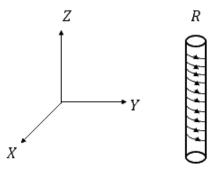
Date of Exam: 9th January (Shift II)

Time: 2:30 pm - 5:30 pm

Subject: Physics

1. An electron gun is placed inside a long solenoid of radius R on its axis. The solenoid has $n \frac{\text{turns}}{\text{length}}$ and carries a current i. The electron gun shoots an electron along the radius of solenoid with speed v. If the electron does not hit the surface of the solenoid, maximum possible value of v is (all symbols have their standard meaning):

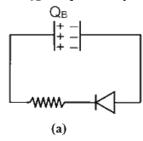


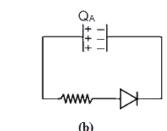
a.
$$\frac{2Re\mu_0in}{m}$$
c.
$$\frac{Re\mu_0in}{m}$$

b.
$$\frac{Re\mu_0 in}{2m}$$

d.
$$\frac{Re\mu_0in}{4m}$$

Two identical capacitors A and B, charged to the same potential 5 V are connected in two different circuit as shows below at time t = 0. If the charges on capacitors A and Bat time t=CR is Q_A and Q_B respectively, then (Here e is the base of natural logarithm)





a.
$$CV$$
, $\frac{CV}{e}$
c. $\frac{CV}{e}$, $\frac{VC}{2}$

b.
$$\frac{cV}{e}$$
, $\frac{cV}{2e}$
d. $\frac{cV}{e}$, CV

d.
$$\frac{CV}{e}$$
, CV

3. For the four sets of three measured physical quantities as given below. Which of the following options is correct?

(i)
$$A_1 = 24.36, B_1 = 0.0724, C_1 = 256.2$$

(ii)
$$A_2 = 24.44$$
, $B_2 = 16.08$, $C_2 = 240.2$

$$(iii)$$
 $A_3 = 25.2, B_3 = 19.2812, C_3 = 236.183$

$$(iv) A_4 = 25, B_4 = 236.191, C_4 = 19.5$$

a.
$$A_4 + B_4 + C_4 < A_1 + B_1 + C_1 =$$

 $A_2 + B_2 + C_2 = A_3 + B_3 + C_3$

b.
$$A_1 + B_1 + C_1 = A_2 + B_2 + C_2 =$$

 $A_3 + B_3 + C_3 = A_4 + B_4 + C_4$

c.
$$A_1 + B_1 + C_1 < A_3 + B_3 + C_3 < A_2 + B_2 + C_2 < A_4 + B_4 + C_4$$

d.
$$A_4 + B_4 + C_4 < A_1 + B_1 + C_1 < A_3 + B_3 + C_3 < A_2 + B_2 + C_2$$

4. A particle starts from the origin at t=0 with an initial velocity of $\vec{u}=3\hat{\imath}$ from origin and moves in the x-y plane with a constant acceleration $\vec{a}=(6\hat{\imath}+4\hat{\jmath})$ m/s². The x-coordinate of the particle at the instant when its y –coordinated is 32 m is D meters. The value of D is:

5. A spring mass system (mass m, spring constant k and natural length l) rest in equilibrium on a horizontal disc. The free end of the spring is fixed at the center of the disc. If the disc together with spring mass system, rotates about its axis with an angular velocity ω ($k >>> m\omega^2$), the relative change in the length of the spring is best given by the option:

a.
$$\frac{m\omega^2}{3k}$$

C.
$$\frac{m\omega^2}{k}$$

b.
$$\sqrt{\frac{2}{3}} \left(\frac{m\omega^2}{k} \right)$$

d.
$$\frac{2m\omega^2}{k}$$

d.
$$\frac{2m\omega^2}{k}$$

6. A small circular loop of conducting wire has radius α and carries current i. It is placed in a uniform magnetic field B perpendicular to its plane such that when rotated slightly about its diameter and released, its starts performing simple harmonic motion of time period T. If the mass of the loop is m then :

