

# Climate Change - Geological Perspective at Stockholms Universitet

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## 1 Reading assignment: Earth's Climate Chapter 4

### 1.1 Key terms

**Greenhouse era:** times when no ice sheets are present

**Icehouse era:** times when ice sheets are present

**Faint young Sun paradox:** the mystery why the Earth's climate has remained relatively stable throughout most of the planet's history, even though the Sun shone 25% to 30% more faintly 4.55 Byr than today.

**Thermostat:** thermostat's role is to mitigate extreme temperature by reacting to hot temperature with cooling down the system (e.g. house) and to cold by heating up. We don't know what the Earth's thermostat was through the history, recompensating for the faint young Sun. Candidates include chemical weathering and life.

**Silicate materials:** examples include quartz and feldspar. Silicate materials typically are made of positively charged cations ( $\text{Na}^{+1}$ ,  $\text{K}^{+1}$ ,  $\text{Fe}^{+2}$ ,  $\text{Mg}^{+2}$ ,  $\text{Al}^{+3}$  and  $\text{Ca}^{+2}$ ) that are chemically bonded to negatively charged  $\text{SiO}_4$  (silicate) structures.

**Chemical weathering feedback:** chemical weathering creates a negative feedback in the climate. Since chemical weathering is strongly correlated to temperature and precipitation, we can distinguish two causal chains:

- initial change  $\rightarrow$  warmer climate  $\rightarrow$  increased temperature, precipitation, vegetation  $\rightarrow$  increased chemical weathering  $\rightarrow$  increased  $\text{CO}_2$  removal by weathering  $\rightarrow$  reduction of initial warming
- initial change  $\rightarrow$  colder climate  $\rightarrow$  decreased temperature, precipitation, vegetation  $\rightarrow$  decreased chemical weathering  $\rightarrow$  decreased  $\text{CO}_2$  removal by weathering  $\rightarrow$  reduction of initial cooling

**Gaia hypothesis:** in its weakest and commonly accepted form, it states that as life-forms gradually developed in complexity, they played a progressively greater role in chemical weathering and its control of Earth's climate. In its most extreme version, it states that life evolved for the purpose of regulating Earth's climate.

**Snowball Earth hypothesis:** the hypothesis that Earth was once nearly frozen, around 715 to 640 million years ago. Climate scientists have found evidence that glaciers existed on several continents during that time. Some believe these continents were located in the tropics then, but it's hard to locate them back in time.

### 1.2 Review questions

#### 1.2.1 Why is Venus so much warmer than Earth today?

Its atmosphere has 96%  $\text{CO}_2$  (compared to Earth's 0.2%), creating a much stronger greenhouse effect, trapping much more heat.

#### 1.2.2 What factors explain why Earth is habitable today?

Small greenhouse effect adding only  $32^\circ\text{C}$  to average temperature in Earth's atmosphere.

#### 1.2.3 Why does the faint young Sun pose a paradox?

Astrophysical models of the Sun's evolution indicate it was 25% to 30% weaker early in Earth's history. Climate model simulations show that the weaker sun would have resulted in a completely frozen Earth for more than half of its early history if the atmosphere had the same composition as it does today.

Primitive life forms date back to at least 3.5 Byr ago, and their presence on Earth is incompatible with a completely frozen planet at that time.

#### 1.2.4 What evidence suggests that Earth has always had a long term thermostat regulating its climate?

The faint young Sun paradox, the specific evidence being prevalence of water-deposited sedimentary rocks throughout Earth's early history.

#### 1.2.5 Why is volcanic input of $\text{CO}_2$ to Earth's atmosphere not a candidate for its thermostat?

Volcanic processes are driven by the heat sources located deep in the Earth's interior and are well removed from contact with (and reactions to) climate system.

#### 1.2.6 What climate factors affect the removal of $\text{CO}_2$ from the atmosphere by chemical weathering?

Temperature: weathering rates roughly double for each  $10^\circ\text{C}$  increase in temperature.

Precipitation: increased rainfall boosts the level of groundwater held in soils, and the water combines with  $\text{CO}_2$  to form carbonic acid and enhance the weathering process.

Vegetation: plants extract  $\text{CO}_2$  from the atmosphere through photosynthesis, and deliver it to soils, where it combines with groundwater to form carbonic acid. It enhances the rate of chemical breakdown of minerals. Presence of vegetation is estimated to increase the rate of chemical weathering by a factor of 2 to 10.

### 1.2.7 Where did the extra CO<sub>2</sub> from Earth's early atmosphere go?

Sediments and rocks.

### 1.2.8 What arguments support and oppose the Gaia hypothesis that life is Earth's true thermostat?

Critics say that too many of the active roles played by organisms in the biosphere today are relatively recent developments in Earth's history. They also point out that the very late appearance of shell-bearing oceanic organisms near 540 million years ago means that life had played no obvious role in transferring the products of chemical weathering on land to the seafloor for the preceding 4 Byr.

Supporters claim that critics underestimate the role of primitive life-forms such as algae in the ocean and microbes on land in Earth's earlier history.

Marine organisms that created oxygen through photosynthesis long ago are believed to have enabled the development of oxygen-rich atmosphere 2.4 Byr.

## 2 Lecture 1: The controls of climate on geological timescales

**Time imbalance:** Coal takes hundreds of millions of years to accumulate from fossils, but takes decades of burning to release. Accumulation happens on **geological** timescale and release at **antropogenic** timescale.

