

Mathematical modelling for the next-generation all-solid-state battery: Nucleation criterion at the (SE|SSE)^(*)-interface

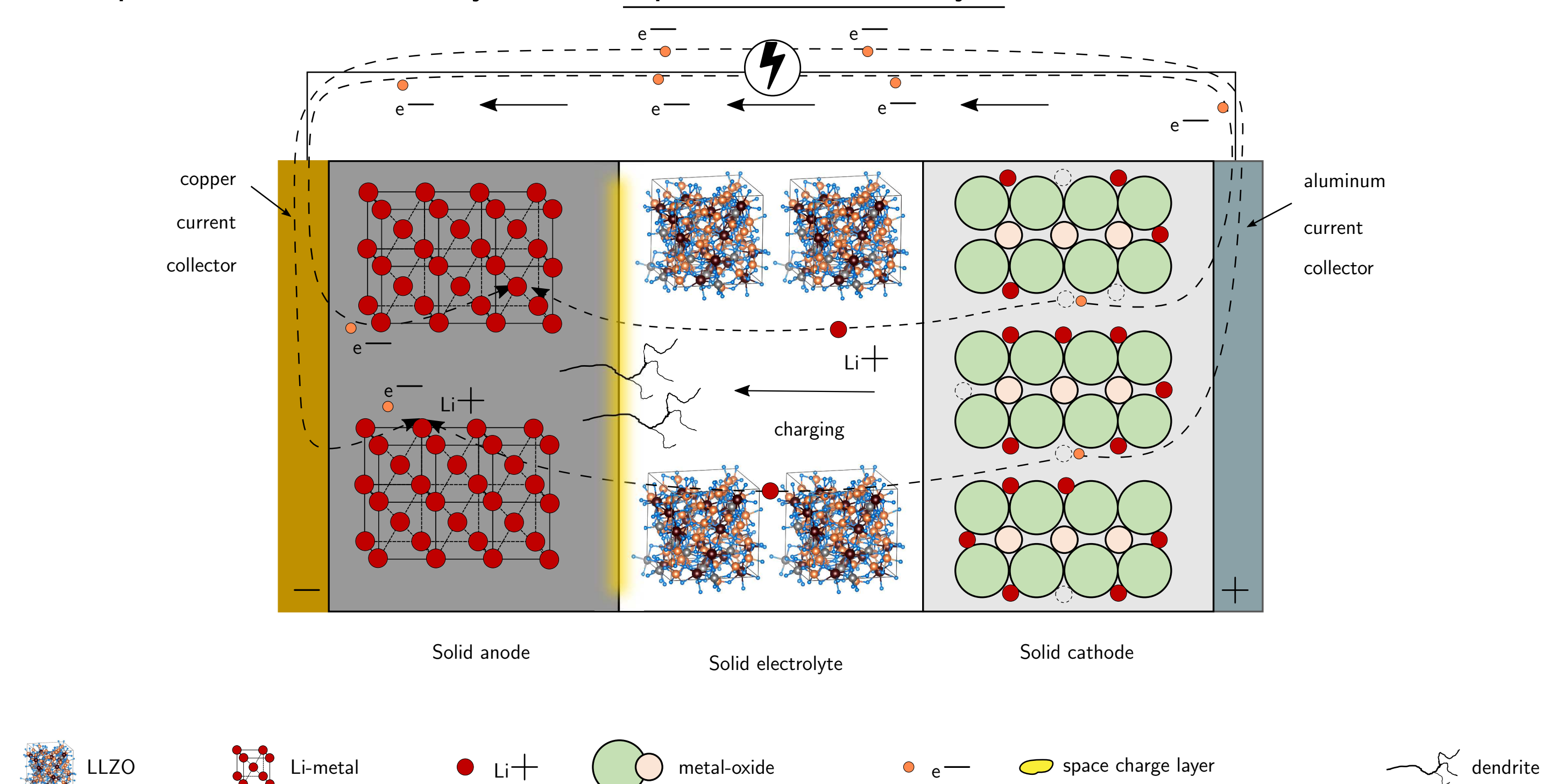
Tuan Vo^{a,b,†}, Claas Hüter^b, Stefanie Braun^a