a.
$$T = \sqrt{\frac{\pi M}{iB}}$$

c. $T = \sqrt{\frac{\pi M}{2iB}}$

c.
$$T = \sqrt{\frac{\pi M}{2iB}}$$

b.
$$T = \sqrt{\frac{2\pi M}{iB}}$$

b.
$$T = \sqrt{\frac{2\pi M}{iB}}$$

d. $T = \sqrt{\frac{2M}{iB}}$

7. A small spherical droplet of density d is floating exactly half immersed in a liquid of density ρ and surface tension T. The radius of droplet is (take note that the surface tension applied an upward force on droplet)

a.
$$r = \sqrt{\frac{2T}{3(\rho+d)g}}$$
c.
$$r = \sqrt{\frac{T}{(d-\rho)g}}$$

c.
$$r = \sqrt{\frac{T}{(d-\rho)g}}$$

b.
$$r = \sqrt{\frac{T}{(\rho+d)g}}$$

d.
$$r = \sqrt{\frac{3T}{(2d-\rho)g}}$$

d.
$$r = \sqrt{\frac{3T}{(2d-\rho)g}}$$

8. A wire of length L and mass per unit length $6 \times 10^{-3} \, kg/m$ is put under tension of 540 N. Two consecutive frequencies that it resonates at are: 420 Hz and 490 Hz. Then L in meter is

a. 8.1 m

b. 2.1 m

c. 1.1 m

d. 5.1 m

9th Jan (Shift 2, Physics)

Page | 6

9. A plane electromagnetic wave is propagating along the direction $\frac{\hat{\imath}+\hat{\jmath}}{\sqrt{2}}$, with the polarization along the direction \hat{k} . The correct form of the magnetic field of the wave would be (here B_0 is an appropriate constant)

a.
$$B_0 \frac{\hat{\imath} - \hat{\jmath}}{\sqrt{2}} cos \left(\omega t - k \left(\frac{\hat{\imath} + \hat{\jmath}}{\sqrt{2}} \right) \right)$$

b.
$$B_0 \frac{i+j}{\sqrt{2}} cos \left(\omega t + k \left(\frac{i+j}{\sqrt{2}} \right) \right)$$

c.
$$B_0 \frac{j-\hat{\iota}}{\sqrt{2}} cos \left(\omega t + k \left(\frac{\hat{\iota}+\hat{\jmath}}{\sqrt{2}}\right)\right)$$

d.
$$B_0 \hat{k} cos \left(\omega t - k \left(\frac{\hat{\imath} + \hat{\jmath}}{\sqrt{2}} \right) \right)$$

10. Two gases-Argon (atomic radius 0.07 *nm*, atomic weight 40) and Xenon (atomic radius 0.1 *nm*, atomic weight 140) have the same number density and are at the same temperature. The ratio of their respective mean free time is closest to

11. Two steel wires having same length are suspended from a ceiling under the same load. If the ratio of their energy stored per unit volume is 1: 4, the ratio of their diameters is:

a.
$$\sqrt{2}: 1$$

b.
$$1:\sqrt{2}$$

12. Planets A has a mass M and radius R. Planet B has the mass and half the radius of planet A. If the escape velocities from the planets A and B are v_A and v_B respectively, then surfaces is $\frac{v_A}{v_B} = \frac{n}{4}$, the value of n is:

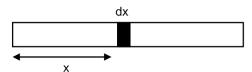
a. 3

b. 2

c. 4

d. 5

13. A rod of length L has non-uniform linear mass density given by $\rho(x) = \left(a + b\left(\frac{x}{L}\right)^2\right)$, Where a and b are constants and $0 \le x \le L$. The value of x for the center of mass of the rod is at:



a. $\frac{3L}{2} \left(\frac{2a+b}{3a+b} \right)$
c. $\frac{3L}{4} \left(\frac{a+b}{3a+b} \right)$

