treatment. The amount of solvent required in the analysis was also substantially decreased. This method, which offers rapid and high-throughput dye analysis, is suitable to be applied in routine tests for screening of banned dyes in foods (Fig. 3).

3.2.3. Limit of quantitation

The limit of quantitation (LOQ) in this study was determined by the sample concentration, which produced a peak area 10 times greater than noise. The LOQ was significantly improved by the LC-MS/MS method in comparison with the traditional HPLC. The obtained LOQ of dyes ranged from approximately 0.001 to 1 mg/kg (Table 4). Some dyes, such as Fast Green FCF, New Coccine, Erythrosine, Allura Red AC, and Amarath, had higher LOQs at 1 mg/kg because of their weak ionization.

3.3. Application to real samples

The developed method in this study was applied for the determination of 20 dyes in six commercial syrup-preserved fruit and 18 commercial chili powder products collected from supermarkets in Taipei City, Taiwan. The results are shown in Table 5. Tartrazine, Sunset Yellow, and New Coccine were detected in two, three, and two of six syrup-preserved fruit samples, respectively, and they are labeled on the packages. However, Amaranth, which is an illegal food dye in certain countries but declared illegal in Taiwan, was detected in one of six syrup-preserved fruit samples. This sample was imported and the use of Amaranth is not indicated on its package. No dyes were detected in 18 chili powder samples.

Table 4 $-$ Limit of quantitation (LOQ) of 20 dyes.			
Compound	LOQ (mg/kg)		
Sudan Orange G	0.01		
Dimethyl Yellow	0.005		
Sudan I	0.01		
Sudan II	0.005		
Sudan Red G	0.001		
Para Red	0.01		
Sudan III	0.025		
Sudan Red 7B	0.001		
Sudan IV	0.05		
Sudan Red B	0.02		
Fast Garnet GBC	0.01		
Fast Green FCF	1		
Tartrazine	0.5		
New Coccine	1		
Indigo Carmine	0.25		
Brilliant Blue FCF	0.5		
Sunset Yellow FCF	0.25		
Erythrosine	1		
Allura Red AC	1		
Amaranth	1		

Table 5 — Quantitation results for dyes in positive chili powders analyzed by LC-MS/MS.				
Sample	Dyes	Concentration (mg/kg)		
Syrup-preserved fruit (S1)	Tartrazine	250.8		
	Sunset Yellow	22.8		
Syrup-preserved fruit (S2)	Tartrazine	597.8		
	Sunset Yellow	15.2		
Syrup-preserved fruit (S3)	Amaranth	169.5		
	Sunset Yellow	29.3		
	New Coccine	214.6		
Syrup-preserved fruit (S3)	New Coccine	43.6		
LC-MS/MS = liquid chromatography/tandem mass spectrometry.				

4. Conclusion

Monitoring of synthetic dyes in foods is very important in both domestic and imported foods. This study presents a suitable analysis method for the extraction, detection, and quantitation of 20 dyes using LC-MS/MS in chili powder and syrup-preserved fruit products. The newly developed preparation procedure, through the extraction of acetonitrile, is rapid and simple and offers very good recovery and precise results. The 20-minute LC-MS/MS method under MRM mode is able to detect all target compounds in a single run with an LOQ between 0.001 and 1 mg/kg. Overall, the LC-MS/MS method can be applied in routine dye testing and surveillance programs for the control of the presence of dyes in chili powders and syrup-preserved fruits.

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