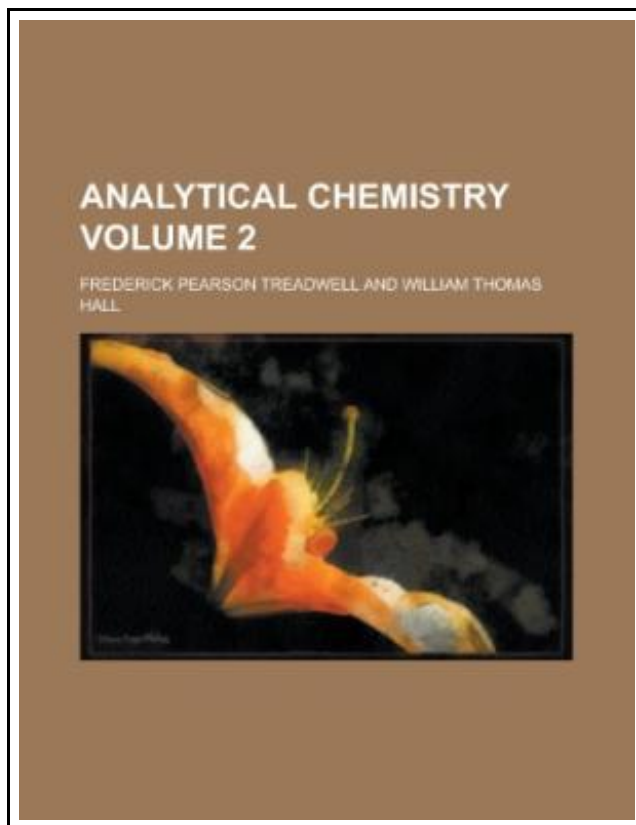


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Rarebooksclub.com, United States, 2013. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can usually download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1904 edition. Excerpt: .lacking in Fig. 55.) The right end of the tube is left open for the time being. A slow current of oxygen is passed through the apparatus and the furnace is lighted. At first the flame is turned low and the whole tube is heated equally. Gradually the temperature is raised, until, with the tiles covering the tube, the copper oxide is at a dull-red heat. Usually some water condenses in the right-hand end of the tube; this is expelled by carefully holding a hot tile under the tube. When all the water is removed, and when the presence of oxygen can be detected at the right end of the tube (by its igniting a glowing splinter), this end of the tube is closed with a rubber stopper through which an open calcium chloride tube is placed. The burners are now turned down and the oxygen current discontinued. All of the flames are extinguished after some time except those under the right half of the tube. While the tube is cooling, the calcium chloride tube and the potash bulb (or soda-lime tubes) are weighed (the stoppers being replaced immediately after the weighing) and from 0.15-0.2 gm. of the substance is weighed into a porcelain or platinum boat. If the substance is a difficultly-volatile oil it is weighed into a The oxygen must be free from hydrogen. Commercial oxygen often contains the latter, in which case it is necessary to pass the gas through...



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