- b. $\frac{3L}{4} \left(\frac{2a+b}{3a+b} \right)$ d. $\frac{4L}{3} \left(\frac{a+b}{3a+b} \right)$

14. A particle of mass m is projected with a speed u from the ground at angle $\theta = \frac{\pi}{3}$ is w.r.t. horizontal (x-axis). When it has reached its maximum height, it collides completely inelastically with another particle of the same mass and velocity $u \hat{\imath}$. The horizontal distance covered by the combined mass before reaching the ground is :

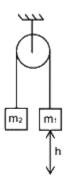
a.
$$\frac{3\sqrt{3}u^2}{8g}$$

b.
$$\frac{2\sqrt{2}u^2}{g}$$

$$c. \quad \frac{5u^2}{8g}$$

d.
$$\frac{3\sqrt{2}u^2}{4g}$$

15. A uniformly thick wheel with moment of inertia I and radius R is free to rotate about its center of mass (see fig). A massless string is wrapped over its rim and two blocks of massless string is wrapped over its rim and two blocks of masses m_1 and m_2 ($m_1 > m_2$) are attached to the ends of string. The system is released from rest. The angular speed of the wheel when m_1 descend by a distance h is :



a.
$$\frac{1}{R} \sqrt{\frac{2(m_1 - m_2)gh}{(m_1 + m_2) + \frac{I}{R^2}}}$$

C.
$$\frac{gh}{R} \sqrt{\frac{(m_1 - m_2)}{m_1 + m_2 + \frac{1}{R}}}$$

b.
$$\frac{gh}{R} \sqrt{\frac{(m_2 + m_1)}{m_1 + m_2 + \frac{1}{R^4}}}$$

d.
$$\frac{1}{R} \sqrt{\frac{2(m_2 + m_1)gh}{m_1 + m_2 + \frac{I}{R^2}}}$$

16. The energy required to ionise a hydrogen like ion in its ground state is 9 Rydbergs. What is the wavelength of the radiation emitted when the electron in this ion jumps from the second excited state to the ground stare?

a. 8.6

b. 11.4

c. 24.2

d. 35.8

17. There is a small source of light at some depth below the surface of water (refractive index $\frac{4}{3}$) in a tank of large cross sectional surface area. Neglecting any reflection from the bottom and absorption by water, percentage of light that emerges out of surface is (nearly): [Use the fact that surface area of a spherical cap of height h and radius of curvature r is $2\pi rh$]

a. 17 %

b. 34 %

c. 50 %

d. 21 %

9th Jan (Shift 2, Physics)

Page | 12

18. An electron of mass m and magnitude of charge |e| initially at rest gets accelerated by a constant electric field E. The of charge of de-Broglie wavelength of this electron at time t ignoring relativistic effects is

a.
$$-\frac{|e|Et}{h}$$

$$C. -\frac{h}{|e|Et^2}$$

b.
$$\frac{h}{|e|E\sqrt{t}}$$

b.
$$\frac{h}{|e|E\sqrt{t}}$$
d.
$$-\frac{2ht^2}{|e|E}$$

- 19. In LC circuit the inductance $L=40\,mH$ and $C=100\,\mu F$. If a voltage $V(t)=10\,sin\,(314t)$ is applied to the circuit, the current in the circuit is given as
 - a. 10 cos (314t)

b. 0.52 *cos* (314*t*)

c. 0.52 sin (314t)

d. 5.2 *cos* (314*t*)