^aDepartment of Mathematics, Applied and Computational Mathematics (ACoM), RWTH Aachen University, Schinkelstraße 02, 52062 Aachen, Germany

^bInstitute of Energy and Climate Research (IEK-2), Forschungszentrum Jülich, Wilhelm-Johnen-Straße, 52428 Jülich, Germany

Next-generation All-solid-state battery (ASSB)

Rechargeable Lithium-ion battery (LIB) stays at the heart of electric vehicles, portable electric devices, and energy storage systems [1]. Nowadays, LIB helps human life become more efficient as well as helps to cope with global environment issues thanks to its zero emission. However, conventional LIB is sensible to temperature and pressure, easily prone to flammable and explosive. This is mainly due to liquid-based electrolyte found in such conventional LIBs.



All-solid-state battery (ASSB) is, therefore, one of promising candidates to overcome bottlenecks of conventional LIBs. It is because the solid-state electrolyte (SSE) in ASSB, e.g. SSE made of the highly ionic-conductive polycrystalline LLZO, is highly stable towards temperature and pressure. Nevertheless, metallic Lithium dendrite triggered at (SE|SSE)-interface is the main drawback as these dendritic threads extrapolate into grain boundary network of SSE, causing degradation of ionic conductivity and the likelihood of short-circuit.

The next-generation all-solid-state battery should, consequentially, be able to cope with micro-dendritic threads at the (SE|SSE)-interface, and hence, to foresee nucleation points caused by propagations of these dendrites, when possible.

This poster is aimed to model the nucleation by taking into consideration polarization, structural tensor, locally non-uniform electric field, and Griffith analysis.

$$a_{\text{Griffith}} := a^* = \arg \min_{a \in \mathbb{R}} \left(\iiint_{\Omega} f(a, \mathbf{u}; \lambda, \mu, \mathbf{d} \otimes \mathbf{d}) d\Omega - \iint_{\Gamma} f(a; \gamma) d\Gamma \right) \Big|_{\mathbf{u}^{(s)}} \quad (1)$$

where

Modelling goal: Interface analysis + Numerical modelling

Two main goals to model the solid electrolyte part of the all-solid-state battery is as follows:

1. To capture the **preferred direction** behaviour of the solid electrolyte due to electric potential.
2. To satisfy **thermodynamic consistency**:
 - Conservation of mass, linear & angular momentum and energy for the solid electrolyte.
 - Entropy inequality is guaranteed with sharper conditions, which lead to constitutive equation.

$$a_{\text{Griffith}} := a^* = \arg \min_{a \in \mathbb{R}} \left(\iiint_{\Omega} f(a, \mathbf{u}; \lambda, \mu, \mathbf{d} \otimes \mathbf{d}) d\Omega - \iint_{\Gamma} f(a; \gamma) d\Gamma \right) \Big|_{\mathbf{u}^{(s)}}$$

