

Mathematical modelling for the next-generation all-solid-state battery: Nucleation interface

Tuan Vo^{a,b}, Claas Hüter^b, Stefanie Braun^a

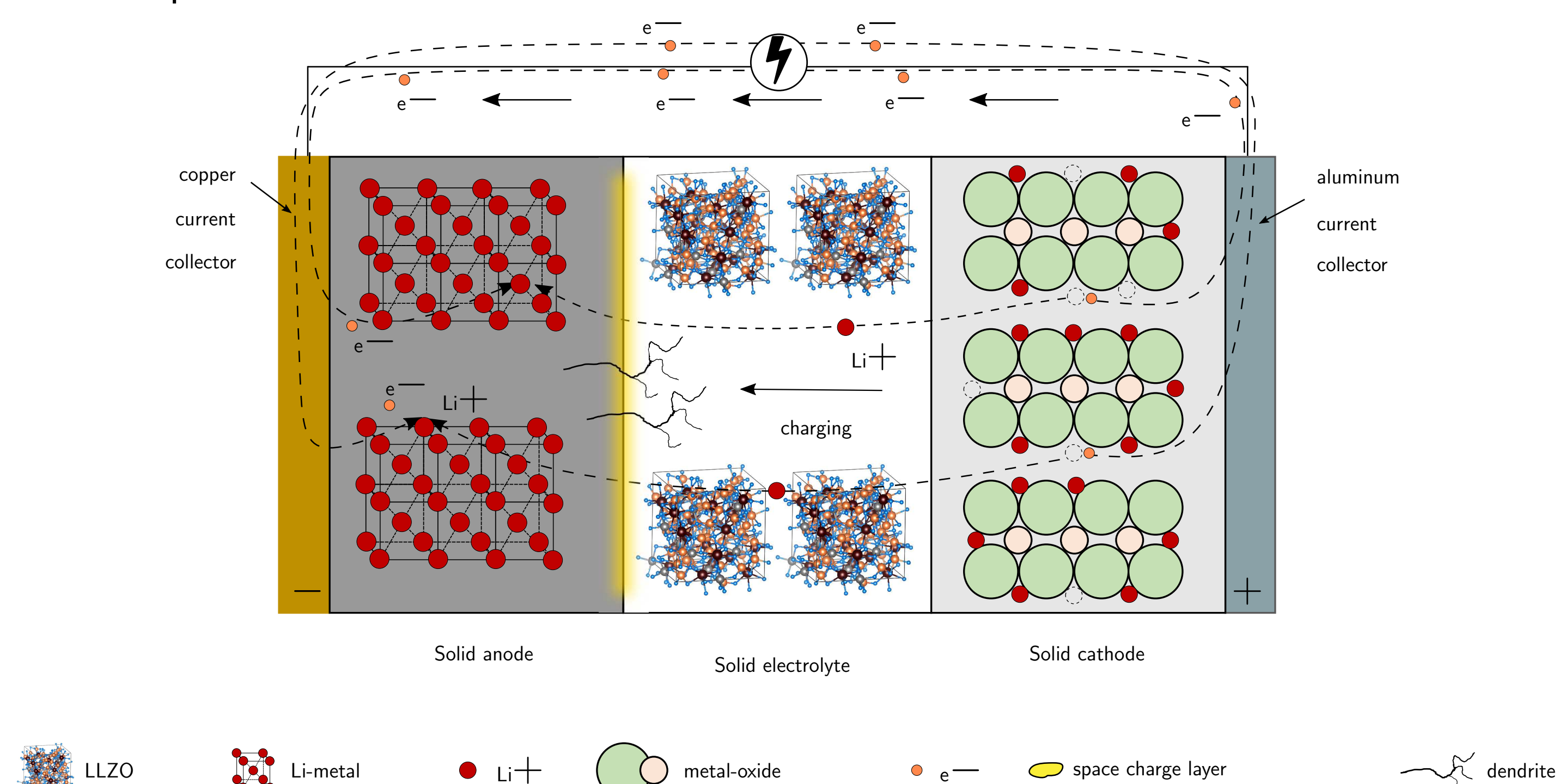
(1) Department of Mathematics, Applied and Computational Mathematics (ACoM), RWTH Aachen University, Schinkelstraße 02, 52062 Aachen, Germany
(2) Institute of Energy and Climate Research (IEK-2), Forschungszentrum Jülich, Wilhelm-Johnen-Straße, 52428 Jülich, Germany