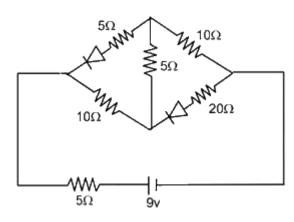
20. The current (i) in the network is

a. 0 *A*

c. 0.2 *A*

b. 0.3 *A*

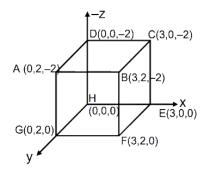
d. 0.6 *A*



21. Starting at temperature 300 K, one mole of an ideal diatomic gas ($\gamma = 1.4$) is first compressed adiabatically from volume V_1 to $V_2 = \frac{V_1}{16}$. It is then allowed to expand isobarically to volume $2V_2$. If all the processes are the quasi-static then the final temperature of the gas (${}^{0}K$) is (to the nearest integer)

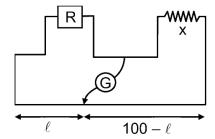
9th Jan (Shift 2, Physics)

22. An electric field $\vec{E}=4x\hat{\imath}-(y^2+1)\hat{\jmath}\,N/C$, passes through the box shown in figure. The flux of the electric field through surface ABCD and BCGF are marked as ϕ_1 and ϕ_2 , then difference between $(\phi_1-\phi_2)$ is $(\frac{Nm^2}{C})$

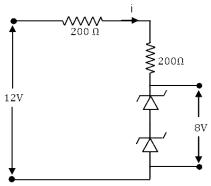


23. In a Young's double slit experiment 15 fringes are observed on a small portion of the screen when light of wavelength $500 \ nm$ is used. 10 fringes are observed on the same section of the screen when another light source of wavelength λ is used. Then the value of λ is (nm))

24. In a meter bridge experiment S is a standard resistance. R is a resistance wire. It is found that balancing length is l=25 cm. If R is replaced by a wire of half length and half diameter that of R of same material, then the balancing l (in cm) will now be



25. The circuit shown below is working as a $8\,V$ dc regulated voltage source. When $12\,V$ is used as input, the power dissipated (in mW) in each diode id; (considering both zener diode are identical).



Date: 9th January 2020

Time: 02:30 PM - 05:30 PM

Subject: Chemistry

- 5 g of Zinc is treated separately with an excess of 1.
- dilute hydrochloric acid and I.
- II. aqueous sodium hydroxide.

The ratio of the volumes of H₂ evolved in these two reactions is:

- a. 2:1
- c. 1:1

- b. 1:2
- d. 1:4

- The solubility product of $Cr(OH)_3$ at 298 K is 6×10^{-31} . The concentration of hydroxide ions in a 2. saturated solution Cr(OH)₃ will be:
 - a. $(18 \times 10^{-31})^{1/4}$
 - c. $(2.22 \times 10^{-31})^{1/4}$

- b. $(18 \times 10^{-31})^{1/2}$ d. $(4.86 \times 10^{-29})^{1/4}$

- 3. Among the statements (a)-(d), the correct ones are :
 - a) Lithium has the highest hydration enthalpy among the alkali metals.
 - b) Lithium chloride is insoluble in pyridine.
 - c) Lithium cannot form ethynide upon its reaction with ethyne.
 - d) Both lithium and magnesium react slowly with H₂O.
 - a. (a), (b) and (d) only

b. (b) and (c) only

c. (a), (c) and (d) only

d. (a) and (d) only

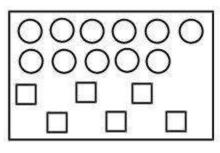
- 4. The first and second ionization enthalpies of a metal are 496 and 4560 kJ mol $^{-1}$ respectively. How many moles of HCl and H_2SO_4 , respectively, will be needed to react completely with 1 mole of metal hydroxide?
 - a. 1 and 2

b. 1 and 0.5

c. 1 and 1

d. 2 and 0.5

5. In the figure shown below reactant A (represented by the square) is in equilibrium with product B (represented by circle). The equilibrium constant is:



- a. 1
- c. 8

- b. 2
- d. 4

- 6. The correct order spin-only magnetic moments of the following complexes is :
 - I. $[Cr(H_2O)_6]Br_2$
 - II. Na₄[FeCN₆]
 - III. Na₃[Fe(C₂O₄)₃] ($\Delta_0 > P$)
 - IV. $(Et_4N)_2[CoCl_4]$
 - a. (III)>(I)>(II)>(IV)
 - c. (I)>(IV)>(III)>(II)