Contact

Tuan Vo

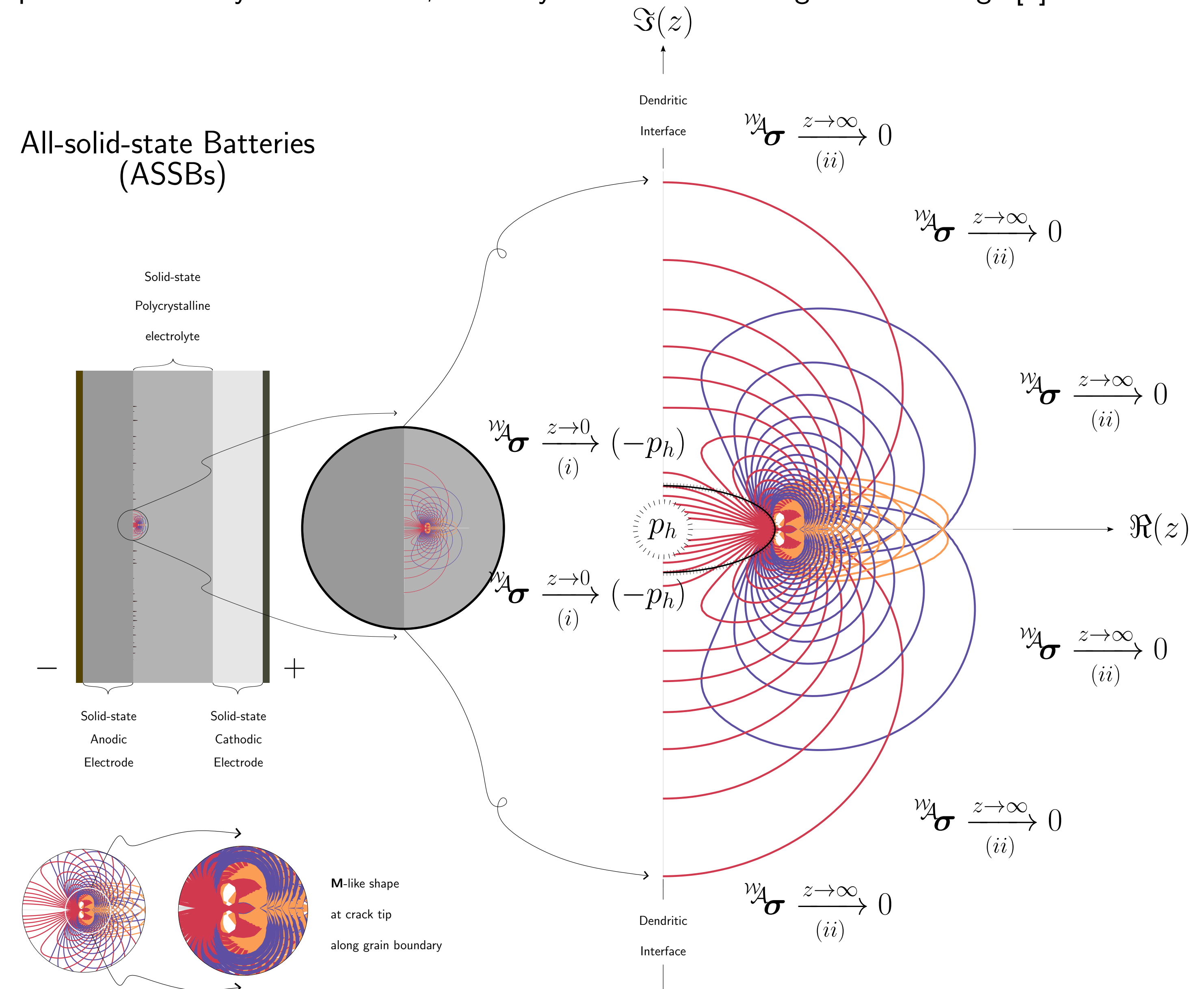
Applied and Computational Mathematics (ACoM)
RWTH Aachen University
vo@acom.rwth-aachen.de



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Mathematical model

Interface between solid electrode and solid-state electrolyte (SE|SSE) taking place at space charge layer (SCL) [2] found in all-solid-state lithium-ion batteries (ASSLiBs) critically exhibits mechanical and electrochemical instability [3]. This evidence points directly to the fact that the soft metallic lithium negative electrode is erroneously prone to triggering dendritic by-products of silvery lithium metal, under cycles of electric charge and discharge [4].



Besides, polycrystalline garnet-typed solid-state electrolyte such as LLZO exhibit grain boundaries and various sizes and shapes of grains under microscopic observation. Therefore, this type of microstructure distinctively leads to nuance destruction of ceramic-like materials. Consequentially, dendritic by-products contribute to degradation of ionic conductivity and trace along grain boundaries in SSE. This phenomenon, notwithstanding, is predicted, quantified, and controlled based on analysing the multi-scale coupled problem subjected to conditions of Griffith criterion.

References

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- (*) (Solid electrode | Solid-state electrolyte)