Next-generation All-solid-state battery (ASSB)

Rechargeable Lithium-ion battery (LIB) stays at the heart of every energy storage system and electric vehicle. Undoubtedly, LIB benefits human life efficiently as well as friendly-environment. Besides, a more advanced LIB, so-called **all-solid-state battery** (ASSB), is introduced recently as ASSB is expected with non-inflammation and non-explosion as seen in common LIBs. Yet, defect due to polarization is one natural phenomenon of **solid electrolyte** (SE) to be tackled.

$$a_{\text{Griffith}} := a^* = \arg \min_{a \in \mathbb{R}} \left(\iint_{\Omega} f(a, \mathbf{u}; \lambda, \mu, \mathbf{d} \otimes \mathbf{d}) d\Omega - \iint_{\Gamma} f(a; \gamma) d\Gamma \right) \Big|_{\mathbf{u}^{(s)}}$$

This poster is aimed to model the polarized SE with the use of **structural tensor**. A typical LIB includes three main components: cathode, anode and electrolyte. Different types of LIB have a variation of constitutive material composed of battery. An ASSB means that the three main components are **all made of solid material**.



Modelling goal: Interface analysis + Numerical modelling

Two main goals to model the solid electrolyte part of the all-solid-state battery is as follows:

1. To capture the **preferred direction** behaviour of the solid electrolyte due to electric potential.
2. To satisfy **thermodynamic consistency**:
 - Conservation of mass, linear & angular momentum and energy for the solid electrolyte.
 - Entropy inequality is guaranteed with sharper conditions, which lead to constitutive equation.

$$a_{\text{Griffith}} := a^* = \arg \min_{a \in \mathbb{R}} \left(\iint_{\Omega} f(a, \mathbf{u}; \lambda, \mu, \mathbf{d} \otimes \mathbf{d}) d\Omega - \iint_{\Gamma} f(a; \gamma) d\Gamma \right) \Big|_{\mathbf{u}^{(s)}}$$