- b. (III)>(I)>(IV)>(II)
- d. (II) (IV)>(III)

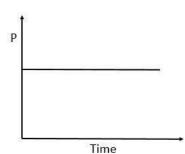
- 7. The true statement amongst the following
 - a. S is a function of temperature but ΔS is not a function of temperature.
 - b. Both ΔS and S are functions of temperature.
 - c. Both S and Δ S are not functions of temperature.
 - d. S is not a function of temperature but ΔS is a function of temperature.

- 8. The reaction of $H_3N_3B_3Cl_3$ (A) with LiBH₄ in tetrahydrofuran gives inorganic benzene (B). Furthur, the reaction of (A) with (C) leads to $H_3N_3B_3(Me)_3$. Compounds (B) and (C) respectively, are:
 - a. Boron nitride, MeBr
 - c. Borazine, MeBr

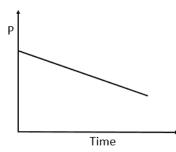
- b. Diborane, MeMgBr
- d. Borazine, MeMgBr

9. A mixture of gases O_2 , H_2 and CO are taken in a closed vessel containing charcoal. The graph that represents the correct behaviour of pressure with time is:

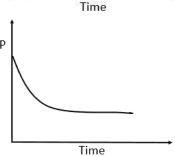
a.



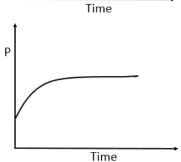
b.



c.



d.



- 10. The isomer(s) of [Co(NH₃)₄Cl₂] that has/have a Cl-Co-Cl angle of 90°, is/are:
 - a. cis only

b. trans only

c. meridional and trans

d. cis and trans

- 11. Amongst the following, the form of water with lowest ionic conductance at 298 K is:
 - a. distilled water

b. sea water

- c. saline water used for intra venous injection
- d. water from a well

- 12. The number of sp² hybrid orbitals in molecule of benzene is:
 - a. 18

b. 24

c. 6

d. 12

13. Which of the following reactions will not produce a racemic product?

а

b.
$$(CH_3)_2$$
-CH-CH= CH_2

c.

d.
$$CH_3CH_2-CH=CH_2$$
 HBr

14. Which of the following has the shortest C–Cl bond?

a.
$$Cl - CH = CH_2$$

c.
$$Cl - CH = CH - OCH_3$$

b.
$$Cl - CH = CH - CH_3$$

d.
$$Cl - CH = CH - NO_2$$

- 15. Biochemical oxygen demand (BOD) is the amount of oxygen required (in ppm):
 - a. for the photochemical breakdown of waste present in $1m^3$ volume of a water body
 - b. by anaerobic bacteria to break-down inorganic waste present in a water body.
 - c. by bacteria to break-down organic waste in a certain volume of water sample.
 - d. for sustaining life in a water body

- 16. Which polymer has chiral, monomer(s)?
 - a. Buna-N
 - c. Nylon 6,6

- b. Neoprene
- d. PHBV

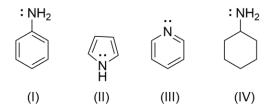
17. A, B and C are three biomolecules. The results of the tests performed on them are given below:

	Molisch's Test	Barfoed Test	Biuret Test
A	Positive	Negative	Negative
В	Positive	Positive	Negative
С	Negative	Negative	Positive

A, B and C are respectively

a. A=Lactose B=Glucose C=Albumin
 b. A=Lactose B=Glucose C=Alanine
 c. A=Lactose B=Fructose C=Alanine
 d. A=Glucose B=Sucrose C=Albumin

18. The decreasing order of basicity of the following amines is:



a. I > II > III > IV

b. IV > III > I > II

c. II > I > III > IV

d. IV > I > II > III

19.

