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**39.1.15 - Meat and Meat Products**

**AOAC Official Method 928.08**  
**Nitrogen in Meat**

**Kjeldahl Method**  
**First Action 1928**  
**Final Action 1974**

***A. Reagents***

(a) *Kjeldahl catalyst*.—15 g  $\text{K}_2\text{SO}_4$  + 0.7 g  $\text{HgO}$  (commercially prepared catalysts are available containing pumice, if desired).

(b) *Sulfuric acid*.— $\text{H}_2\text{SO}_4$ , ACS.

(c) *NaOH solution*.—Prepare 1200 mL  $\text{NaOH}$  (1 + 1). Let stand until clear (ca 10 days).

(d) *Metallic zinc*.—Powder, ACS, to be used if catalyst does not contain pumice.

(e) *Indicator solution*.—Fleisher Methyl Purple, or equivalent.

(f) *Acid potassium phthalate*.—NIST Standard.

(g) *Standardized NaOH solution*.—0.2000 0.0004M. Add 108 mL  $\text{NaOH}$  (1 + 1) to  $\text{CO}_2$ -free  $\text{H}_2\text{O}$  and dilute to 10 L. Standardize against potassium acid phthalate, using phenolphthalein indicator.

(h) *Standardized acid solution*.—0.2000 0.0004M.  $\text{HCl}$  or 0.0002M  $\text{H}_2\text{SO}_4$

(i) *Hydrochloric acid*.—Dilute 178 mL 35-37%  $\text{HCl}$  to 10 L. Standardize against standard  $\text{NaOH}$  solution, and adjust strength accordingly.

(j) *Sulfuric acid*.—Dilute 55 mL 98%  $\text{H}_2\text{SO}_4$  to 10 L. Standardize against standard  $\text{NaOH}$  solution, and adjust strength accordingly.

(k) *Sodium hydroxide.—Sodium thiosulfate solution.*—Dissolve 460 g  $\text{Na}_2\text{S}_2\text{O}_3 \cdot 5\text{H}_2\text{O}$  in  $\text{H}_2\text{O}$ ; dilute to 1 L with  $\text{H}_2\text{O}$ , and add this solution to 15,250 g NaOH dissolved in 14,250 mL  $\text{H}_2\text{O}$ . This will yield 20 L of 50% (w/v) NaOH solution. If other volumes are desired, adjust weights of NaOH and  $\text{Na}_2\text{S}_2\text{O}_3 \cdot 5\text{H}_2\text{O}$  accordingly. Specific gravity of final solution should be 1.45. 40 g  $\text{K}_2\text{S}$  or  $\text{Na}_2\text{S}$  per liter of final solution may be used in lieu of  $\text{Na}_2\text{S}_2\text{O}_3 \cdot 5\text{H}_2\text{O}$ . (*Note:* Prepared in excess to precipitate residual Hg.)

Standardize each standard solution with primary standard (*see* [Appendix A](#), Standard Solutions and Certified Reference Materials) and check one against the other. Test all reagents before using by blank determination with 2 g sugar, which ensures partial reduction of any nitrates present and provides organic material to assure acid requirements and digestion characteristics similar to those encountered in meat standards.

### ***B. Determination***

Place weighed test portion (2.0-2.2 g) in digestion flask. Add 0.7 g HgO or 0.65 g metallic Hg, 15 g powdered  $\text{K}_2\text{SO}_4$  or anhydrous  $\text{Na}_2\text{SO}_4$ , and 40 mL  $\text{H}_2\text{SO}_4$ . If test portion >2.2 g is used, increase  $\text{H}_2\text{SO}_4$  by 10 mL for each g test portion used. Place flask in inclined position and heat gently until frothing ceases (if necessary, add small amount of paraffin to reduce frothing); boil briskly until solution clears, and then 30 min longer (2 h if organic material is present).

Cool, add ca 200 mL  $\text{H}_2\text{O}$ , cool <25°C, add 25 mL of the sulfide of thiosulfate solution, and mix to precipitate Hg. Add few Zn granules to prevent bumping, tilt flask, and add layer of NaOH without agitation. (For each 10 mL  $\text{H}_2\text{SO}_4$  used, or its equivalent in diluted  $\text{H}_2\text{SO}_4$ , add 15 g solid NaOH or enough solution to make contents strongly alkaline.) (Thiosulfate or sulfide solution may be mixed with the NaOH solution before addition to flask.) Immediately connect flask to distilling bulb on condenser, and, with tip of condenser immersed in standardized acid and 5-7 drops indicator in receiver, rotate flask to mix contents thoroughly; then heat until all  $\text{NH}_3$  has distilled (150 mL distillate). Remove receiver, wash tip of condenser, and titrate excess standardized acid in distillate with standardized NaOH solution. Correct for blank determination on reagents.

$$\text{N, \%} = \frac{[(\text{mL standardized acid} \times \text{M acid}) - (\text{mL standardized NaOH} \times \text{M NaOH})]}{1.4007/\text{g test portion}}$$

### **References:**

*JAOC* **11**, 408(1928).

## **Alternative II (Copper-Based Catalyst) First Action 1993**

(Applicable to determination of nitrogen in meat and meat products.)

Results of the interlaboratory study supporting the acceptance of the method:

Ground beef ( protein value = 17%)

$s_r = 0.22$ ;  $s_R = 0.26$ ;  $RSD_r = 1.30\%$ ;  $RSD_R = 1.51\%$

Canned ham ( protein value = 18%)

$s_r = 0.23$ ;  $s_R = 0.27$ ;  $RSD_r = 1.27\%$ ;  $RSD_R = 1.48\%$

Smoked ham ( protein value = 15%)

$s_r = 0.20$ ;  $s_R = 0.24$ ;  $RSD_r = 1.36\%$ ;  $RSD_R = 1.58\%$

Pork sausage ( protein value = 10.5%)

$s_r = 0.16$ ;  $s_R = 0.19$ ;  $RSD_r = 1.54\%$ ;  $RSD_R = 1.80\%$

Cooked sausage ( protein value = 12%)

$s_r = 0.18$ ;  $s_R = 0.21$ ;  $RSD_r = 1.47\%$ ;  $RSD_R = 1.72\%$

Dry cured ham ( protein value = 23%)

$s_r = 0.28$ ;  $s_R = 0.31$ ;  $RSD_r = 1.16\%$ ;  $RSD_R = 1.36\%$

### ***C. Principle***

Test sample is digested in  $H_2SO_4$ , using  $CuSO_4$  as catalyst, converting N to  $NH_3$  which is distilled and titrated.

### ***D. Apparatus and Reagents***

See **A**, except a copper catalyst is used in place of mercury catalyst.

(a) *Copper catalyst*.—Prepare using 15.0 g  $K_2SO_4$  and 0.45 g  $CuSO_4$ . As boiling aid, 0.1 g pumice may be added. (Propac 15p, Alfie Packers, Inc., Omaha, NE, USA, or other commercially prepared catalysts are suitable.)

### ***E. Preparation of test samples***

See [983.18](#) (see 39.1.01).

### ***F. Determination***

See **B**, except in paragraph 1, after solution clears, boil 45 min longer (2 h for test portions containing organic material).

***G. Calculations***

See **B**. (*Note:* Alternative II yields results that are, on average, 99.0% of results from Alternative I.)

**Reference:**

*J. AOAC Int.* **77**, 1542(1994).