Contact

Tuan Vo

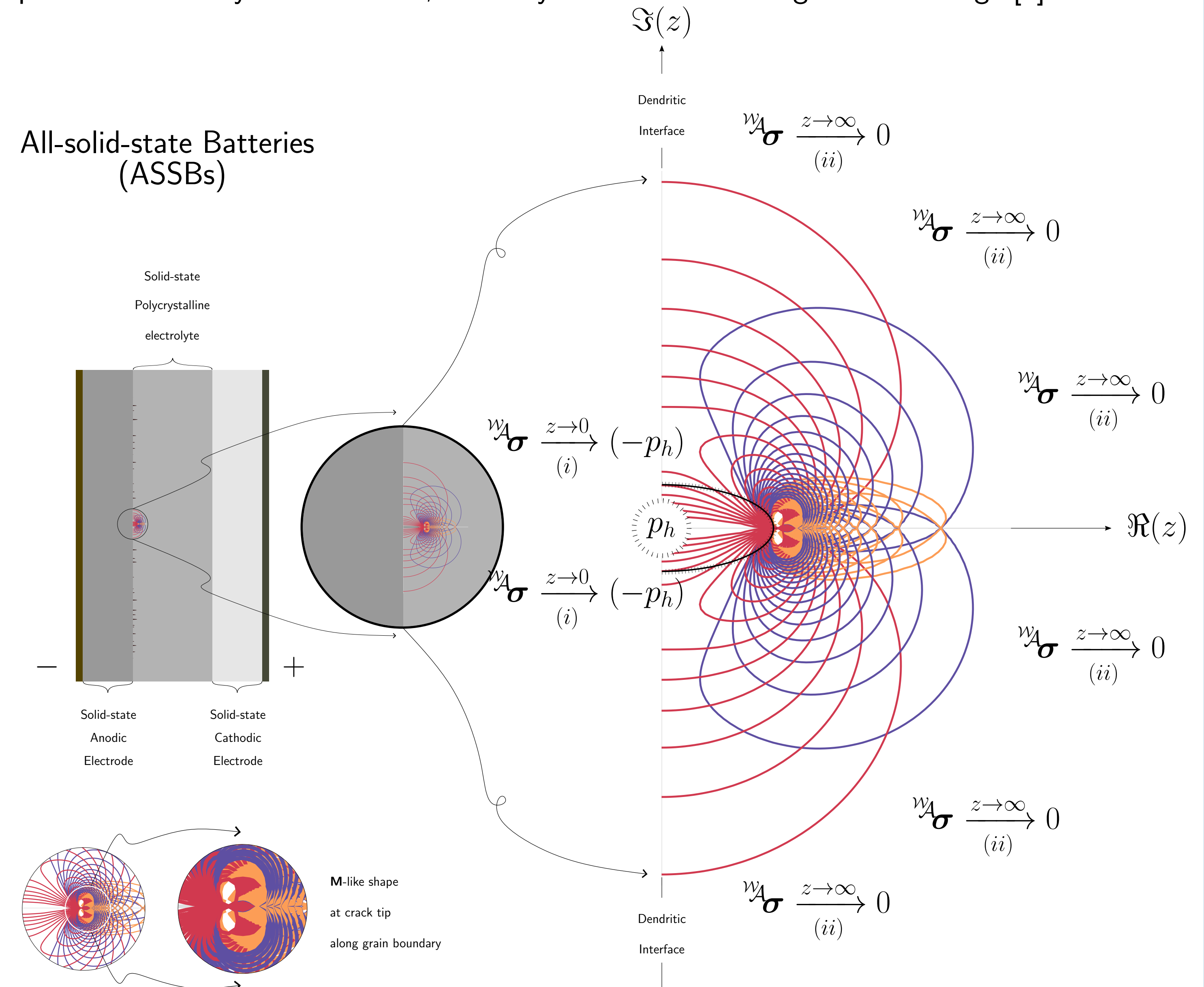
Applied and Computational Mathematics (ACoM)
RWTH Aachen University
vo@acom.rwth-aachen.de



Scan me

Mathematical model

Interface between solid electrode and solid-state electrolyte (SE|SSE) taking place at space charge layer (SCL) [2] found in all-solid-state lithium-ion batteries (ASSLiBs) critically exhibits mechanical and electrochemical instability [3]. This evidence points directly to the fact that the soft metallic lithium negative electrode is erroneously prone to triggering dendritic by-products of silvery lithium metal, under cycles of electric charge and discharge [4].



Besides, polycrystalline garnet-typed solid-state electrolyte such as LLZO exhibit grain boundaries and various sizes and shapes of grains under microscopic observation. Therefore, this type of microstructure distinctively leads to nuance destruction of ceramic-like materials. Consequentially, dendritic by-products contribute to degradation of ionic conductivity and trace along grain boundaries in SSE. This phenomenon, notwithstanding, is predicted, quantified, and controlled based on analysing the multi-scale coupled problem subjected to conditions of Griffith criterion.

References

- [1] **T. Vo** *Modeling the swelling phenomena of li-ion battery cells based on a numerical chemo-mechanical coupled approach*. Master thesis, Robert Bosch Batteries Systems GmbH, 2018.
- [2] **S. Braun**, C. Yada and A. Latz. *Thermodynamically consistent model for Space-Charge-Layer formation in a solid electrolyte*. Journal of Physical Chem., 119, 22281-22288, 2015.
- [3] **C. Hüter**, S. Fu, M. Finsterbusch, E. Figgemeier, L. Wells, and R. Spatschek. *Electrode-electrolyte interface stability in solid state electrolyte system: influence of coating thickness under varying residual stresses*. AIMS materials Science, 4(4):867-877, 2017.
- [4] **S. Kim**, J. S. Kim, L. Miara, Y. Wang, S. K. Jung, S. Y. Park, Z. Song, H. King, M. Badding, J. M. Chang, V. Roev, G. Yoon, R. Kim, J. H. Kim, K. Yoon, D. Im, and K. Kang. "High-energy and durable lithium metal batteries using garnet-type solid electrolytes with tailored lithium-metal compatibility," Nature Communications, 13(1):1883, Apr 2022.