# **9.2.36 - Metals and Other Elements at Trace Levels in Foods / Single Element Methods**

# **AOAC Official Method 973.36 Titanium in Cheese**

# Spectrophotometric Method First Action 1973 Final Action 1976

#### A. Standard Solution

*Titanium dioxide standard solution.*—0.1 mg/mL. Accurately weigh 50 mg TiO<sub>2</sub> and transfer to 250 mL beaker; add 15 g anhydrous Na<sub>2</sub>SO<sub>4</sub> and 50 mL H<sub>2</sub>SO<sub>4</sub>. Add boiling chips, cover with watch glass, and heat to bp on hot plate to dissolve. Cool, and cautiously add 100 mL H<sub>2</sub>O with stirring. (Warm on steam bath if solution becomes cloudy.) Cool, transfer solution to 500 mL volumetric flask containing 200 mL H<sub>2</sub>O, and dilute to volume with H<sub>2</sub>O.

### B. Preparation of Material

Weigh, to nearest 0.1 g, 10 g prepared test sample, <u>955.30</u> (*see* 33.7.02), into 100 mL Pt dish and char under IR lamp. Place in cold furnace and ignite at 850°C to white ash.

Cool, add ca 1.5 g anhydrous Na<sub>2</sub>SO<sub>4</sub> and 10 mL H<sub>2</sub>SO<sub>4</sub>, cover with watch glass, and bring to bp on hot plate to dissolve. Turn heat off and let cool on hot plate. Cautiously rinse cover, carefully add ca 30 mL H<sub>2</sub>O, and mix with stirring rod to disperse any insoluble salts. Heat on steam bath if insoluble material forms cake on bottom of dish.

Transfer quantitatively to 100 mL volumetric flask with aid of ca 40 mL  $H_2O$ . If solution is cloudy, heat on steam bath or in boiling  $H_2O$  bath to clarify. Cool, and dilute to volume with  $H_2O$ .

## C. Preparation of Standard Curve

Transfer 0, 1, 2, 3, 4, and 5 mL  $TiO_2$  standard solution to separate 5 mL glass-stoppered graduates (or volumetric flasks) and dilute to volume with  $H_2SO_4$  (1 + 9).

Add 0.2 mL 30%  $H_2O_2$ , mix, and determine A on recording spectrophotometer in 1.0 cm cells from 650 to 325 nm against 0.2 mL 30%  $H_2O_2$  in 5.0 mL  $H_2SO_4$  (1 + 9). Determine A at maximum, ca 408 nm, and prepare standard curve.

#### D. Determination

Transfer 3.0 mL test solution to 5 mL glass-stoppered graduate (or volumetric flask), dilute to volume with  $H_2SO_4$  (1 + 9), and continue as in C, beginning "Add 0.2 mL 30%  $H_2O_2$ , . . . ".

Determine mg  $TiO_2$  in test solution from standard curve, and calculate as percent  $TiO_2$ .

#### **Reference:**

JAOAC **56**, 535(1973).

CAS-7440-32-6 (titanium)

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