

Boron

Boron is a [chemical element](#) with the [symbol](#) **B** and [atomic number](#) 5. Produced entirely by [cosmic ray spallation](#) and [supernovae](#) and not by [stellar nucleosynthesis](#), it is a low-abundance element in the [Solar System](#) and in the [Earth's crust](#).^[11] It constitutes about 0.001 percent by weight of Earth's crust.^[12] Boron is concentrated on Earth by the water-solubility of its more common naturally occurring compounds, the [borate minerals](#). These are mined industrially as [evaporites](#), such as [borax](#) and [kernite](#). The largest known boron deposits are in [Turkey](#), the largest producer of boron minerals.

Elemental boron is a [metalloid](#) that is found in small amounts in [meteoroids](#) but chemically uncombined boron is not otherwise found naturally on Earth. Industrially, very pure boron is produced with difficulty because of refractory contamination by carbon or other elements. Several [allotropes of boron](#) exist: [amorphous](#) boron is a brown powder; crystalline boron is silvery to black, extremely hard (about 9.5 on the [Mohs scale](#)), and a poor [electrical conductor](#) at room temperature. The primary use of elemental boron is as [boron filaments](#) with applications similar to [carbon fibers](#) in some high-strength materials.

Boron is primarily used in chemical compounds. About half of all boron consumed globally is an additive in [fiberglass](#) for insulation and structural materials. The next leading use is in [polymers](#) and [ceramics](#) in high-strength, lightweight structural and refractory materials. [Borosilicate glass](#) is desired for its greater strength and thermal shock resistance than ordinary soda lime glass. Boron as [sodium perborate](#) is used as a [bleach](#). A small amount of boron is used as a [dopant](#) in [semiconductors](#), and [reagent](#) intermediates in the [synthesis of organic fine chemicals](#). A few boron-containing organic pharmaceuticals are used or are in study. Natural boron is composed of two stable isotopes, one of which ([boron-10](#)) has a number of uses as a neutron-capturing agent.

In biology, [borates](#) have low toxicity in mammals (similar to [table salt](#)), but are more toxic to [arthropods](#) and are used as [insecticides](#). [Boric acid](#) is mildly antimicrobial, and several natural boron-containing organic antibiotics are known.^[13] Boron is an essential plant nutrient and boron compounds such as borax and [boric acid](#) are used as [fertilizers](#) in [agriculture](#), although it's only required in small amounts, with excess being toxic. Boron compounds play a strengthening role in the cell walls of all plants. There is no consensus on whether boron is an essential nutrient for mammals, including humans, although there is some evidence it supports bone health.