The compound [P] is:

a.

c.

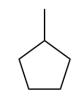
b.

20. In the following reaction A is:

A
$$(i)$$
 Br₂, hv (ii) KOH (alc.)

(iii) O₃ (iv) (CH₃)₂S (v) NaOH(aq) + Δ

a.



c.



b.



d.

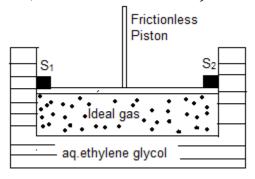
21. The sum of total number of bonds between chromium and oxygen atoms in chromate and dichromate ions is —

22. A sample of milk splits after 60 min. at 300K and after 40 min at 400K when the population of lactobacillus acidophilus in it doubles . The activation energy (in kJ/mol) for this process is closest to ----- .

(Given, R = 8.3 J mol⁻¹K⁻¹), $\ln(\frac{2}{3})$ = 0.4, e^{-3} = 4.0)

23. One litre of sea water (d = $1.03 \frac{g}{cm^3}$) contains 10.3 mg of O_2 gas. Determine the concentration of O_2 in ppm:

24. A cylinder containing an ideal gas (0.1 mol of 1.0 dm 3) is in thermal equilibrium with a large volume of 0.5 molal aqueous solution of ethylene glycol at it freezing point. If the stoppers S_1 and S_2 (as shown in the figure) suddenly withdrawn, the volume of the gas in liters after equilibrium is achieved will be ----(Given, K_f (water)=2.0 K kg mol $^{-1}$, R = 0.08 dm 3 atm K^{-1} mol $^{-1}$)



25. Consider the following reactions

A
$$\frac{\text{(i) CH}_3\text{MgBr}}{\text{(ii) H}_3\text{O}^+}$$
 B $\frac{\text{Cu}}{573 \text{ K}}$ 2-methyl-2-butene

The mass percentage of carbon in A is:

Date of Exam: 9th January 2020 (Shift 2)

Time: 2:30 P.M. to 5:30 P.M.

Subject: Mathematics

1. If $A = \{x \in \mathbb{R} : |x| < 2\}$ and $B = \{x \in \mathbb{R} : |x - 2| \ge 3\}$ then:

a.
$$A - B = [-1,2]$$

c. $A \cup B = \mathbf{R} - (2,5)$

b.
$$B - A = \mathbf{R} - (-2.5)$$

d.
$$A \cap B = (-2, -1)$$

2. If 10 different balls has to be placed in 4 distinct boxes at random, then the probability that two of these boxes contain exactly 2 and 3 balls is :

a.
$$\frac{965}{2^{10}}$$

b.
$$\frac{945}{2^{10}}$$

c.
$$\frac{945}{2^{11}}$$

d.
$$\frac{965}{2^{11}}$$

- 3. If $x = 2\sin\theta \sin 2\theta$ and $y = 2\cos\theta \cos 2\theta$, $\theta \in [0,2\pi]$, then $\frac{d^2y}{dx^2}$ at $\theta = \pi$ is:
 - a. $-\frac{3}{8}$

b. $\frac{3}{4}$

c. $\frac{3}{2}$

d. $-\frac{3}{4}$

Answer: (Bonus)

- 4. Let f and g be differentiable functions on R, such that $f \circ g$ is the identity function. If for some $a, b \in R$, g'(a) = 5 and g(a) = b, then f'(b) is equal to :
 - a. $\frac{2}{5}$

b. 5

c. 1

d. $\frac{1}{5}$

5. In the expansion of $\left(\frac{x}{\cos\theta} + \frac{1}{x\sin\theta}\right)^{16}$, if l_1 is the least value of the term independent of x when $\frac{\pi}{8} \le \theta \le \frac{\pi}{4}$ and l_2 is the least value of the term independent of x when $\frac{\pi}{16} \le \theta \le \frac{\pi}{8}$, then the ratio l_2 : l_1 is equal to :

