## Copper

**Copper** is a <u>chemical element</u> with the <u>symbol</u> **Cu** (from <u>Latin</u>: *cuprum*) and <u>atomic number</u> 29. It is a soft, malleable, and <u>ductile</u> metal with very high <u>thermal</u> and <u>electrical conductivity</u>. A freshly exposed surface of pure copper has a <u>pinkish-orange color</u>. Copper is used as a conductor of heat and electricity, as a <u>building material</u>, and as a constituent of various metal <u>alloys</u>, such as <u>sterling silver</u> used in <u>jewelry</u>, <u>cupronickel</u> used to make marine hardware and <u>coins</u>, and <u>constantan</u> used in strain gauges and thermocouples for temperature measurement.

Copper is one of the few metals that can occur in nature in a directly usable metallic form (<u>native metals</u>). This led to very early human use in several regions, from c. 8000 BC. Thousands of years later, it was the first metal to be <u>smelted</u> from sulfide ores, c. 5000 BC; the first metal to be cast into a shape in a mold, c. 4000 BC; and the first metal to be purposefully alloyed with another metal, <u>tin</u>, to create <u>bronze</u>, c. 3500 BC. [5]

In the <u>Roman era</u>, copper was mined principally on <u>Cyprus</u>, the origin of the name of the metal, from <u>aes cyprium</u> (metal of Cyprus), later corrupted to <u>cuprum</u> (Latin). <u>Coper</u> (<u>Old English</u>) and <u>copper</u> were derived from this, the later spelling first used around 1530.

Commonly encountered compounds are copper(II) salts, which often impart blue or green colors to such minerals as <u>azurite</u>, <u>malachite</u>, and <u>turquoise</u>, and have been used widely and historically as pigments.

Copper used in buildings, usually for roofing, oxidizes to form a green <u>verdigris</u> (or <u>patina</u>). Copper is sometimes used in <u>decorative art</u>, both in its elemental metal form and in compounds as pigments. Copper compounds are used as <u>bacteriostatic agents</u>, <u>fungicides</u>, and wood preservatives.

Copper is essential to all living organisms as a trace <u>dietary mineral</u> because it is a key constituent of the respiratory enzyme complex <u>cytochrome c oxidase</u>. In <u>molluscs</u> and <u>crustaceans</u>, copper is a constituent of the blood pigment <u>hemocyanin</u>, replaced by the iron-complexed <u>hemoglobin</u> in fish and other <u>vertebrates</u>. In humans, copper is found mainly in the liver, muscle, and bone. The adult body contains between 1.4 and 2.1 mg of copper per kilogram of body weight.