

AOAC Official Method 931.08

Formaldehyde in Food

First Action 1931

(See also [964.21](#) [see 44.5.14].)

A. Preparation of Test Portion

If food is solid or semisolid, macerate 100 g with 100 mL H₂O in mortar. Transfer to 800 mL Kjeldahl flask, acidify with H₃PO₄, add 1 mL excess, connect with condenser through trap, and slowly distil 50 mL. For milk, dilute 100 mL with 100 mL H₂O, and acidify and distil as for solids. With other liquid foods, acidify 200 mL and distil as for solids.

B. Chromotropic Acid Test

(a) *Reagent*.—Prepare saturated solution of 1,8-dihydroxynaphthalene-3,6-disulfonic acid (ca 500 mg/100 mL) in ca 72% H₂SO₄ (pour 150 mL H₂SO₄ into 100 mL H₂O and cool). Solution is light straw-colored.

(b) *Test*.—Place 5 mL reagent in test tube and add, with mixing, 1 mL distillate, **A**. Place in boiling H₂O bath 15 min, and observe during heating period. Presence of HCHO is indicated by appearance of light to deep purple (depth of color depending on amount of HCHO present).

Reference:

Z. Anal. Chem. **110**, 22(1937).

C. Hehner-Fulton Test

(a) *Oxidizing solution*.—To cold H₂SO₄ add, in small portions, equal volume saturated Br₂-H₂O, cooling throughout operation.

(b) *Test*.—To 6 mL cold H₂SO₄ add 5 mL distillate, **A**, slowly and with cooling. Place 5 mL mixture in test tube, and add, slowly and with cooling, 1 mL aldehyde-free milk, then 0.5 mL oxidizing solution. Mix. Purplish-pink indicates HCHO.

Reference:

Ind. Eng. Chem. Anal. Ed. **3**, 199(1931).

CAS-50-00-0 (formaldehyde)