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Polarimetry

By United States Bureau of Standards

Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1914 Excerpt: .+ 2 NaOH = Pb(NO₃)₂.Pb(OH)₂ + 2NaNO₃ The precipitate is washed free of soluble impurities and mixed to a cream for use in clarification. The clarification may also be performed by forming the basic lead nitrate in the solution to be polarized. This is done by first adding a measured quantity of the lead nitrate solution 1 to 10 cc, according to the darkness of the sample, and then an exactly equal quantity of the alkaline solution. The excess of alkali must be avoided. The whole is then shaken and made to volume. The latter procedure gives the better clarification, but introduces a considerable quantity of soluble salts, which may affect the polarization. The defects of the basic nitrate are in general 43 Herles: Zs. Zuckerind Bohmen, 13, p. 559; 14, p. 343; 21, p. 189. those of the basic acetate. The volume of the precipitate...



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