Average Earth surface temperature is around 15°C.

### 2.1 Climate factors

Earth absorbs sunlight and radiates heat energy back into space. These 3 factors control the process:

- solar radiation
- albedo effect
- greenhouse effect

#### Solar radiation

Some prerequisites for calculations:

**Stefan-Boltzmann law** describes the intensity of the thermal radiation emitted by matter in terms of that matter's temperature. Formula is  $E = \sigma T^4$ , where  $\sigma = 5.670367 \times 10^{-8} W.m^{-2}.K^{-4}$

**Solar radiation** constant, in other words, the amount of energy emitted by the Sun is  $3.87 \times 10^{26} W^1$ .

**Solar constant**  $S_0$  describes the amount of energy received by a given area one astronomical unit<sup>2</sup> away from the Sun. Let's calculate it:

$$d_{Earth} = 149,597,870,700m$$

<sup>1</sup>When an object's velocity is held constant at one meter per second against a constant opposing force of one newton, the rate at which work is done is one watt:  $1W = 1kg \cdot m^2 \cdot s^{-3}$

<sup>2</sup>roughly equal to average distance Sun-Earth

Solar constant  $S_0 = \frac{Q}{4\pi d^2} = 1362 W.m^{-2}$ . Since Earth is not flat, but is a rotating sphere, this number is divided by 4, so the effective energy received from Solar radiation is  $342 W.m^{-2}$ .

Now from Stefan-Boltzmann's law, we can calculate the temperature:

$$E = \sigma T^4$$

$$E = 342 W.m^{-2}$$

$$T = (E \cdot \sigma^{-1})^{1/4} = 6^\circ$$

Now let's compare with values for Venus:

$$d_{Venus} = 108 \times 10^9 m$$

$$E_{Venus} = 658 W.m^{-2}$$

$$T_{Venus} = -55^\circ$$

#### Albedo

Black seat: low albedo, white cat: high albedo

Venus has albedo effect of  $\alpha = 77\%$

Earth has albedo effect of  $\alpha = 30\%$

Of course, Earth's albedo is much harder to calculate because the terrain varies a lot, compared to Venus which has a relatively uniform surface.

Venus radiates back to space  $658 W.m^{-2} \cdot 77\% = 504 W.m^{-2}$ . Earth radiates back to space  $342 W.m^{-2} \cdot 30\% = 103 W.m^{-2}$ .

Taking into account albedo effect, Venus' surface temperature should be  $-46^\circ$  and Earth's  $-18^\circ$ .

#### Greenhouse effect

Earth: greenhouse effect increases temperature by  $32^\circ$ .

Let's calculate how much the temperature increased due to greenhouse effect since the preindustrial era, knowing that CO<sub>2</sub>'s content in atmosphere increased from 285ppm to 425ppm.

$$\Delta T = 4.38 \ln \frac{CO_{2\text{present day}}}{CO_{2\text{preindustrial}}} = 4.38 \ln \frac{425\text{ppm}}{285\text{ppm}} = 1.75^\circ$$

### 2.2 Earth's temperature summary

$$\begin{matrix} 6^\circ & + & -24^\circ & + & 32^\circ \\ \text{Solar radiation} & & \text{Albedo} & & \text{Greenhouse cases} \end{matrix}$$

### 2.3 Faint Young Sun paradox

We have fossils from 3.5 Byr ago. Earliest fossils are stromatolites<sup>3</sup>.

Assuming the same percentage of CO<sub>2</sub> in the atmosphere, the average temperature on Earth at that time (3.5 Byr ago) should have been around  $0^\circ$  (due to lower solar radiation), meaning no running water, which precludes the possibility of life.

### 2.4 Source of CO<sub>2</sub> on geological timescales

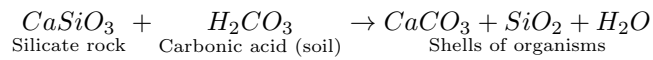
Volcanoes

<sup>3</sup>Stromatolites are layered sedimentary formations created mainly by photosynthetic microorganisms such as cyanobacteria, sulfate-reducing bacteria and Pseudomonadota (formerly proteobacteria).

## 2.5 Earth's thermostat – chemical weathering

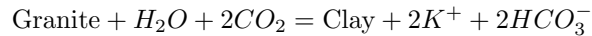
Hydrolysis is the main mechanism for removing CO<sub>2</sub> from the atmosphere. Three key ingredients are minerals that make typical continental rocks, water derived from rain, and CO<sub>2</sub> derived from the atmosphere.

The central equation for chemical weathering is:



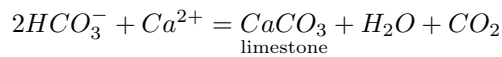
## 2.6 Chemical weathering of silicate rocks

First stage of chemical weathering happens under the influence of rain:



During the weathering, carbon dioxide switches from being a greenhouse gas to being a solute.

The bicarbonate ions are then carried by rivers and eventually end up in seas and oceans. In the ocean, bicarbonates find calcium which they react with, and make limestone, which is calcium carbonate.



We took 2 molecules of carbon from the atmosphere, and return only one, the other one is deposited in limestone. Thus the precipitation limestone is a sink.

Q: Can chemical weathering of silicate rocks compensate for anthropogenic CO<sub>2</sub> emissions?

A: No, it is way too slow.