a. 16:1

b. 8:1

c. 1:8

d. 1:16

6. Let $a, b \in \mathbb{R}$, $a \neq 0$, such that the equation, $ax^2 - 2bx + 5 = 0$ has a repeated root α , which is also a root of the equation $x^2 - 2bx - 10 = 0$. If β is the root of this equation, then $\alpha^2 + \beta^2$ is equal to:

a. 24c. 26

b. 25

d. 28

7. Let a function $f: [0,5] \to \mathbf{R}$, be continuous, f(1) = 3 and \mathbf{F} be defined as:

 $F(x)=\int_1^x t^2 \,g(t)dt$, where $g(t)=\int_1^t f(u)du$ Then for the function ${\it F}$, the point x=1 is

a. a point of inflection.

b. a point of local maxima

c. a point of local minima.

d. not a critical point

8. Let [t] denotes the greatest integer $\leq t$ and $\lim_{x\to 0} x\left[\frac{4}{x}\right] = A$. Then the function, $f(x) = [x^2]\sin \pi x$ is discontinuous, when x is equal to

a.
$$\sqrt{A+1}$$

b.
$$\sqrt{A}$$

c.
$$\sqrt{A+5}$$

d.
$$\sqrt{A+21}$$

9. Let
$$a - 2b + c = 1$$
,
If $f(x) = \begin{vmatrix} x+a & x+2 & x+1 \\ x+b & x+3 & x+2 \\ x+c & x+4 & x+3 \end{vmatrix}$, then:

a.
$$f(-50) = 501$$

b.
$$f(-50) = -1$$

c.
$$f(50) = 1$$

d.
$$f(50) = -501$$

10. Given:
$$f(x) = \begin{cases} x, & 0 \le x < \frac{1}{2} \\ \frac{1}{2}, & x = \frac{1}{2} \\ 1 - x, & \frac{1}{2} < x \le 1 \end{cases}$$

and $g(x) = \left(x - \frac{1}{2}\right)^2$, $x \in \mathbb{R}$. Then the area (in sq. units) of the region bounded by the curves y = f(x) and y = g(x) between the lines 2x = 1 to $2x = \sqrt{3}$ is:

a.
$$\frac{\sqrt{3}}{4} - \frac{1}{3}$$

b.
$$\frac{1}{3} + \frac{\sqrt{3}}{4}$$

c.
$$\frac{1}{2} + \frac{\sqrt{3}}{4}$$

d.
$$\frac{1}{2} - \frac{\sqrt{3}}{4}$$

11. The following system of linear equations

$$7x + 6y - 2z = 0$$
,
 $3x + 4y + 2z = 0$
 $x - 2y - 6z = 0$, has

- a. infinitely many solutions, (x, y, z) satisfying y = 2z
- b. infinitely many solutions (x, y, z) satisfying x = 2z
- c. no solution
- d. only the trivial solution

- 12. If $p \to (p \land \sim q)$ is false. Then the truth values of p and q are respectively
 - a. F, T

c. F, F

b. T, Fd. T, T

- 13. The length of minor axis (along y-axis) of an ellipse of the standard form is $\frac{4}{\sqrt{3}}$. If this ellipse touches the line x + 6y = 8, then its eccentricity is :
 - a. $\frac{1}{2}\sqrt{\frac{5}{3}}$

b. $\frac{1}{2}\sqrt{\frac{11}{3}}$

c. $\sqrt{\frac{5}{6}}$

d. $\frac{1}{3}\sqrt{\frac{11}{3}}$

14. If z be a complex number satisfying |Re(z)| + |Im(z)| = 4, then |z| cannot be:

a.
$$\sqrt{7}$$

b.
$$\sqrt{\frac{17}{2}}$$

c.
$$\sqrt{10}$$

d.
$$\sqrt{8}$$

15. If
$$x = \sum_{n=0}^{\infty} (-1)^n \tan^{2n}\theta$$
 and $y = \sum_{n=0}^{\infty} \cos^{2n}\theta$, where $0 < \theta < \frac{\pi}{4}$, then:

