### Ammonia Fountain

## **Purpose**

To demonstrate the solubility of a gas and the effect of a gas pressure difference.

#### **Materials**

A large glass container stopper assembly

1 liter round bottom flask 2 hole stopper to fit round bottom

phenolphthalein glass tube

ring stand and iron ring disposable pipet bulbs (2)

ammonia gas disposable pipet

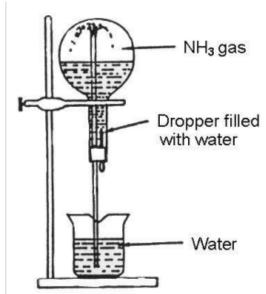
#### **Procedure**

1. The set-up for the apparatus is sketched below: (the disposable pipet must be fine tipped

and the flask dry.)

2. Fill the beaker three-quarters with water and add a few drops of phenolphthalein indicator.

- 3. Fill the dropper with water and insert it into the stopper assembly.
- 4. Invert the flask and fill the flask with ammonia gas.
  - a. Use a lecture bottle of ammonia gas.
  - b. Alternative: React and equal amount of NH<sub>4</sub>Cl<sub>(5)</sub> and NaOH<sub>(5)</sub> in a dray flask fitted with a one hole stopper and glass tube.
- 5. Collect enough gas to turn red litmus paper blue at the mouth of the inverted round bottom flask.
- 6. Seal the round bottom flask with the stopper assembly. Seal the long glass tube with a disposable pipet bulb.
- 7. Place the stopper assembly extending from the flask into the beaker. Make certain the glass tube is just above the bottom of the beaker.
- 8. To begin the reaction remove the pipet bulb from the glass tube while it is submerged. Squirt water from the dropper into the flask.
- 9. The water will rise in the tube and will spray into the flask as a pink fountain.



### **Additional Information**

For a Holiday (red & green) version, add the following to the water:

10 mL bromothymol blue

5 mL methyl red

5 drops 6M HCl

# **Questions for the Students**

- 1. What is the pressure of the ammonia in the flask before the water is added? How do you know?
- 2. What happens to the pressure of the ammonia gas when the water is added? Why?
- 3. At the end of the demonstration was the flask filled completely with water? What does this tell you?
- 4. Why is the water pink when it enters the flask?

# **Disposal**

Solution can be poured down the drain.