## Feb 20 Absorption Lab Data

 $SI = bottom \ of \ absorption \ column$ 

S2 = middle of absorption column

S3 = gas outlet from absorption column

Data with Heater Off: [temperature set point between 9°C and 11°C]

	Sample 1	Sample 2	Sample 3	Sample 4
S1 CO <sub>2</sub> Concentration	29.3%	27.8%	27.7%	27.5%
S2 CO <sub>2</sub> Concentration	29.1%	27.7%	27.4%	27.3%
S3 CO <sub>2</sub> Concentration	28.3%	27.1%	27.1%	27.0%
$CO_2$ Gauge Temp (°C)	20	19.5	19	19
$H_2O$ Gauge Temp (°C)	16	15.5	15	15
Temp on Screen (°C)	13	12	12	11
Mix Outlet Temp (°C)	20	19	19.5	19

Data with Heater On: [temperature set point at 26°C]

	Sample 5	Sample 6
S1 CO <sub>2</sub> Concentration	27.5%	27.5%
S2 CO <sub>2</sub> Concentration	27.4%	27.3%
S3 CO <sub>2</sub> Concentration	26.8%	26.8%
CO <sub>2</sub> Gauge Temp (°C)	20	20
$H_2O$ Gauge Temp (°C)	17	17
Temp on Screen (°C)	13	14
Mix Outlet Temp (℃)	24	25

## Notes:

- Took a long time for the operating temperature (of inlet  $CO_2$  and  $H_2O$  streams) to drop. Held at  $13^{\circ}C$  for  $\sim$ 20 mins before first set of samples was taken, then dropped slightly for subsequent samples (temperature set point was between  $9^{\circ}C$  and  $11^{\circ}C$  for the entirety of the initial sample collection time)
- K2 pressure decreased from -0.1 to -0.5 atm between first and second set of samples

o Sample 1 data is less reliable than that of 2, 3, and 4

\*\*Additional note: we don't know the composition of liquid stream into absorption unit-- needs to be approximated or solved for