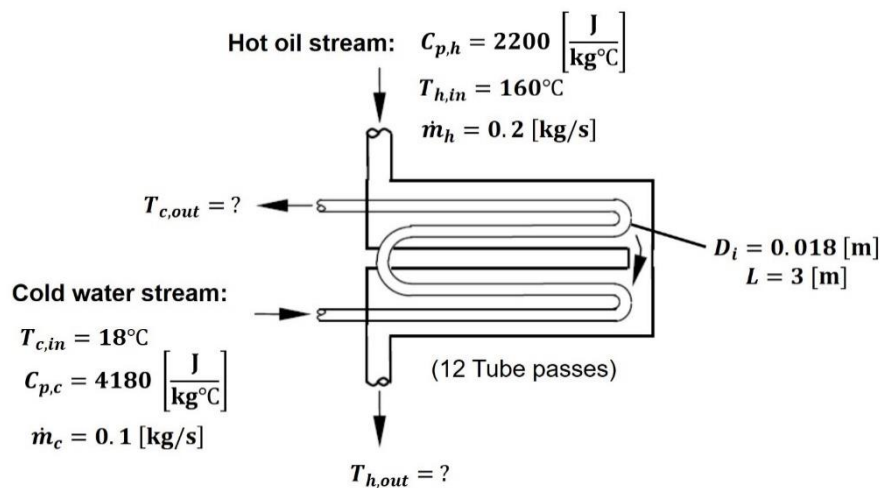


WORKED EXAMPLE

EFFECTIVENESS-NTU METHOD

Hot oil ($C_p = 2200 \text{ J/kg}\cdot^\circ\text{C}$) is to be cooled by water ($C_p = 4180 \text{ J/kg}\cdot^\circ\text{C}$) in a 2-shell-pass and 12-tube-pass heat exchanger. The tubes are thin-walled and are made of copper with a diameter of 1.8 cm. The length of each tube pass in the heat exchanger is $L = 3 \text{ m}$, and the overall heat transfer coefficient is $U = 340 \text{ W/m}^2\cdot^\circ\text{C}$. Water flows through the tubes at a total rate of 0.1 kg/s , and the oil through the shell at a rate of 0.2 kg/s . The water and the oil enter at temperatures of 18°C and 160°C , respectively.

Determine the rate of heat transfer in the heat exchanger and the outlet temperatures of the water and the oil.



SOLUTION

Objective is to determine \dot{Q} and unknown outlet temperatures.

Since we don't know \dot{Q} and have insufficient information to infer Q or the outlet temperatures, then this is our clue that we need to use the ε -NTU method.

We can get the heat transfer rate from (equation 23, lecture notes):

$$\varepsilon = \frac{\dot{Q}}{\dot{Q}_{\max}}$$

where,

$$\dot{Q}_{\max} = C_{\min}(T_{h,in} - T_{c,in})$$

and

$$C_{\min} = \min\{\dot{m}_c C_{p,c}; \dot{m}_h C_{p,h}\}$$

With,

$$\dot{m}_c C_{p,c} = 0.1 \times 4180 = 418 \text{ [W/}^\circ\text{C]}$$

and

$$\dot{m}_h C_{p,h} = 0.2 \times 2200 = 440 \text{ [W/}^\circ\text{C]}$$

$$\therefore C_{\text{minimum}} = \dot{m}_c C_{p,c} = 418 \text{ [W/}^\circ\text{C]}$$

and

$$\begin{aligned} Q_{\text{max}} &= C_{\text{minimum}}(T_{h,\text{in}} - T_{c,\text{in}}) \\ &= 418 \times (160 - 18) \\ &= 59,356 \text{ [W]} \end{aligned}$$

Next, we need to get ε , and noting that $\varepsilon = f(C, NTU)$, we also have to find the heat capacity ratio C from:

$$C = \frac{C_{\text{minimum}}}{C_{\text{maximum}}} = \frac{418}{440} = 0.95$$

and number of transfer units (NTU) from:

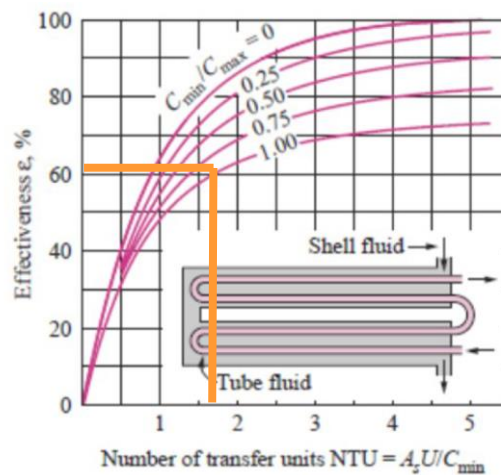
$$NTU = \frac{UA}{C_{\text{minimum}}} = \frac{340 \times 2.04}{418} = 1.659$$

where the surface area, A , was calculated from:

$$A = \pi DL \times n = \pi \times 0.018 \times 3 \times 12 = 2.04 \text{ [m}^2\text{]}$$

and the parameter $n = 12$ is the number of tube passes.

Now, from the effectiveness chart (Figure 13-26(d), Appendix C, lecture notes) for $\varepsilon = f(0.95, 1.659)$ we get $\varepsilon \sim 0.61$.



(d) Two-shell passes and 4, 8, 12, ... tube passes

The actual heat transfer rate is then:

$$Q = \varepsilon Q_{\max} = 0.61 \times 59356 = 36207.2 \text{ [W]}$$

The outlet temperatures then follow from:

$$Q = \dot{m}_h c_{p,h} (\Delta T_h)$$

$$\therefore T_{h,\text{out}} = T_{h,\text{in}} - \frac{Q}{\dot{m}_h c_{p,h}} = 160 - \frac{36207}{440} = 77.7 \text{ [}^\circ\text{C]}$$

and, similarly:

$$T_{c,\text{out}} = T_{c,\text{in}} + \frac{Q}{\dot{m}_c c_{p,c}} = 18 + \frac{36207}{418} = 104.6 \text{ [}^\circ\text{C]}$$