a.
$$y(1+x) = 1$$

b.
$$x(1-y) = 1$$

c.
$$y(1-x) = 1$$

b.
$$x(1-y) = 1$$

d. $x(1+y) = 1$

16. If
$$\frac{dy}{dx} = \frac{xy}{x^2 + y^2}$$
; $y(1) = 1$; then a value of x satisfying $y(x) = e$ is:

a.
$$\sqrt{3}e$$

b.
$$\frac{1}{2}\sqrt{3}e$$

c.
$$\sqrt{2}e$$

d.
$$\frac{e}{\sqrt{2}}$$

17. If one end of focal chord *AB* of the parabola $y^2 = 8x$ is at $A\left(\frac{1}{2}, -2\right)$, then the equation of tangent to it at *B* is

a.
$$x + 2y + 8 = 0$$

c.
$$x - 2y + 8 = 0$$

b.
$$2x - y - 24 = 0$$

d.
$$2x + y - 24 = 0$$

- 18. Let a_n be the n^{th} term of a G.P. of positive terms. If $\sum_{n=1}^{100} a_{2n+1} = 200$ and $\sum_{n=1}^{100} a_{2n} = 100$ then $\sum_{n=1}^{200} a_n$ is equal to:
 - a. 300

b. 175

c. 225

d. 150

19. A random variable \boldsymbol{X} has the following probability distribution:

X	1	2	3	4	5
P(X)	K^2	2 <i>K</i>	K	2 <i>K</i>	$5K^2$

Then P(X > 2) is equal to:

a. $\frac{7}{12}$

b. $\frac{23}{36}$

c. $\frac{1}{36}$

d. $\frac{1}{6}$

20. If
$$\int \frac{d\theta}{\cos^2 \theta \ (\tan 2\theta + \sec 2\theta)} = \lambda \tan \theta + 2 \log_e |f(\theta)| + C$$
 where C is constant if integration, then the ordered pair $(\lambda, f(\theta))$ is equal to:

a.
$$(-1, 1 - \tan \theta)$$

b.
$$(-1, 1 + \tan \theta)$$

c.
$$(1, 1 + \tan \theta)$$

d.
$$(1, 1 - \tan \theta)$$

21. Let \vec{a} , \vec{b} and \vec{c} be three vectors such that $|\vec{a}| = \sqrt{3}$, $|\vec{b}| = 5$, \vec{b} . $\vec{c} = 10$ and the angle between \vec{b} and \vec{c} is $\frac{\pi}{3}$. If \vec{a} is perpendicular to vector $\vec{b} \times \vec{c}$, then $|\vec{a} \times (\vec{b} \times \vec{c})|$ is equal to _____

22. If $C_r = {}^{25}C_r$ and $C_0 + 5 \cdot C_1 + 9 \cdot C_2 + \dots + 101 \cdot C_{25} = 2^{25} \cdot k$ then k is equal to_____.

23. If the curves $x^2 - 6x + y^2 + 8 = 0$ and $x^2 - 8y + y^2 + 16 - k = 0$, (k > 0) touch each other at a point, then the largest value of k is _____.

24. The number of terms common to the A.P.'s 3, 7, 11, ... 407 and 2, 9, 16, ... 709 is _____.

25. If the distance between the plane, 23x - 10y - 2z + 48 = 0 and the plane containing the lines $\frac{x+1}{2} = \frac{y-3}{4} = \frac{z+1}{3}$ and $\frac{x+3}{2} = \frac{y+2}{6} = \frac{z-1}{\lambda}$, $(\lambda \in R)$ is equal to $\frac{k}{\sqrt{633}}$, then k is equal to