Cobalt

Cobalt is a <u>chemical element</u> with the <u>symbol</u> **Co** and atomic number 27. Like <u>nickel</u>, cobalt is found in the Earth's crust only in chemically combined form, save for small deposits found in alloys of natural <u>meteoric iron</u>. The <u>free element</u>, produced by reductive <u>smelting</u>, is a hard, lustrous, silver-gray metal.

Cobalt-based blue pigments (cobalt blue) have been used since ancient times for jewelry and paints, and to impart a distinctive blue tint to glass, but the color was later thought to be due to the known metal bismuth. Miners had long used the name kobold ore (German for goblin ore) for some of the blue-pigment-producing minerals; they were so named because they were poor in known metals, and gave poisonous arsenic-containing fumes when smelted. In 1735, such ores were found to be reducible to a new metal (the first discovered since ancient times), and this was ultimately named for the kobold.

Today, some cobalt is produced specifically from one of a number of metallic-lustered ores, such as <u>cobaltite</u> (CoAsS). The element is, however, more usually produced as a by-product of <u>copper</u> and <u>nickel</u> mining. <u>The copper belt</u> in the <u>Democratic Republic of the Congo</u> (DRC) and <u>Zambia</u> yields most of the global cobalt production. World production in 2016 was 116,000 tonnes (according to <u>Natural Resources Canada</u>), and the DRC alone accounted for more than 50%. [4]

Cobalt is primarily used in <u>lithium-ion batteries</u>, and in the manufacture of <u>magnetic</u>, wear-resistant and high-strength <u>alloys</u>. The compounds cobalt silicate and <u>cobalt(II) aluminate</u> (CoAl₂O₄, cobalt blue) give a distinctive deep blue color to <u>glass</u>, <u>ceramics</u>, <u>inks</u>, <u>paints</u> and <u>varnishes</u>. Cobalt occurs naturally as only one stable <u>isotope</u>, cobalt-59. <u>Cobalt-60</u> is a commercially important radioisotope, used as a <u>radioactive tracer</u> and for the production of high-energy <u>gamma rays</u>.

Cobalt is the active center of a group of <u>coenzymes</u> called <u>cobalamins</u>. <u>Vitamin B₁₂</u>, the best-known example of the type, is an essential <u>vitamin</u> for all animals. Cobalt in inorganic form is also a <u>micronutrient</u> for <u>bacteria</u>, algae, and fungi.