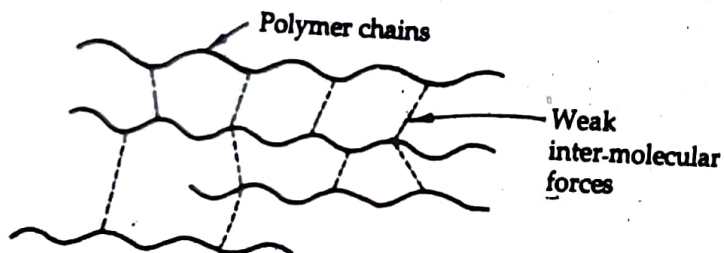


Thermoplastics

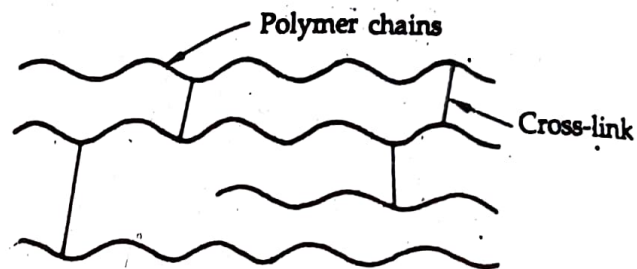
1. They are formed either by addition or by condensation polymerisation reactions.
2. They have either linear or branched structures.
3. Adjacent polymer chains are held together by either van der Waals forces, or by dipole-dipole forces or by H-bonds.



4. They soften on heating and stiffen on cooling.
5. Low molecular weight thermoplastics are soluble in their suitable solvents.
6. They can be remoulded, re-shaped and re-used.
7. They can be reclaimed from waste, i.e., they can be recycled.
8. During moulding of thermoplastics, there is no change in their chemical composition.
9. They are tough materials.
10. Examples : PE, PP, PVC, PMMA, PS, PTFE, Nylons, Polyesters etc.

Thermosets

- They are formed by condensation polymerisation reactions.
- They have three dimensional, cross-linked network structure.
- Adjacent polymer chains are held together by strong covalent bonds called cross-links.



They do not soften on heating.

- They are generally insoluble in any solvent.
- They can't be re-moulded and hence cannot be re-used.
- They cannot be re-claimed from waste. They cannot be recycled.
- They undergo chemical changes such as further polymerisation and cross-linking during moulding process.
- They are brittle materials.
- Examples : PF, UF, MF, Epoxy, XLPE etc.